

Article

Inkjet-Printed Electron Transport Layers for Perovskite Solar Cells

Dongli Lu ¹, Wei Zhang ², Lars Kloo ² and Liubov Belova ^{1,*}

¹ Department of Materials Science and Engineering, KTH Royal Institute of Technology, SE-10044 Stockholm, Sweden; donglil@kth.se

² Department of Chemistry, Applied Physical Chemistry, KTH Royal Institute of Technology, SE-10044 Stockholm, Sweden; wzha@kth.se (W.Z.); larsa@kth.se (L.K.)

* Correspondence: lyuba@kth.se

Abstract: Inkjet printing emerged as an alternative deposition method to spin coating in the field of perovskite solar cells (PSCs) with the potential of scalable, low-cost, and no-waste manufacturing. In this study, the materials TiO₂, SrTiO₃, and SnO₂ were inkjet-printed as electron transport layers (ETLs), and the PSC performance based on these ETLs was optimized by adjusting the ink preparation methods and printing processes. For the mesoporous ETLs inkjet-printed from TiO₂ and SrTiO₃ nanoparticle inks, the selection of solvents for dispersing nanoparticles was found to be important and a cosolvent system is beneficial for the film formation. Meanwhile, to overcome the low current density and severe hysteresis in SrTiO₃-based devices, mixed mesoporous SrTiO₃/TiO₂ ETLs were also investigated. In addition, inkjet-printed SnO₂ thin films were fabricated by using a cosolvent system and the effect of the SnO₂ ink concentrations on the device performance was investigated. In comparison with PSCs based on TiO₂ and SrTiO₃ ETLs, the SnO₂-based devices offer an optimal power conversion efficiency (PCE) of 17.37% in combination with a low hysteresis. This work expands the range of suitable ETL materials for inkjet-printed PSCs and promotes the commercial applications of inkjet printing techniques in PSC manufacturing.



Citation: Lu, D.; Zhang, W.; Kloo, L.; Belova, L. Inkjet-Printed Electron Transport Layers for Perovskite Solar Cells. *Materials* **2021**, *14*, 7525. <https://doi.org/10.3390/ma14247525>

Academic Editor: Fabrice Goubard

Received: 25 October 2021

Accepted: 2 December 2021

Published: 8 December 2021

Publisher's Note: MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Copyright: © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

Keywords: inkjet printing; electron transport layers; perovskite solar cells; TiO₂; SrTiO₃; SnO₂; cosolvent system

1. Introduction

Electron transport layers (ETLs), which effectively collect photo-generated electrons from the light-absorbing perovskite material and transport these electrons to the conductive contact layer, are critical for fabricating efficient perovskite solar cells (PSCs). TiO₂ is the most widely used electron transport material in the community of PSCs because of its chemical robustness, abundance, low cost, and good chemical and thermal stability, as well as good conduction band alignment with the perovskite [1,2]. A bilayer ETL consisting of a compact TiO₂ (c-TiO₂) film and a mesoporous TiO₂ (mp-TiO₂) layer is preferred for highly efficient PSCs [3,4]. Such a morphology was shown to offer a power conversion efficiency (PCE) exceeding 25% [5–7]. However, concerns regarding performance degradation of TiO₂-based devices under long-time UV illumination [8] together with hysteresis and recombination problems caused by inefficient charge transfer at the TiO₂/perovskite interface [1,9] remain. Although much effort was devoted to overcoming these issues and to promoting the device performance to new levels by employing doping [10], graphene/TiO₂ composites [11,12], surface passivation [13], and interface engineering [14,15], the phenomenon of scan-direction hysteresis when using TiO₂ ETLs in PSCs is still difficult to suppress [16]. Investigation of alternative electron transport materials may be a more effective strategy than tedious optimization efforts involving TiO₂.

SrTiO₃ is one of those alternative electron transport materials. As compared to that of TiO₂, SrTiO₃ exhibits a similar band gap of 3.2 eV and, despite its slightly higher conduction band energy level, good band alignment with the perovskite material. This is beneficial for a high open-circuit voltage (V_{OC}) of SrTiO₃-based PSCs [17]. The high electron mobility (5–8 cm²/(V·s)) of the materials may promote charge transport and help to reduce recombination losses [18]. There were already some reports focused on the applications of SrTiO₃ ETLs in PSCs. It was an early stage reported that devices employing commercial SrTiO₃ nanoparticles (average diameter \leq 100 nm) exhibited higher V_{OC} and lower short-circuit current density (J_{SC}) than PSCs based on TiO₂ because of the large particle size of the SrTiO₃ nanoparticles [17]. Later, graphene/SrTiO₃ composites [19] and TiO₂/SrTiO₃ composites [20,21] were applied as ETLs to compensate for the low current density and retain the advantage of the high open-circuit voltage. Afterwards, PSCs based on a compact SrTiO₃ ETL with a smaller particle size showed an improved stability and enhanced electron transport in the SrTiO₃ ETL [18]. With the help of doping [8] and mixtures with another ETL [22], both the photovoltaic performance and stability of SrTiO₃-based PSCs were improved.

SnO₂ is considered to represent a promising alternative electron transport material compared to TiO₂ because of its favorable electrical and chemical properties [23–25]. Its high electron mobility (100–200 cm²/(V·s)) and deep conduction band facilitate electron extraction and transport [26]. SnO₂ also exhibits a wide bandgap (3.8 eV) and extensive stability [27,28] under UV illumination. After SnO₂ was directly used as the ETL in a planar PSC in 2015 [25], to further improve the photovoltaic performance and stability of SnO₂-based PSCs, efforts were made to adjust the electrical properties and to passivate the ETL/perovskite interface, such as employing mesoporous SnO₂ [9,29,30], doping [31–35], a gradient interlayer [16], titanium (IV) chloride (TiCl₄) treatment [36], or graphene [37].

The inkjet technology offers a solution to the emerging demands of additive patterning of functional multilayers and device components. The technology offers mask-free, cost-effective, and direct patterning processes, which are easily scalable to meter format. Therefore, the inkjet printing method shows a large potential for the up-scaling manufacturing of perovskite solar cells associated with less material consumption and negligibly small waste, in contrast to the conventionally employed spin-coating method. Inkjet printing was applied to fabricate functional layers relevant to PSCs, such as carrier transport layers [38–40], electrodes [41], and perovskite absorbers [42–46]. In this work, we focus on the inkjet-printed ETLs for PSCs. TiO₂ ETLs were intensively investigated but not for inkjet-printed ETLs used in PSCs. Huckaba et al. fabricated inkjet-printed mp-TiO₂ for PSCs and the champion cell offered a PCE of 18.29% [38]. Recently, Buffiere et al. reported that PSCs with inkjet-printed c-TiO₂ ETLs exhibited a PCE of 13.7% [39]. High-performance SnO₂ ETLs are usually prepared by spin coating [47], atomic layer deposition [48], electron beam evaporation [49] or dual-fuel combustion [50] methods, while there are few reports [40] focused on inkjet-printed SnO₂. For the application of inkjet-printed SnO₂ in PSCs, there is as far as we know only one recent report by Rohnacher et al., in which their best device displayed an efficiency of 18.8% [40]. To the best of our knowledge, there is no publication on inkjet-printed SrTiO₃ ETL based PSCs. As mentioned before, the intrinsic properties of TiO₂, such as the low mobility and instability under UV illumination, limit the photovoltaic performance of TiO₂-based PSCs. Therefore, further research on inkjet-printed ETLs is still necessary for improving PCEs of inkjet-printed PSCs, and thus to make them comparable with or even superior to the spin-coated alternatives, and eventually promote the commercialization of inkjet-printed PSCs. Investigations of inkjet-printed SrTiO₃ and SnO₂ could expand the selection range of inkjet-printed ETLs for PSCs and pave the way for fully inkjet-printed and efficient PSCs.

In this study, we present inkjet printing processes for different ETLs for the application in PSCs by optimizing the ink design, the film uniformity and the device performance of devices based on these printed functional layers. The importance of the solvents used for dispersing TiO₂ and SrTiO₃ nanoparticles is investigated, and cosolvent systems are found to contribute to uniform film formation of mesoporous TiO₂ and SrTiO₃ ETLs. SrTiO₃-based PSCs show higher PCE than TiO₂-based devices. However, they suffer from lower current density and severe scan-direction hysteresis. Subsequently, TiO₂ nanoparticles are introduced into the mesoporous SrTiO₃ (mp-SrTiO₃) layer to solve these issues by optimizing the TiO₂ nanoparticle concentration in the SrTiO₃ ink. The cosolvent system is also applied to print SnO₂ ETLs and the effects of the precursor ink concentration on the device performance of SnO₂-based PSCs are studied. Finally, an optimal PCE of 17.37% is achieved.

2. Materials and Methods

2.1. Materials

All reagents and solvents were purchased and used as received without further purification. Titanium (IV) isopropoxide (Ti(OCH(CH₃)₂)₄, 97%) and titanium (IV) oxide powder (TiO₂, ≥99.5%, 21 nm) were purchased from Sigma–Aldrich (Darmstadt, Germany). Strontium titanate nanoparticles (SrTiO₃, 99.95%, 100 nm) were obtained from US Research Nanomaterials (Houston, TX, USA). Tin (IV) oxide (15% in H₂O colloidal dispersion) was purchased from Alfa Aesar (Kandel, Germany). Lead iodide (PbI₂, 99.99%) and lead bromide (PbBr₂, >98.0%) were purchased from TCI (Tokyo, Japan). Formamidinium iodide (FAI, CH(NH₂)₂I, >98%) and methylammonium bromide (MABr, CH₃NH₃Br, >98%) were obtained commercially. Spiro-OMeTAD (99.8%) was purchased from Borun New Material Technology (Ningbo, China). Bis(trifluoromethane)sulfonimide lithium salt (LiTFSI, 99.95%), FK209 (Co(III) TFSI salt, 98%) and 4-*tert*-butylpyridine (TBP, 98%) were purchased from Sigma–Aldrich (Darmstadt, Germany).

2.2. Set-Up of Inkjet Printer

An inkjet station was constructed and used as a drop-on-demand inkjet printer in our lab. This system was designed for printheads from XaarJet to achieve flexible inkjet printing. In this work, XJ126/50 (126 active nozzles with a drop volume of 50 pL) printheads were used for printing TiO₂ nanoparticle inks and XJ126/80 (126 active nozzles with a drop volume of 80 pL) printheads for other inks. More details about the technical information of the two printheads are shown in Table S1. The ejection voltage for all inks was set to 20 V and the printing frequency to 283.46 Hz. In this work, the inkjet printing of electron transport layers was operated under ambient conditions.

2.3. Device Fabrication

Before cleaning, fluorine-doped tin oxide (FTO) glass substrates (14 Ω/sq, Pilkington TEC) were cut into pieces with the size of 25 × 15 mm. One edge of each piece was etched with Zn powder and 2M HCl aqueous solution. Subsequently, these substrates were successively sonicated in a detergent solution (5% deconex in water), deionized water, acetone, and 2-propanol for 15 min. The clean substrates were stored in 2-propanol before use.

Before inkjet printing, the substrates were placed on a preheated printing stage at 60 °C. A compact TiO₂ layer was inkjet-printed from a 0.125 M solution of titanium isopropoxide dissolved in 2-isopropoxyethanol, and then annealed at 450 °C for 45 min. On the top of the compact TiO₂ layer, the mesoporous TiO₂ layer was also inkjet-printed with an ink made by dispersing TiO₂ nanoparticles in a mixture of ethanol and ethylene glycol (9/1, *v/v*). After printing, the substrate was annealed at 500 °C for 45 min. For SrTiO₃-based devices, the compact TiO₂ layer was fabricated using the same procedure as for the TiO₂ based devices. The mesoporous SrTiO₃ layer was inkjet-printed with a suspension ink of SrTiO₃ nanoparticles dispersed in a mixed solvent (ethanol/ethylene glycol = 9/1, *v/v*). Thereafter,

the film was annealed at 500 °C for 45 min. The inkjet-printed mp-SrTiO₃/TiO₂ ETL was prepared based on the same ink preparation and printing procedure as for the SrTiO₃ ETL, except that a small amount of TiO₂ was added into the SrTiO₃-based inks. For SnO₂-based devices, a single SnO₂ ETL was used instead of the bilayer ETLs. The single layer was inkjet-printed using inks of diluted 15% tin oxide colloidal dispersion in a mixture of deionized water and ethylene glycol (9/1, *v/v*). A small amount of Triton X100 was added to adjust the surface tension of the SnO₂ inks. Afterwards, the printed SnO₂ film was annealed in a furnace at 220 °C for 1 h.

For reference TiO₂-based devices, the fabrication procedure of the bilayer TiO₂ ETLs in the literature [51] was followed. The compact TiO₂ layer was deposited by spray pyrolysis at 450 °C with a solution of 0.2 M titanium (IV) isopropoxide and 2 M acetylacetonate in isopropyl alcohol. Thereafter, the diluted TiO₂ nanoparticle solution (Dyesol DSL 30NR-T, TiO₂ paste/absolute ethanol = 1/5.5) was spin-coated at 4500 rpm for 30 s. The substrate was immediately transferred to a hotplate at 80 °C for 15 min and annealed at 500 °C for 30 min.

For the preparation of the perovskite precursors and the deposition of the perovskite layers, we followed the procedure described in the literature [51]. The perovskite precursor was prepared by dissolving 1.1 M PbI₂, 1 M FAI, 0.2 M PbBr₂ and 0.2 M MABr in a mixed solvent (N,N-dimethylformamide/dimethyl sulfoxide = 4/1, *v/v*). Seventy-five μL of the perovskite precursor was spread onto the substrate and spin-coated at 4500 rpm for 30 s, and subsequently 125 μL of chlorobenzene was sprayed onto the perovskite film during 15 s. The resulting perovskite film was immediately dried on a hotplate at 100 °C for 30 min. A hole transport layer was spin-coated at 4000 rpm for 30 s using a solution consisting of 70 mM Spiro-OMeTAD, 20 mM LiTFSI, 200 mM TBP, and 2 mM FK 209 Co(III) TFSI in chlorobenzene. An 80 nm Au electrode was deposited under vacuum by thermal evaporation (Edwards Auto 306). The final devices are displayed in Figure S1.

2.4. Characterization

XRD traces of the TiO₂ compact layers were recorded by a grazing incidence X-ray diffractometer (Siemens D5000, Siemens, Munich, Germany) employing Cu K α radiation ($\lambda = 1.5406 \text{ \AA}$). The morphology of the printed films, the cross-section images and the thickness of devices were studied by a combined focused ion beam/scanning electron microscope (FIB/SEM, FEI Nova 600 Nanolab, FEI Company, Eindhoven, The Netherlands). The working area of the solar cells was defined by a mask of 0.126 cm² and the active area was illuminated under an AM 1.5G solar simulator (Newport 91160-1000) with an incident light density of 100 mW/cm². Photocurrent density-voltage (*J-V*) characteristics were collected by a Keithley 2400 source-measure unit.

3. Results and Discussion

Bilayer TiO₂ electron transport layers improve the device performance according to previous studies [3,4]. In this work, we applied the bilayer structure for different ETL materials consisting of a compact blocking layer and a mesoporous scaffold layer, as shown in Figure 1. Although considerable attempts were made to print the mesoporous TiO₂ ETL for perovskite solar cells and dye-sensitized solar cells [38,52–55], less attention was paid to inkjet printing of other ETL materials, such as SrTiO₃ and SnO₂. In this work, we deposited the bilayer TiO₂ ETLs by inkjet printing and optimized the printing processes by monitoring the photovoltaic performance of the inkjet-printed devices, and thereafter applied the optimized printing procedure for inkjet printing of mp-SrTiO₃ and SnO₂ ETLs.

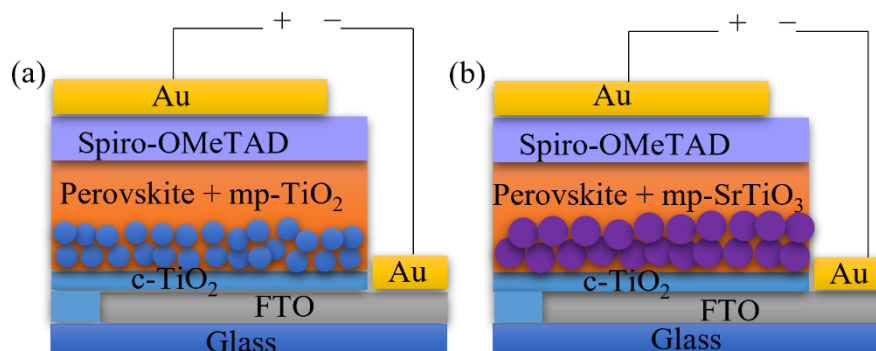


Figure 1. Architecture of (a) mp-TiO₂- and (b) mp-SrTiO₃-based PSC devices.

A c-TiO₂ layer may effectively work as a hole blocking layer, which helps to reduce recombination losses and thus to improve photovoltaic performance, and thus commonly used in high performance PSCs [3,4,56,57]. Figure 2 shows that a uniform and pinhole-free c-TiO₂ layer with a thickness of 50 nm can be successfully deposited by inkjet printing, and that the compact TiO₂ layer crystallized with an anatase crystal structure.

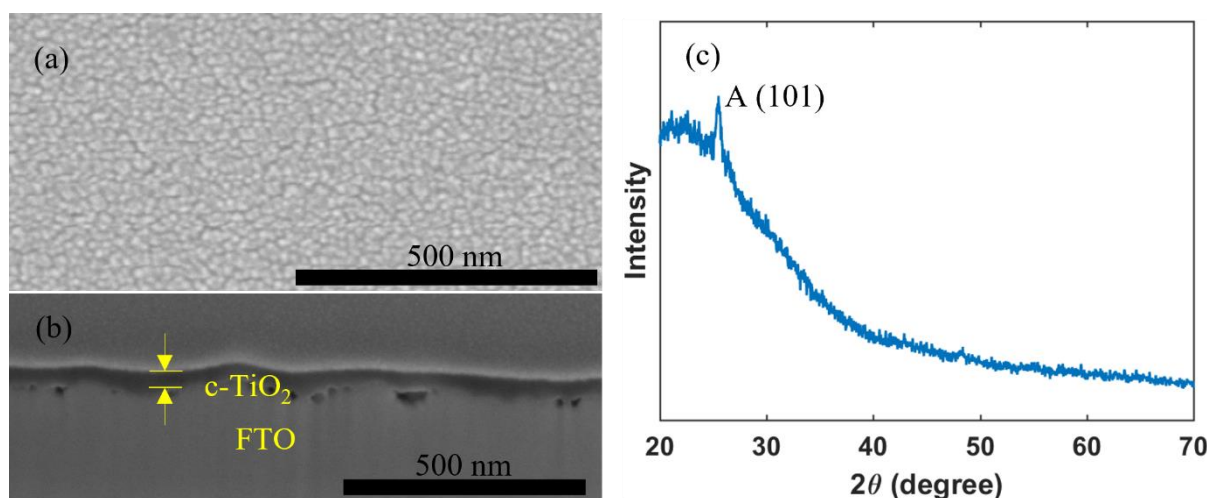


Figure 2. (a) Top view; (b) cross section SEM images; and (c) XRD pattern of an inkjet-printed c-TiO₂ layer.

A major challenge for inkjet printing of high-quality thin films is to eliminate the ‘coffee-ring’ effect, which often occurs when the solvent of a drop containing dispersed solids evaporates and the nonvolatile solid components assemble at the periphery of the drop due to the outward capillary flow and pinned contact line [58,59]. To suppress the coffee-ring effect and control the film morphology, intense efforts were made in the literature and different approaches were proposed, such as solvent composition engineering [60,61], contact line depinning [59,62], evaporation temperature adjustment [63,64], and particle modification [65]. For inkjet printing, the optimization of the ink properties is the most common strategy to improve the film uniformity. Therefore, the selection of solvents for the solutions, and especially for nanoparticle dispersions, is crucial. The emphasis was focused on the physicochemical properties of the solvents, such as the viscosity and surface tension [66,67], which control the ink printability, film formation, and clogging mitigation of the nozzles within the printhead. A cosolvent system is widely used to control the morphology because of the inward Marangoni flow induced by the surface-tension gradients [53,54]. In the present work, we optimized the solvent composition for nanoparticle suspensions aimed at high device performance. The viscosity and surface tension of the inks were in the range of 1–25 mPa·s and 20–50 mN·m⁻¹. Table 1 and Figure 3 demonstrate the effect of the solvent 2-isopropoxyethanol (IPE) and the mixed solvent ethanol: ethylene

glycol (EtOH:EG = 9:1, *v:v*), used for dispersing TiO₂ nanoparticles, on the photovoltaic performance of TiO₂-based devices. For each set of solvents, 12 devices were manufactured. The average and champion photovoltaic parameters, including V_{OC} , J_{SC} , fill factor (FF) and PCE, are listed in Table 1 and the statistical distribution is illustrated in Figure 3. The series resistance (R_s) and shunt resistance (R_{sh}) are also estimated from the inverse of the slope of the J - V curves at the region of the V_{oc} and the J_{sc} , respectively. Although Device S1 displayed a higher champion PCE than Device S2 (Figure S2), Device S2 offered a slightly higher average PCE of 10.97%, as compared to 10.64% of Device S1. From Figure 4a,b, the difference in the drying mechanics of the solvent IPE and EtOH:EG leads to also a difference in the microstructure (Figure S3) of the resulting mp-TiO₂ film. A thicker ETL could allow more perovskite to be absorbed to absorb more incident light but may also lead to a larger series resistance causing Device S1 to exhibit a higher J_{SC} and a lower FF than Device S2. With a higher TiO₂ nanoparticle ink concentration, Device S3 showed an increase in J_{SC} and a decrease in FF due to the formation of a thicker mesoporous TiO₂ layer, as compared to Device S2. Eventually, as seen in Figure 3, although there is no considerable difference in PCE between Device S1, S2, and S3, Device S3 showed the best reproducibility ascribed to the microstructure of mp-TiO₂ (Figure 4c) inkjet-printed with TiO₂ inks based on the mixed solvent EtOH:EG.

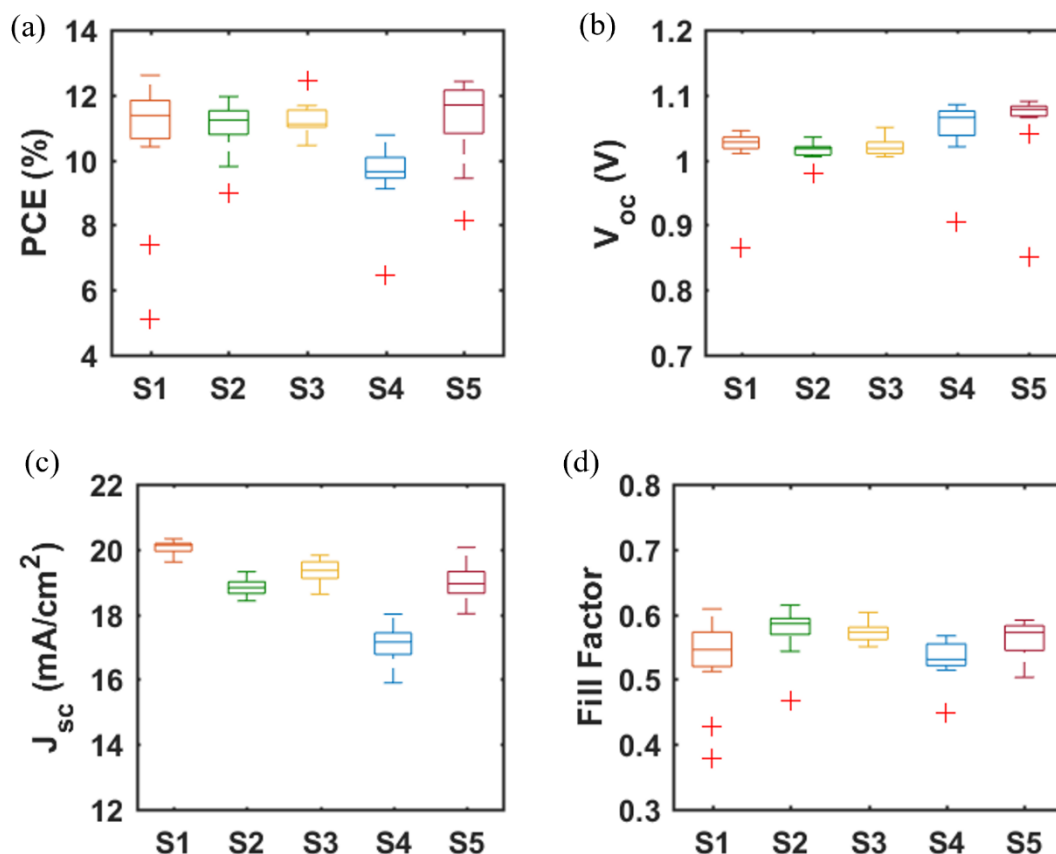


Figure 3. (a) PCE, (b) V_{OC} , (c) J_{SC} , and (d) FF of PSCs based on mp-TiO₂ and mp-SrTiO₃ ETLs inkjet-printed from nanoparticle inks with IPE or EtOH:EG as solvents. Data are obtained from 12 devices for each type of PSC. (Boxplots, red + signs represent outliers).

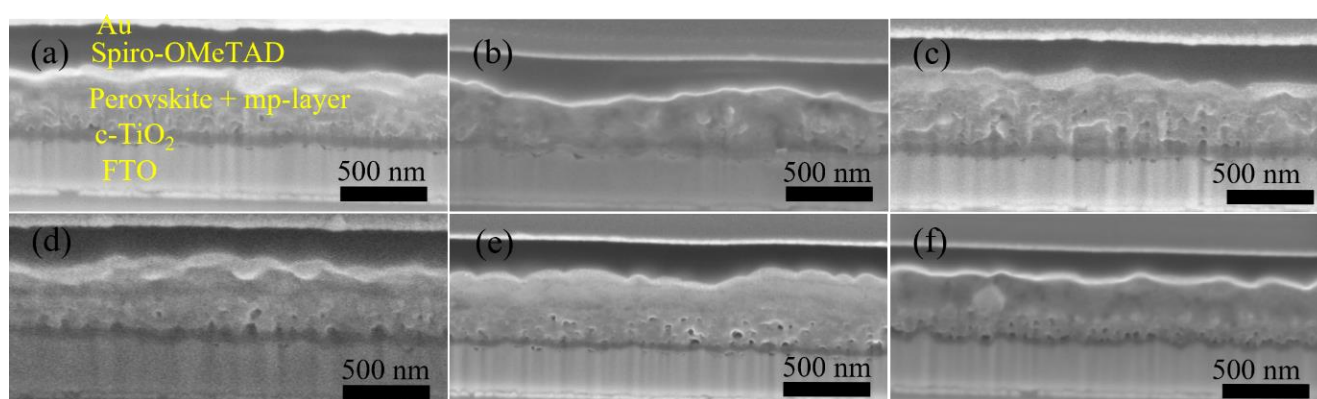


Figure 4. Cross-section SEM images of Device (a) S1, (b) S2, (c) S3, (d) S4, (e) S5, and (f) a spin-coated, TiO₂-based PSC.

Table 1. Photovoltaic parameters from the current density-voltage (*J*-*V*) characteristics of mp-TiO₂- and mp-SrTiO₃-based PSCs. Each average value is based on 12 devices.

	Ink Concentration	Solvent		PCE (%)	V_{OC} (V)	J_{SC} (mA/cm ²)	FF	R_s (Ω·cm ²)	R_{sh} (kΩ·cm ²)
S1	0.25M TiO ₂	IPE	average	10.64 ± 2.21	0.99 ± 0.11	20.06 ± 0.20	0.53 ± 0.07		
			champion	12.60	1.04	19.83	0.61	7.8	2.0
S2	0.25M TiO ₂	EtOH:EG	average	10.97 ± 0.84	1.01 ± 0.01	18.83 ± 0.26	0.57 ± 0.04		
			champion	11.94	1.04	18.95	0.61	8.0	0.9
S3	0.35M TiO ₂	EtOH:EG	average	11.26 ± 0.49	1.02 ± 0.01	19.30 ± 0.36	0.57 ± 0.01		
			champion	12.42	1.05	19.63	0.60	7.7	2.0
S4	0.25M SrTiO ₃	IPE	average	9.54 ± 1.07	1.05 ± 0.05	17.08 ± 0.57	0.53 ± 0.03		
			champion	10.76	1.08	17.48	0.57	12	0.4
S5	0.25M SrTiO ₃	EtOH:EG	average	11.25 ± 1.30	1.06 ± 0.07	18.99 ± 0.63	0.56 ± 0.03		
			champion	12.41	1.08	20.06	0.57	16	1.0

Furthermore, we fabricated SrTiO₃-based Device S4 and Device S5 by inkjet printing of SrTiO₃ nanoparticle inks with IPE and EtOH:EG as solvents, respectively. From Figure 3, it can be noted that Device S5 exhibits a higher efficiency with substantial improvements in all photovoltaic parameters, as compared to that of Device S4. A capping layer of perovskite on top of the ETL is beneficial for the improved performance because direct contact between ETL and the hole transport layer thereby is avoided, and thus recombination losses reduced. As seen in Figure 4d,e, the more porous structure of the mp-SrTiO₃ films in Device S4 (Figure S3) results in a thinner capping perovskite layer, thereby causing more pronounced recombination losses and consequently accounting for the lower device performance.

Figure 3 also demonstrates that the photovoltaic performance of the SrTiO₃-based Device S5 is comparable to that of TiO₂-based devices. The V_{OC} of SrTiO₃-based solar cells is higher than that of the TiO₂-based counterparts. A reason for the higher photovoltages is that a higher conduction band edge energy level and a smaller band edge offset of the SrTiO₃ (Figure 5) will result in a higher V_{OC} [17,18]. Another reason is that a capping perovskite layer is clearly observed in SrTiO₃-based devices (Figure 4d,e) but not in TiO₂-based devices (Figure 4b,c), thereby reducing the direct shunt loss path and yielding a higher V_{OC} . The lower J_{SC} obtained from SrTiO₃-based devices as compared to that of TiO₂-based devices is because of the limited loading of perovskite, caused by the smaller effective surface area of mp-SrTiO₃ owing to the larger SrTiO₃ nanoparticles [17,68].

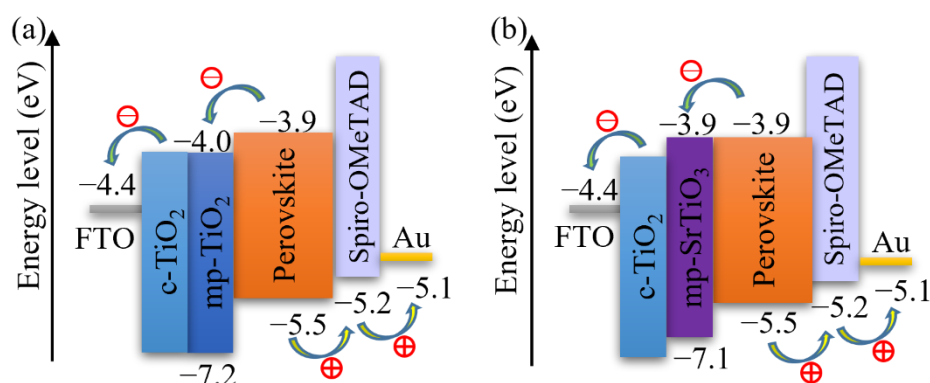


Figure 5. Schematics of the energy levels of (a) mp-TiO₂- and (b) mp-SrTiO₃-based PSCs.

The photovoltaic performance of mp-SrTiO₃-based PSCs was further optimized. As seen in Table 2 and Figure 6, with lower concentrations of SrTiO₃ in the nanoparticle inks, the performance of mp-SrTiO₃-based PSCs is improved with higher V_{OC} , J_{SC} and FF . The main reason can be attributed to the ink concentrations which influence the thickness and microstructure of the inkjet-printed mp-SrTiO₃ layer (Figure 7). Furthermore, inkjet-printed SrTiO₃-based devices based on a 0.15 M SrTiO₃ ink showed a PCE of 14.56%. Higher V_{OC} is obtained for PSCs manufactured from 0.15 M SrTiO₃ inks as compared to that of TiO₂-based devices, as mentioned before. However, the recorded J_{SC} is much lower since SrTiO₃ nanoparticles with a large size express a smaller effective surface area, which limits the overall interfacial area between the perovskite layer and the ETL [20]. Therefore, this can result in an inefficient charge transfer in SrTiO₃-based devices accounting for the pronounced scan-direction hysteresis observed, as shown in Figure 8, and the large difference in the overall performance for the two opposite scan directions. This dependence on the scan direction is much more pronounced in devices made from 0.15 M SrTiO₃ inks than in TiO₂-based ones. Another reason could be that SrTiO₃-based cells demonstrate a larger R_s than TiO₂-based ones. A higher R_s causes larger charge loss at high bias voltages and results in an inefficient charge transfer in the forward scan direction. In addition, the resulting SrTiO₃ ETL (Figure 7c) printed from 0.15 M SrTiO₃ inks is quite thin at some positions, which may cause the direct contact between the compact layer and the perovskite layer and then lead to recombination losses, thereby resulting in the pronounced hysteresis in SrTiO₃-based PSCs.

Table 2. Photovoltaic parameters from J - V characteristics of mp-SrTiO₃-based PSCs inkjet-printed with inks of different concentrations. Each average value is based on 9 devices.

Sample		PCE (%)	V_{OC} (V)	J_{SC} (mA/cm ²)	FF	R_s (Ω·cm ²)	R_{sh} (kΩ·cm ²)
0.35M SrTiO ₃	average	9.81 ± 1.34	1.03 ± 0.02	17.51 ± 1.05	0.54 ± 0.03	11	0.6
	champion	12.04	1.07	18.99	0.59		
0.25M SrTiO ₃	average	10.33 ± 1.28	1.04 ± 0.03	18.06 ± 1.00	0.55 ± 0.05	11	3.3
	champion	11.98	1.07	17.79	0.63		
0.15M SrTiO ₃	average	12.75 ± 0.97	1.07 ± 0.01	18.77 ± 0.84	0.63 ± 0.03	9.7	5.0
	champion	14.56	1.08	20.16	0.67		

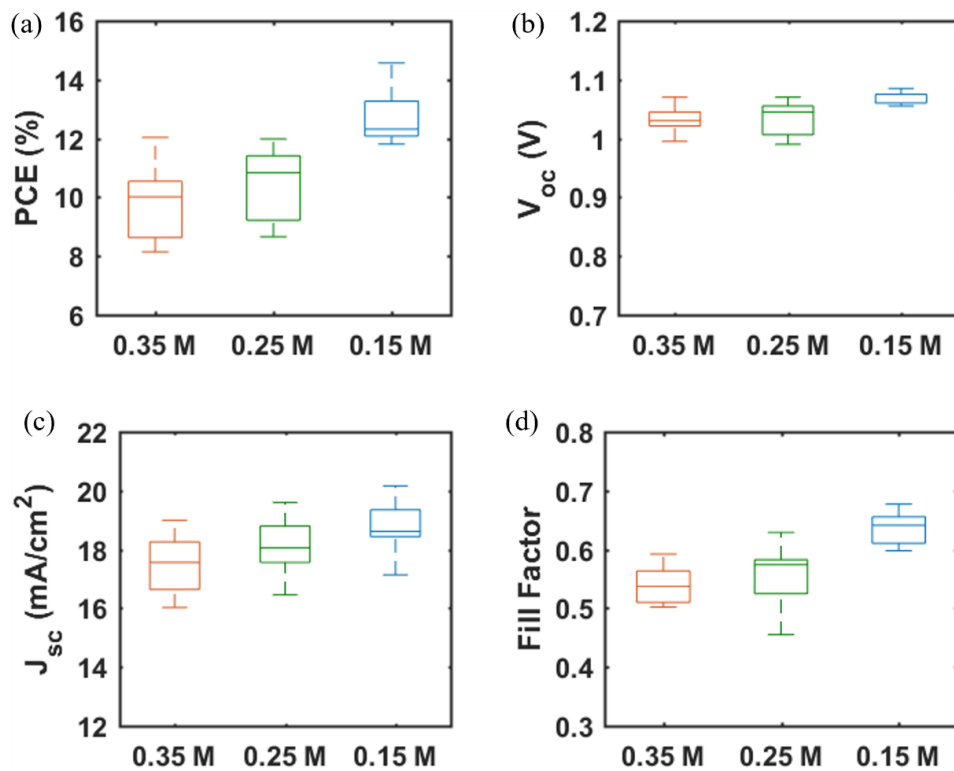


Figure 6. (a) PCE, (b) V_{oc} , (c) J_{sc} , and (d) FF of PSCs based on mp-SrTiO₃ ETLs inkjet-printed with SrTiO₃ nanoparticle inks of different concentrations. Data are obtained from 9 devices of each type.

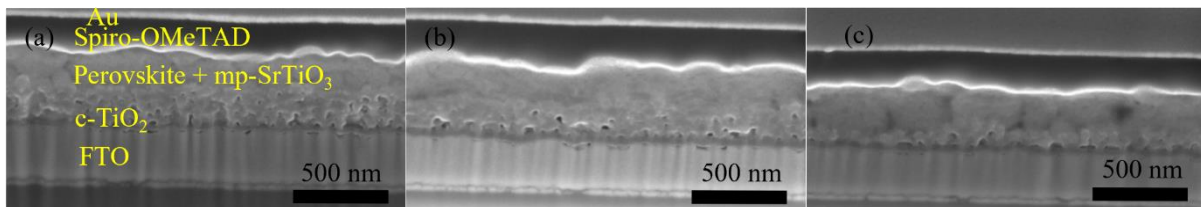


Figure 7. Cross-section SEM images of SrTiO₃-based PSCs inkjet-printed using (a) 0.35 M, (b) 0.25 M, and (c) 0.15 M inks.

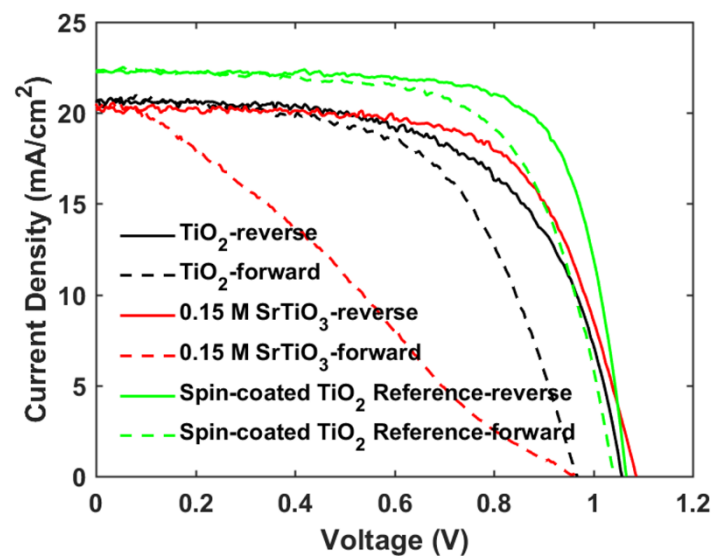


Figure 8. Reverse and forward scan J - V curves of mp-TiO₂- and mp-SrTiO₃-based PSCs.

To increase the J_{SC} and suppress the hysteresis observed in SrTiO₃-based PSCs, a small amount of TiO₂ nanoparticles was added to the SrTiO₃ nanoparticle inks to generate a mixed mesoporous SrTiO₃/TiO₂ ETL [21]. From Table 3 and Figure 9, the photovoltaic performance of SrTiO₃/TiO₂-based PSCs is improved as compared to SrTiO₃-based PSCs, and the devices display a highest PCE of 15.73% with $V_{OC} = 1.11$ V, $J_{SC} = 20.99$ mA/cm² and $FF = 0.68$ (TiO₂ concentration is 10 wt.% with respect to SrTiO₃). The V_{OC} shows a negligible change, while both J_{SC} and FF increase and then decrease with TiO₂ concentration ranging from 0 to 20%. Firstly, with introducing the smaller TiO₂ nanoparticles to the SrTiO₃ ETL, the interfacial area between the perovskite layer and the ETL will increase, and thus the J_{SC} is expected to increase. The smaller TiO₂ nanoparticles could also fill the pores between the larger SrTiO₃ nanoparticles and as well at the surface of the mesoporous ETL, which is beneficial for increasing the FF . With the TiO₂ concentration further increased from 10% to 20%, the electrical resistance of the SrTiO₃/TiO₂ ETL will increase, resulting in a decrease in J_{SC} and FF .

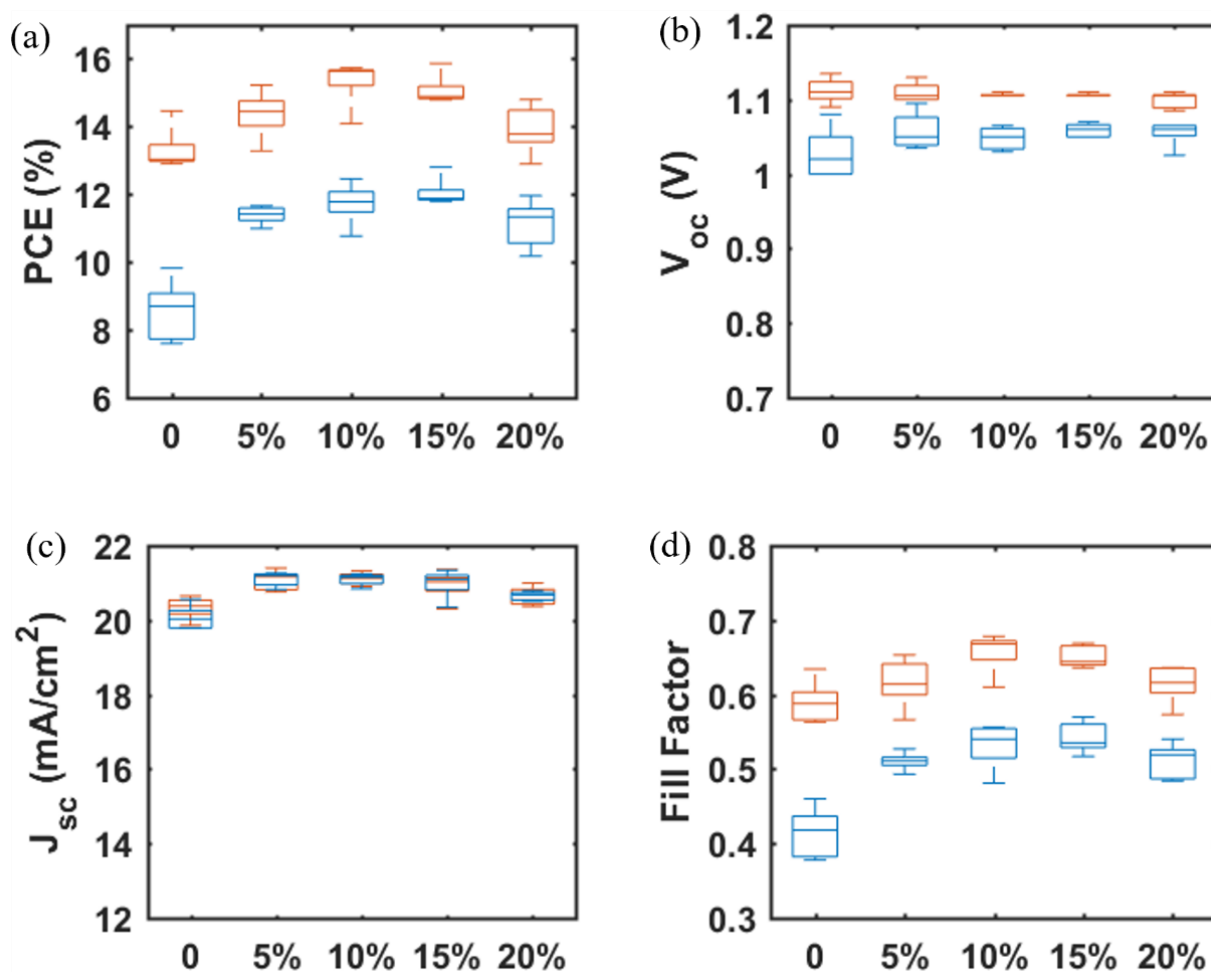


Figure 9. TiO₂ concentration-dependent photovoltaic parameters (a) PCE, (b) V_{OC} , (c) J_{SC} , and (d) FF for mp-SrTiO₃/TiO₂-based PSCs. Red boxes and blue boxes represent data from reverse and forward scan directions, respectively.

Table 3. Average photovoltaic parameters for mp-SrTiO₃/TiO₂-based PSCs inkjet-printed with inks of different TiO₂ concentrations. Each average value is based on 5 devices.

Sample	Scan Direction	PCE (%)	V _{OC} (V)	J _{sc} (mA/cm ²)	FF	R _s (Ω·cm ²)	R _{sh} (kΩ·cm ²)
SrTiO ₃ /0% TiO ₂	reverse	13.31 ± 0.65	1.11 ± 0.02	20.33 ± 0.30	0.59 ± 0.03	9.7	2.0
	forward	8.54 ± 0.90	1.03 ± 0.03	20.06 ± 0.32	0.41 ± 0.03		
	champion	14.46	1.12	20.38	0.63		
SrTiO ₃ /5% TiO ₂	reverse	14.37 ± 0.70	1.11 ± 0.01	21.00 ± 0.25	0.62 ± 0.03	8.3	1.4
	forward	11.39 ± 0.26	1.06 ± 0.02	21.10 ± 0.19	0.51 ± 0.01		
	champion	15.23	1.10	21.09	0.65		
SrTiO ₃ /10% TiO ₂	reverse	15.34 ± 0.70	1.11 ± 0.00	21.11 ± 0.17	0.66 ± 0.03	7.0	2.5
	forward	11.73 ± 0.61	1.05 ± 0.02	21.09 ± 0.16	0.53 ± 0.03		
	champion	15.73	1.10	20.99	0.68		
SrTiO ₃ /15% TiO ₂	reverse	15.07 ± 0.45	1.11 ± 0.00	20.94 ± 0.39	0.65 ± 0.01	7.3	3.3
	forward	12.05 ± 0.42	1.06 ± 0.01	20.98 ± 0.38	0.54 ± 0.02		
	champion	15.86	1.11	21.36	0.67		
SrTiO ₃ /20% TiO ₂	reverse	13.93 ± 0.72	1.10 ± 0.01	20.62 ± 0.25	0.61 ± 0.03	8.7	1.0
	forward	11.12 ± 0.69	1.06 ± 0.02	20.64 ± 0.12	0.51 ± 0.02		
	champion	14.80	1.11	20.99	0.64		

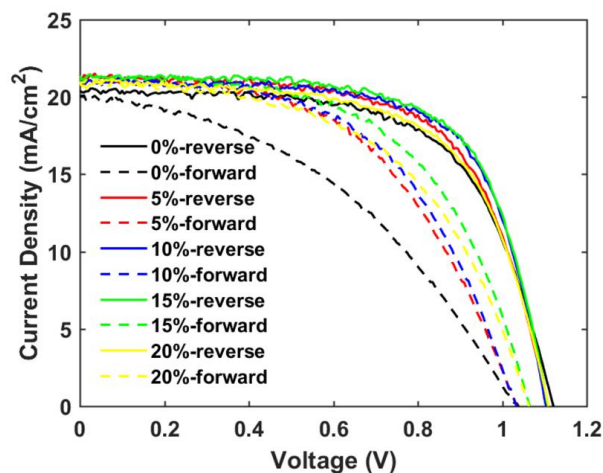
The main objective of the formulating nanoparticle mixtures, the severe hysteresis problem, is indeed found to become suppressed for the mixed SrTiO₃/TiO₂-based PSCs, as shown in Table 4 and Figure 10. A hysteresis index is normally used to describe the performance difference between the reverse and forward scan direction and is defined by

$$HI = \frac{PCE|reverse - PCE|forward}{PCE|reverse} \quad (1)$$

where $PCE|reverse$ and $PCE|forward$ represent the power conversion efficiency from the reverse and forward scan directions, respectively. The hysteresis index decreases by 34% when the TiO₂ concentration is increased from 0% to 10%, and the charge-transfer ability of the mp-SrTiO₃/TiO₂ ETL apparently is improved.

Table 4. Average hysteresis index for mp-SrTiO₃/TiO₂-based PSCs inkjet-printed with inks of different TiO₂ concentrations.

Sample	SrTiO ₃ /0% TiO ₂	SrTiO ₃ /5% TiO ₂	SrTiO ₃ /10% TiO ₂	SrTiO ₃ /15% TiO ₂	SrTiO ₃ /20% TiO ₂
Hysteresis index	0.36	0.21	0.24	0.20	0.20

**Figure 10.** Reverse and forward scan J - V curves of champion cells of mp-SrTiO₃/TiO₂-based PSCs inkjet-printed with inks of different TiO₂ concentrations.

Furthermore, we fabricated PSCs based on an inkjet-printed SnO₂ ETL layer with a configuration as illustrated in Figure 11. Also in this system, we used a cosolvent system for printing the SnO₂ thin films from the commercial SnO₂ colloidal dispersion. Water works as the main solvent, and ethylene glycol and Triton are used to control the drying properties and to adjust the viscosity and surface tension for high-quality printing results. The details for preparing the SnO₂ ink are described in the experimental section. The top SEM image of the inkjet-printed SnO₂ thin film is shown in Figure S4. By optimizing the SnO₂ thickness through changes of the ink concentration, we obtained a best performing cell with a PCE of 17.37% with a V_{OC} of 1.10V, J_{SC} of 21.13 mA/cm², and FF of 0.75, and the J - V characteristics are shown in Figure 12a. We also recorded the steady-state current density of the champion cell at the maximum power point $V_{mpp} = 0.925$ V. As seen in Figure 12b, a steady-state current density of 21.05 mA/cm² and a stabilized PCE of 17.30% were obtained, which agree well with the J_{SC} and PCE extracted from the J - V experiments. The average photovoltaic parameters of PSCs based on SnO₂ ETLs printed with inks of different concentrations are shown in Table 5. When the SnO₂ ink concentration is increased from 0.375% to 3%, the PCE (Figure S5) of SnO₂-based PSCs decreases continuously. As seen in the cross-sectional SEM images in Figure 13, the SnO₂ thickness increases from 30 nm to 130 nm when the ink concentration is increased from 0.375% to 3%. Too thick SnO₂ films result in large sheet resistances, hampering charge transfer, thereby leading to lower PCEs.

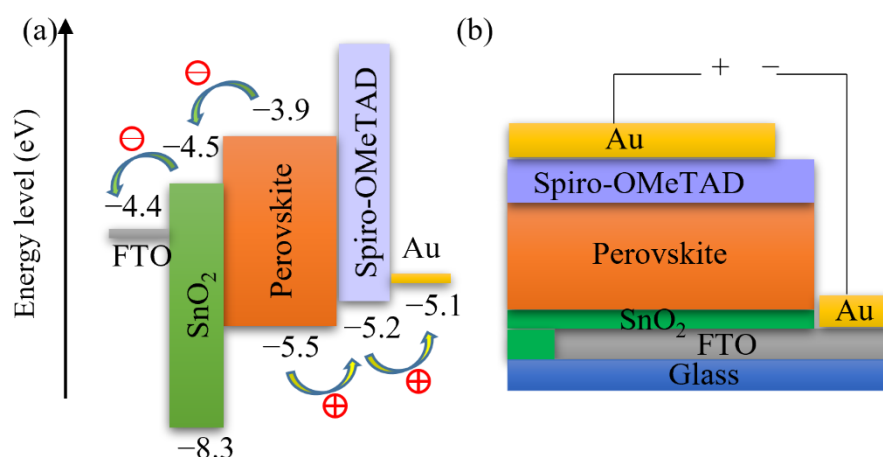


Figure 11. Schematics of (a) energy levels and (b) configuration of SnO₂-based PSCs.

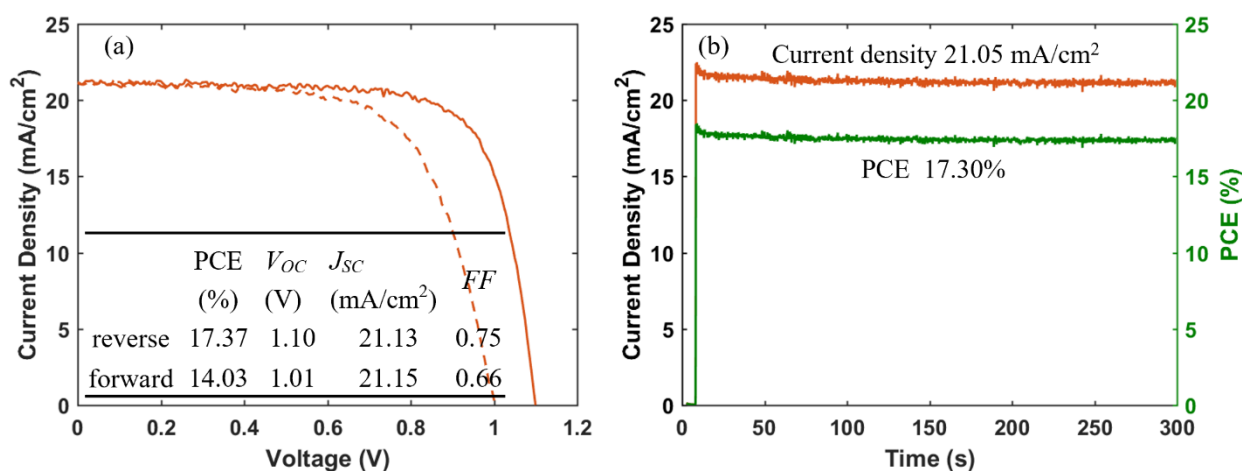


Figure 12. (a) J - V curves (solid and dashed lines represent data from reverse and forward scan, respectively), and (b) steady-state current density (at a bias of 0.925 V) and PCE of champion cell of PSCs based on inkjet-printed SnO₂ ETLs.

Table 5. Photovoltaic parameters for PSCs based on inkjet-printed SnO₂ ETLs prepared from different ink concentrations. Averages are based on different number of cells, as indicated in second column.

Cells	Numbers of Devices	Scan Direction	PCE (%)	V _{OC} (V)	J _{sc} (mA/cm ²)	FF	Hysteresis Index	R _s (Ω·cm ²)	R _{sh} (kΩ·cm ²)
0.375%	14	reverse	16.61 ± 0.70	1.10 ± 0.01	20.85 ± 0.26	0.73 ± 0.02	0.16 ± 0.03	4.9	5.0
		forward	13.93 ± 0.66	1.02 ± 0.01	21.01 ± 0.24	0.65 ± 0.03			
		champion	17.37	1.10	21.13	0.75			
0.75%	7	reverse	15.57 ± 0.66	1.06 ± 0.00	20.42 ± 0.36	0.72 ± 0.02	0.21 ± 0.02	5.7	2.5
		forward	12.29 ± 0.45	0.94 ± 0.01	20.42 ± 0.45	0.64 ± 0.02			
		champion	16.53	1.06	20.96	0.74			
1.5%	8	reverse	11.30 ± 0.77	0.95 ± 0.03	18.81 ± 0.46	0.63 ± 0.03	0.13 ± 0.04	9.5	1.1
		forward	9.85 ± 0.48	0.85 ± 0.02	18.73 ± 0.56	0.62 ± 0.03			
		champion	12.59	0.99	19.58	0.65			
3%	7	reverse	10.32 ± 0.81	0.91 ± 0.04	17.82 ± 0.45	0.64 ± 0.05	0.06 ± 0.03	8.5	2.0
		forward	9.73 ± 0.63	0.85 ± 0.03	17.72 ± 0.52	0.65 ± 0.04			
		champion	11.70	0.94	18.25	0.68			

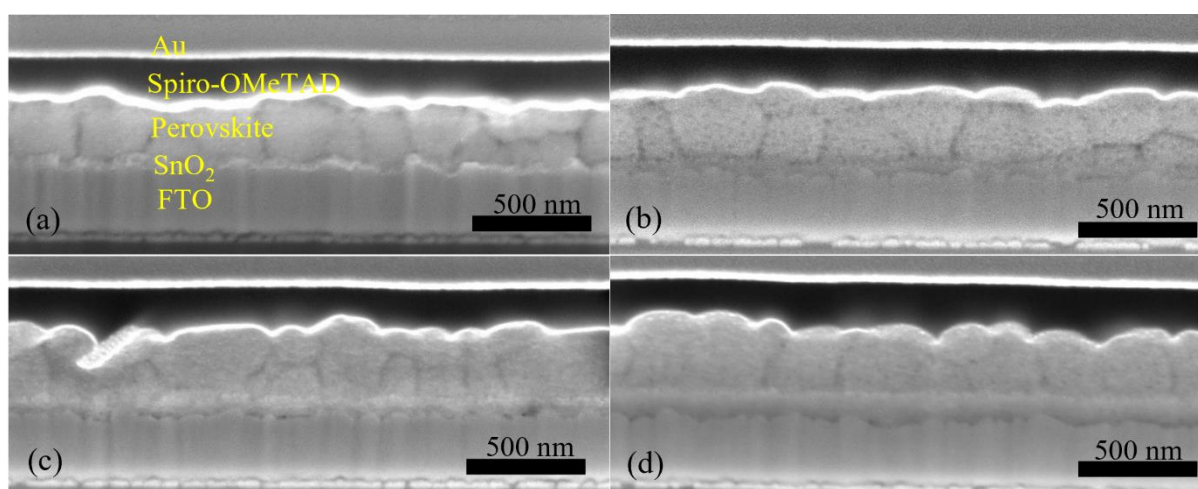


Figure 13. Cross-sectional SEM images for PSCs based on inkjet-printed SnO₂ ETLs prepared from (a) 0.375%, (b) 0.75%, (c) 1.5%, and (d) 3% precursor inks.

As compared to that of TiO₂- and SrTiO₃-based PSCs, the SnO₂-based congeners show less hysteresis, as well as better performance because of the superior electrical properties of the SnO₂ material, such as high electron mobility [26]. Therefore, highly efficient perovskite solar cells based on inkjet-printed SnO₂ ETLs should be attainable after a thorough optimization.

4. Conclusions

In this work, we developed the inkjet printing processes for TiO₂, SrTiO₃, and SnO₂ ETLs. By optimizing the PSC performance based on the printed ETLs, the drying properties of cosolvent inks are beneficial for uniform film formation of the mesoporous TiO₂ and SrTiO₃ ETLs. Although SrTiO₃-based PSCs perform better than TiO₂-based ones, the former devices suffer from low current density and severe scan-direction hysteresis. PSCs based on the mixed mesoporous SrTiO₃/TiO₂ ETL can mitigate these problems and offer a higher PCE than the SrTiO₃-only based devices. Furthermore, the strategy using cosolvent inks was also applied for inkjet printing of SnO₂ ETLs. The best performing SnO₂-based PSCs displayed an optimal PCE of 17.37% and low scan-direction hysteresis. In summary, the printing processes for different ETL materials were developed and will be used for fully inkjet-printed PSCs in our future work.

Supplementary Materials: The following are available online at <https://www.mdpi.com/article/10.3390/ma14247525/s1>, Figure S1: Device geometry of perovskite solar cells, Figure S2: J-V curves of champion PSCs based on mp-TiO₂ and mp-SrTiO₃ ETLs inkjet-printed from nanoparticle inks with IPE or EtOH:EG as solvents, Figure S3: SEM images of mesoporous layers inkjet-printed from

(a) TiO₂ inks with the solvent IPE, (b) TiO₂ inks with the solvent EtOH:EG, (c) SrTiO₃ inks with the solvent IPE, and (d) SrTiO₃ inks with the solvent EtOH:EG, Figure S4: SEM image of inkjet-printed SnO₂ thin film, Figure S5: *J-V* curves of champion PSCs based on inkjet-printed SnO₂ ETLs prepared from 0.375%, 0.75%, 1.5%, and 3% precursor inks, Table S1: Technical information of the printhead XJ126/50 and XJ126/80. These parameters are obtained from the Xaar126 data sheet.

Author Contributions: Conceptualization, D.L. and L.B.; methodology, D.L. and W.Z.; validation, D.L. and W.Z.; formal analysis, D.L.; investigation, D.L., W.Z. and L.B.; resources, L.K. and L.B.; data curation, D.L.; writing—original draft preparation, D.L.; writing—review and editing, D.L., W.Z., L.K. and L.B.; visualization, D.L.; supervision, L.B. and L.K.; project administration, L.B.; funding acquisition, D.L. and L.B. All authors have read and agreed to the published version of the manuscript.

Funding: This research was funded by the China Scholarship Council (CSC), the Swedish Research Council (VR), grant number 2017-04875, and the Carl Trygger Foundation.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: The data presented in this study are available on request from the corresponding author.

Acknowledgments: Authors acknowledge Resource Center for Coordination of Electron Microscopy at KTH and Anastasia Riazanova (Dept. of Fibre and Polymer Technology) for the help with part of SEM analysis.

Conflicts of Interest: The authors declare no conflict of interest. The funders had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, or in the decision to publish the results.

References

1. Tavakoli, M.M.; Giordano, F.; Zakeeruddin, S.M.; Grätzel, M. Mesoscopic Oxide Double Layer as Electron Specific Contact for Highly Efficient and UV Stable Perovskite Photovoltaics. *Nano Lett.* **2018**, *18*, 2428–2434. [CrossRef]
2. Lee, Y.; Paek, S.; Cho, K.T.; Oveisi, E.; Gao, P.; Lee, S.; Park, J.-S.; Zhang, Y.; Humphry-Baker, R.; Asiri, A.M.; et al. Enhanced charge collection with passivation of the tin oxide layer in planar perovskite solar cells. *J. Mater. Chem. A* **2017**, *5*, 12729–12734. [CrossRef]
3. Jeon, N.J.; Noh, J.H.; Kim, Y.C.; Yang, W.S.; Ryu, S.; Seok, S.I. Solvent engineering for high-performance inorganic–organic hybrid perovskite solar cells. *Nat. Mater.* **2014**, *13*, 897–903. [CrossRef] [PubMed]
4. Jeong, I.; Park, Y.H.; Bae, S.; Park, M.; Jeong, H.; Lee, P.; Ko, M.J. Solution-Processed Ultrathin TiO₂ Compact Layer Hybridized with Mesoporous TiO₂ for High-Performance Perovskite Solar Cells. *ACS Appl. Mater. Interfaces* **2017**, *9*, 36865–36874. [CrossRef]
5. Jung, E.H.; Jeon, N.J.; Park, E.Y.; Moon, C.S.; Shin, T.J.; Yang, T.-Y.; Noh, J.H.; Seo, J. Efficient, stable and scalable perovskite solar cells using poly(3-hexylthiophene). *Nat. Cell Biol.* **2019**, *567*, 511–515. [CrossRef]
6. Kim, M.; Kim, G.-H.; Lee, T.K.; Choi, I.W.; Choi, H.W.; Jo, Y.; Yoon, Y.J.; Kim, J.W.; Lee, J.; Huh, D.; et al. Methylammonium Chloride Induces Intermediate Phase Stabilization for Efficient Perovskite Solar Cells. *Joule* **2019**, *3*, 2179–2192. [CrossRef]
7. NREL. Best Research-Cell Efficiency Chart. Available online: <https://www.nrel.gov/pv/cell-efficiency.html> (accessed on 12 January 2021).
8. Tsvetkov, N.; Moon, B.C.; Lee, J.Y.; Kang, J.K. Controlled Synthesis of Nanocrystalline Nb:SrTiO₃ Electron Transport Layers for Robust Interfaces and Stable High Photovoltaic Energy Conversion Efficiency in Perovskite Halide Solar Cells. *ACS Appl. Energy Mater.* **2020**, *3*, 344–351. [CrossRef]
9. Yi, H.; Wang, D.; Mahmud, M.A.; Haque, F.; Upama, M.B.; Xu, C.; Duan, L.; Uddin, A. Bilayer SnO₂ as Electron Transport Layer for Highly Efficient Perovskite Solar Cells. *ACS Appl. Energy Mater.* **2018**, *1*, 6027–6039. [CrossRef]
10. Giordano, F.; Abate, A.; Baena, J.-P.C.; Saliba, M.; Matsui, T.; Im, S.H.; Zakeeruddin, S.M.; Nazeeruddin, M.K.; Hagfeldt, A.; Graetzel, M. Enhanced electronic properties in mesoporous TiO₂ via lithium doping for high-efficiency perovskite solar cells. *Nat. Commun.* **2016**, *7*, 10379. [CrossRef] [PubMed]
11. Chen, L.; Zhou, Y.; Tu, W.; Li, Z.; Bao, C.; Dai, H.; Yu, T.; Liu, J.; Zou, Z. Enhanced photovoltaic performance of a dye-sensitized solar cell using graphene–TiO₂ photoanode prepared by a novel in situ simultaneous reduction-hydrolysis technique. *Nanoscale* **2013**, *5*, 3481–3485. [CrossRef]
12. Wang, J.T.-W.; Ball, J.M.; Barea, E.M.; Abate, A.; Alexander-Webber, J.A.; Huang, J.; Saliba, M.; Mora-Sero, I.; Bisquert, J.; Snaith, H.J.; et al. Low-Temperature Processed Electron Collection Layers of Graphene/TiO₂ Nanocomposites in Thin Film Perovskite Solar Cells. *Nano Lett.* **2014**, *14*, 724–730. [CrossRef] [PubMed]

13. Wojciechowski, K.; Stranks, S.D.; Abate, A.; Sadoughi, G.; Sadhanala, A.; Kopidakis, N.; Rumbles, G.; Li, C.-Z.; Friend, R.H.; Jen, A.K.-Y.; et al. Heterojunction Modification for Highly Efficient Organic–Inorganic Perovskite Solar Cells. *ACS Nano* **2014**, *8*, 12701–12709. [[CrossRef](#)] [[PubMed](#)]
14. Li, T.; Rui, Y.; Zhang, X.; Shi, J.; Wang, X.; Wang, Y.; Yang, J.; Zhang, Q. Anatase TiO₂ nanorod arrays as high-performance electron transport layers for perovskite solar cells. *J. Alloys Compd.* **2020**, *849*, 156629. [[CrossRef](#)]
15. Choi, J.; Song, S.; Hörantner, M.T.; Snaith, H.J.; Park, T. Well-Defined Nanostructured, Single-Crystalline TiO₂ Electron Transport Layer for Efficient Planar Perovskite Solar Cells. *ACS Nano* **2016**, *10*, 6029–6036. [[CrossRef](#)] [[PubMed](#)]
16. Hou, Y.; Chen, X.; Yang, S.; Li, C.; Zhao, H.; Yang, H.G. A Band-Edge Potential Gradient Heterostructure to Enhance Electron Extraction Efficiency of the Electron Transport Layer in High-Performance Perovskite Solar Cells. *Adv. Funct. Mater.* **2017**, *27*, 1700878. [[CrossRef](#)]
17. Bera, A.; Wu, K.W.; Sheikh, A.; Alarousu, E.; Mohammed, O.F.; Wu, T. Perovskite Oxide SrTiO₃ as an Efficient Electron Transporter for Hybrid Perovskite Solar Cells. *J. Phys. Chem. C* **2014**, *118*, 28494–28501. [[CrossRef](#)]
18. Neophytou, M.; de Bastiani, M.; Gasparini, N.; Aydin, E.; Ugur, E.; Seitkhan, A.; Moruzzi, F.; Choaie, Y.; Ramadan, A.J.; Troughton, J.R.; et al. Enhancing the Charge Extraction and Stability of Perovskite Solar Cells Using Strontium Titanate (SrTiO₃) Electron Transport Layer. *ACS Appl. Energy Mater.* **2019**, *2*, 8090–8097. [[CrossRef](#)]
19. Wang, C.; Tang, Y.; Hu, Y.; Huang, L.; Fu, J.; Jin, J.; Shi, W.; Wang, L.; Yang, W. Graphene/SrTiO₃nanocomposites used as an effective electron-transporting layer for high-performance perovskite solar cells. *RSC Adv.* **2015**, *5*, 52041–52047. [[CrossRef](#)]
20. Hu, Y.; Wang, C.; Tang, Y.; Huang, L.; Fu, J.; Shi, W.; Wang, L.; Yang, W. Three-dimensional self-branching anatase TiO₂ nanorods with the improved carrier collection for SrTiO₃-based perovskite solar cells. *J. Alloys Compd.* **2016**, *679*, 32–38. [[CrossRef](#)]
21. Okamoto, Y.; Fukui, R.; Fukazawa, M.; Suzuki, Y. SrTiO₃/TiO₂ composite electron transport layer for perovskite solar cells. *Mater. Lett.* **2017**, *187*, 111–113. [[CrossRef](#)]
22. Mahmoudi, T.; Wang, Y.; Hahn, Y.-B. SrTiO₃/Al₂O₃-Graphene Electron Transport Layer for Highly Stable and Efficient Composites-Based Perovskite Solar Cells with 20.6% Efficiency. *Adv. Energy Mater.* **2020**, *10*, 1903369. [[CrossRef](#)]
23. Yeom, E.J.; Shin, S.S.; Yang, W.S.; Lee, S.J.; Yin, W.; Kim, D.; Noh, J.H.; Ahn, T.K.; Seok, S.I. Controllable synthesis of single crystalline Sn-based oxides and their application in perovskite solar cells. *J. Mater. Chem. A* **2017**, *5*, 79–86. [[CrossRef](#)]
24. Jiang, Q.; Zhao, Y.; Zhang, X.; Yang, X.; Chen, Y.; Chu, Z.; Ye, Q.; Li, X.; Yin, Z.; You, J. Surface passivation of perovskite film for efficient solar cells. *Nat. Photonics* **2019**, *13*, 460–466. [[CrossRef](#)]
25. Ke, W.; Fang, G.; Liu, Q.; Xiong, L.; Qin, P.; Tao, H.; Wang, J.; Lei, H.; Li, B.; Wan, J.; et al. Low-Temperature Solution-Processed Tin Oxide as an Alternative Electron Transporting Layer for Efficient Perovskite Solar Cells. *J. Am. Chem. Soc.* **2015**, *137*, 6730–6733. [[CrossRef](#)] [[PubMed](#)]
26. Jiang, Q.; Zhang, L.; Wang, H.; Yang, X.; Meng, J.; Liu, H.; Yin, Z.; Wu, J.; Zhang, X.; You, J. Enhanced electron extraction using SnO₂ for high-efficiency planar-structure HC(NH₂)₂PbI₃-based perovskite solar cells. *Nat. Energy* **2017**, *2*, 16177. [[CrossRef](#)]
27. Anaraki, E.H.; Kermanpur, A.; Steier, L.; Domanski, K.; Matsui, T.; Tress, W.; Saliba, M.; Abate, A.; Grätzel, M.; Hagfeldt, A.; et al. Highly efficient and stable planar perovskite solar cells by solution-processed tin oxide. *Energy Environ. Sci.* **2016**, *9*, 3128–3134. [[CrossRef](#)]
28. Zhu, Z.; Bai, Y.; Liu, X.; Chueh, C.-C.; Yang, S.; Jen, A.K.-Y. Enhanced Efficiency and Stability of Inverted Perovskite Solar Cells Using Highly Crystalline SnO₂ Nanocrystals as the Robust Electron-Transporting Layer. *Adv. Mater.* **2016**, *28*, 6478–6484. [[CrossRef](#)] [[PubMed](#)]
29. Xiong, L.; Qin, M.; Chen, C.; Wen, J.; Yang, G.; Guo, Y.; Ma, J.; Zhang, Q.; Qin, P.; Li, S.; et al. Fully High-Temperature-Processed SnO₂ as Blocking Layer and Scaffold for Efficient, Stable, and Hysteresis-Free Mesoporous Perovskite Solar Cells. *Adv. Funct. Mater.* **2018**, *28*, 1706276. [[CrossRef](#)]
30. Li, Y.; Zhu, J.; Huang, Y.; Liu, F.; Lv, M.; Chen, S.; Hu, L.; Tang, J.; Yao, J.; Dai, S. Mesoporous SnO₂ nanoparticle films as electron-transporting material in perovskite solar cells. *RSC Adv.* **2015**, *5*, 28424–28429. [[CrossRef](#)]
31. Ren, X.; Yang, D.; Yang, Z.; Feng, J.; Zhu, X.; Niu, J.; Liu, Y.; Zhao, W.; Liu, S.F. Solution-Processed Nb:SnO₂ Electron Transport Layer for Efficient Planar Perovskite Solar Cells. *ACS Appl. Mater. Interfaces* **2017**, *9*, 2421–2429. [[CrossRef](#)]
32. Yang, G.; Lei, H.; Tao, H.; Zheng, X.; Ma, J.; Liu, Q.; Ke, W.; Chen, Z.; Xiong, L.; Qin, P.; et al. Reducing Hysteresis and Enhancing Performance of Perovskite Solar Cells Using Low-Temperature Processed Y-Doped SnO₂Nanosheets as Electron Selective Layers. *Small* **2017**, *13*, 1601769. [[CrossRef](#)] [[PubMed](#)]
33. Roose, B.; Johansen, C.M.; Dupraz, K.; Jaouen, T.; Aebi, P.; Steiner, U.; Abate, A. A Ga-doped SnO₂ mesoporous contact for UV stable highly efficient perovskite solar cells. *J. Mater. Chem. A* **2018**, *6*, 1850–1857. [[CrossRef](#)]
34. Jung, E.H.; Chen, B.; Bertens, K.; Vafaie, M.; Teale, S.; Proppe, A.H.; Hou, Y.; Zhu, T.; Zheng, C.; Sargent, E.H. Bifunctional Surface Engineering on SnO₂ Reduces Energy Loss in Perovskite Solar Cells. *ACS Energy Lett.* **2020**, *5*, 2796–2801. [[CrossRef](#)]
35. Zhang, M.; Wu, F.; Chi, D.; Shi, K.; Huang, S. High-efficiency perovskite solar cells with poly(vinylpyrrolidone)-doped SnO₂ as an electron transport layer. *Mater. Adv.* **2020**, *1*, 617–624. [[CrossRef](#)]
36. Huang, X.; Hu, Z.; Xu, J.; Wang, P.; Wang, L.; Zhang, J.; Zhu, Y. Low-temperature processed SnO₂ compact layer by incorporating TiO₂ layer toward efficient planar heterojunction perovskite solar cells. *Sol. Energy Mater. Sol. Cells* **2017**, *164*, 87–92. [[CrossRef](#)]
37. Zhao, X.; Tao, L.; Li, H.; Huang, W.; Sun, P.; Liu, J.; Liu, S.; Sun, Q.; Cui, Z.; Sun, L.; et al. Efficient Planar Perovskite Solar Cells with Improved Fill Factor via Interface Engineering with Graphene. *Nano Lett.* **2018**, *18*, 2442–2449. [[CrossRef](#)]

38. Huckaba, A.J.; Lee, Y.; Xia, R.; Paek, S.; Bassetto, V.C.; Oveisi, E.; Lesch, A.; Kinge, S.; Dyson, P.J.; Girault, H.; et al. Inkjet-Printed Mesoporous TiO₂ and Perovskite Layers for High Efficiency Perovskite Solar Cells. *Energy Technol.* **2019**, *7*, 317–324. [[CrossRef](#)]
39. Buffiere, M.; Ali, K.; Fares, E.; Samara, A.; Shetty, A.R.; Al Hassan, O.; Belaidi, A. Inkjet-Printed Compact TiO₂ Electron Transport Layer for Perovskite Solar Cells. *Energy Technol.* **2020**, *8*, 2000330. [[CrossRef](#)]
40. Rohnacher, V.; Ullrich, F.; Eggers, H.; Schackmar, F.; Hell, S.; Salazar, A.; Huck, C.; Hernandez-Sosa, G.; Paetzold, U.W.; Jaegermann, W.; et al. Analytical Study of Solution-Processed Tin Oxide as Electron Transport Layer in Printed Perovskite Solar Cells. *Adv. Mater. Technol.* **2021**, *6*, 2000282. [[CrossRef](#)]
41. Xie, M.; Lu, H.; Zhang, L.; Wang, J.; Luo, Q.; Lin, J.; Ba, L.; Liu, H.; Shen, W.; Shi, L.; et al. Fully Solution-Processed Semi-Transparent Perovskite Solar Cells with Ink-Jet Printed Silver Nanowires Top Electrode. *Sol. RRL* **2018**, *2*, 1700184. [[CrossRef](#)]
42. Li, S.-G.; Jiang, K.-J.; Su, M.-J.; Cui, X.-P.; Huang, J.-H.; Zhang, Q.-Q.; Zhou, X.-Q.; Yang, L.-M.; Song, Y.-L. Inkjet printing of CH₃NH₃PbI₃ on a mesoscopic TiO₂ film for highly efficient perovskite solar cells. *J. Mater. Chem. A* **2015**, *3*, 9092–9097. [[CrossRef](#)]
43. Gheno, A.; Huang, Y.; Bouclé, J.; Ratier, B.; Rolland, A.; Even, J.; Vedraïne, S. Toward Highly Efficient Inkjet-Printed Perovskite Solar Cells Fully Processed Under Ambient Conditions and at Low Temperature. *Sol. RRL* **2018**, *2*, 1800191. [[CrossRef](#)]
44. Mathies, F.; Eggers, H.; Richards, B.S.; Hernandez-Sosa, G.; Lemmer, U.; Paetzold, U.W. Inkjet-Printed Triple Cation Perovskite Solar Cells. *ACS Appl. Energy Mater.* **2018**, *1*, 1834–1839. [[CrossRef](#)]
45. Eggers, H.; Schackmar, F.; Abzieher, T.; Sun, Q.; Lemmer, U.; Vaynzof, Y.; Richards, B.; Hernandez-Sosa, G.; Paetzold, U.W. Inkjet-Printed Micrometer-Thick Perovskite Solar Cells with Large Columnar Grains. *Adv. Energy Mater.* **2020**, *10*, 1903184. [[CrossRef](#)]
46. Schackmar, F.; Eggers, H.; Frericks, M.; Richards, B.S.; Lemmer, U.; Hernandez-Sosa, G.; Paetzold, U.W. Perovskite Solar Cells with All-Inkjet-Printed Absorber and Charge Transport Layers. *Adv. Mater. Technol.* **2021**, *6*, 2000271. [[CrossRef](#)]
47. Jung, K.-H.; Seo, J.-Y.; Lee, S.; Shin, H.; Park, N.-G. Solution-processed SnO₂ thin film for a hysteresis-free planar perovskite solar cell with a power conversion efficiency of 19.2%. *J. Mater. Chem. A* **2017**, *5*, 24790–24803. [[CrossRef](#)]
48. Wang, C.; Zhao, D.; Grice, C.R.; Liao, W.; Yu, Y.; Cimaroli, A.; Shrestha, N.; Roland, P.J.; Chen, J.; Yu, Z.; et al. Low-temperature plasma-enhanced atomic layer deposition of tin oxide electron selective layers for highly efficient planar perovskite solar cells. *J. Mater. Chem. A* **2016**, *4*, 12080–12087. [[CrossRef](#)]
49. Ma, J.; Zheng, X.; Lei, H.; Ke, W.; Chen, C.; Chen, Z.; Yang, G.; Fang, G. Highly Efficient and Stable Planar Perovskite Solar Cells with Large-Scale Manufacture of E-Beam Evaporated SnO₂ Toward Commercialization. *Sol. RRL* **2017**, *1*, 1700118. [[CrossRef](#)]
50. Liu, X.; Tsai, K.-W.; Zhu, Z.; Sun, Y.; Chueh, C.-C.; Jen, A.K.-Y. A Low-Temperature, Solution Processable Tin Oxide Electron-Transporting Layer Prepared by the Dual-Fuel Combustion Method for Efficient Perovskite Solar Cells. *Adv. Mater. Interfaces* **2016**, *3*, 1600122. [[CrossRef](#)]
51. Zhang, W.; Liu, P.; Sadollahkhani, A.; Li, Y.; Zhang, B.; Zhang, F.; Safdari, M.; Hao, Y.; Hua, Y.; Kloo, L. Investigation of Tri-phenylamine (TPA)-Based Metal Complexes and Their Application in Perovskite Solar Cells. *ACS Omega* **2017**, *2*, 9231–9240. [[CrossRef](#)]
52. Bernacka-Wojcik, I.; Wojcik, P.J.; Aguas, H.; Fortunato, E.; Martins, R. Inkjet printed highly porous TiO₂ films for improved electrical properties of photoanode. *J. Colloid Interface Sci.* **2016**, *465*, 208–214. [[CrossRef](#)]
53. Cherrington, R.; Hughes, D.J.; Senthilarasu, S.; Goodship, V. Inkjet-Printed TiO₂ Nanoparticles from Aqueous Solutions for Dye-Sensitized Solar Cells (DSSCs). *Energy Technol.* **2015**, *3*, 866–870. [[CrossRef](#)]
54. Oh, Y.; Yoon, H.G.; Lee, S.-N.; Kim, H.-K.; Kim, J. Inkjet-Printing of TiO₂ Co-Solvent Ink: From Uniform Ink-Droplet to TiO₂ Photoelectrode for Dye-Sensitized Solar Cells. *J. Electrochem. Soc.* **2011**, *159*, B34–B38. [[CrossRef](#)]
55. Kunugi, Y.; Shimoyama, Y.; Umezue, S. Fabrication of Dye-sensitized Solar Cells Using Electrostatic Inkjet Printing. *J. Photopolym. Sci. Technol.* **2013**, *26*, 383–385. [[CrossRef](#)]
56. Peng, B.; Jungmann, G.; Jäger, C.; Haarer, D.; Schmidt, H.-W.; Thelakkat, M. Systematic investigation of the role of compact TiO₂ layer in solid state dye-sensitized TiO₂ solar cells. *Co-Ord. Chem. Rev.* **2004**, *248*, 1479–1489. [[CrossRef](#)]
57. Wu, Y.; Yang, X.; Chen, H.; Zhang, K.; Qin, C.; Liu, J.; Peng, W.; Islam, A.; Bi, E.; Ye, F.; et al. Highly compact TiO₂ layer for efficient hole-blocking in perovskite solar cells. *Appl. Phys. Express* **2014**, *7*, 052301. [[CrossRef](#)]
58. Deegan, R.D.; Bakajin, O.; Dupont, T.F.; Huber, G.; Nagel, S.R.; Witten, T.A. Capillary flow as the cause of ring stains from dried liquid drops. *Nature* **1997**, *389*, 827. [[CrossRef](#)]
59. Li, Y.-F.; Sheng, Y.-J.; Tsao, H.-K. Evaporation Stains: Suppressing the Coffee-Ring Effect by Contact Angle Hysteresis. *Langmuir* **2013**, *29*, 7802–7811. [[CrossRef](#)] [[PubMed](#)]
60. Kim, D.; Jeong, S.; Park, B.K.; Moon, J. Direct writing of silver conductive patterns: Improvement of film morphology and conductance by controlling solvent compositions. *Appl. Phys. Lett.* **2006**, *89*, 264101. [[CrossRef](#)]
61. Park, J.; Moon, J. Control of Colloidal Particle Deposit Patterns within Picoliter Droplets Ejected by Ink-Jet Printing. *Langmuir* **2006**, *22*, 3506–3513. [[CrossRef](#)]
62. Sun, D.; Chen, C.; Zhang, J.; Wu, X.; Chen, H.; Guo, T. High performance inkjet-printed metal oxide thin film transistors via addition of insulating polymer with proper molecular weight. *Appl. Phys. Lett.* **2018**, *112*, 012102. [[CrossRef](#)]
63. Soltman, D.; Subramanian, V. Inkjet-Printed Line Morphologies and Temperature Control of the Coffee Ring Effect. *Langmuir* **2008**, *24*, 2224–2231. [[CrossRef](#)]
64. Li, Y.; Yang, Q.; Li, M.; Song, Y. Rate-dependent interface capture beyond the coffee-ring effect. *Sci. Rep.* **2016**, *6*, 24628. [[CrossRef](#)]

65. Yunker, P.J.; Still, T.; Lohr, M.A.; Yodh, A.G. Suppression of the coffee-ring effect by shape-dependent capillary interactions. *Nat. Cell Biol.* **2011**, *476*, 308–311. [[CrossRef](#)] [[PubMed](#)]
66. Tang, Z.; Fang, K.; Bukhari, M.N.; Song, Y.; Zhang, K. Effects of Viscosity and Surface Tension of a Reactive Dye Ink on Droplet Formation. *Langmuir* **2020**, *36*, 9481–9488. [[CrossRef](#)] [[PubMed](#)]
67. Zhu, Z.; Zhang, J.; Guo, D.; Ning, H.; Zhou, S.; Liang, Z.; Yao, R.; Wang, Y.; Lu, X.; Peng, J. Functional Metal Oxide Ink Systems for Drop-on-Demand Printed Thin-Film Transistors. *Langmuir* **2020**, *36*, 8655–8667. [[CrossRef](#)] [[PubMed](#)]
68. Phoon, B.L.; Lai, C.W.; Pan, G.-T.; Yang, T.C.-K.; Juan, J.C. One-pot hydrothermal synthesis of strontium titanate nanoparticles photoelectrode using electrophoretic deposition for enhancing photoelectrochemical water splitting. *Ceram. Int.* **2018**, *44*, 9923–9933. [[CrossRef](#)]