

Microshear bond strength of dual-cure resin cement in zirconia after different cleaning techniques: an *in vitro* study

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PURPOSE. This study aimed to compare the microshear bond strength (μ SBS) of dual-cure resin cement in CAD-CAM zirconia after different cleaning techniques.

MATERIALS AND METHODS. Fifty discs of zirconia-based ceramic from Ivoclar Vivadent were embedded in acrylic resin. The discs were divided into five groups according to the cleaning methods used: Group 1: drying with spraying + sandblasting with Al_2O_3 ; Group 2: washed with water and dried with spraying + sandblasting with Al_2O_3 ; Group 3: washed with distilled water and dried with spraying + sandblasting with Al_2O_3 + zirconium oxide (Ivoclean); Group 4: washed with distilled water and dried with spraying + sandblasting with Al_2O_3 + potassium hydroxide (Zirclean); and Group 5: washed with distilled water and dried with spraying + sandblasting with Al_2O_3 + 1% NaClO. All of the groups were contaminated with artificial saliva for 1 minute and then cleaned. Statistical analyses were performed using ANOVA and Tukey's tests. **RESULTS.** There were statistically significant differences among all groups for μ SBS ($P < .05$). The group treated with zirconium oxide (Group 3) showed the highest μ SBS (18.75 ± 0.23 MPa). **CONCLUSION.** When applied to zirconia, the cleaning methods affected the bonding with resin cement differently. [J Adv Prosthodont 2021;13:237-45]

KEYWORDS

Zirconia; Ceramics; Dental cements; Cleaning; Mechanical test

INTRODUCTION

Metal-free dental prostheses are used widely due to their biocompatibility and superior aesthetic properties.^{1,2} Zirconia is one of the most commonly used materials for such prostheses, which can be used for bridge infrastructures, monolithic crowns, and implant abutments in dentistry.^{3,4} One of the

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most significant advantages imparted by metal-free prosthesis is the marginal adaptation, which avoids microleakage and increases retention.⁵ In this context, zirconia-based ceramics have shown suitable properties due to their aesthetic appearance and biocompatible properties compared to metal and porcelain restorations.³ The reduced laboratory costs for CAD-CAM metal-free restorations have assisted with their availability.⁶ All restorative materials, including zirconia-based materials, must be tested clinically to determine whether they have adequate marginal adaptation or require occlusal adjustment. Previous studies have shown that contamination of zirconia with saliva, blood, or silicone reduces the bond strength between the zirconia and the resin cement.⁷⁻¹¹ The luting procedure, a cleaning process for zirconia, could be an effective method to improve the bonding and maintain the reliability of prosthetic treatment.^{12,13}

Despite the excellent properties that zirconia offers to dentistry, this ceramic also displays significant challenges in its bonding ability. Zirconia offers lower adhesion compared to other ceramics, such as feldspathic porcelain and lithium disilicate, due to zirconia's higher crystalline phase content.^{14,15} The most commonly used method to improve the bond strength between resin cements and zirconia is with aluminum oxide (Al_2O_3 , with 50 to 100 μm particles) by sandblasting.¹⁶⁻¹⁸ However, this process may produce cracks, making zirconia more susceptible to fractures.¹⁹ To overcome this issue, chemical bonds with zirconia have been proposed by incorporating -methacryloyloxydecyl dihydrogen phosphate (10-MDP) into primers and resin cements to promote ionic and hydrogen bonding to zirconia.^{4,20-23}

Besides the concerns with bonding due to zirconia's high crystallinity phase, steps for cleaning zirconia prior to cementation have been investigated to eliminate saliva and organic contaminants and generate proper chemical adhesion.^{8,9,11} Some studies²⁴⁻²⁸ have examined the benefits of cleaning agents such as those with zirconium oxide (commercially available as Ivoclean) and sodium hypochlorite at 1% (NaClO). Other substances could be used to remove organic residue, such as potassium hydroxide. However, no studies have analyzed the effects of potassium hy-

droxide (commercially available as Zirclean) in this regard. After different cleaning techniques, this study compares the microshear bond strength (μSBS) of dual-cure resin cement in CAD-CAM zirconia. The null hypothesis to be tested is that using chemical agents for cleaning after sandblasting and posterior application of primer does not influence the μSBS to zirconia.

MATERIALS AND METHODS

This study was performed *in vitro*, following the standards of and the guidance from the Ethics Committee of the Universidad Científica Del Sur, Perú. A pilot study was performed before the current study. A summarized flowchart of the study design is presented in Fig. 1.

The Specimen preparation was fifty discs of zirconia (IPS MT A1, Ivoclar Vivadent AG, Schaan, Liechtenstein) measuring 11 mm in diameter and 3 mm thick were used in this study. The manufacturing process utilized a CAD/CAM milling machine (Wieland Dental ZENOTEC mini, Ivoclar Vivadent, Lindenstr.2,75175, Pforzheim, Germany). The sintering process was performed at 1500°C to 1700°C according to the manufacturer's instructions.²⁹ One surface of each disc was tested and sandblasted with Al_2O_3 at 50 μm for 15 s at 10 mm and 2.5 bar. The samples were cleaned ultrasonically in isopropyl alcohol for 3 min, rinsed with distilled water, and dried.³⁰ Subsequently, the ceramic surfaces were embedded in a polyvinyl chloride (PVC) tube (25 mm diameter and 40 mm height) using a self-cured acrylic resin.¹⁶ All of the materials tested in this study are shown in Table 1. The zirconia discs were divided into five groups ($n = 10$ per group). All of them were subjected to contamination with artificial saliva (Salival, LUSA laboratory, Lima, Perú). Artificial saliva was used due to its higher standardization compared to human saliva and no need for approval by an ethics committee. The artificial saliva used in this research has the same composition as previous studies.^{26,31} Afterward, all the samples for each group were prepared as follows (Fig. 2).

Group 1 (control group): The samples were contaminated with artificial saliva for 1 min, then dried with a spray for 15 s and sandblasted with Al_2O_3 at 50 μm for 20 s at 10 mm at 2.5 bar.

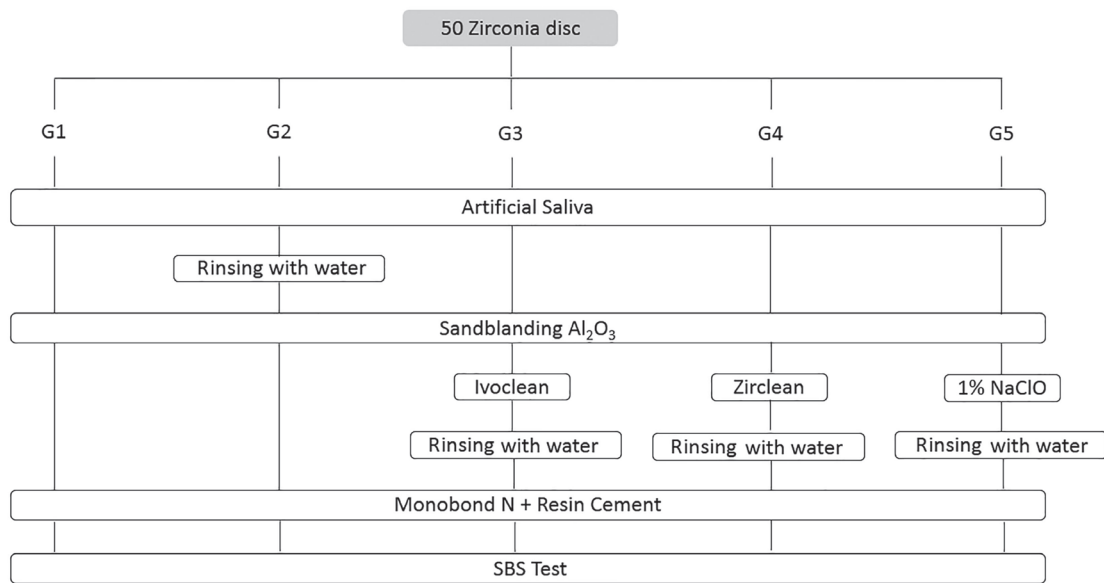


Fig. 1. Flowchart of the study indicating the five groups tested via μ SBS.

Table 1. Description of the commercial products materials name, composition, and manufacturers used in the study

Materials	Composition	Manufacturer	Lot no.
IPS e-max ZirCAD MT A1 - Ivoclar Vivadent AG	ZrO ₂ (86.0 - 93.5%), Y ₂ O ₃ (> 6.5 % - ≤ 8.0 %), HF (≤ 5.0%), Al ₂ O ₃ ≤ 1.0%, other oxides ≤ 1.0%	Ivoclar Vivadent, Schaan, Lichtenshtein	W31929
Al ₂ O ₃	50 μ m aluminum oxide powder	Bio-Art Equipamentos Odontológicos Ltda, São Carlos, SP, Brazil	19-00586
Ivoclean	Sodium hydroxide, ZrO ₂ , water, polyethylene glycol, pigments	Ivoclar Vivadent, Schaan, Lichtenshtein	X36169
Zirclean	Potassium hydroxide	Bisco, Schaumburg, IL, USA	REF B-7351
Monobond N	Ethanol, 3-(trimethoxysilyl) propyl methacrylate, methacrylated phosphoric acid ester, disulfide methacrylate	Ivoclar Vivadent, Liechtenstein	Y19262
SET PP Self-adhesive, self-etching, fluoride-releasing resin cement	35% methacrylate 65% inorganic filler	SeT PP, SDI, Victoria, Australia	S19031091
Artificial Saliva	Sodium chloride 0.084 g, potassium chloride 0.120 g, calcium chloride dihydrate 0.015 g, magnesium chloride hexahydrate 0.005 g, carboxymethylcellulose sodium 0.375 g, propylene glycol 4,000 g, propylparaben 0.010 g, distilled water qs. 100.00 mL	Salival, LUSA laboratory, Lima, Perú	1080218

Group 2: The samples were contaminated with artificial saliva for 1 min and then rinsed with water-spraying for 15 s, dried with a spray for 15 s, and sandblasted with Al_2O_3 at 50 μm for 20 s at 10 mm at 2.5 bar.

Group 3: The samples were contaminated with artificial saliva for 1 min and then rinsed with water-spraying for 15 s, dried with spray for 15 s, and sandblasted with Al_2O_3 at 50 μm for 20 s at a distance of 10 mm at 2.5 bar. Zirconium oxide (Ivoclean) was applied with a microbrush. It was allowed to react for 20 s on the surface, which then was rinsed with water spray for 30 s and air dried for another 10 s according

to the manufacturer's instructions.

Group 4: Treated the same as Group 3 until sandblasting. Then, potassium hydroxide (Zirclean) was applied. The product was allowed to react for 20 s, after which rinsing with water spray continued for 30 s, followed by air-drying for 10 s according to the manufacturer's instructions.

Group 5: Treated the same way as Group 3 until sandblasting. Then, sodium hypochlorite (1% NaClO) was applied. The product was allowed to react for 20 s, and the samples were rinsed with water spray for 30 s and air dried for 10 s.

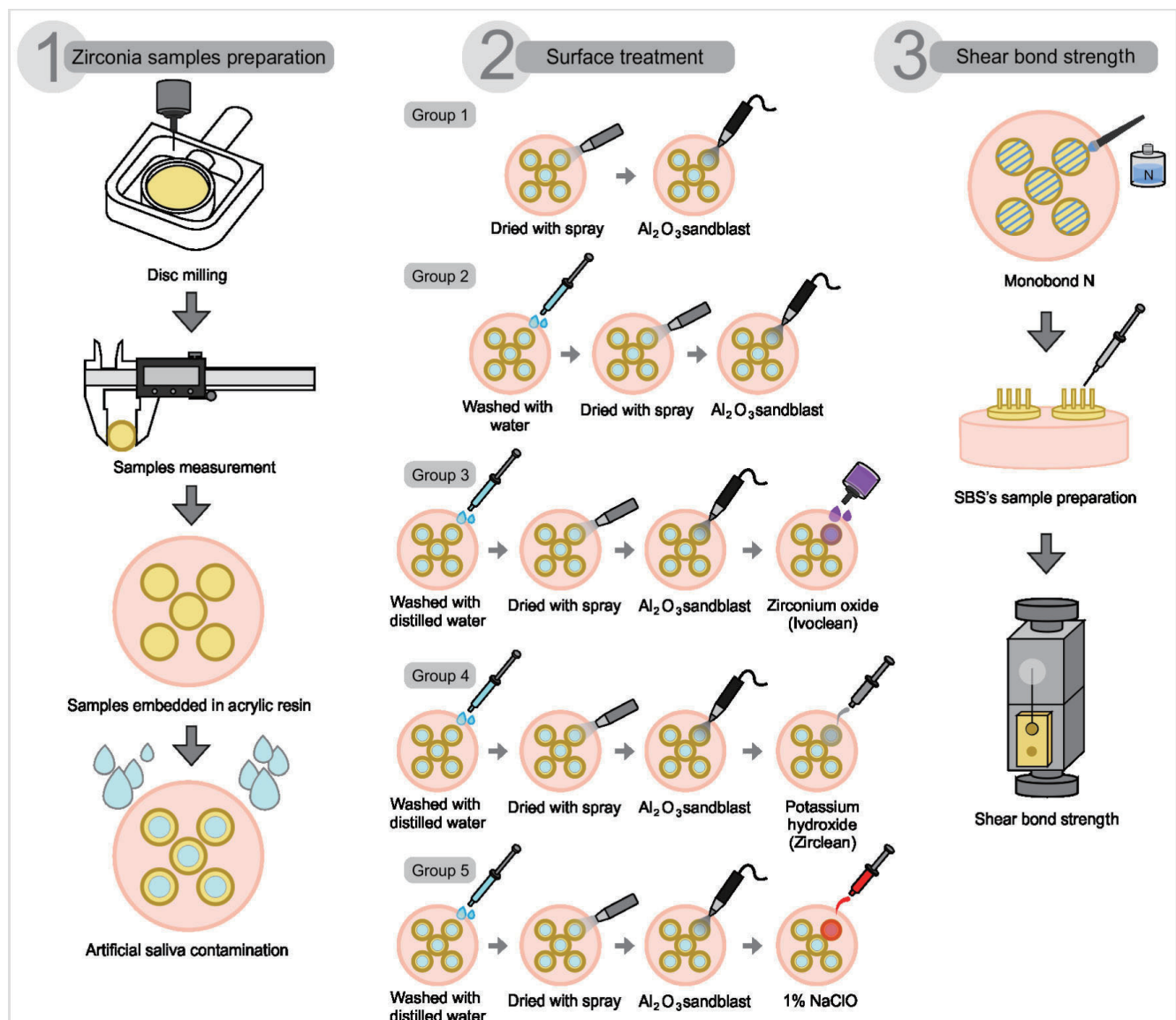


Fig. 2. Illustration evidencing the step-by-step of the present study design.

Immediately after all of the groups were prepared with their respective cleaning methods, Monobond Plus primer (Ivoclar Vivadent) was applied to all specimens with a microbrush for 60 s. Then, the surface was gently dried with air.³²

After each cleaning and conditioning treatment, three Tygon plastic tubes (1 mm internal diameter and 3 mm thick) were filled with the resin cement SET PP SDI (dual-cure resin cement SeT PP, SDI, Victoria, Australia). The manipulation was performed according to the manufacturer's recommendation. The dual-cure resin cement was photoactivated for 20 s with 1100 mW/cm (Elipar™ DeepCure-L 3M™ Oral Care, St. Paul, MN, USA). After 10 min, the tubes were removed using a no. 11 scalpel blade to expose the cement cylinders. The specimens were stored in distilled water for 24 h at 37°C.³³

The sample unit consisted of 10 disks per group (five total groups) of zirconia ceramics with three cement cylinders each (Tygon, Medical Accessories, Lima, Perú). Microshear bond strength (μ SBS) was evaluated using a universal testing machine in all of the disks (Universal testing machine, LG CMT-5L number 7419, Seoul, South Korea). A steel hook was used to pull each resin cement cylinder, applying a shear force of 0.75 mm/min until fracture.^{26,34,35}

After SBS testing, the debonded zirconia ceramic specimens were examined under microscopy (Andonstar ADSM201, Shenzhen Andonstar Technology Co. Ltd., Shenzhen, Guangdong, China) and magnification of 1600 \times . The failures were classified as adhesive failures between the zirconia disc and cement cylinders, cohesive failures in cement cylinders, or mixed failures. The failure pattern was calculated as percentages of each group.

All of the statistical analyses were performed using SPSS version 23 (SPSS Inc., Armonk, NY, USA). The data distributions were analyzed with the Shapiro-Wilk test, and the results of the μ SBS tests were analyzed via a one-way analysis of variance (ANOVA) and Tukey post-hoc test. A 5% significance level was used.

RESULTS

Fig. 3 shows the outcomes as mean and microshear

values of MPa. The group treated with Ivoclean (Group 3) showed the highest μ SBS (18.75 ± 0.23 MPa), while the group washed with water, dried with spray, and sandblasted (Group 2) showed the lowest value (10.22 ± 0.10 MPa) ($P < .05$).

The groups presented the following decreasing order of μ SBS levels: Group 3, Group 4, Group 5, Group 1, and Group 2 ($P < .05$). The use of any cleaning agents tested (Groups 3, 4, and 5) increased the μ SBS in comparison to only sandblasting (Group 1) or rinsing with water and sandblasting (Group 2) ($P < .05$).

The failure modes are presented in Table 2. Cohesive failure was observed among the most groups. Only one group showed a low incidence of adhesive failure. Mixed failures were not predominant for the zirconium ceramics.

DISCUSSION

Previous studies investigated the effects of cleaning agents for zirconia, such as water, 37% phosphoric acid, and NaClO, because of the decreased bond strength when saliva contaminants are present on ceramic surfaces.^{8,9} Zirconia has strong adsorption potential for phosphorus, which is found in saliva.

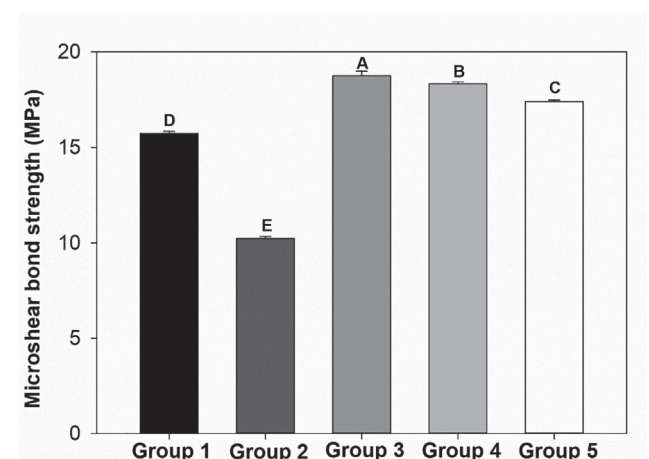


Fig. 3. Results of μ SBS of dual-cure resin cement on zirconia after treatment with one of the five cleaning techniques applied. The results are displayed in mean and standard deviation values of MPa. Different letters indicate a statistically significant difference among groups ($P < .05$).

Consequently, it is adsorbed on the surface of zirconia. Due to this reason, saliva contamination is not removed only with a water wash.⁹ In the current research, the use of cleaning agents improved the μ SBS compared to no cleaning agents. These agents are recommended to remove saliva contamination on the inner surface of restorations before cementation because they are able to eliminate and remove organic materials, magnesium, and carbonate ions from zirconia's surface.^{24,32} Thus, the null hypothesis must be rejected.

The main components of saliva are water (99%), glycoproteins, mucins, enzymes, immunoglobulins, nitrogenous products, and electrolytes,³⁶ which are similar to artificial saliva.³¹ Saliva is adsorbed onto zirconia easily, and other elements can contaminate the ceramic's surface, such as carbon compounds.¹² Organic solvents (acetone or isopropanol) have already been tested as cleaning agents, but they have shown incomplete removal of organic contaminants.¹² Phosphoric acid at 37% showed a better cleaning effect but lower bond strength, in comparison to Ivoclean.^{32,37} Neither water nor phosphoric acid solution at 37% fully removes saliva contaminants because phosphoric acid has phosphates that are compatible with the phosphates in saliva, which bind with zirconia and decreases the adhesion force.^{9,24} Besides the sandblasting being essential for increasing the surface roughness, the best outcomes from luting zirconia involve combining physical (sandblasting) and chemical (chemical agents such as 10-MDP) approaches.³⁸ However, sandblasting should be used

with caution because excessive abrasion will weaken the zirconia. For this reason, it is reasonable to investigate how to improve the interactions between bonding agents and zirconia by removing improper elements from saliva contaminants.

Ceramics can be challenging materials for adhesion, depending on their composition and physicochemical surface properties. Contamination could occur at different steps, from the try-in procedures to the final luting of the restorations.¹¹ In this context, few studies have compared cleaning agents, and no studies have evaluated the effect of potassium hydroxide on the zirconia's surfaces. According to the manufacturer (Bisco), Zirclean is a strong alkaline gel formulated with potassium hydroxide. It is a strong alkaline solution that can clean contaminants from restorations increasing the adhesive strength. NaClO also has been tested due to its well-known dissolution properties and consequent use as an endodontic irrigation solution. This solution has been used as an endodontic disinfectant due to its effective antimicrobial and tissue dissolving capabilities.^{39,40} Although all tested cleaning methods obtained better results than the group cleaned with water or sandblasting only, Ivoclean showed the highest μ SBS, with a statistically significant difference. The rationale for the easy interaction between Ivoclean and the phosphorus is present in saliva.

There are different mechanisms of action to obtain a clean surface. According to manufacturer Ivoclar Vivadent, Ivoclean's main components are sodium hydroxide and ZrO_2 . This hyper-saturated solution of

Table 2. Failure pattern for tested specimens (in percentage)

Group	Failure Model		
	Cohesive	Adhesive	Mixed
Group 1: drying with spray + sandblast with Al_2O_3	88	9	3
Group 2: washed with water and dried with spray + sandblast with Al_2O_3	40	50	10
Group 3: washed with distilled water and dried with spray + sandblast with Al_2O_3 + zirconium oxide (Ivoclean)	100	0	0
Group 4: washed with distilled water and dried with spray + sandblast with Al_2O_3 + potassium hydroxide (Zirclean)	100	0	0
Group 5: washed with distilled water and dried with spray + sandblast with Al_2O_3 + 1% NaClO	98	2	0

zirconium particles removes contaminants from saliva, which are attracted to the Ivoclean chemical solution.^{24,32} Therefore, the cleaning agent consists of an alkaline solution with a suspension of particles, a hydrophilic character, and a low contact angle on zirconia. The contaminants react with the sodium hydroxide and are removed after the Ivoclean is adsorbed on the surface.^{8,24} Even though the primary mechanism of all the tested commercial cleaning agents seems to be an acid-base reaction on the zirconia's surfaces, Ivoclean achieved the highest μ SBS. This finding corroborates a previous study showing better outcomes for Ivoclean than common agents such as phosphoric acid at 37%.⁹ It was not possible to identify why Groups 1 and 2 showed a statistical difference, in that the only difference was the use of water before sandblasting for Group 2. However, it is possible to conclude that bond strength improved when a set of measures was used beyond just sandblasting or water and sandblasting

In the present study, the failure mode which was predominantly cohesive was found in a complementary way to the SBS test. It indicates that the stresses generated during the micro-shear test occurred at the level of the resin cement and not at the level of the materials' adhesion.

Further research could focus on the longitudinal effects of cleaning agents on the bonding to zirconia. Moreover, a more in-depth understanding of cleaning agents' mechanism on zirconia surfaces and chemical analyses of the cleaning agents, including pH analysis, are needed to determine why some commercial available materials provide better immediate outcomes.

CONCLUSION

The cleaning methods applied to zirconia affected the bonding with the dual-cure resin cement differently. The cleaning methods involving zirconium oxide led to the highest μ SBS results, followed by potassium hydroxide.

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To my mother Carmen my angel in heaven. In addition

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