



Cite this: *RSC Adv.*, 2019, 9, 35401

Computational investigations of the binding mechanism of novel benzophenone imine inhibitors for the treatment of breast cancer†

Amneh Shtaiwi,^a Rohana Adnan,^b Melati Khairuddean^b and Shafi Ullah Khan^c

4-Hydroxytamoxifen (4-OHT), the most common hormone used for the treatment of breast cancer, is a selective estrogen receptor modulator (SERM) inhibitor that acts as an antagonist in breast tissue and a partial agonist in the endometrium. However, the detailed molecular mechanism of 4-OHT structure modification has not been well investigated to date. Herein, molecular docking, molecular dynamics simulations and free energy calculations were performed to explore the mechanisms of the molecular interactions between newly designed benzophenone imines (BIs) and the three forms apo, antagonist and agonist of the human estrogen receptor hER α . The proposed inhibitors were designed by replacing the triarylethylene estrogenic scaffold found in 4-OHT with Schiff base triarylimine derivatives. The antiestrogen scaffold *i.e.* the *O*-alkyl side chain in 4-OHT was developed by incorporating an alanine amino acid side chain functionality into the triarylimine scaffold. Docking results reveal that the newly designed BIs bind to the hydrophobic open pocket of the apo and antagonist hER α conformations with higher affinity as compared to the natural and synthetic estrogen estradiol (E2) and 4-OHT. The analysis of the molecular dynamics simulation results based on six different systems of the best docked BI (5c) with hER α receptors demonstrates stable interactions, and the complex undergoes fewer conformational fluctuations in the open apo/antagonist hER α receptors as compared to the case of the closed agonist. In addition, the calculated binding free energies indicate that the main factor that contributes to the stabilization of the receptor–inhibitor complexes is hydrophobic interactions. This study suggests that the development of these Schiff base derivatives may be worth exploring for the preparation of new 4-OHT analogues.

Received 24th June 2019
Accepted 10th September 2019

DOI: 10.1039/c9ra04759j

rsc.li/rsc-advances

Introduction

Breast cancer is the most common cancer among women worldwide and in the Asia-Pacific region, accounting for 18% of all cancer cases in 2012. Although the incidence rates of breast cancer remain significantly higher in New Zealand and Australia, in recent years, a rapid increase in the cases of breast cancer has been observed in several Asian countries, particularly Malaysia and Thailand.¹ The incidence and mortality estimates for Malaysia in 2012 alone were 5410 and 2572 cases, respectively.^{1,2} The human estrogen receptor hER α protein, which is located on the surface of tumors, can bind to estrogen (17 β -estradiol, E2) and enhance the growth and spreading of

cancer cells;³ moreover, the estrogen hormone has been identified as a key stimulant in the development of hER α -positive breast cancer, which constitutes around 70–80% of all breast cancers.^{4,5} Breast cancer involves the generation of tumors through estrogen signalling after the binding of estrogen to the hER α receptors; this activates hormone-responsive genes that promote DNA synthesis and cell reproduction;⁶ the estrogen receptor is a nuclear receptor, and these receptors have common structural functions, referred to as domains, including the N-terminal (A/B) domain, the DNA binding domain (C), the hinge domain (D) and the ligand binding domain (LBD, E/F) or the C-terminal domain. The activation function 2 (AF-2) in the ligand binding domain has a ligand-dependent transcription factor.^{7,8} A common treatment of hormone-sensitive breast cancer in the early stage is surgery to remove the tumour, radiotherapy and hormone therapy.⁹ The hormone therapy can be used to remove or block the action of hormones such as estrogen, which has been recognised as a key molecular driver in breast cancers.^{9,10} In the ligand–receptor pathway, the binding of E2 to hER α induces a dynamic conformational change that leads to hER α dimerization and association with co-regulatory proteins with the subsequent transcriptional

^aSchool of Pharmacy, Middle East University, Queen Alia Airport Street, 11118 Amman, Jordan. E-mail: ashtaiwi@meu.edu.jo

^bSchool of Chemical Sciences, Universiti Sains Malaysia, 11800 Penang, Malaysia. E-mail: r_adnan@usm.my; Tel: +6046533262

^cSchool of Pharmacy, Monash University Malaysia, Jalan Lagoon Selatan, Bandar Sunway, 47500 Subang Jaya, Malaysia. E-mail: shafiullahpharmD@gmail.com

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c9ra04759j



activation of estrogen-responsive genes.¹¹ Selective estrogen receptor modulators (SERMs), such as the benzothiophene derivative 4-hydroxy tamoxifen (**4-OHT**), act as competitive blockers of the estrogen-hER α binding and have been successfully used in the treatment of hER α -positive breast cancer.¹² When the ligands bind to hER α , many changes occur, particularly in the position of helix12 (H12) at the ligand binding site, and these changes differ when different ligands bind to the receptor. Moreover, the transcriptional activity differs when different ligands bind to hER α .¹³ Upon the binding of **E2** to the hER α ligand binding domain, estradiol rests in a binding cavity within the hydrophobic core of the LBD formed by the helices H3, H6, H8, H11 and H12.¹⁴ The substrate forms a series of specific hydrogen bonds *via* the two hydroxyl groups. The phenolic hydroxyl group from the A ring forms direct hydrogen bonds with the carboxylate of a glutamic acid residue in H3 (Glu353) and the arginine residue in H6 (Arg394). On the other hand, the 17 β -hydroxyl group in the D ring forms a hydrogen bond with a single histidine residue in H11 (His524) (see Fig. 1). The hydrophobic core of **E2** also plays a role in the binding of **E2** with the hydrophobic residues of the ER-LBD, which forms close contacts with the alanine and phenylalanine amino acid residues that bind the ligand.^{14,15}

The result of the hydrophobic and polar binding modes of **E2** is the folding of H12 across H3 and H11, leading to the agonist conformation and thus enhancing gene transcription.^{15,16} When agonists, such as estradiol, bind to ER (see Fig. 2a), the ligand is trapped within a hydrophobic binding cavity formed by the helices H3 (blue), H6 (grey) and H11 (green).¹⁷ This allows the inner hydrophobic surface of H12 (red) to fold across H3 and H11 and cap the entrance of the cavity.¹⁸ Conversely, antagonists, such as the synthetic antiestrogen **4-OHT**, have polar or bulky steric side-chains and occupy the same binding cavity as agonists; however, they force H12 to move towards the open binding site. This allows H12 (red) to overlap with the H3 (blue) and H5 (orange) regions (see Fig. 2b) and occupy the surface where the co-activator protein should bind.^{13,17,18}

The extended apo conformation of the NR LBDs was first described in retinoic X receptor- α (RXR α), where H12 was extended away from the surface of the ligand binding core and did not have any hydrophobic interactions with the LBD.¹⁹

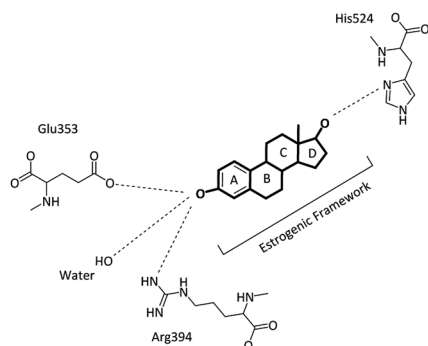


Fig. 1 Schematic showing the binding mode of **E2** in the hER α -LBD complex.

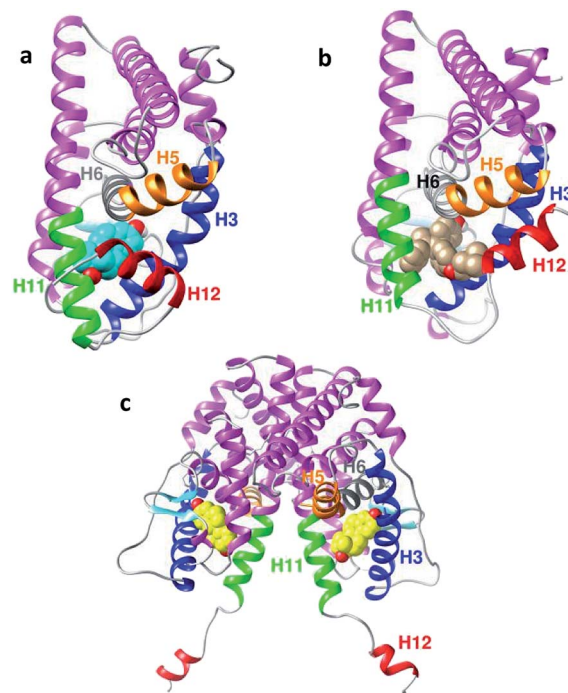


Fig. 2 (a) Backbone of the agonist conformation of hER α LBD (PDB ID: 1G50) in the complex with estradiol **E2** (cyan). (b) Antagonist conformation of the hER α LBD (PDB ID: 3ERT) in the complex with **4-OHT** (grey). (c) Apo conformation (PDB ID: 1A52) in the complex with estradiol **E2** (yellow). Important helices are highlighted: H3 (blue), H5 (orange), H6 (grey), H11 (green) and H12 (red).

Similarly, the apo-form of the human estrogen receptor hER α (PDB ID: 1A52 (see Fig. 2c) employs this type of extended conformation,²⁰ where H12 interacts with the other monomer of the dimer binding site. Moreover, H11 and H12 in two neighboring monomers bond *via* disulphide bonds. Therefore, H12 in the first monomer interacts with the binding site of the second monomer to result in these cross-monomer interactions.^{21,22}

SERMs, such as **4-OHT**, can deactivate the estrogen signaling pathway *via* competitive binding to the ER, causing a conformational change in the subsequently formed ER dimer. This involves the shifting of H12 to an adjacent coactivator site (AF-2), thus blocking the binding of the co-activator; this significantly reduces the level of estrogen-regulated gene transcription.¹⁶

Antiestrogens

4-OHT has been used as a front-line endocrine therapy for breast cancer in pre-menopausal and post-menopausal women for the last 40 years. Moreover, it is used in the treatment of male breast cancer.²³ At the hER α binding site, **4-OHT** occupies the same hydrophobic binding pocket as **E2**, involving the helices H3, H6, H8 and H11 (see Fig. 3). Similar to the A-ring of **E2**, the phenolic hydroxyl group of **4-OHT** interacts with Glu353 and Arg394.²⁴ The dimethylaminoethoxy group, which is the side-chain of **4-OHT**, lies in a narrow channel between H3 and

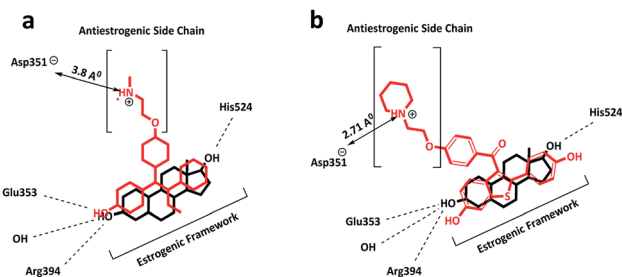


Fig. 3 Schematic showing the overlay of the binding modes of (a) E2 with 4-OHT and (b) E2 with raloxifene in the hER α -LBD. The amino acids that interact with E2 and SERMs and the hydrogen bonding formation are shown.

H11, and the tertiary amine of the chain is placed near an amino acid residue, Asp351.¹⁶ These strong interactions prevent the hydrophobic inner surface of H12 from entering the region and folding over the binding pocket, thereby disrupting the coactivator surface and forcing the H12 orientation towards an open/antagonist conformation. For this reason, the majority of SERMs possess an alkylaminoethoxy side-chain that contributes to the blocking of the transcription of estrogen-dependent genes in breast tissue.²⁵

The interactions of the side-chain terminal in 4-OHT with the Asp351 amino acid residue in the LBD differ from those in other SERMs, such as the raloxifene side-chain, as the alkylaminoethoxy side-chain is significantly stronger in raloxifene as compared to that in 4-OHT. The side-chain adopts a position much closer to the Asp351 residue, with a distance of 2.7 Å (as compared to 3.8 Å in other SERMs) and this contributes to the improved shielding of Asp351 from H12 binding and an increased antagonistic effect.

The effects of side-chain and Asp351 amino acid interactions on the enhanced antagonistic properties were further demonstrated by amino acid substitution experiments.²⁶

The mutation of Asp351 to glutamate results in an increased distance between the piperidine nitrogen and the protein residue, which subsequently results in an increase in the agonist effect.²⁶

On the other hand, SERMs, such as 4-OHT and raloxifene, share the same estrogenic framework binding mode in hER α and act as antagonists in some tissues, whereas they show agonist properties in others, as presented in Table 1.²⁷ Fulvestrant is a selective estrogen receptor down-regulator (SERD), a type of inhibitor with a bulky hydrophobic alkyl-sulfinyl side chain that binds to hER α and causes protein degradation. It is

Table 1 Estrogen behavior of various ligands in different tissues based on preclinical studies

Compound	Uterus	Bone	Cholesterol
Estradiol ²⁸	Agonist	Agonist	Agonist
Tamoxifen ²⁸	Partial agonist	Agonist	Agonist
Raloxifene ²⁷	Antagonist	Agonist	Agonist
Fulvestrant ²⁹	Antagonist	Antagonist	Antagonist

used to treat estrogen receptor-sensitive breast cancer, along with older classes of drugs and aromatase inhibitors.²⁸ Fulvestrant has been found to alter the antagonistic behaviour of full antiestrogen as compared to the previously described SERMs. Fulvestrant causes the complete disruption of the AF2 domain and deactivation of the AF1 domain; moreover, it does not demonstrate agonistic behaviour in any tissue as compared to other SERMs (see Table 1).²⁹

Although 4-OHT is a front-line treatment for breast cancer, resistance and increased risk of endometrial and uterus cancers are important clinical problems that have been associated with 4-OHT.^{30,31} Therefore, novel small molecule inhibitors with the ability to overcome antiestrogen resistance while limiting the adverse side effects are valuable pharmaceutical targets. This study describes new approaches to obtain these inhibitors through the incorporation of imine derivatives with alanine side chain functionalities into the antiestrogen scaffolds to generate analogues of 4-OHT. In this study, novel benzophenone imines (BIs) were designed by replacing the triarylethylene estrogenic scaffold found in 4-OHT with Schiff base triarylimine derivatives. On the other hand, the antiestrogenic tail *O*-alkyl side chain in 4-OHT was replaced by incorporating the alanine amino acid side chain functionality into the triarylimine scaffolds.

Recently, bisphenols,³² cyclic amides,³³ diphenylamines,³⁴ and Schiff bases³⁵ have been studied as estrogen receptor ligands. Schiff bases were first designed as selective ligands for the estrogen receptors α and β , where they behaved as ER α agonists and ER β antagonists.³⁵ However, the development of Schiff bases as hER α antagonists through the incorporation of side chains to mimic the 4-OHT backbone has never been reported. The presence of alkyl chains of varying lengths in tamoxifen and various SERMs has been shown to increase the activity of Schiff bases as selective estrogen receptor down-regulators (SERDs), leading to enhanced antagonist effects in MCF-7 cells and anti-breast cancer activity.^{36–38}

In this article, we report the binding interactions of newly designed Schiff bases with hER α using molecular docking and molecular dynamics simulations. In addition, we explored the stability and structural changes of the newly designed ligands following their complexation with three hER α forms, *i.e.* agonist, antagonist and apo conformations, to provide new information that might be useful in the development of new inhibitors with improved anti-estrogenic properties to treat breast cancer.

Methods

Strategies for the design of new benzophenone imines

The development of new BI hybrids with strong affinity for hER α could be achieved using core scaffolds that mimicked the binding interactions of known antiestrogens. In this study, the non-steroidal SERM triphenylethylene backbone of 4-OHT was chosen as a suitable scaffold due to its high affinity for hER α . The triarylethylene framework in tamoxifen, ospemifene, toremifene and many SERMs forms a backbone, which is responsible for mimicking the effect of natural estrogen, that can bind to the

hER as an agonist or antagonist.³⁹ The triarylethylene framework acts as an estrogen agonist, and the two alkyl chains attached to this framework are responsible for its antagonist behavior.^{39,40} Based on these observations, the contributions of polar imine functional groups in the presence of an alanine amino acid side chain (see Fig. 4) towards the anti-breast cancer activities of the novel triarylimines were examined.

Molecular preparations of the hER α structures and ligands

The X-ray crystal structures of apo, agonist and antagonist human estrogen receptor (hER α), PDB ID: 1A52, 1G50 and 3ERT were retrieved from the Protein Data Bank (<http://www.rcsb.org>). In the protein preparation step, all the water molecules were removed except for those involved in the stabilization of the protein–ligand complex. The ligands, *i.e.* 17 β -estradiol (**E2**) bound to apo 1A52, **4-OHT** bound to the antagonist 3ERT and **E2** bound to agonist 1G50, were removed from the crystal structures of the host molecules. Prior to conducting the docking calculations using Autodock 4.2.6, hydrogen atoms were added to the protein structures, and the missing residues of the proteins were fixed. Then, the hydrogen bonds were optimized to relax the strains using the AMBER FF99SB-ILDN force field⁴¹ program with the steepest descent algorithm for 1000 steps. The 3-D structures of the ligands were built using the Gaussview program,⁴² and the energy minimizations for the ligands were performed using the semiempirical PM3 method⁴³ available in Gaussian 2009.⁴⁴ In the second docking protocol, the preparation of proteins was carried out using the protein preparation utilities of Discovery Studio Client

16.1.0. Similarly, all water molecules were removed except for those involved in the stabilization of the protein–ligand complex. Hydrogen atoms were added to the protein structure to obtain the correct ionization and tautomeric states. The active site was selected around the co-crystal ligand within the protein structure using the Make Receptor 3.3.0 tool in OpenEye Suite.⁴⁵ In the ligand preparation step, the structures of all nine benzophenone imine derivatives were generated using ChemDraw Professional v15, and energy minimizations of the compounds were carried out using the ligand preparation utilities available in Discovery Studio Client 16.1.0; moreover, the maximum conformers were generated using the OMEGA 2.5.1 tools of OpenEye Suite.⁴⁶

Molecular docking

The molecular docking calculations were performed using two approaches. The first approach uses a Lamarckian Genetic Algorithm available in Autodock 4.2.6.^{47,48} Docking simulations were applied for the three hER α conformations, *i.e.* 17 β -estradiol bound to the apo (1A52.pdb), **4-OHT** bound to the antagonist (3ERT.pdb), and 17 β -estradiol bound to the agonist (1G50.pdb) structure, to ensure reproducibility in predicting the binding sites, binding conformations and binding affinities. The receptors and the ligands were prepared using the Autodock Tools program. Gasteiger partial charges were computed for each molecule. Rotatable bonds were defined as free for the ligands and rigid for the receptors. The grid maps of the docking simulation for the apo system 1A52 protein were prepared using a grid box of 70 \times 70 \times 65 Å and centred on the *x*, *y*, and *z* coordinates (107.27, 13.94, and 96.38), respectively. For the antagonist system 3ERT protein, the grid maps were prepared using an 80 \times 80 \times 80 Å grid box along the axes (30.01, -1.90, and 24.46), respectively. Finally, for the agonist system, the 1G50 protein was centred on the coordinates 105.28, 15.09, and 23.94 with the grid box size of 70 \times 70 \times 65 Å, respectively. Grid spacings were set at 0.375 Å for all systems. In this study, each docked compound was derived from 100 independent docking runs that were set to terminate after a maximum of 25 \times 10⁶ energy evaluations, a maximum number of 27 000 generations, and a mutation rate of 0.02; moreover, the population size was set to use 300 randomly placed individuals. A total of 100 independent docking runs were carried out for each docking system. For the second comparative docking approach, molecular docking calculations were performed using the two utilities FRED and HYBRID within the OEDocking tool of OpenEye Suite.⁴⁹ The docking protocol was validated by redocking the crystal structures of **E2** and **4-OHT** present in the respective proteins and found to have the RMSD values lower than 2 Å. After the docking validation protocol, all nine benzophenone imines were docked into the active sites of the target proteins, and a minimum 10 poses were generated. After the completion of the docking calculations, the best poses of each compound were selected based on the lowest FRED Chemguass4 and HYBRID Chemguass4 scores. The best possible conformations were then selected and analysed using the LigPlot and Chimera (<http://www.cgl.ucsf.edu/chimera>) programs.^{50,51} The newly designed BI ligands including the

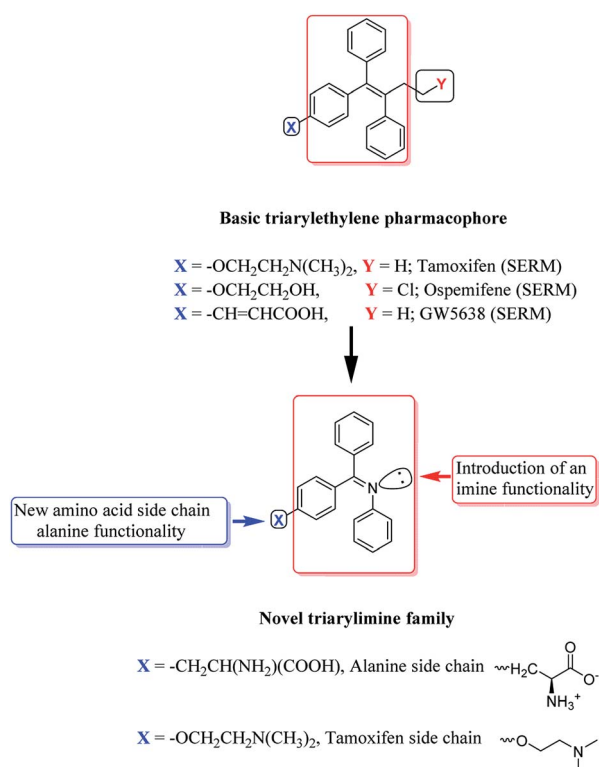


Fig. 4 The design of novel benzophenone imine inhibitors.

natural and synthetic substrates used in this study are shown in Fig. 5. The newly designed BIs were verified by the Lipinski's rule of five (RO5, molecular weight less than 500, $\log P$ or coefficient partition between -5 and 5 , less than five hydrogen bond donors and less than ten hydrogen bond acceptors),⁵² as presented in Table 2.

Molecular dynamics simulations

A 100 ns MD simulation was performed for each of the six models obtained from the best benzophenone imines BIs **5c** with the three hER α conformations apo 1A52, antagonist 3ERT and agonist 1G50, as shown in Table 3. All simulations were performed using the GROMACS 5.0.7 software package with the AMBER FF99SB-ILDN force field.⁴¹ The starting conformations of the protein–ligand were obtained from the lowest binding energy conformations *via* the docking study. The topology parameters of the hER α conformations were created using the GROMACS program, whereas the topology of ligand **5c** was generated using ACPYPE available in the Amber-Tools package.⁵³ The systems were solvated with the TIP3P water model in a cubic box with the spacing distance of 1.2 nm around the surface.⁵⁴ To neutralize the systems, counter ions were added to balance the charge of the proteins. Then, the systems were minimized using the steepest descent method for 6000 steps, followed by the Berendsen thermostat equilibration run in the NVT (constant number of particles, volume and temperature) ensemble for 200 ps at 300 K.⁵⁵ Then, the production runs were performed using the Parrinello–Rahman barostat in the NPT ensemble (constant number of particles, pressure and temperature) for 1 ns at 1 bar and 300 K.⁵⁶ After the temperature and pressure adjustments, MD simulation runs were performed for the six different models for 100 ns each (see Table 3). The simulations were performed at 1 bar and 300 K without position restraints on the solute. The cutoffs for the Coulomb and van der Waals interactions were set to 12 Å and updated every 2 fs. The particle mesh Ewald (PME) method was used to correct the electrostatic interactions.⁵⁷ The LINCS algorithm was used to constrain the bonds with hydrogen atoms.⁵⁸ All simulations were computed with a time step of 2 fs, and the coordinates were obtained every 500 steps. Molecular graphics images were produced using VMD⁵⁹ and PyMOL.⁶⁰ The

Table 2 Computed standard properties of 4-hydroxytamoxifen (4-OHT) and benzophenone imine (BI) derivatives based on the Lipinski's rule of five

Ligand	Molecular weight	H-Bond donors	H-Bond acceptors	$\log P$
4-OHT	387	1	3	4.28
1c	360	4	5	3.51
2c	378	4	5	3.65
3c	394	4	5	3.39
4c	374	4	5	3.82
5c	376	5	6	3.22
6c	344	3	4	3.81
7c	284	5	5	1.76
8c	360	1	4	4.50
9c	428	4	5	4.64

graphs were prepared using the xmgrace software (<http://plasma-gate.weizmann.ac.il/Grace/>). The results of the MD simulation trajectories were analysed using root mean square deviation (RMSD), root mean square fluctuation (RMSF), structure stability, transition path and hydrogen bond analysis.

Free energy calculations

Free energy calculations were performed using the molecular mechanics Poisson–Boltzmann surface area (MM-PBSA) method available in the GROMACS software package prepared using the *g_mmpbsa* tool.⁶¹ In this study, the last 20 ns of all the simulations of the complexes were chosen as the equilibrium part of the trajectory for energy analysis. MM-PBSA was applied to predict the average binding free energies using a Python script, *MmPbSaStat.py*, available in the *g_mmpbsa* package. Moreover, two output files, *summary_energy.dat* and *full_energy.dat*, were obtained; *summary_energy.dat* contains the average and standard deviations of all energetic components including the binding energy, polar solvation energy, ΔG_{polar} , solvent-accessible surface area (SASA), $\Delta G_{\text{nonpolar}}$, electrostatic interaction, ΔE_{elec} , and van der Waals interaction, E_{vdw} , whereas *full_energy.dat* contains the values of the energetic terms as a function of time, which have been plotted using the xmgrace software. On the other hand, to calculate the average contribution of the residues to the binding energy, the Python script *MmPbSaDecomp.py* was

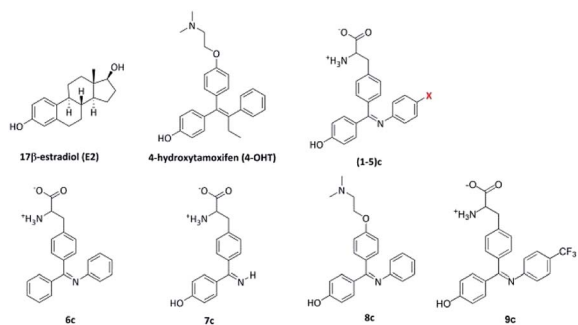


Fig. 5 Structures of 17 β -estradiol E2, synthetic estrogen 4-OHT and the newly designed ligands used in this study, where X for 1, 2, 3, 4 and 5 is H, F, Cl, CH₃ and OH groups, respectively.

Table 3 The six different models used in 100 ns MD simulations for each model of apo 1A52, antagonist 3ERT and agonist 1G50 with and without ligand **5c**

Model no.	Initial conformation	Ligand
Ap	Apo_1A52	None
ap_c	Apo_1A52	5c
An	Antagonist_3ERT	None
an_c	Antagonist_3ERT	5c
Ag	Agonist_1G50	None
ag_c	Agonist_1G50	5c

used, and the results, including the binding energy for each residue, were plotted using the *energyMapIn.dat* file with the *xmgrace* software. Furthermore, the *energy2bfac* tool was used to visualize the energy contribution of each residue with its structure using the VMD program.

Results and discussion

Docking studies

Molecular docking of benzophenone imines (BIs) (1–9)c was studied with the three hER α forms *i.e.* apo, antagonist and agonist binding sites to compare the binding interactions and free binding energies of the benzophenone imine derivatives with those of the three receptors (see Table 4). It is interesting to observe that the binding affinities of BIs (1–6)c are higher than those of natural E2 and synthetic 4-OHT in the apo and

antagonist models except for the diarylimine 7c and the *O*-alkyl chain scaffold 8c, which have lower binding affinities.

Table 4 shows that BIs (1–6)c, 4-OHT and E2 display single cluster docking conformations, which means that the ligands are docked in the same orientation. On the other hand, the diarylimine 7c and the *o*-alkyl chain scaffold 8c showed multiple cluster docking conformations (see Fig. 7). A comparison of the docked poses of 100 conformations of E2 at the apo hER α binding site with its original crystal structure is shown in Fig. 6a. It can be observed that E2 engages in hydrophobic interactions along with hydrogen bonding interactions with Arg394, Glu353 and His524. The docked pose of 4-OHT with apo hER α is shown in Fig. 6b.

It has been observed that the estrogenic scaffold in 4-OHT has the same binding mode as that of E2; moreover, the *O*-alkyl side chain in 4-OHT is responsible for the interaction

Table 4 Binding free energies (in kcal mol⁻¹), inhibition constants (in nM) and number of clusters of the three hER α conformations interacting with E2, 4-OHT and benzophenone imines (1–9)c

Ligand	Autodock		Calculated K_i	FRED		HYBRID	
	Cluster	Binding energy, ΔG		Binding energy, ΔG	Binding energy, ΔG		
Apo hERα-ligand							
E2	100	-9.67	81.36	-16.53	-16.83		
4-OHT	100	-10.30	28.06	-15.24	-14.37		
1c	100	-11.67	2.81	-18.52	-10.34		
2c	100	-11.68	2.73	-12.32	-10.43		
3c	100	-11.30	5.22	-12.32	-13.84		
4c	100	-11.59	3.21	-17.32	-10.06		
5c	100	-11.89	1.91	-18.16	-15.14		
6c	100	-11.28	5.43	-16.46	-10.04		
7c	92	-9.39	131.2	-12.80	-10.72		
8c	96	-11.18	5.16	-16.13	-10.36		
9c	—	—	—	-14.81	-12.70		
Antagonist hERα-ligand							
E2	100	-10.23	27.86	-14.69	-14.95		
4-OHT	100	-10.84	11.40	-18.02	-17.72		
1c	100	-10.29	28.84	-16.92	-15.77		
2c	100	-10.30	28.06	-13.03	-9.67		
3c	100	-10.64	15.77	-16.74	-16.18		
4c	100	-10.85	11.14	-16.58	-16.49		
5c	100	-10.92	9.95	-17.91	-17.32		
6c	100	-9.81	63.99	-15.18	-13.75		
7c	92	-7.86	1730	-12.92	-12.12		
8c	82	-9.99	47.96	-17.60	-17.48		
9c	—	—	—	-16.40	-14.44		
Agonist hERα-ligand							
E2	100	-10.81	12.00	-18.31	-18.31		
4-OHT	48	-7.31	4350	-8.95	-6.71		
1c	13	-6.35	22 180	-10.63	-9.82		
2c	6	-5.52	90 310	-8.75	-8.87		
3c	37	-5.03	207 180	-9.03	-9.18		
4c	47	-5.14	170 180	-9.44	-9.19		
5c	19	-5.00	217 020	-9.57	-9.59		
6c	18	-6.43	19 210	-12.28	-9.91		
7c	11	-7.40	3770	-14.01	-11.62		
8c	13	-8.12	1120	-10.55	-9.33		
9c	—	—	—	-7.88	-7.77		

with Asp351, which is responsible for the antiestrogen properties. For the newly designed Schiff bases, the hydroxyl groups in all BI derivatives except **6c** form hydrogen bonding interactions with the Glu353 and/or Arg394 amino acid residues, thus mimicking the binding modes of **E2** and **4-OHT**. Moreover, the hydroxyl group in the imine aromatic ring forms a hydrogen bond with His524 at the other site of the LBD. On the other hand, the carboxylic acid groups in the **1–7c** side chains, along with the *N*-alkyl side chain in **8c**, form hydrogen bonds with Asp351. The best docked conformer, **5c**, formed a hydrogen bond with Asp351 that was shorter (2.13 Å) than the bond observed in **4-OHT** (3.29 Å, see Fig. 6c); this indicated that the BI derivatives might form stronger hydrogen bonds with Asp351 and decrease the agonistic activity.

This means that triarylimines in the **1–9c** framework along with triarylethylene in **4-OHT** act as agonists mimicking the effect of the natural estrogen **E2**, which can bind to ER as an agonist or antagonist. On the other hand, the two side chains, alanine in **1–7c** and the *O*-alkyl chain in **8c**, attached to triarylimine along with the *O*-alkyl side chain in **4-OHT** are responsible for its antagonist behaviour. In addition, the docking studies of the BI derivatives revealed that polar (Glu353, Arg394, His524, Asp351 and Lys529), aromatic (Trp383 and Phe404) and non-polar (Leu346, Leu349, Leu384, Leu391, Leu525, Met388, Ala350, Thr347, Trp383, Phe404 and Gly521) amino acid residues played important roles in the stabilization of the hER α -BIs complexes, as presented in Table 5.

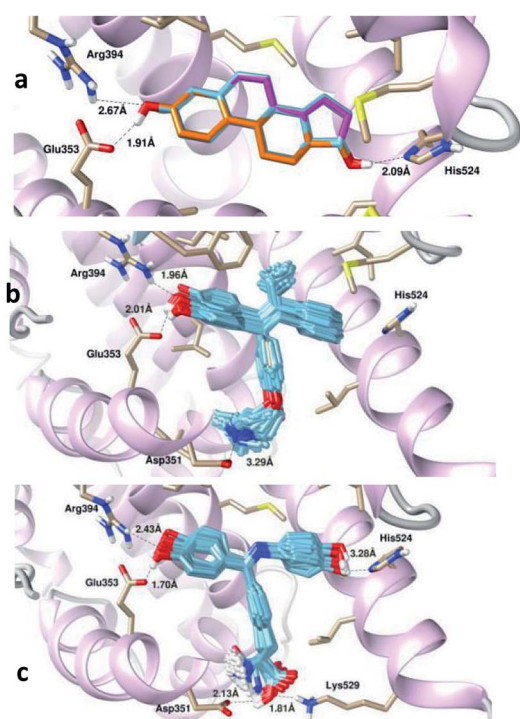


Fig. 6 Superimpositions of the 100 conformations of (a) **E2**, (b) **4-OHT** and (c) **5c** docked in the binding pocket of the apo hER α complex. The important bond lengths involved in the hydrogen bond formation are highlighted.

Interestingly, the docking poses for BIs **1–6c** display a single mode interaction with the antagonist hER α -LBD, as shown in Fig. 7b, which confirms the selectivity of the apo hER α -**1–6c** clusters. This can be attributed to fact that the newly designed triarylimines with alanine side chains display significant structural fitting to the apo and antagonist binding sites, thus mimicking the effects of the triarylethylene and *O*-alkyl side chain frameworks in **4-OHT**.

The docking results of the antagonist model show that the binding free energies of BIs **1–6c** to antagonist hER α are in the same range as that of **4-OHT**, -10.84 kcal mol $^{-1}$, whereas **5c** has the lowest binding free energy of -10.92 kcal mol $^{-1}$. The antagonist hER α -**7c** shows a higher binding energy of -7.86 kcal mol $^{-1}$, and this is in agreement with the apo hER α -**7c** complex, which also shows the highest binding energy of -9.39 kcal mol $^{-1}$ but with multiple clusters.

The results were also compared with those of other docking protocols, *i.e.* FRED and HYBRID Chemgauss4, and the results are tabulated in Table 4. Generally, the results obtained from the three different docking protocols showed different trends for the binding affinities of all the investigated ligands. However, the three docking protocols agree that the antagonist hER α -**5c** has highest binding affinities. FRED performs a systematic and non-stochastic examination of all possible protein–ligand poses and filters for shape complementarity and chemical feature alignment before selecting and optimizing the poses using the Chemgauss4 scoring function.⁴⁹ The HYBRID program, on the other hand, uses the information present in both the structure of the protein and the bound ligand to enhance the docking performance. The results are also consistent with the calculated binding energies obtained using Autodock, which show that the binding affinities of the apo and antagonist hER α complexes are stronger than that of the agonist complex.

It has been observed that **4-OHT** is engaged in hydrogen bonding interactions with the Glu353, Arg394 and Asp351 amino acid residues (see Fig. 8a), whereas **5c**, which has two hydroxyl groups, forms four hydrogen bonds with Glu353, Arg394, His524 and Asp351, as shown in Fig. 8b. Previously, Celik and his co-workers highlighted the importance of hydrogen bond formation with the Glu353 and Arg394 amino acid residues in their docking study on **E2** and **4-OHT**.⁶² It was observed that BIs formed hydrophobic contacts mainly with leucines (Leu346, Leu384, Leu387, Leu391 and Leu525) and methionines (Met343, Met388 and Met421). Other amino acid residues, such as Thr347, Ala350, Trp383, Gly521 and Phe404, also form hydrophobic contacts with **5c**, as shown in Table 5. The ligand forms four hydrogen bonds with Glu353, Arg394, His524 and Asp351. In both the apo and the antagonist models, **5c** and **7c** reveal the lowest hydrophobic interactions due to the hydrophilic nature of **5c**, whereas the small diaryl ligand **7c** contributes to fewer interactions with amino acid residues at the binding site. On the other hand, **4c** has the highest hydrophobic interactions due to the hydrophobic methyl group in the imine aryl ring.

Furthermore, the side chain tails in the BIs **1–7c** ligands are highly flexible, as noted from the docking poses with the

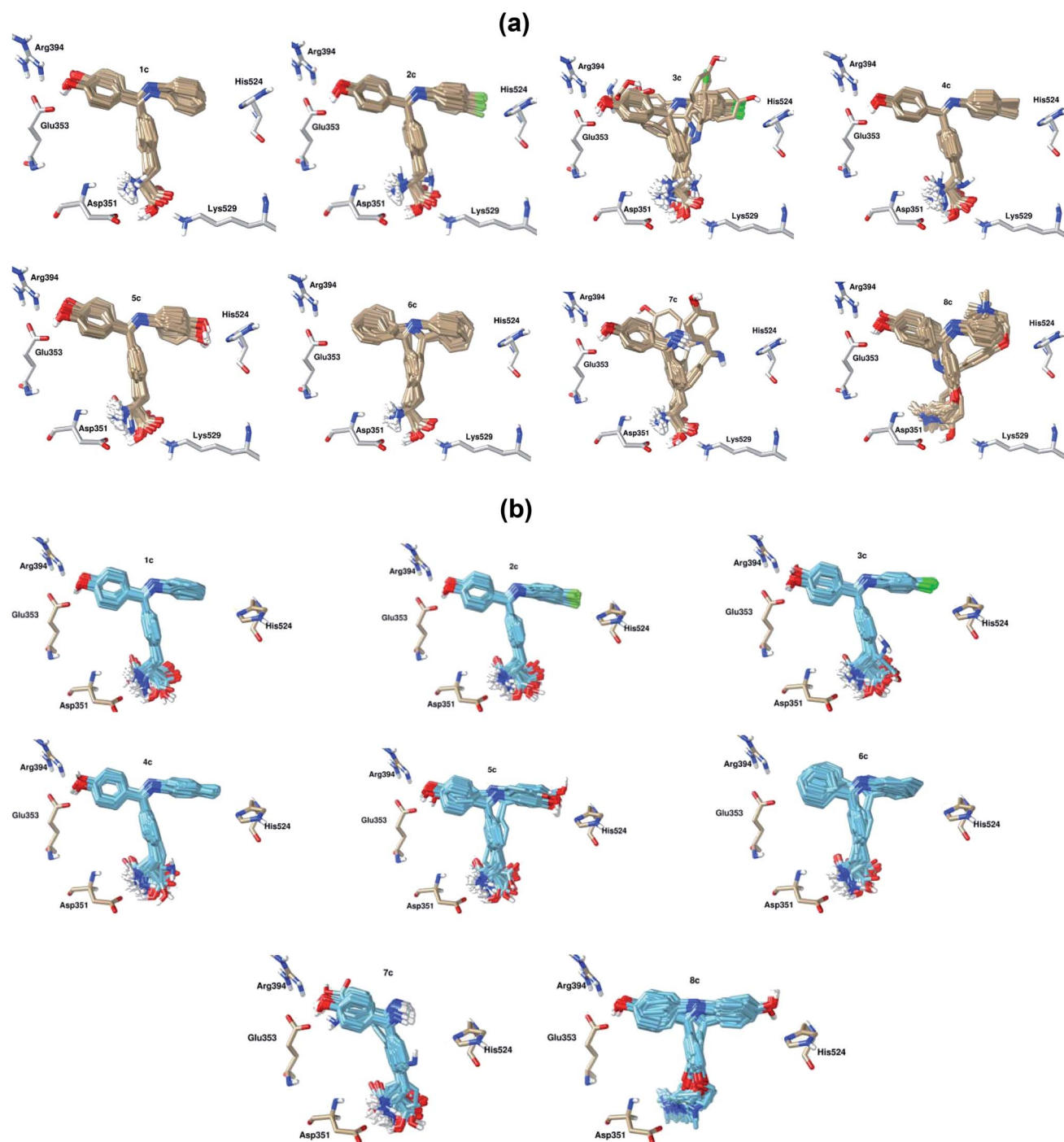


Fig. 7 Superimpositions of 100 conformations for BIs 1–6c docked in the same orientation in (a) apo hER α -BIs and (b) antagonist hER α -BIs. 6c and 7c show multiple cluster docking conformations in both the apo and the antagonist models. Important amino acid residues are highlighted.

antagonist hER α -LBD as compared to the case of the apo system, as shown in Fig. 7. The NH₂ groups in the side chains of BIs 1–7c form hydrogen bonds with the Asp351 amino acid residue in both the apo and the antagonist systems. Moreover, the carboxylic acid group (COOH) forms a hydrogen bond with the Lys529 amino acid residue in the apo system only. This is due to the difference in the Lys529 amino acid positions in the apo and antagonist systems. Lys529 is located at the end of H11 in the apo system and faces the binding site. However, Lys529 in

the antagonist system points out of the binding pocket, thus preventing the formation of hydrogen bonds with the COOH group in the BI ligands. The agonist hER α forms stable complexes with small ligands such as E2, and this allows H12 to cover the binding site and restricts the movement of the ligand. The natural and synthetic substrates E2 and 4-OHT were redocked to the hER α agonist binding site to compare the interaction energies and cluster distributions in the close LBD towards the bulky BI SERMs-like ligands, and the results are

Table 5 Hydrophobic residues of the apo 1A52 and antagonist 3ERT hER α conformations interacting with E2, 4-OHT and BIs (1–9)c

Ligand	Hydrophobic interactions	
	Apo hER α -ligand	Antagonist hER α -ligand
E2	Phe404, Met388, Ile424, Gly521, Leu384, Leu387, Leu391, Leu428, Leu525	4-OHT Leu346, Leu384, Leu387, Leu391, Leu525, Met343, Met421, Thr347, Ala350, Trp383, Gly521, Phe404
1c	Leu346, Leu384, Leu387, Leu391, Leu525, Trp383, Gly521, Met388, Met421, Ala350, Thr347, His524	Leu346, Leu384, Leu387, Leu391, Leu525, Met343, Met421, Thr347, Ala350, Trp383, Gly521
2c	Leu346, Leu384, Leu387, Leu391, Leu525, Trp383, Gly521, Met388, Met421, Ala350, Thr347, His524, Phe404	Leu346, Leu384, Leu387, Leu391, Leu525, Met343, Met421, Thr347, Ala350, Trp383, Gly521
3c	Leu346, Leu349, Leu384, Leu525, Gly521, Met388, Met421, Ala350, Thr347, Phe404	Leu346, Leu384, Leu391, Ile424, Met421, Thr347, Ala350, Trp383, Gly521
4c	Leu346, Leu349, Leu384, Leu525, Ile424, Met388, Met421, Ala350, Thr347, Trp383, Phe404, His524, Gly521	Leu346, Leu384, Leu391, Leu525, Ile424, Met343, Met388, Met421, Thr347, Ala350, Trp383, Gly420, Gly521, His524
5c	Leu346, Leu349, Leu384, Leu391, Leu525, Met388, Ala350, Thr347, Trp383, Phe404, Gly521	Leu346, Leu349, Leu525, Met343, Ala350, Thr347, Trp383, Gly521
6c	Leu346, Leu387, Leu384, Leu391, Leu525, Ala350, Gly521, Met388, Met421, Trp383	Leu346, Leu349, Leu391, Ala350, Thr347, Trp383, Gly521, Met343, Met421, Phe404
7c	Leu346, Leu349, Leu384, Leu387, Leu391, Leu525, Ala350, Thr347, Trp383, Phe404	Leu346, Leu384, Leu387, Ala350, Thr347, Trp383
8c	Leu346, Leu354, Leu384, Leu387, Leu391, Leu525, Met388, Met421, Thr347, Ala350, Gly521, Trp383	Leu346, Leu384, Leu387, Leu391, Leu525, Met343, Met421, Thr347, Ala350, Trp383, Gly521
9c	Ala350, Asp351, Glu353, Leu387, Leu391, Ile424, Met421, Gly521, His524, Leu525	Met343, Leu346, Thr347, Ala350, Glu353, Leu387, Leu391, Leu384, Glu421, Ile424, Met421, Gly521, His524, Leu525

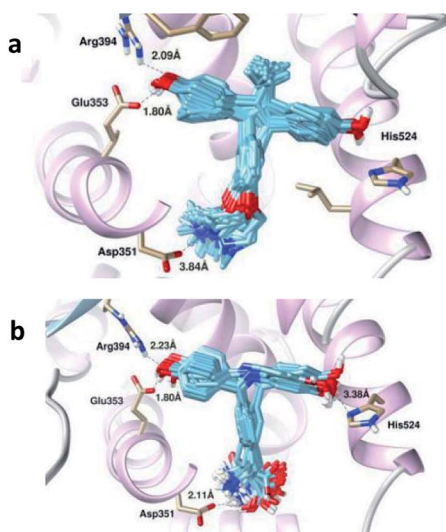


Fig. 8 Superimpositions of the 100 conformations of (a) 4-OHT and (b) 5c docked in the binding pocket of the antagonist hER α complex. The important bond lengths involved in the hydrogen bond formation are highlighted.

tabulated in Table 4. By comparing the binding energies between E2, 4-OHT and the BIs, we have found that E2 forms single clusters in the LBDs of all three hER α forms (see Fig. 9a). Moreover, both 4-OHT and BIs exist in multiple clusters in the agonist form and generally have higher binding energies as compared to E2. In addition, the interactions of the BIs and 4-OHT in the hER α binding pocket involve hydrophobic interactions with the helices H3 (blue), H6 (grey) and H11 (green) (see Fig. 9b). On the other hand, the synthetic antagonist 4-OHT and the newly designed BIs interact with the open/apo and open/antagonist hER α with higher binding affinities. This suggests that the binding poses of the newly designed BIs adopt 4-OHT-like modes in the binding pocket of the antagonist hER α .

Molecular dynamics studies

The flexibility of the hER α binding site is an important factor to investigate the conformational changes of the apo, antagonist and agonist forms of hER α upon its binding with BIs. Molecular dynamics (MD) simulations have provided insights into the dynamic fluctuations of the hER α apo, antagonist and agonist forms upon binding with natural E2 or synthetic 4-OHT.^{62–65}

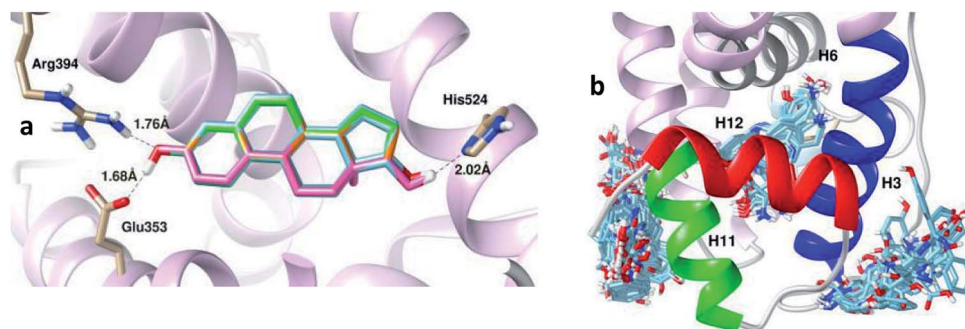


Fig. 9 Superimpositions of the 100 conformations of (a) E2 and (b) 5c docked in the binding pocket of the hER α agonist complex.

Furthermore, the dynamics of H12 at the binding site in the apo and antagonist systems have been reported to be profoundly different.^{63,65}

Simulations were conducted on each of the six different models to determine the best candidate among the benzophenone imines, 5c, with the three hER α conformations. Previously, Fratev investigated the transition of the apo form to either the agonist or the antagonist state,⁶³ whereas Celik and co-workers investigated the binding of E2 in the presence and absence of co-activator proteins.⁶²

Herein, we report the results of the simulations of the apo state to study its behavior as an open binding site towards 4-OHT-like ligands and compare their stabilities with the antagonist and agonist forms. The RMSD values showed that the antagonist and agonist models did not reach stable RMSD values until 20 ns. This is in agreement with previous MD simulations involving the agonist and antagonist forms of hER α , which shows stable RMSD values only after 20 ns.⁶⁵ The average and maximum RMSDs of 2.8 and 4.0 Å for the antagonist form were larger than those for the agonist form, which had the values of 1.8 and 2.8 Å, respectively (see Fig. 10). Moreover, relatively large fluctuations of the antagonist form as compared to those of the agonist form were observed in the longest MD simulations reported to date⁶⁵ and in the 5 ns MD simulations reported by Celik and co-workers.⁶² The larger RMSD value fluctuation from 0 to 35 ns in the free and bound apo forms appears to have occurred due to the large fluctuation in the extended H12 region, and the values started to stabilize after 35 ns until the end of the simulation. These results demonstrate that 5c promotes the stability of the apo hER α complex as compared to the free apo protein. This is also in agreement with a previous study, which states that the apo ER α monomer exhibits high conformational flexibility with respect to H12, thus affecting the stability of the overall hER α structure as compared to the folding of the H12 conformation in the antagonist and agonist forms.⁶³ Upon comparing the RMSD values for the three systems, the results suggest that the complexation of 5c to the apo and antagonist hER α is more favourable and provides stability to the protein structure. On the contrary, the close/agonist hER-5c complex has higher RMSD values as compared to its free agonist form due to the position of H12

that covers the binding site and restricts the movement of the ligand. Variation in the gyration radii (R_g) values of the three systems was expected due to the differences in the folding structure of H12 (ESI Fig. S1[†]). H12 in the apo free and complex forms is unfolding and extending away from the binding site. However, in the antagonist conformation, H12 shifts to an adjacent coactivator site. In the agonist form, H12 covers the ligand binding site with the highest level of folding as compared to the apo and antagonist forms.

Therefore, R_g of the agonist was stable from the beginning to the end of the simulation. Moreover, the R_g values for the antagonist model continued to fluctuate until 60 ns.

For the extended apo state, the R_g values started to fluctuate between 2.03 and 1.85 nm during the simulation time. The fluctuations of the free and complex systems were high at the beginning of the simulation time and then started to decrease to around 1.9 ± 0.05 nm. This result supports the changes observed in the RMSD plot, which show high fluctuations at the beginning of the simulation time.

To illustrate and compare the flexibilities and conformational changes of the three free and bound models, the RMSFs were analyzed, and the results are presented in Fig. 11. The RMSFs of the free and complex antagonist systems are similar to those of the apo system. However, lower flexibility regions in the beginning for H4 and H10 were observed in the antagonist system. On the other hand, the higher regions of the antagonist systems as compared to those of the apo system involve the Asp531 amino acid residue in the end of H11. In general, the fluctuations in the free and complex forms of the apo system are quite similar to each other and to those in the antagonist systems, in which the fluctuation profile of the free antagonist form is similar to that of its complex system.

However, the agonist complex differs greatly from the apo and antagonist open systems. Further analysis reveals that most of the amino acid residues in the agonist complex have highest RMSF values throughout the simulation period. This can be explained by the close LBD of the agonist estrogen receptor conformation with H12, which restricts the ligand movement.

As a whole, amino acid fluctuations in the antagonist forms, *i.e.*, free and complex systems, are located in the residue number 526–535 and 545–550, corresponding to the loops that

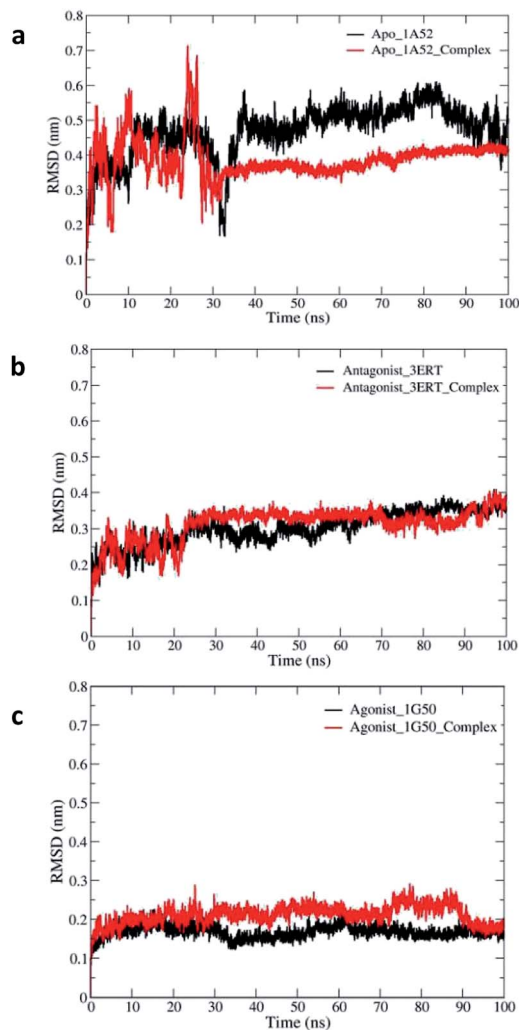


Fig. 10 The RMSDs of the backbone atoms of free and bound hER α throughout the simulation times for the (a) apo, (b) antagonist and (c) agonist models.

are attached to both H12 terminals. These fluctuations in the loops around H12 were also observed in the hER α -4-OHT simulation study.⁶⁴ A high dynamic region was also observed in a loop preceding the H9 involving residues 460–466. The dynamic behavior observed in this region has also been reported in many agonist/antagonist simulation studies.^{66,67} Simulations of the apo systems show slightly different RMSF values in the same region as compared to those of the antagonist forms. Moreover, the free agonist systems have similar dynamic regions as those of the apo and antagonist systems, with decreased fluctuations in a loop preceding H9 (residues 460–466), and this agrees well with a previous simulation study involving hER α .⁶⁵

To study the conformational flexibility and binding of the 5c ligand at the three binding sites of the apo, antagonist and agonist forms of hER α , the RMSDs for 5c in the three systems were analyzed and are presented in Fig. 12a. The calculated ligand RMSDs reached stable values after approximately 20–100

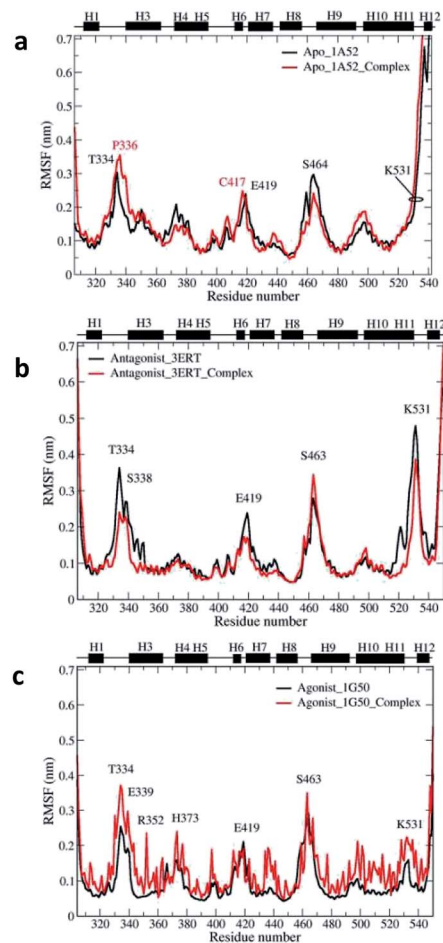


Fig. 11 RMSF profiles of the free and complex hER α throughout the simulation period for the (a) apo, (b) antagonist and (c) agonist models.

ns and stabilized at 0.1 ± 0.05 nm in both the apo and the antagonist hER α complexes.

However, in the agonist hER α complex, higher RMSDs of 0.17 ± 0.02 nm were observed for the ligand. This high variation for the ligand in the agonist system is in agreement with the increasing conformational fluctuations of the agonist receptor amino acid residues in the RMSF plot. The images showing the dynamics of 5c in the binding pocket of the apo, antagonist and agonist systems are shown in Fig. 12. The binding orientation of 5c shows relatively similar variations in the open LBD apo and antagonist forms. The amino group in the alanine side chain binds with the Asp351 amino acid residue, whereas the carboxylic acid functional group in 5c interacts with the Lys529 amino acid residue. In contrast, relatively higher variation in the alanine side chain was observed in the agonist system, as shown in Fig. 12d. In addition, the amino group in the alanine side chain along with the carboxylic acid functional group in 5c bind in opposite orientations in the agonist system as compared to that in the apo and antagonist systems. This prevents the hydrogen bond formation between 5c and the Lys529 amino acid residue in the agonist system.

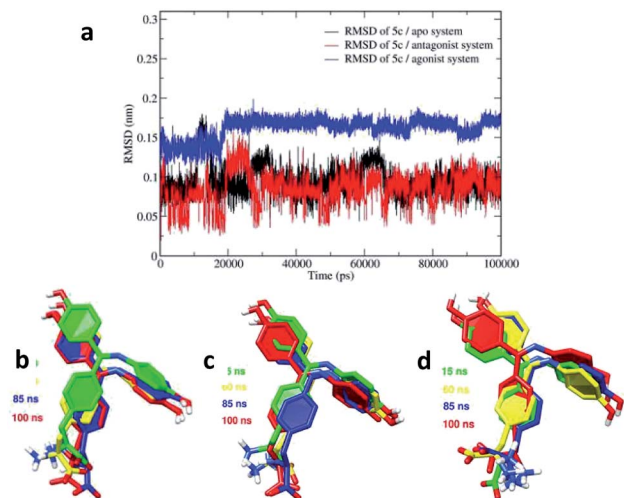


Fig. 12 Conformational changes of the 5c ligand throughout the simulation period. (a) The RMSDs of 5c in the binding pockets of the apo, antagonist and agonist forms. Ligand dynamics images in the binding pockets obtained at different simulation times for the (b) apo, (c) antagonist and (d) agonist systems.

Transition path analysis

The dynamics of the apo, antagonist and agonist hER α complexes at different simulation times were studied and are presented in Fig. 13. Further images were analyzed for the apo complex with an extended H12 conformation, as shown in Fig. 13a. The largest changes occur very quickly, involving H12, which moves from the unfolded form at the beginning of the simulation time to the folded form, and this is actually in agreement with the high fluctuation observed in the RMSD plot during the first 30 ns. Moreover, no transitions to either the agonist or the antagonist form were observed for the folding of H12 until the end of the simulation time; this was also in agreement with a previous study reported by Fratev.⁶³ The author claimed that the apo hER α monomer did not show any transition from the unfolded conformation to either an agonist or an antagonist state, and the transition could only be achieved in the dimer form.

Hydrogen bond analysis

The percentages of hydrogen bond occupancy between the three hER α forms and 5c during the simulation period were calculated using the Python script *readHBmap.py* and the *gmx h-bond* program in the GROMACS database. The hydrogen bonds were determined based on the acceptor–donor atom distances of less than 3.5 Å and the acceptor–H–donor angles greater than 120°. ⁶⁸

Table 6 shows the percentages of the occupancy of H-bonds between 5c and the amino acid residues of the hER α receptor in the three systems. The labelling of the atoms involved in the hydrogen bond formation for the three systems is shown in the ESI Fig. S2.† The highest number of amino acid residues participating in the hydrogen bond formation among the three systems was in the agonist hER α –5c complex. It was observed that the hydrogen bond formation in the agonist complex

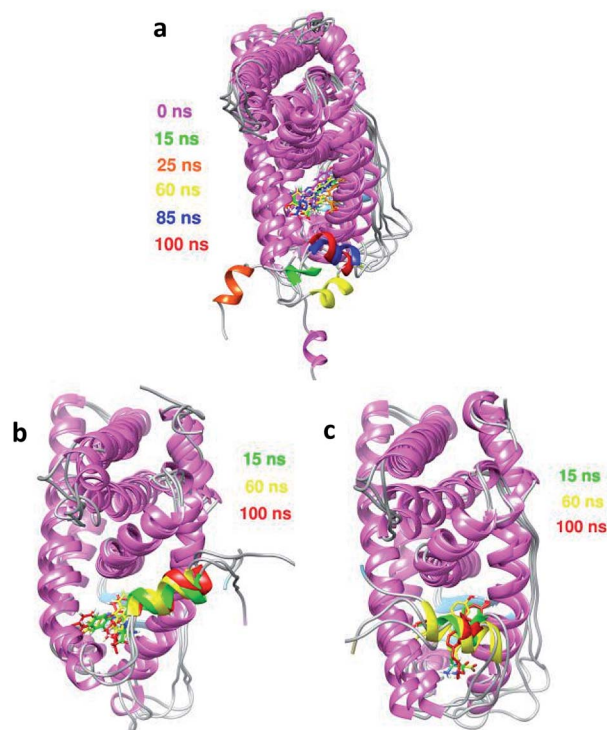


Fig. 13 Overlays of the images of the conformational dynamics taken at different simulation times for (a) apo, (b) antagonist and (c) agonist hER α bound to 5c.

involved His524, Glu353 and Thr347 with more than 50% occupancy. Sporadically, hydrogen bond formation with less than 50% occupancy was observed with Leu346, Met342, Met343, Gly344 and Val533. This result is expected for the close binding site of the agonist system, which restricts the motion of the ligand, allowing high numbers of amino acids to form hydrogen bonds with 5c, as shown in Table 6. In the apo and agonist cases, hydrogen bonding interactions with Glu353 and His524 were observed with high percentages of occupancy (96.8% and 97.3%, respectively).

In contrast, the His524 amino acid residue in the antagonist receptor repositioned to the opposite site during the simulation period, and the percentage of occupancy decreased to 2.0%. The MD results further confirmed that the hydrogen bond formation between Glu353 and 5c played a key role in the ligand–receptor interaction and was found in all three systems.

Free energy calculations

To gain an insight into the contributions of the amino acid residues to the binding energy and predict the average binding energies for the apo, antagonist and agonist complexes, free energy calculations were performed, and the obtained binding free energy components and the results are shown in Table 7. 5c is bound to the apo and antagonist receptors with the strong binding affinities of -68.68 kJ mol⁻¹ and -59.96 kJ mol⁻¹, respectively. The calculated free energies, ΔG_{calc} , of 5c towards the different hER α forms are in the order of agonist > antagonist > apo. These results are consistent with the calculated binding

Table 6 H-bond occupancy between the amino acid residues of hER α and **5c** in the three systems

Complex	Donor and acceptor	Occupation (%)
Ap_c	Glu353@OE2-5c@H3	45.6
	Glu353@OE1-5c@H3	51.2
	His524@ND1-5c@H	97.3
	Thr347@OG1-5c@H1	7.9
	5c@OXT-	22.0
	Lys529@HZ1	
	5c@O-Lys529@HZ1	22.7
	5c@O2-Arg394@H21	3.7
	5c@OXT-	1.6
	Thr374@HG1	
An_c	5c@O-Thr374@HG1	1.0
	Glu353@OE2-5c@H3	52.0
	Glu353@OE1-5c@H3	47.8
Ag_c	5c@O2-Arg394@H21	2.0
	Glu353@OE2-5c@H3	46.0
	Glu353@OE1-5c@H3	30.1
	His524@ND1-5c@H	93.2
	5c@OXT-	25.4
	Thr347@HG1	
	5c@O-Thr347@HG1	58.5
	5c@OXT-Thr347@H	17.0
	5c@O-Thr347@H	41.3
	5c@OXT-Leu346@H	33.2
5c@O-Leu346@H	8.1	
Met342@O-5c@H1	31.7	
5c@OXT-Gly344@H	26.2	
5c@O-Gly344@H	17.8	

energies obtained from the docking results in this study, which show that the binding affinities of the apo and antagonist complexes are stronger than that of the agonist complex. From the contribution of the calculated energy components of the binding free energies shown in Table 7, the main driving force for hER α -5c binding is hydrophobic nonpolar interactions. The polar interactions contributed unfavorably to the binding of the ligand to hER α .

Indeed, the electrostatic and van der Waals interactions contribute favourably towards the binding of 5c to hER α and are compensated by the large polar solvation energy.

Table 8 summarizes the calculated ΔG_{calc} values along with the experimental ΔG_{exp} values of the natural estradiol **E2** and synthetic **4-OHT** obtained from previous studies.^{69,70} Compared

Table 7 Calculated binding free energies (in kJ mol⁻¹) and their components based on the MM-GBSA method for the three hER α -5c complexes

Energy components	Apo	Antagonist	Agonist
van der Waals	-177.21	-145.19	-108.12
Electrostatic	-139.07	-100.61	-73.37
Polar solvation	267.86	201.35	148.62
SASA	-20.25	-15.51	-12.37
Total binding energy	-68.68	-59.96	-45.23

to the natural estrogen **E2** and synthetic **4-OHT** inhibitor, **5c** is bound to the hER α by strong interactions. The calculated binding affinities of the three complexes were in good agreement with the experimental values of **E2** and **4-OHT**. Surprisingly, the calculated binding affinity of the antagonist complex, -59.96 kJ mol⁻¹, is in excellent agreement with the experimental value of the antagonist hER α -**4-OHT** (-59.70 kJ mol⁻¹) obtained by Liu and his co-workers.⁷⁰ Although the calculated binding affinities of **E2** and **4OHT** with hER α obtained by Liu and his co-workers were high as compared to the experimental ΔG_{exp} values,⁷⁰ our calculated binding free energies, ΔG_{calc} , are in very good agreement with the experimental data, especially the calculated value for the antagonist complex hER α -**4-OHT** (-59.70 kJ mol⁻¹). ΔG_{calc} of the agonist hER α -**5c** complex (-45.23 kJ mol⁻¹) is consistent with the calculated and experimental binding energies, -51.21 kJ mol⁻¹ and -51.88 kJ mol⁻¹, respectively, reported by Lipzig and his co-workers.⁶⁹ The fact that the total binding energies, ΔG_{calc} , for the different hER α conformations are comparable to the experimental ΔG_{exp} values reported in the literature reflects the similarity in terms of the chemical and physical properties of newly designed Schiff base ligands, such as **5c**.

A detailed profile of the binding energy contributions was analysed using the MM-PBSA method. Fig. 14 shows the mapping of the energy contributions and the intermolecular ligand-receptor per-residue interaction spectra of the three complexes. During the initial transition to the stable hER α -**5c** forms, the energy arose mainly from the binding with residues H3, H5 and H11 in the three receptors. The hydrophobic interactions arose from the Leu525, Leu346, Leu387, Leu384, Met343, Met421 and Ala350 amino acid residues in both the apo and antagonist complexes. On the other hand, the nonpolar amino acid residues of the agonist hER α -**5c** complex interactions mainly arose from Leu346, Leu525, Leu391, Met388, Met421 and Ile424, with the strongest interactions from the hydrophobic residue Leu525. In addition, amino acid residues Glu353, Arg394, Lys529 and Asp531 made obvious polar contributions to **5c** with high binding energy values. This indicates that the polar amino acid residues and the hydrogen bond interactions destabilize the ligand-receptor interactions during the simulation. On the other hand, hydrophobic

Table 8 Comparison of the calculated (ΔG_{calc}) and experimental (ΔG_{exp}) total binding free energy values (in kJ mol⁻¹) for **E2** and **4-OHT** complexed with hER α as well as with hER α -5c

Complex	From previous studies ^{a,b}		In this study	
	ΔG_{calc}	ΔG_{exp}	Complex	ΔG_{calc}
hER α - E2	-51.21 (ref. 65)	-51.88 (ref. 65)	Apo	-68.68
hER α - E2	-127.61 (ref. 64)	-57.44 (ref. 64)	Antagonist	-59.96
hER α - 4-OHT	-166.94 (ref. 64)	-59.70 (ref. 64)	Agonist	-45.23

^a a ,⁶⁹ b ,⁷⁰ and ΔG_{calc} values were calculated using the linear interaction energy (LIE) approximation method. ^b a ,⁶⁹ b ,⁷⁰ and ΔG_{calc} values were calculated using the molecular mechanics Poisson-Boltzmann surface area (MM-PBSA) method.

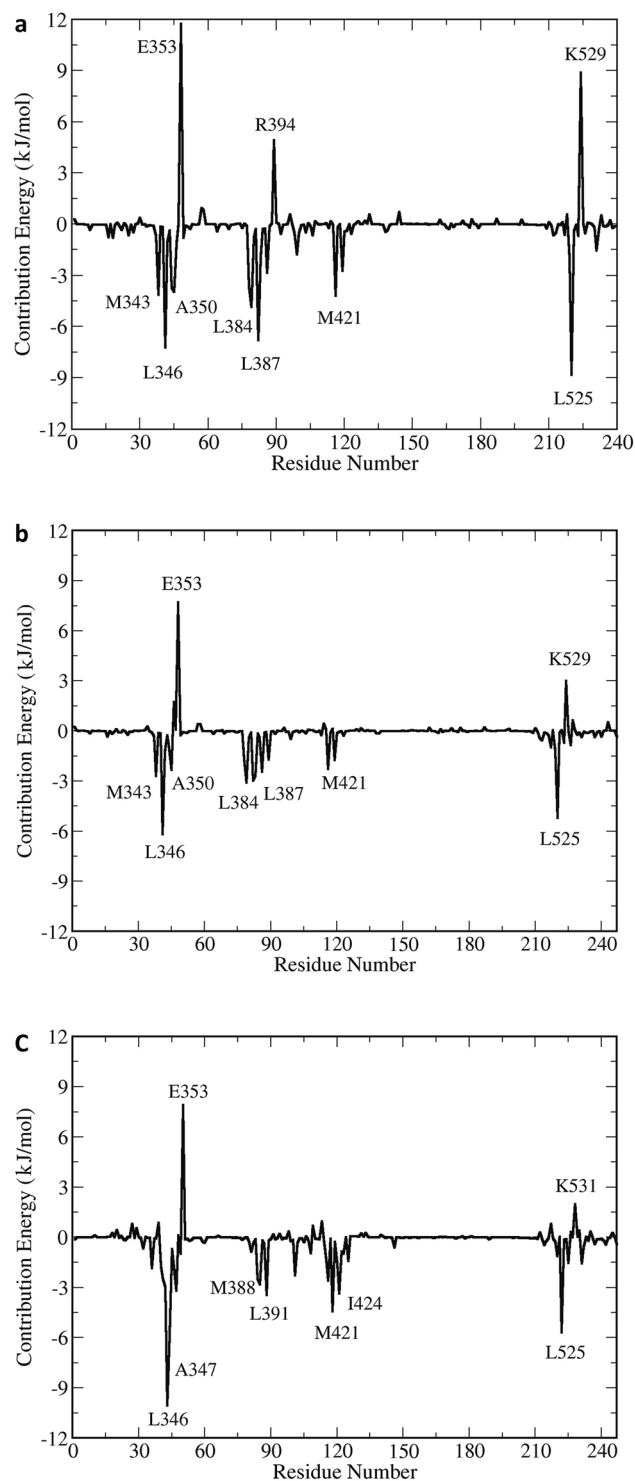


Fig. 14 The mapping of the energy contributions on the structure of hER α -5c and the intermolecular ligand-receptor spectra of the (a) apo, (b) antagonist and (c) agonist forms.

residues, especially Leu525, form strong interactions with 5c, which forms face-to-face interactions with the imine ring; also, Leu387 forms interactions with the second hydroxyl aromatic ring. This suggests that the stability of the 5c ligand in hER α is achieved *via* hydrophobic interactions.

Overall, the obtained results show that BIs can form stable complexes with hER α through the open H12 binding pockets of apo and antagonist hER α . Meanwhile, the open binding pocket of the agonist form restricts the movement of the bulky BIs, such that 5c binds to the receptor with lower affinity and selectivity.

Conclusions

Herein, the mechanisms of the binding of the newly designed benzophenone imine (BI) Schiff base ligands to the apo, antagonist and agonist forms of human estrogen receptor (hER α) were investigated using molecular docking, molecular dynamics simulations and molecular mechanics Poisson-Boltzmann surface area (MM-PBSA) energy calculations. A total of nine proposed ligands were docked, and the results demonstrated that the newly designed Schiff bases could bind to the hydrophobic open pockets of the apo and antagonist hER α conformations with high affinity, mimicking the behavior of the synthetic inhibitor 4-hydroxytamoxifen (4-OHT). Furthermore, the docking poses of the newly designed Schiff base ligands display single modes of interaction with both the open apo and antagonist hER α -LBD, as denoted by single clusters similar to the natural estradiol (E2) and the synthetic inhibitor 4-OHT. The molecular dynamics simulation results of the best docked benzophenone imine derivatives 5c with the three receptors (apo, antagonist and agonist hER α) demonstrate that 5c binds well to the hydrophobic pockets of the three systems. The flexibility and conformational change analysis based on the RMSD, RMSF and ligand RMSD values revealed stable interactions with fewer conformational fluctuations for 5c with the open apo/antagonist hER α receptors as compared to the closed agonist binding site. These results are consistent with the calculated binding energies obtained from the docking results, which show that the binding affinities of the apo and antagonist hER α -5c complexes are higher than that of the agonist hER α -5c complex. The results also prove that the newly designed Schiff base mimics the behavior of the synthetic antagonist 4-OHT. The occupancy of hydrogen bonds obtained from the MD results further confirmed that the hydrogen bond formation between Glu353 and 5c played a key role in the ligand-receptor interactions and displayed highest hydrogen bond formation among the three systems. The MM-PBSA binding free energy calculation results further confirmed the stability of the hER α -5c systems in the order of apo > antagonist > agonist. These results are consistent with the calculated binding energies obtained from the docking results. In addition, based on the contribution of the calculated energy components of the total binding free energies highlighted herein, the main driving force for the hER α -5c binding is hydrophobic nonpolar interactions. Polar interactions contributed unfavourably to the binding of the ligand. Finally, the high binding affinity for the newly designed Schiff base ligand with hER α suggests that this Schiff base T-shaped C=N family may be worth exploring in the development of new drugs for breast cancer treatment.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The authors gratefully acknowledge the financial support received from the Universiti Sains Malaysia through the USM Fellowship Scheme under the Institute of Postgraduate Studies and Ministry of Higher Education through FRGS Grant No. 203/PKIMIA/6711558. Khan thanks OpenEye Scientific Software, Inc., for a free academic license of the OpenEye Toolkits.

References

- 1 D. R. Youlden, S. M. Cramb, C. H. Yip and P. D. Baade, *Cancer Biol. Med.*, 2014, **11**, 101–115.
- 2 G. Agarwal, P. Pradeep, V. Aggarwal, C.-H. Yip and P. S. Cheung, *World J. Surg.*, 2007, **31**, 1031–1040.
- 3 G. M. Clark, *Breast Cancer*, 1995, **2**, 79–89.
- 4 W. F. Anderson, N. Chatterjee, W. B. Ershler and O. W. Brawley, *Breast Cancer Res. Treat.*, 2002, **76**, 27–36.
- 5 S. R. Johnston and M. Dowsett, *Nat. Rev. Cancer*, 2003, **3**, 821–831.
- 6 N. Platet, A. M. Cathiard, M. Gleizes and M. Garcia, *Crit. Rev. Oncol. Hematol.*, 2004, **51**, 55–67.
- 7 J.-F. Arnal, C. Fontaine, A. Abot, M.-C. Valera, H. Laurell, P. Gourdy and F. Lenfant, *Steroids*, 2013, **78**, 576–582.
- 8 M. Parker, *Biochem. Soc. Symp.*, 1998, **63**, 45–50.
- 9 L. Downey, R. Livingston and A. Stopeck, *J. Am. Geriatr. Soc.*, 2007, **55**, 1636–1644.
- 10 R. Blamey, *Eur. J. Cancer*, 2003, **39**, 286–294.
- 11 Q. Zhou and N. E. Davidson, *Cancer Biol. Ther.*, 2006, **5**, 848–849.
- 12 B. L. Riggs and L. C. Hartmann, *N. Engl. J. Med.*, 2003, **348**, 618–629.
- 13 W. Bourguet, P. Germain and H. Gronemeyer, *Trends Pharmacol. Sci.*, 2000, **21**, 381–388.
- 14 G. M. Anstead, K. E. Carlson and J. A. Katzenellenbogen, *Steroids*, 1997, **62**, 268–303.
- 15 A. M. Brzozowski, A. C. Pike, Z. Dauter, R. E. Hubbard, T. Bonn, O. Engström, L. Öhman, G. L. Greene, J.-Å. Gustafsson and M. Carlquist, *Nature*, 1997, **389**, 753–758.
- 16 S. Saha Roy and R. K. Vadlamudi, *Int. J. Breast Cancer*, 2012, **2012**, 1–9.
- 17 A. C. Pike, A. M. Brzozowski, R. E. Hubbard, T. Bonn, A. G. Thorsell, O. Engström, J. Ljunggren, J. Å. Gustafsson and M. Carlquist, *EMBO J.*, 1999, **18**, 4608–4618.
- 18 U. Egner, N. Heinrich, M. Ruff, M. Gangloff, A. Mueller-Fahrnow and J. M. Wurtz, *Med. Res. Rev.*, 2001, **21**, 523–539.
- 19 W. Bourguet, M. Ruff, P. Chambon, H. Gronemeyer and D. Moras, *Nature*, 1995, **375**, 377–382.
- 20 D. M. Tanenbaum, Y. Wang, S. P. Williams and P. B. Sigler, *Proc. Natl. Acad. Sci. U. S. A.*, 1998, **95**, 5998–6003.
- 21 M. R. Batista and L. Martínez, *Biophys. J.*, 2013, **105**, 1670–1680.
- 22 C. Watanabe, H. Watanabe and S. Tanaka, *Biochim. Biophys. Acta Protein Proteonomics*, 2010, **1804**, 1832–1840.
- 23 W.-C. Park and V. C. Jordan, *Trends Mol. Med.*, 2002, **8**, 82–88.
- 24 A. K. Shiau, D. Barstad, P. M. Loria, L. Cheng, P. J. Kushner, D. A. Agard and G. L. Greene, *Cell*, 1998, **95**, 927–937.
- 25 V. C. Jordan, *J. Med. Chem.*, 2003, **46**, 1081–1111.
- 26 H. Liu, W.-C. Park, D. J. Bentrem, K. P. McKian, A. De Los Reyes, J. A. Loweth, J. M. Schafer, J. W. Zapf and V. C. Jordan, *J. Biol. Chem.*, 2002, **277**, 9189–9198.
- 27 S. Fuqua, J. Russo, S. Shackney and M. Stearns, *Postgrad. Med. J.*, 2001, 3–10.
- 28 C. I. Lee, A. Goodwin and N. Wilcken, *Cochrane Database Syst. Rev.*, 2017, 1–59.
- 29 H. U. Bryant and W. H. Dere, *Proc. Soc. Exp. Biol. Med.*, 1998, **217**, 45–52.
- 30 A. Ring and M. Dowsett, *Endocr. Relat. Cancer*, 2004, **11**, 643–658.
- 31 S. Sommer and S. A. Fuqua, *Semin. Cancer Biol.*, 2001, **11**, 339–352.
- 32 L. Li, Q. Wang, Y. Zhang, Y. Niu, X. Yao and H. Liu, *PLoS One*, 2015, **10**, e0120330.
- 33 S. R. Stauffer, J. Sun, B. S. Katzenellenbogen and J. A. Katzenellenbogen, *Bioorg. Med. Chem.*, 2000, **8**, 1293–1316.
- 34 K. Ohta, Y. Chiba, T. Ogawa and Y. Endo, *Bioorg. Med. Chem. Lett.*, 2008, **18**, 5050–5053.
- 35 Z.-Q. Liao, C. Dong, K. E. Carlson, S. Srinivasan, J. C. Nwachukwu, R. W. Chesnut, A. Sharma, K. W. Nettles, J. A. Katzenellenbogen and H.-B. Zhou, *J. Med. Chem.*, 2014, **57**, 3532–3545.
- 36 G. Kaur, M. P. Mahajan, M. K. Pandey, P. Singh, S. R. Ramiseti and A. K. Sharma, *Eur. J. Med. Chem.*, 2014, **86**, 211–218.
- 37 G. Kaur, M. P. Mahajan, M. K. Pandey, P. Singh, S. R. Ramiseti and A. K. Sharma, *Bioorg. Med. Chem. Lett.*, 2016, **26**, 1963–1969.
- 38 T. Shoda, M. Kato, R. Harada, T. Fujisato, K. Okuhira, Y. Demizu, H. Inoue, M. Naito and M. Kurihara, *Bioorg. Med. Chem.*, 2015, **23**, 3091–3096.
- 39 S. Ray, *Drugs Future*, 2004, **29**, 185–203.
- 40 M. R. Schneider, R. W. Hartmann, F. Sinowatz and W. Amselgruber, *J. Cancer Res. Clin. Oncol.*, 1986, **112**, 258–265.
- 41 K. Lindorff-Larsen, S. Piana, K. Palmo, P. Maragakis, J. L. Klepeis, R. O. Dror and D. E. Shaw, *Proteins*, 2010, **78**, 1950–1958.
- 42 R. Dennington, T. Keith and J. Millam, *GaussView, version 5*, 2009.
- 43 J. J. Stewart, *J. Comput. Chem.*, 1989, **10**, 221–264.
- 44 M. J. Frisch, G. W. Trucks, H. B. Schlegel, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota,

- R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery Jr, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, *Gaussian 09*, Gaussian, Inc., Wallingford CT, 2009.
- 45 OpenEye Scientific Software, Santa Fe, NM, <http://www.eyesopen.com>.
- 46 P. C. Hawkins, A. G. Skillman, G. L. Warren, B. A. Ellingson and M. T. Stahl, *J. Chem. Inf. Model.*, 2010, **50**, 572–584.
- 47 G. M. Morris, D. S. Goodsell, R. S. Halliday, R. Huey, W. E. Hart, R. K. Belew and A. J. Olson, *J. Comput. Chem.*, 1998, **19**, 1639–1662.
- 48 G. M. Morris, R. Huey, W. Lindstrom, M. F. Sanner, R. K. Belew, D. S. Goodsell and A. J. Olson, *J. Comput. Chem.*, 2009, **30**, 2785–2791.
- 49 *OEDocking 3.2.2*, OpenEye Scientific Software, Santa Fe, NM, 2017, <http://www.eyesopen.com>.
- 50 E. F. Pettersen, T. D. Goddard, C. C. Huang, G. S. Couch, D. M. Greenblatt, E. C. Meng and T. E. Ferrin, *J. Comput. Chem.*, 2004, **25**, 1605–1612.
- 51 A. C. Wallace, R. A. Laskowski and J. M. Thornton, *Protein Eng. Des. Sel.*, 1995, **8**, 127–134.
- 52 C. A. Lipinski, *Drug Discov. Today Technol.*, 2004, **1**, 337–341.
- 53 A. W. S. da Silva and W. F. Vranken, *BMC Res. Notes*, 2012, **5**, 367–374.
- 54 W. L. Jorgensen, J. Chandrasekhar, J. D. Madura, R. W. Impey and M. L. Klein, *J. Phys. Chem.*, 1983, **79**, 926–935.
- 55 H. J. Berendsen, J. P. Postma, W. F. van Gunsteren and J. Hermans, *Intermol. Forces*, 1981, **14**, 331–342.
- 56 M. Parrinello and A. Rahman, *J. Appl. Phys.*, 1981, **52**, 7182–7190.
- 57 M. J. Abraham and J. E. Gready, *J. Comput. Chem.*, 2011, **32**, 2031–2040.
- 58 B. Hess, H. Bekker, H. J. Berendsen and J. G. Fraaije, *J. Comput. Chem.*, 1997, **18**, 1463–1472.
- 59 W. Humphrey, A. Dalke and K. Schulten, *J. Mol. Graphics*, 1996, **14**, 33–38.
- 60 W. L. DeLano, *CCP4 Newsletter On Protein Crystallography*, 2002, vol. 40, pp. 82–92.
- 61 R. Kumari, R. Kumar, O. S. D. D. Consortium and A. Lynn, *J. Chem. Inf. Model.*, 2014, **54**, 1951–1962.
- 62 L. Celik, J. D. D. Lund and B. Schiött, *Biochemistry*, 2007, **46**, 1743–1758.
- 63 F. Fratev, *Phys. Chem. Chem. Phys.*, 2015, **17**, 13403–13420.
- 64 T. D. McGee, J. Edwards and A. E. Roitberg, *Int. J. Environ. Res. Public Health*, 2008, **5**, 111–114.
- 65 H. L. Ng, *J. Mol. Graphics Modell.*, 2016, **69**, 72–77.
- 66 S. Chakraborty, A. S. Levenson and P. K. Biswas, *BMC Struct. Biol.*, 2013, **13**, 27–38.
- 67 J. Zeng, W. Li, Y. Zhao, G. Liu, Y. Tang and H. Jiang, *J. Phys. Chem. B*, 2008, **112**, 2719–2726.
- 68 L. M. Gregoret, S. D. Rader, R. J. Fletterick and F. E. Cohen, *Proteins*, 1991, **9**, 99–107.
- 69 M. M. Van Lipzig, A. M. Ter Laak, A. Jongejan, N. P. Vermeulen, M. Wamelink, D. Geerke and J. H. Meerman, *J. Med. Chem.*, 2004, **47**, 1018–1030.
- 70 J.-Y. Liu and S. D. Mooney, *Int. J. Biochem. Mol. Biol.*, 2011, **2**, 190–198.