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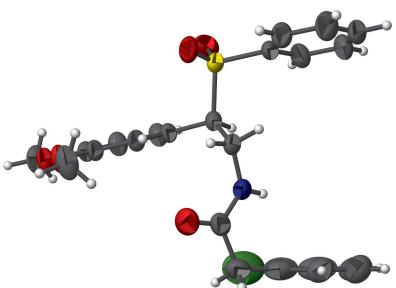
N-[2-(3,4-Dimethoxyphenyl)-2-(phenylsulfanyl)-ethyl]-2-(2-fluorophenyl)acetamide

Marco A. Ovalle, José A. Romero and Gerardo Aguirre*

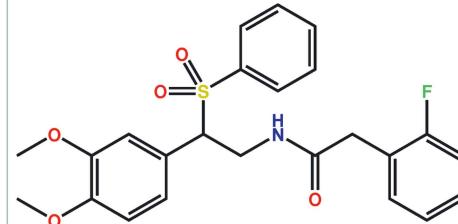
Tecnológico Nacional de México/Instituto Tecnológico de Tijuana, Centro de Graduados e Investigación en Química. Apartado Postal 1166, Tijuana, B.C., Mexico. *Correspondence e-mail: gaguirre@tectijuana.mx

The title compound, $C_{24}H_{24}FNO_3S$, is an intermediate in the synthesis of fluorine containing isoquinoline alkaloids, which crystallizes in the triclinic space group $P\bar{1}$ with one molecule in the asymmetric unit. The structure presents a racemic mixture of enantiomers. The C—S—C—C torsion angle between the benzene ring system and the sulfonyl benzene ring is $-178.5(1)^\circ$. In the crystal, N—H···O hydrogen bonds between neighbouring molecules form chains of molecules along the a -axis direction.

3D view



Chemical scheme



Structure description

Sulfur- and fluorine-containing molecules play important roles in medicinal chemistry. Sulfur-containing compounds often show a variety of biological activities and serve important functions in applications in the pharmaceutical industry (Bernardi *et al.*, 1985). A variety of sulfur-containing molecules have been isolated from natural sources and play major roles in drug discovery and development. The role of fluorine in drug design and development is expanding rapidly, as more is learned about the unique properties associated with this unusual element and how to deploy it in the pharmaceutical industry. The introduction of fluorine into a molecule can influence conformation, pK_a , intrinsic potency, membrane permeability, metabolic pathways, and pharmacokinetic properties (Gillis *et al.*, 2015). Various sulfur- and fluorine-containing molecules have been studied for their applications in medicinal chemistry, with those containing a sulfone group emerging with promising results. Some examples are the recently reported thiochroman-4-one derivatives (Vargas *et al.*, 2017), in which structure–activity relationships have been studied and it has been found that the vinyl sulfone and fluorine moieties play important roles in the biological activity of the molecules (Fig. 1). The literature also reveals that these types of compounds also serve as neuroprotective agents (Woo *et al.*, 2014), exhibit



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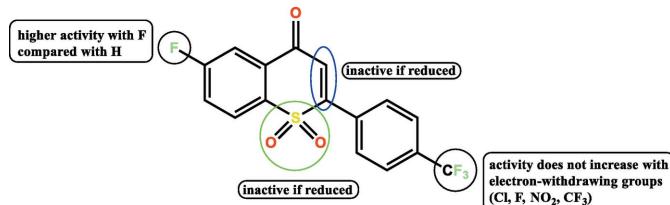


Figure 1
Structure–activity relationships of thiochroman-4-one derivatives

antituberculosis activity (Tiwari *et al.*, 2015) and inhibit prostate cancer (Goa & Spencer, 1998).

Continuing our interest in developing new sulfur- and fluorine-containing C17 S1 C7 C1 biologically active alkaloids, we report here the synthesis and characterization of the title compound (Fig. 2) as a racemic mixture. The torsion angle between the benzene ring system and the sulfonyl benzene ring is $-178.5(1)^\circ$. The C11–C10–C9 angle [117.8(2) $^\circ$] is slightly widened in comparison to an sp^3 -hybridized carbon atom; this is probably due to an attractive interaction between the fluorine on the benzene ring and the hydrogen atoms on the benzyl carbon. In the crystal, N–H \cdots O hydrogen bonds between neighbouring molecules form chains of molecules along the *a*-axis direction (Table 1; Fig. 3).

Synthesis and crystallization

The title compound was synthesized by the oxidation of *N*-(2-(3,4-dimethoxyphenyl)-2-(phenylthio)ethyl)-2-(2-fluorophenyl)acetamide (0.229 g, 0.54 mmol) treated with NaIO₄ (0.264, 1.23 mmol) in water (6 ml). The reaction mixture was stirred for 2 h in reflux and then was allowed to cool down at room temperature prior to extractions with DCM (3×20 ml). The solvent in the combined organic layer was removed under vacuum and purified by flash chromatography on silica gel (DCM/MeOH 95:5) to give a pale-yellow solid in 67% yield (0.163 g, 0.36 mmol).

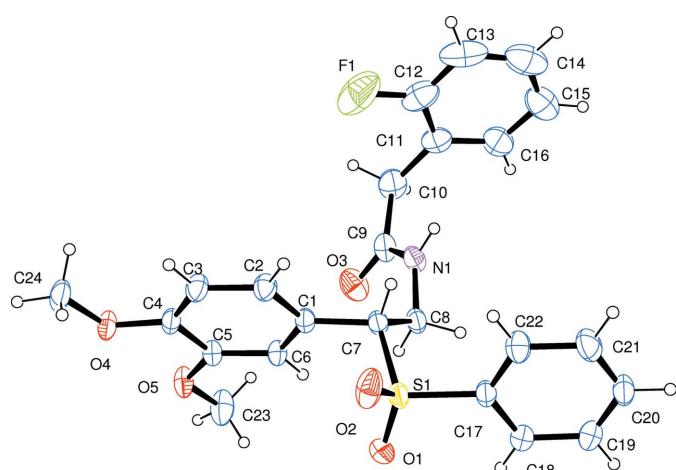


Figure 2
The molecular structure of the title compound with displacement ellipsoids drawn at the 30% probability level.

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
N1–H1 \cdots O4 ⁱ	0.86	2.46	3.207 (2)	145
N1–H1 \cdots O5 ⁱ	0.86	2.50	2.945 (3)	113

Symmetry code: (i) $x - 1, y, z$.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₂₄ H ₂₄ FNO ₅ S
M_r	457.50
Crystal system, space group	Triclinic, $P\bar{1}$
Temperature (K)	293
a, b, c (Å)	8.3751 (3), 9.7640 (4), 15.3592 (5)
α, β, γ ($^\circ$)	97.676 (3), 93.885 (3), 115.319 (4)
V (Å ³)	1114.18 (8)
Z	2
Radiation type	Cu $K\alpha$
μ (mm ⁻¹)	1.68
Crystal size (mm)	0.21 \times 0.12 \times 0.07
Data collection	
Diffractometer	Rigaku SuperNova, Dual, Cu at zero, AtlasS2
Absorption correction	Multi-scan (<i>CrysAlis PRO</i> ; Rigaku OD, 2015)
T_{\min}, T_{\max}	0.816, 1.000
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	14171, 4663, 4030
R_{int}	0.022
(sin θ/λ) _{max} (Å ⁻¹)	0.631
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.052, 0.151, 1.08
No. of reflections	4663
No. of parameters	291
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å ⁻³)	0.37, -0.30

Computer programs: *CrysAlis PRO* (Rigaku OD, 2015), *SIR2004* (Burla *et al.*, 2007), *SHELXL* (Sheldrick, 2015), *Mercury* (Macrae *et al.*, 2020) and *publCIF* (Westrip, 2010).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

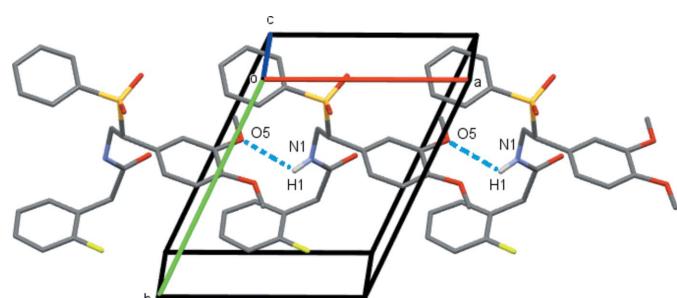


Figure 3
Crystal packing of the title compound viewed along the *c* axis. Intermolecular hydrogen bonds are shown as dashed blue lines. Hydrogen atoms not involved in hydrogen bonding are omitted for clarity.

Funding information

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full crystallographic data

IUCrData (2021). **6**, x201663 [https://doi.org/10.1107/S2414314620016636]

N-[2-(3,4-Dimethoxyphenyl)-2-(phenylsulfanyl)ethyl]-2-(2-fluorophenyl)-acetamide

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N-[2-(3,4-Dimethoxyphenyl)-2-(phenylsulfanyl)ethyl]-2-(2-fluorophenyl)acetamide

Crystal data

$C_{24}H_{24}FNO_5S$
 $M_r = 457.50$
Triclinic, $P\bar{1}$
 $a = 8.3751 (3)$ Å
 $b = 9.7640 (4)$ Å
 $c = 15.3592 (5)$ Å
 $\alpha = 97.676 (3)^\circ$
 $\beta = 93.885 (3)^\circ$
 $\gamma = 115.319 (4)^\circ$
 $V = 1114.18 (8)$ Å³

$Z = 2$
 $F(000) = 480$
 $D_x = 1.364 \text{ Mg m}^{-3}$
Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å
Cell parameters from 6250 reflections
 $\theta = 2.9\text{--}76.6^\circ$
 $\mu = 1.68 \text{ mm}^{-1}$
 $T = 293 \text{ K}$
Irregular, translucent intense colourless
0.21 × 0.12 × 0.07 mm

Data collection

Rigaku SuperNova, Dual, Cu at zero, AtlasS2
diffractometer
Radiation source: micro-focus sealed X-ray
tube, SuperNova (Cu) X-ray Source
Mirror monochromator
Detector resolution: 5.1980 pixels mm⁻¹
 ω scans
Absorption correction: multi-scan
(CrysAlisPro; Rigaku OD, 2015)

$T_{\min} = 0.816$, $T_{\max} = 1.000$
14171 measured reflections
4663 independent reflections
4030 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.022$
 $\theta_{\max} = 76.8^\circ$, $\theta_{\min} = 2.9^\circ$
 $h = -10 \rightarrow 10$
 $k = -12 \rightarrow 12$
 $l = -19 \rightarrow 17$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.052$
 $wR(F^2) = 0.151$
 $S = 1.08$
4663 reflections
291 parameters
0 restraints

Primary atom site location: structure-invariant
direct methods
Hydrogen site location: inferred from
neighbouring sites
H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0751P)^2 + 0.4124P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.37 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.30 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. The C-bound H atoms were placed in calculated positions and treated as riding: C—H = 0.95–0.99 Å with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{Cmethyl})$ and $1.2U_{\text{eq}}(\text{C})$ for other H atoms.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.39652 (6)	0.22548 (6)	0.61360 (4)	0.04490 (17)
F1	0.6743 (4)	0.9600 (3)	0.84275 (19)	0.1440 (11)
O1	0.4440 (2)	0.1166 (2)	0.64885 (15)	0.0685 (5)
O2	0.4379 (2)	0.2591 (2)	0.52804 (11)	0.0733 (6)
O3	0.6887 (3)	0.5399 (3)	0.93514 (12)	0.0724 (5)
O4	1.24722 (18)	0.61837 (19)	0.69344 (11)	0.0505 (4)
O5	1.09028 (19)	0.4413 (2)	0.80129 (10)	0.0524 (4)
N1	0.4765 (2)	0.5359 (2)	0.83493 (12)	0.0482 (4)
H1	0.421805	0.586998	0.818120	0.058*
C1	0.7007 (2)	0.4691 (2)	0.69245 (13)	0.0387 (4)
C2	0.7860 (3)	0.5660 (3)	0.63531 (15)	0.0480 (5)
H2	0.720985	0.596306	0.597378	0.058*
C3	0.9689 (3)	0.6187 (3)	0.63396 (15)	0.0493 (5)
H3	1.024558	0.683868	0.595053	0.059*
C4	1.0682 (2)	0.5755 (2)	0.68960 (14)	0.0416 (4)
C5	0.9826 (3)	0.4781 (2)	0.74848 (13)	0.0401 (4)
C6	0.8009 (2)	0.4252 (2)	0.74868 (13)	0.0400 (4)
H6	0.744532	0.359023	0.787008	0.048*
C7	0.5010 (2)	0.4060 (2)	0.69177 (13)	0.0395 (4)
H7	0.464425	0.478089	0.667997	0.047*
C8	0.4351 (3)	0.3881 (3)	0.78194 (14)	0.0443 (5)
H8A	0.307058	0.324890	0.773090	0.053*
H8B	0.490585	0.336217	0.813264	0.053*
C9	0.5952 (3)	0.5977 (3)	0.90877 (15)	0.0517 (5)
C10	0.6108 (4)	0.7481 (3)	0.95980 (18)	0.0684 (7)
H10A	0.599310	0.736563	1.021142	0.082*
H10B	0.730226	0.827165	0.958892	0.082*
C11	0.4797 (4)	0.8064 (3)	0.92863 (16)	0.0598 (6)
C12	0.5128 (6)	0.9088 (4)	0.8713 (2)	0.0853 (10)
C13	0.3893 (8)	0.9597 (4)	0.8421 (2)	0.1011 (14)
H13	0.416268	1.030105	0.803685	0.121*
C14	0.2253 (7)	0.9011 (5)	0.8724 (3)	0.1040 (13)
H14	0.139973	0.931991	0.853658	0.125*
C15	0.1876 (6)	0.8014 (5)	0.9281 (3)	0.0930 (11)
H15	0.076739	0.762740	0.947768	0.112*
C16	0.3134 (4)	0.7560 (4)	0.9562 (2)	0.0728 (8)
H16	0.285373	0.687787	0.995874	0.087*

C17	0.1652 (3)	0.1620 (2)	0.61424 (14)	0.0426 (4)
C18	0.0725 (3)	0.0374 (3)	0.65450 (19)	0.0578 (6)
H18	0.130805	-0.013477	0.679429	0.069*
C19	-0.1083 (4)	-0.0101 (3)	0.6570 (2)	0.0741 (8)
H19	-0.171943	-0.092894	0.684308	0.089*
C20	-0.1935 (3)	0.0647 (3)	0.6194 (2)	0.0682 (7)
H20	-0.315207	0.031227	0.620524	0.082*
C21	-0.1021 (3)	0.1872 (4)	0.58042 (19)	0.0648 (7)
H21	-0.161059	0.238211	0.556282	0.078*
C22	0.0794 (3)	0.2369 (3)	0.57638 (16)	0.0552 (6)
H22	0.141613	0.319261	0.548562	0.066*
C23	1.0135 (3)	0.3553 (4)	0.86799 (19)	0.0692 (8)
H23A	0.927690	0.253193	0.840542	0.104*
H23B	0.955450	0.405134	0.902463	0.104*
H23C	1.105425	0.349513	0.905972	0.104*
C24	1.3384 (3)	0.7113 (3)	0.6318 (2)	0.0645 (7)
H24A	1.287198	0.656772	0.572378	0.097*
H24B	1.462304	0.733861	0.640809	0.097*
H24C	1.326957	0.805723	0.640687	0.097*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0308 (2)	0.0492 (3)	0.0513 (3)	0.0168 (2)	0.00980 (19)	-0.0007 (2)
F1	0.155 (2)	0.1078 (18)	0.128 (2)	0.0098 (17)	0.0651 (18)	0.0350 (16)
O1	0.0445 (9)	0.0507 (9)	0.1138 (16)	0.0277 (8)	0.0078 (9)	0.0042 (9)
O2	0.0508 (10)	0.0919 (14)	0.0495 (10)	0.0093 (9)	0.0176 (8)	-0.0057 (9)
O3	0.0679 (12)	0.0981 (15)	0.0585 (11)	0.0487 (11)	-0.0068 (9)	0.0035 (10)
O4	0.0292 (7)	0.0595 (9)	0.0602 (9)	0.0132 (6)	0.0144 (6)	0.0205 (7)
O5	0.0329 (7)	0.0771 (11)	0.0533 (9)	0.0249 (7)	0.0114 (6)	0.0270 (8)
N1	0.0403 (9)	0.0568 (11)	0.0489 (10)	0.0267 (8)	0.0018 (7)	-0.0023 (8)
C1	0.0294 (9)	0.0432 (10)	0.0413 (10)	0.0145 (8)	0.0053 (7)	0.0057 (8)
C2	0.0393 (10)	0.0534 (12)	0.0534 (12)	0.0199 (9)	0.0061 (9)	0.0189 (10)
C3	0.0398 (11)	0.0536 (12)	0.0552 (12)	0.0164 (9)	0.0138 (9)	0.0239 (10)
C4	0.0288 (9)	0.0451 (10)	0.0465 (11)	0.0115 (8)	0.0105 (8)	0.0080 (8)
C5	0.0311 (9)	0.0493 (11)	0.0390 (10)	0.0166 (8)	0.0069 (7)	0.0088 (8)
C6	0.0316 (9)	0.0484 (11)	0.0404 (10)	0.0161 (8)	0.0112 (7)	0.0121 (8)
C7	0.0314 (9)	0.0442 (10)	0.0438 (10)	0.0180 (8)	0.0066 (7)	0.0062 (8)
C8	0.0333 (9)	0.0504 (11)	0.0499 (11)	0.0197 (8)	0.0105 (8)	0.0045 (9)
C9	0.0410 (11)	0.0665 (14)	0.0448 (12)	0.0226 (10)	0.0062 (9)	0.0053 (10)
C10	0.0630 (16)	0.0700 (17)	0.0566 (15)	0.0229 (13)	-0.0015 (12)	-0.0123 (12)
C11	0.0762 (17)	0.0491 (13)	0.0452 (12)	0.0237 (12)	0.0045 (11)	-0.0052 (10)
C12	0.112 (3)	0.0577 (16)	0.0650 (18)	0.0184 (17)	0.0204 (17)	0.0056 (13)
C13	0.174 (5)	0.0602 (19)	0.067 (2)	0.053 (2)	-0.006 (2)	0.0107 (15)
C14	0.144 (4)	0.094 (3)	0.085 (3)	0.075 (3)	-0.010 (3)	-0.011 (2)
C15	0.099 (3)	0.106 (3)	0.087 (2)	0.063 (2)	0.0086 (19)	0.001 (2)
C16	0.082 (2)	0.0758 (19)	0.0651 (17)	0.0404 (16)	0.0142 (14)	0.0053 (14)
C17	0.0301 (9)	0.0480 (11)	0.0444 (11)	0.0151 (8)	0.0050 (8)	-0.0017 (8)

C18	0.0417 (12)	0.0467 (12)	0.0840 (17)	0.0176 (10)	0.0137 (11)	0.0130 (12)
C19	0.0440 (13)	0.0555 (15)	0.115 (2)	0.0125 (11)	0.0274 (14)	0.0172 (15)
C20	0.0325 (11)	0.0645 (16)	0.096 (2)	0.0165 (11)	0.0112 (12)	-0.0084 (14)
C21	0.0452 (13)	0.0781 (18)	0.0735 (17)	0.0341 (13)	-0.0057 (11)	0.0041 (14)
C22	0.0416 (11)	0.0683 (15)	0.0540 (13)	0.0228 (11)	0.0021 (9)	0.0127 (11)
C23	0.0491 (13)	0.102 (2)	0.0669 (16)	0.0323 (14)	0.0147 (12)	0.0473 (16)
C24	0.0403 (12)	0.0674 (16)	0.0872 (19)	0.0164 (11)	0.0290 (12)	0.0344 (14)

Geometric parameters (\AA , $^{\circ}$)

S1—O1	1.4378 (19)	C10—H10B	0.9700
S1—O2	1.4264 (18)	C10—C11	1.515 (4)
S1—C7	1.813 (2)	C11—C12	1.372 (4)
S1—C17	1.765 (2)	C11—C16	1.382 (4)
F1—C12	1.353 (5)	C12—C13	1.398 (6)
O3—C9	1.221 (3)	C13—H13	0.9300
O4—C4	1.367 (2)	C13—C14	1.385 (6)
O4—C24	1.428 (3)	C14—H14	0.9300
O5—C5	1.363 (2)	C14—C15	1.332 (6)
O5—C23	1.423 (3)	C15—H15	0.9300
N1—H1	0.8600	C15—C16	1.371 (5)
N1—C8	1.443 (3)	C16—H16	0.9300
N1—C9	1.338 (3)	C17—C18	1.386 (3)
C1—C2	1.380 (3)	C17—C22	1.377 (3)
C1—C6	1.394 (3)	C18—H18	0.9300
C1—C7	1.513 (2)	C18—C19	1.387 (3)
C2—H2	0.9300	C19—H19	0.9300
C2—C3	1.394 (3)	C19—C20	1.371 (4)
C3—H3	0.9300	C20—H20	0.9300
C3—C4	1.379 (3)	C20—C21	1.358 (4)
C4—C5	1.403 (3)	C21—H21	0.9300
C5—C6	1.383 (3)	C21—C22	1.393 (3)
C6—H6	0.9300	C22—H22	0.9300
C7—H7	0.9800	C23—H23A	0.9600
C7—C8	1.530 (3)	C23—H23B	0.9600
C8—H8A	0.9700	C23—H23C	0.9600
C8—H8B	0.9700	C24—H24A	0.9600
C9—C10	1.518 (4)	C24—H24B	0.9600
C10—H10A	0.9700	C24—H24C	0.9600
O1—S1—C7	107.85 (11)	C11—C10—H10B	107.9
O1—S1—C17	107.54 (11)	C12—C11—C10	124.0 (3)
O2—S1—O1	119.44 (13)	C12—C11—C16	115.0 (3)
O2—S1—C7	107.41 (11)	C16—C11—C10	121.0 (3)
O2—S1—C17	108.68 (11)	F1—C12—C11	116.5 (4)
C17—S1—C7	105.03 (9)	F1—C12—C13	120.1 (4)
C4—O4—C24	117.09 (18)	C11—C12—C13	123.3 (4)
C5—O5—C23	117.34 (16)	C12—C13—H13	121.3

C8—N1—H1	118.2	C14—C13—C12	117.4 (3)
C9—N1—H1	118.2	C14—C13—H13	121.3
C9—N1—C8	123.6 (2)	C13—C14—H14	119.4
C2—C1—C6	118.69 (18)	C15—C14—C13	121.2 (4)
C2—C1—C7	120.93 (18)	C15—C14—H14	119.4
C6—C1—C7	120.33 (17)	C14—C15—H15	120.2
C1—C2—H2	119.7	C14—C15—C16	119.6 (4)
C1—C2—C3	120.55 (19)	C16—C15—H15	120.2
C3—C2—H2	119.7	C11—C16—H16	118.3
C2—C3—H3	119.6	C15—C16—C11	123.5 (3)
C4—C3—C2	120.83 (19)	C15—C16—H16	118.3
C4—C3—H3	119.6	C18—C17—S1	118.95 (18)
O4—C4—C3	125.98 (18)	C22—C17—S1	120.22 (17)
O4—C4—C5	115.10 (18)	C22—C17—C18	120.8 (2)
C3—C4—C5	118.92 (18)	C17—C18—H18	120.5
O5—C5—C4	115.25 (17)	C17—C18—C19	119.0 (2)
O5—C5—C6	124.92 (18)	C19—C18—H18	120.5
C6—C5—C4	119.83 (18)	C18—C19—H19	120.0
C1—C6—H6	119.4	C20—C19—C18	120.1 (3)
C5—C6—C1	121.17 (18)	C20—C19—H19	120.0
C5—C6—H6	119.4	C19—C20—H20	119.6
S1—C7—H7	107.1	C21—C20—C19	120.7 (2)
C1—C7—S1	107.66 (13)	C21—C20—H20	119.6
C1—C7—H7	107.1	C20—C21—H21	119.8
C1—C7—C8	116.05 (16)	C20—C21—C22	120.5 (2)
C8—C7—S1	111.53 (14)	C22—C21—H21	119.8
C8—C7—H7	107.1	C17—C22—C21	118.9 (2)
N1—C8—C7	111.25 (18)	C17—C22—H22	120.5
N1—C8—H8A	109.4	C21—C22—H22	120.5
N1—C8—H8B	109.4	O5—C23—H23A	109.5
C7—C8—H8A	109.4	O5—C23—H23B	109.5
C7—C8—H8B	109.4	O5—C23—H23C	109.5
H8A—C8—H8B	108.0	H23A—C23—H23B	109.5
O3—C9—N1	123.5 (2)	H23A—C23—H23C	109.5
O3—C9—C10	120.3 (2)	H23B—C23—H23C	109.5
N1—C9—C10	116.2 (2)	O4—C24—H24A	109.5
C9—C10—H10A	107.9	O4—C24—H24B	109.5
C9—C10—H10B	107.9	O4—C24—H24C	109.5
H10A—C10—H10B	107.2	H24A—C24—H24B	109.5
C11—C10—C9	117.8 (2)	H24A—C24—H24C	109.5
C11—C10—H10A	107.9	H24B—C24—H24C	109.5

Hydrogen-bond geometry (\AA , °)

$D—H\cdots A$	$D—H$	$H\cdots A$	$D\cdots A$	$D—H\cdots A$
N1—H1 \cdots O4 ⁱ	0.86	2.46	3.207 (2)	145

N1—H1···O5 ⁱ	0.86	2.50	2.945 (3)	113
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Symmetry code: (i) $x-1, y, z$.