

Supplementary Information

Asymmetric Hydrogenation of 1,1-Diarylethylenes and Benzophenones Through a Relay Strategy

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Table of Contents

Supplementary Methods.....	2
1 General Information	2
2 Investigation of Arene Exchange	2
2.1 The regioisomeric ratios (r.r.) of 1a complexed with different metal species.....	2
2.2 Selective forming of Cr(CO) ₃ -complexed benzophenones.....	3
2.3 The preparation of substrates	9
3 Asymmetric Hydrogenation of 1,1-Diarylethylenes and Benzophenones	12
3.1 Synthesis of ruthenium-catalysts	12
3.2 Asymmetric hydrogenation of 1a-Cr with different ruthenium-catalysts	13
3.3 General procedure for asymmetric hydrogenation of benzophenones and 1,1-diarylethylenes	13
4 Computational Studies.....	29
5 NMR Chromatograms of Compounds	31
Supplementary References	104

Supplementary Methods

1 General Information

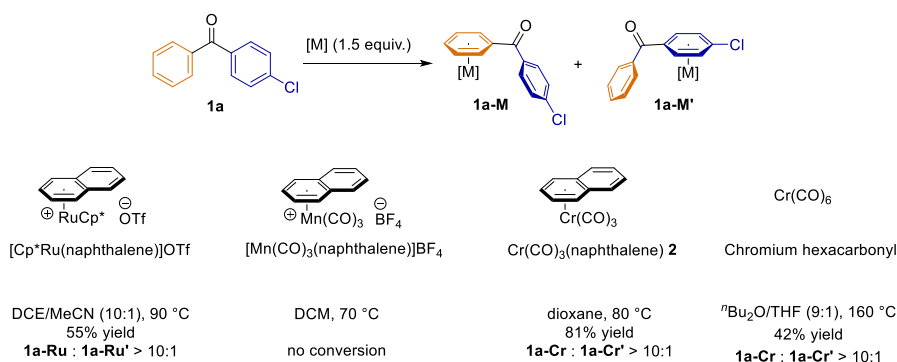
Solvents: Anhydrous toluene was freshly distilled from sodium, other anhydrous solvents were all purchased from J&K Scientific Co., Ltd, stored in glovebox, and used without further purification.

Chromatography: Analytical thin layer chromatography was performed on 0.25 mm silica gel 60-F254. Visualization was carried out with UV light. Flash column chromatography was performed on 200–300 mesh silica gel, which purchased from Yantai Jiangyou Co., China.

Spectroscopy and Instruments: ^1H NMR was recorded on Bruker AVANCE NEO instrument (500 MHz, 600 MHz). Chemical shifts were quoted in parts per million (ppm) referenced with tetramethylsilane ($\delta = 0.00$ ppm) or solvent peak ($\delta = 7.26$ ppm in chloroform- d , $\delta = 2.05$ ppm in Acetone- d_6). ^{13}C NMR spectra were recorded on Bruker AVANCE NEO instrument (126 MHz, 151 MHz), and were fully decoupled by broad band proton decoupling. Chemical shifts were reported in ppm referenced to the center peak of a triplet at 77.00 ppm of chloroform- d . ^{19}F NMR spectra were recorded on Bruker AVANCE NEO instrument (471 MHz, 565 MHz). The following abbreviations (or combinations thereof) were used to explain multiplicities: s = singlet, d = doublet, dd = doublet of doublets, t = triplet, m = multiplet, br = broad. Coupling constants, J , were reported in Hertz unit (Hz). High-resolution mass spectra (HRMS) were recorded on a Waters Mass spectrometer using ESI/APCI-TOF (electrospray ionization-time of flight). Low-resolution mass spectras were recorded on Agilent 5977B GC/MSD. The enantiomeric excess value was detected on Agilent Infinity II SFC or HPLC system with Chiralpak columns.

2 Investigation of Arene Exchange

2.1 The regioisomeric ratios (r.r.) of **1a** complexed with different metal species



Supplementary Fig. 1 The regioisomeric ratios (r.r.) of **1a** complexed with different metal species.

$\text{Cr}(\text{CO})_6$ was commercially available and used after sublimation.

$[\text{Mn}(\text{CO})_3(\text{naphthalene})]\text{BF}_4$ was prepared by a modified procedure¹. $\text{Mn}(\text{CO})_5\text{Br}$ (1.0 g) and AgBF_4 (1.1 equiv) were dissolved in 50 mL of CH_2Cl_2 and refluxed for 1 h under N_2 in the dark. The naphthalene (2 equiv) in 10 mL of CH_2Cl_2 was then added and the reaction mixture refluxed overnight. After it was cooled to room temperature, the solution was filtered through Celite and concentrated to ca. 20 mL in vacuo and the product precipitated with diethyl ether. The yellow $[\text{Mn}(\text{CO})_3(\text{naphthalene})]\text{BF}_4$ salts were washed repeatedly with ether and dried in vacuo. Yellow solid. ^1H NMR (500 MHz, Acetone- d_6) δ 7.90 (s, 4H), 7.49 (s, 4H).

$\text{Cr}(\text{CO})_3(\text{naphthalene})$ **2** was prepared by a modified procedure². Chromium hexacarbonyl (17.6 g, 0.08 mol), KOH (32.0 g), degassed ethanol (80 ml), degassed n -butanol (80 ml) and degassed water (25 ml) were placed in a 500 ml two-necked flask provided with a stirrer and reflux condenser under N_2 . The contents of the flask were warmed gradually on a metal bath to 150 °C. Chromium hexacarbonyl which sublimed initially, was washed out later by the solvent and no more sublimation occurred during the reaction. The reaction solution turned gradually red. After 4 h of refluxing the mixture was cooled, 150 ml of concentrated aqueous ammonia solution added and the mixture was slowly stirred for 2 h. A yellow precipitate was filtered through a glass filter under the nitrogen atmosphere, washed with aqueous ammonia and small quantities of methanol and ester.

To a cold (0 °C) ether solution (250 mL) of $[(\text{NH}_3)_3\text{Cr}(\text{CO})_3]$ (60 mmol) and naphthalene (72 mmol) was added $\text{BF}_3 \cdot \text{OEt}_2$ (240 mmol). The mixture was then allowed to warm to ambient temperature and stir for 8 days. Excess $\text{BF}_3 \cdot \text{OEt}_2$ was quenched with 1 M HCl solution. The solvent was removed from the dried organic layer and the residual naphthalene sublimed from the crude mixture at 50 °C (15 mbar) to give the crude product. Pure product can be obtained by recrystallization from DCM/hexane. Red solid. ^1H NMR (500 MHz, Chloroform- d) δ 7.56 (dd, $J = 6.5, 3.2$ Hz, 2H), 7.41 (dd, $J = 6.5, 3.2$ Hz, 2H), 6.13 (dd, $J = 4.8, 2.8$ Hz, 2H), 5.52 (dd, $J = 5.0, 2.8$ Hz, 2H).

$[\text{Cp}^*\text{Ru}(\text{naphthalene})]\text{OTf}$ was prepared by a modified procedure for $[\text{Cp}^*\text{Ru}(\text{naphthalene})]\text{BF}_4$ ³. A solution of $\text{RuCl}_3 \cdot 3\text{H}_2\text{O}$ (0.970 g, 4.0 mmol) in degassed ethanol (25 mL) was kept under reflux under N_2 in a two-neck round bottom flask until the color of the solution turned green (about 2 h). Pentamethylcyclopentadiene (3.0 mL, 20 mmol) and naphthalene (2.560 g, 20 mmol) were added to the reaction mixture, and the resulting solution was heated at reflux for 6 h. The reaction mixture was worked up in air. After the solvent was evaporated, water (15 mL) was added to the residue, and the solids were filtered off. The filtrate was treated with HOTf (5 mL) and extracted with CH_2Cl_2 (3 x 40 mL). The combined organic extracts

were dried over anhydrous Na₂SO₄ and evaporated to give crude product. Pure product can be obtained by recrystallization from DCM/hexane. Yellow solid. 1.556 g, 76% yield. ¹H NMR (600 MHz, Chloroform-*d*) δ 7.64 (dd, *J* = 6.7, 3.1 Hz, 2H), 7.55 (dd, *J* = 6.7, 3.1 Hz, 2H), 6.59 (dd, *J* = 3.8, 2.1 Hz, 2H), 6.08 (dd, *J* = 4.1, 2.3 Hz, 2H), 1.67 (s, 15H). ¹³C NMR (151 MHz, Chloroform-*d*) δ 130.81, 127.72, 96.99, 93.87, 88.50, 85.47, 9.62. ¹⁹F NMR (565 MHz, Chloroform-*d*) δ -77.98. HRMS-ESI (*m/z*): [M-OTf]⁺ calcd. for [C₂₀H₂₃Ru]⁺, 365.0838, found, 365.0851.

General procedure:

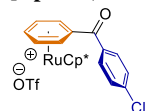
With [Cp*Ru(naphthalene)]OTf: In a glovebox, a solution of [Cp*Ru(naphthalene)]OTf (0.15 mmol, 1.5 equiv.), **1a** (0.1 mmol, 1.0 equiv.) in DCE/MeCN (1 mL/0.1 mL) was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at 90 °C in the dark for 24 h. Yield and regioisomeric ratio were determined by ¹H NMR using 1,1,2,2-tetrachloroethane as the internal standard.

With [Mn(CO)₃(naphthalene)]BF₄: In a glovebox, a solution of [Mn(CO)₃(naphthalene)]BF₄ (0.15 mmol, 1.5 equiv.), **1a** (0.1 mmol, 1.0 equiv.) in DCM (1 mL) was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at 70 °C in the dark for 24 h. Yield and regioisomeric ratio were determined by ¹H NMR using CH₂Br₂ as the internal standard.

With Cr(CO)₃(naphthalene) 2: In a glovebox, a solution of Cr(CO)₃(naphthalene) **2** (0.15 mmol, 1.5 equiv.), **1a** (0.1 mmol, 1.0 equiv.) in 1,4-dioxane (1 mL) was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at 80 °C in the dark for 24 h. Yield and regioisomeric ratio were determined by ¹H NMR using CH₂Br₂ as the internal standard.

With Cr(CO)₆: A flame-dried round bottom flask equipped with a magnetic stirrer and a reflux condenser was charged with **1a** (2.0 mmol, 1.0 equiv), Cr(CO)₆ (3.0 mmol, 1.5 equiv) and anh. ⁿBu₂O and THF (9:1 v/v, 0.2 M) evacuated and backfilled with Ar. The resulting suspension was subjected freeze-pump-thaw cycles (3 × 30 min) and then refluxed (external temperature 160 °C) for 48 h. The solution was then cooled down to room temperature and filtered through a short pad of silica. The silica pad was washed with DCM (3 × 20 mL) and the organic layer was then concentrated in vacuo. Yield and regioisomeric ratio were determined by ¹H NMR using CH₂Br₂ as the internal standard.

[Cp*Ru(4-chlorobenzoylbenzene)]OTf (**1a**-Ru)



White solid. ¹H NMR (600 MHz, Chloroform-*d*) δ 7.81 (d, *J* = 8.2 Hz, 2H), 7.55 (d, *J* = 8.3 Hz, 2H), 6.31 – 6.24 (m, 3H), 6.22 (d, *J* = 5.5 Hz, 2H), 1.94 (s, 15H). ¹³C NMR (151 MHz, Chloroform-*d*) δ 191.45, 141.02, 133.63, 131.00, 129.52, 98.13, 94.96, 89.28, 87.71, 87.43, 10.23. ¹⁹F NMR (565 MHz, Chloroform-*d*) δ -78.03. HRMS-ESI (*m/z*): [M-OTf]⁺ calcd. for [C₂₃H₂₄ClORu]⁺, 453.0554, found, 453.0569.

2.2 Selective forming of Cr(CO)₃-complexed benzophenones

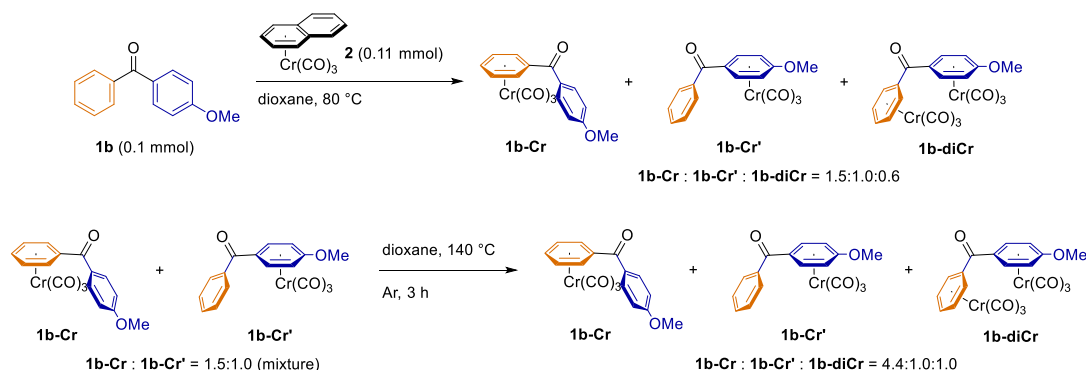
1-Cr (or s-Cr) yield		Ar ¹	Ar ²
78%	1a	Ph	(4-Cl)C ₆ H ₄
75%	s1	Ph	(2-F)C ₆ H ₄
80%	1c	Ph	(3-F)C ₆ H ₄
80%	1d	Ph	(4-F)C ₆ H ₄
73%	1e	Ph	(4-Br)C ₆ H ₄
72%	1y	Ph	(3-CF ₃)C ₆ H ₄
78%	1f	Ph	(4-CF ₃)C ₆ H ₄
82%	1g	Ph	(4-OCF ₃)C ₆ H ₄
60%	1h	Ph	(4-CO ₂ Me)C ₆ H ₄
38%	1i	Ph	[4-C(O)NMe ₂]C ₆ H ₄
58%	s2	Ph	(4-OAc)C ₆ H ₄
35% (1b -Cr' 23%)	1b	Ph	(4-OMe)C ₆ H ₄
80%	1z	Ph	(4-OTf)C ₆ H ₄
82%	1aa	Ph	(3,4-bisCl)C ₆ H ₃
84%	1k	Ph	(3,4-OCH ₂ O)C ₆ H ₃
80%	1ab	Ph	2-naphthyl
ratio > 10 : 1			
1-Cr (or s-Cr) yield		Ar ¹	Ar ²
77%	1l	(3-OMe)C ₆ H ₄	(4-F)C ₆ H ₄
68%	1m	(4-OMe)C ₆ H ₄	(4-Cl)C ₆ H ₄
61%	1n	(4-OMe)C ₆ H ₄	(4-Br)C ₆ H ₄
76%	1o	(4-OMe)C ₆ H ₄	(4-CF ₃)C ₆ H ₄
77%	1p	(4-OMe)C ₆ H ₄	(4-OCF ₃)C ₆ H ₄
70%	s3	(4-OMe)C ₆ H ₄	(3,5-bisCF ₃)C ₆ H ₃
74%	1v	(4-OMe)C ₆ H ₄	(3-F, 5-OMe)C ₆ H ₃
47%	1q	(4-OTBS)C ₆ H ₄	(4-Cl)C ₆ H ₄
80%	1r	(4-NMe ₂)C ₆ H ₄	(4-OMe)C ₆ H ₄
84%	1s	(4-NMe ₂)C ₆ H ₄	(4-CO ₂ Me)C ₆ H ₄
72%	1j	(4-NMe ₂)C ₆ H ₄	phenyl
85%	1u	(2-Me)C ₆ H ₄	(2-CF ₃)C ₆ H ₄
65%	1x	(2-Me)C ₆ H ₄	(4-Cl)C ₆ H ₄
76%	1t	(4-Me)C ₆ H ₄	(4-F)C ₆ H ₄
66%	1ac	(4-vinyl)C ₆ H ₄	(4-F)C ₆ H ₄
85%	1w	[3,4-(CH ₂) ₄]C ₆ H ₃	2-naphthyl
5 : 1 ≥ ratio			

Supplementary Fig. 2 Selective formation of Cr(CO)₃-benzophenone complex.

General procedure:

For ¹H NMR analysis: In a glovebox, a solution of Cr(CO)₃(naphthalene) **2** (0.15 mmol, 1.5 equiv.), **1** (0.1 mmol, 1.0 equiv.) in 1,4-dioxane (1 mL) in a 4 mL glass vial was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at 80 °C in the dark for 24 h. After cooling to room temperature, the reaction solution was analyzed by ¹H NMR directly and the regioisomeric ratios were determined by ¹H NMR.

For product purification: In a glovebox, a solution of $\text{Cr}(\text{CO})_3(\text{naphthalene})$ **2** (0.3 mmol, 1.5 equiv.), **1** (0.2 mmol, 1.0 equiv.) in 1,4-dioxane (2 mL) in a 4 mL glass vial was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at 80 °C in the dark for 24 h. After cooling to room temperature, the solution was evaporated under reduced pressure. The red residue was purified by column chromatography eluting with EtOAc/Petroleum ether (impurities that cannot be separated by column chromatography can be removed by recrystallization from diethyl ether and pentane).



Supplementary Fig. 3 The regioisomeric ratios (r.r.) of **1b** complexed with $\text{Cr}(\text{CO})_3$.

In a glovebox, a solution of $\text{Cr}(\text{CO})_3(\text{naphthalene})$ **2** (0.11 mmol, 1.1 equiv.), **1b** (0.1 mmol, 1.0 equiv.) in 1,4-dioxane (1 mL) was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at 80 °C in the dark for 24 h. The regioisomeric ratio was determined by ^1H NMR.

Then, the solution was evaporated under reduced pressure. The red residue was purified by column chromatography to get the mixture (**1b-Cr** and **1b-Cr'**, 1.5:1.0). In a glovebox, a solution of the mixture, in 1,4-dioxane (0.5 mL) was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at 140 °C in the dark for 3 h. The regioisomeric ratio was determined by ^1H NMR.

4-chlorobenzoylbenzene chromium tricarbonyl (**1a-Cr**)



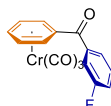
Known compound.⁴ Red solid. 55.2 mg, 78% yield. The regioisomeric ratio was > 10:1. R_f = 0.33 (petroleum ether/ethyl acetate, PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform-*d*) δ 7.74 (d, J = 8.6 Hz, 2H), 7.49 (d, J = 8.6 Hz, 2H), 5.99 (d, J = 6.8 Hz, 2H), 5.64 (t, J = 6.3 Hz, 1H), 5.33 (t, J = 6.5 Hz, 2H). ^{13}C NMR (126 MHz, Chloroform-*d*) δ 230.33, 192.47, 138.91, 134.75, 130.09, 128.96, 95.60, 95.49, 94.57, 89.44. HRMS-APCI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_{10}\text{ClCrO}_4]^+$, 352.9667, found, 352.9663.

2-fluorobenzoylbenzene chromium tricarbonyl (**1s1-Cr**)



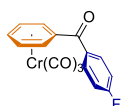
Red solid. 50.3 mg, 75% yield. The regioisomeric ratio was > 10:1. R_f = 0.45 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform-*d*) δ 7.58–7.49 (m, 2H), 7.30 (td, J = 7.6, 0.9 Hz, 1H), 7.22–7.14 (m, 1H), 6.01 (d, J = 6.8 Hz, 2H), 5.67 (t, J = 6.3 Hz, 1H), 5.28 (t, J = 6.6 Hz, 2H). ^{13}C NMR (126 MHz, Chloroform-*d*) δ 230.17, 190.55, 159.01 (d, J = 250.1 Hz), 133.11 (d, J = 8.1 Hz), 129.96 (d, J = 2.9 Hz), 125.35 (d, J = 16.3 Hz), 124.84 (d, J = 3.5 Hz), 116.12 (d, J = 21.4 Hz), 95.59 (d, J = 1.9 Hz), 95.19, 94.65, 88.97. ^{19}F NMR (471 MHz, Chloroform-*d*) δ -111.81. HRMS-APCI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_{10}\text{CrFO}_4]^+$, 336.9963, found, 336.9958.

3-fluorobenzoylbenzene chromium tricarbonyl (**1c-Cr**)



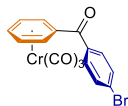
Red solid. 53.7 mg, 80% yield. The regioisomeric ratio was > 10:1. R_f = 0.33 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform-*d*) δ 7.56 (d, J = 7.7 Hz, 1H), 7.53–7.43 (m, 2H), 7.30 (tdd, J = 8.3, 2.5, 0.8 Hz, 1H), 6.02 (d, J = 6.5 Hz, 2H), 5.65 (t, J = 6.2 Hz, 1H), 5.33 (t, J = 6.5 Hz, 2H). ^{13}C NMR (126 MHz, Chloroform-*d*) δ 230.28, 192.29 (d, J = 2.2 Hz), 162.42 (d, J = 249.3 Hz), 138.40 (d, J = 6.5 Hz), 130.38 (d, J = 7.9 Hz), 124.32 (d, J = 3.2 Hz), 119.39 (d, J = 21.3 Hz), 115.63 (d, J = 22.9 Hz), 95.54, 95.15, 94.68, 89.39. ^{19}F NMR (471 MHz, Chloroform-*d*) δ -110.98. HRMS-APCI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_{10}\text{CrFO}_4]^+$, 336.9963, found, 336.9958.

4-fluorobenzoylbenzene chromium tricarbonyl (**1d-Cr**)



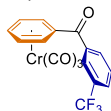
Red solid. 54.1 mg, 80% yield. The regioisomeric ratio was > 10:1. $R_f = 0.33$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform- d) δ 7.84 (dd, $J = 8.5, 5.4$ Hz, 2H), 7.19 (t, $J = 8.5$ Hz, 2H), 6.00 (d, $J = 6.5$ Hz, 2H), 5.63 (t, $J = 6.2$ Hz, 1H), 5.34 (t, $J = 6.4$ Hz, 2H). **^{13}C NMR** (126 MHz, Chloroform- d) δ 230.42, 192.16, 165.30 (d, $J = 254.5$ Hz), 132.65 (d, $J = 3.3$ Hz), 131.33 (d, $J = 9.2$ Hz), 115.85 (d, $J = 21.8$ Hz), 96.11, 95.53, 94.51, 89.48. **^{19}F NMR** (471 MHz, CDCl_3) δ -105.48. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_{10}\text{CrFO}_4]^+$, 336.9963, found, 336.9964.

4-bromobenzoylbenzene chromium tricarbonyl (1e-Cr)



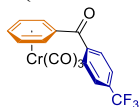
Known compound⁴. Red solid. 57.7 mg, 73% yield. The regioisomeric ratio was > 10:1. $R_f = 0.33$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform- d) δ 7.65 (s, 4H), 5.99 (d, $J = 6.6$ Hz, 2H), 5.64 (t, $J = 6.4$ Hz, 1H), 5.33 (t, $J = 6.5$ Hz, 2H). **^{13}C NMR** (126 MHz, Chloroform- d) δ 230.31, 192.64, 135.20, 131.93, 130.17, 127.41, 95.48, 94.58, 89.43. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_{10}\text{BrCrO}_4]^+$, 396.9162, found, 396.9165.

3-(trifluoromethyl)benzoylbenzene chromium tricarbonyl (1y-Cr)



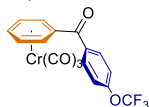
Orange solid. 55.8 mg, 72% yield. The regioisomeric ratio was > 10:1. $R_f = 0.33$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform- d) δ 8.03 (s, 1H), 7.97 (d, $J = 7.6$ Hz, 1H), 7.86 (d, $J = 7.7$ Hz, 1H), 7.66 (t, $J = 7.7$ Hz, 1H), 5.99 (d, $J = 6.4$ Hz, 2H), 5.67 (t, $J = 6.1$ Hz, 1H), 5.34 (t, $J = 6.4$ Hz, 2H). **^{13}C NMR** (126 MHz, Chloroform- d) δ 230.12, 192.43, 137.18, 131.76, 131.23 (q, $J = 33.5$ Hz), 129.36, 128.84 (q, $J = 3.8$ Hz), 125.34 (q, $J = 4.5$ Hz), 123.49 (q, $J = 272.5$ Hz), 95.39, 94.68, 89.42. **^{19}F NMR** (471 MHz, Chloroform- d) δ -62.76. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{10}\text{CrF}_3\text{O}_4]^+$, 386.9931, found, 386.9927.

4-(trifluoromethyl)benzoylbenzene chromium tricarbonyl (1f-Cr)



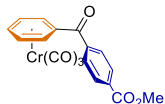
Red solid. 60.0 mg, 78% yield. The regioisomeric ratio was > 10:1. $R_f = 0.33$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform- d) δ 7.86 (d, $J = 8.1$ Hz, 2H), 7.78 (d, $J = 8.2$ Hz, 2H), 5.99 (d, $J = 6.4$ Hz, 2H), 5.67 (t, $J = 6.2$ Hz, 1H), 5.33 (t, $J = 6.5$ Hz, 2H). **^{13}C NMR** (126 MHz, Chloroform- d) δ 230.13, 192.82, 139.74, 133.70 (q, $J = 32.9$ Hz), 128.73, 125.68 (q, $J = 3.7$ Hz), 123.51 (q, $J = 272.7$ Hz), 95.44, 94.78, 94.47, 89.41. **^{19}F NMR** (471 MHz, Chloroform- d) δ -63.06. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{10}\text{CrF}_3\text{O}_4]^+$, 386.9931, found, 386.9931.

4-(trifluoromethoxy)benzoylbenzene chromium tricarbonyl (1g-Cr)



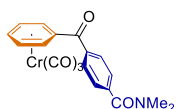
Orange solid. 66.1 mg, 82% yield. The regioisomeric ratio was > 10:1. $R_f = 0.33$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform- d) δ 7.85 (d, $J = 8.7$ Hz, 2H), 7.34 (d, $J = 8.1$ Hz, 2H), 6.00 (d, $J = 6.1$ Hz, 2H), 5.65 (t, $J = 6.2$ Hz, 1H), 5.34 (t, $J = 6.5$ Hz, 2H). **^{13}C NMR** (126 MHz, Chloroform- d) δ 230.31, 192.25, 152.09 (q, $J = 1.5$ Hz), 134.75, 130.64, 120.59, 120.28 (q, $J = 258.9$ Hz), 95.46, 94.59, 89.49. **^{19}F NMR** (471 MHz, Chloroform- d) δ -57.60. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{10}\text{CrF}_3\text{O}_5]^+$, 402.9880, found, 402.9885.

4-(methoxycarbonyl)benzoylbenzene chromium tricarbonyl (1h-Cr)



Red solid. 45.0 mg, 60% yield. The regioisomeric ratio was 6.5:1. $R_f = 0.50$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 3:1 v/v. **^1H NMR** (500 MHz, Chloroform- d) δ 8.17 (d, $J = 8.3$ Hz, 2H), 7.80 (d, $J = 8.3$ Hz, 2H), 6.01 (s, 2H), 5.66 (t, $J = 6.2$ Hz, 1H), 5.32 (t, $J = 6.5$ Hz, 2H), 3.97 (s, 3H). **^{13}C NMR** (126 MHz, Chloroform- d) δ 230.19, 193.21, 166.07, 140.31, 133.21, 129.78, 128.34, 95.53, 94.77, 94.75, 89.34, 52.50. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{13}\text{CrO}_6]^+$, 377.0112, found, 377.0107.

4-(dimethylcarbamoyl)benzoylbenzene chromium tricarbonyl (1i-Cr)



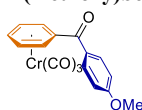
Orange solid. 29.7 mg, 38% yield. The regioisomeric ratio was > 10:1. $R_f = 0.20$ (PE/EA = 1:1 v/v). Elution with PE/EA/DCM = 2:1:1 to 1:1:1 v/v/v. $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.80 (d, $J = 8.0$ Hz, 2H), 7.55 (d, $J = 8.0$ Hz, 2H), 6.02 (d, $J = 6.4$ Hz, 2H), 5.65 (t, $J = 6.2$ Hz, 1H), 5.33 (t, $J = 6.4$ Hz, 2H), 3.15 (s, 3H), 3.00 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 230.31, 193.20, 170.30, 140.10, 137.36, 128.67, 127.22, 95.58, 95.23, 94.67, 89.46, 39.42, 35.30. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{19}\text{H}_{16}\text{CrNO}_5]^+$, 390.0428, found, 390.0418.

4-(acetoxymethyl)benzoylbenzene chromium tricarbonyl (s2-Cr)



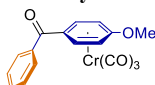
Red solid. 43.6 mg, 58% yield. The regioisomeric ratio was 6.4:1. $R_f = 0.35$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 3:1 v/v. (1.1 equiv **2**) $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.84 (d, $J = 8.6$ Hz, 2H), 7.24 (d, $J = 8.7$ Hz, 2H), 6.02 (d, $J = 6.0$ Hz, 2H), 5.62 (t, $J = 6.2$ Hz, 1H), 5.33 (t, $J = 6.5$ Hz, 2H), 2.35 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 230.45, 192.48, 168.85, 153.87, 133.90, 130.37, 121.88, 96.11, 95.58, 94.48, 89.50, 21.14. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{13}\text{CrO}_6]^+$, 377.0112, found, 377.0105.

4-(methoxymethyl)benzoylbenzene chromium tricarbonyl (1b-Cr)



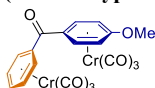
Known compound.⁴ Yellow solid. 24.3 mg, 35% yield. The regioisomeric ratio was 1.5:1.0. $R_f = 0.20$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. (1.1 equiv **2**) $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.85 (d, $J = 8.4$ Hz, 2H), 6.98 (d, $J = 8.4$ Hz, 2H), 6.01 (d, $J = 6.5$ Hz, 2H), 5.59 (t, $J = 6.2$ Hz, 1H), 5.34 (t, $J = 6.4$ Hz, 2H), 3.90 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 230.80, 191.91, 163.38, 131.43, 128.85, 113.89, 97.95, 95.65, 94.28, 89.60, 55.54. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{13}\text{CrO}_5]^+$, 349.0163, found, 349.0156.

4-benzoylanisole chromium tricarbonyl (1b-Cr')



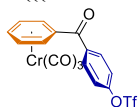
Orange solid. 16.2 mg, 23% yield. $R_f = 0.23$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.72 (d, $J = 7.5$ Hz, 2H), 7.58 (t, $J = 5.1$ Hz, 1H), 7.50 (d, $J = 6.4$ Hz, 2H), 6.19 (d, $J = 6.9$ Hz, 2H), 5.18 (d, $J = 6.9$ Hz, 2H), 3.78 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 230.48, 192.85, 144.16, 136.76, 132.16, 128.55, 128.38, 96.20, 91.42, 76.60, 55.88. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{13}\text{CrO}_5]^+$, 349.0163, found, 349.0158.

(4-methoxyphenyl chromium tricarbonyl)(phenyl chromium tricarbonyl)methanone (1b-diCr)



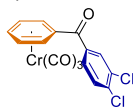
Red solid. 10.6 mg, 11% yield. $R_f = 0.35$ (PE/EA = 4:1 v/v). Elution with PE/EA = 4:1 v/v. $^1\text{H NMR}$ (600 MHz, Chloroform-*d*) δ 6.27 (d, $J = 6.2$ Hz, 2H), 6.06 (d, $J = 5.2$ Hz, 2H), 5.61 (d, $J = 6.6$ Hz, 1H), 5.33 (d, $J = 7.1$ Hz, 2H), 5.20 (d, $J = 5.3$ Hz, 2H), 3.80 (s, 3H). $^{13}\text{C NMR}$ (151 MHz, Chloroform-*d*) δ 230.27, 230.19, 188.53, 143.92, 96.58, 95.43, 94.81, 94.32, 91.02, 89.10, 76.38, 55.97. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{20}\text{H}_{13}\text{Cr}_2\text{O}_8]^+$, 484.9415, found, 484.9421.

4-(((Trifluoromethyl)sulfonyl)oxy)benzoylbenzene chromium tricarbonyl (1z-Cr)



Orange solid. 74.4 mg, 80% yield. The regioisomeric ratio was > 10:1. $R_f = 0.33$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.89 (d, $J = 8.8$ Hz, 2H), 7.43 (d, $J = 8.7$ Hz, 2H), 5.99 (d, $J = 6.1$ Hz, 2H), 5.67 (t, $J = 6.3$ Hz, 1H), 5.35 (t, $J = 6.5$ Hz, 2H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 230.13, 192.01, 151.83, 136.56, 130.80, 121.78, 118.69 (d, $J = 320.8$ Hz), 95.37, 94.70, 94.67, 89.46. $^{19}\text{F NMR}$ (471 MHz, Chloroform-*d*) δ -72.66. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{10}\text{CrF}_3\text{O}_7\text{S}]^+$, 466.9499, found, 466.9501.

3,4-bis(chloromethyl)benzoylbenzene chromium tricarbonyl (1aa-Cr)



Red solid. 63.4 mg, 82% yield. The regioisomeric ratio was > 10:1. $R_f = 0.33$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.88 (d, $J = 1.8$ Hz, 1H), 7.63 (dd, $J = 8.3, 1.9$ Hz, 1H), 7.59 (d, $J = 8.2$ Hz, 1H), 5.98 (d, $J = 6.3$ Hz, 2H), 5.66 (t, $J =$

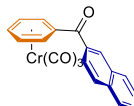
6.2 Hz, 1H), 5.34 (t, J = 6.5 Hz, 2H). ^{13}C NMR (126 MHz, Chloroform- d) δ 230.10, 191.28, 137.07, 136.00, 133.30, 130.78, 130.53, 127.73, 95.36, 94.73, 94.66, 89.36. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_9\text{Cl}_2\text{CrO}_4]^+$, 386.9278, found, 386.9271.

(benzo[d][1,3]dioxole-5-carbonyl)benzene chromium tricarbonyl (1k-Cr)



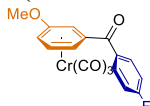
Yellow solid. 61.1mg, 84% yield. The regioisomeric ratio was > 10:1. R_f = 0.55 (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 3:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.44 (dd, J = 8.1, 1.4 Hz, 1H), 7.32 (d, J = 1.3 Hz, 1H), 6.89 (d, J = 8.1 Hz, 1H), 6.08 (s, 2H), 6.01 (d, J = 6.4 Hz, 2H), 5.59 (t, J = 6.2 Hz, 1H), 5.33 (t, J = 6.4 Hz, 2H). ^{13}C NMR (126 MHz, Chloroform- d) δ 230.68, 191.50, 151.67, 148.08, 130.48, 125.18, 109.20, 107.98, 101.97, 97.58, 95.65, 94.37, 89.44. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{11}\text{CrO}_6]^+$, 362.9955, found, 362.9949.

2-naphthoylbenzene chromium tricarbonyl (1ab-Cr)



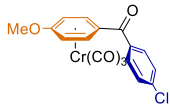
Known compound.⁴ Orange solid. 58.7 mg, 80% yield. The regioisomeric ratio was > 10:1. R_f = 0.35 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 8.39 (s, 1H), 7.97 (t, J = 9.1 Hz, 2H), 7.92 (d, J = 8.0 Hz, 1H), 7.84 (d, J = 8.3 Hz, 1H), 7.61 (dt, J = 20.6, 7.0 Hz, 2H), 6.10 (d, J = 6.5 Hz, 2H), 5.65 (t, J = 6.2 Hz, 1H), 5.35 (t, J = 6.4 Hz, 2H). ^{13}C NMR (126 MHz, Chloroform- d) δ 230.62, 193.56, 135.14, 133.55, 132.16, 129.97, 129.27, 128.70, 128.44, 127.84, 127.12, 124.91, 96.50, 95.94, 94.62, 89.51. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{20}\text{H}_{13}\text{CrO}_4]^+$, 369.0213, found, 369.0208.

3-(4-fluorobenzoyl)anisole chromium tricarbonyl (1l-Cr)



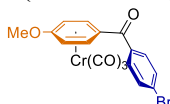
Red solid. 56.5 mg, 77% yield. The regioisomeric ratio was > 10:1. R_f = 0.40 (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 3:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.86 (dd, J = 8.7, 5.3 Hz, 2H), 7.19 (t, J = 8.6 Hz, 2H), 5.69 (dd, J = 2.2, 1.2 Hz, 1H), 5.57 (t, J = 6.6 Hz, 1H), 5.44 (d, J = 6.4 Hz, 1H), 5.41 (dd, J = 6.8, 2.0 Hz, 1H), 3.76 (s, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 231.60, 193.51, 165.40 (d, J = 254.8 Hz), 141.25, 132.69 (d, J = 3.2 Hz), 131.50 (d, J = 9.0 Hz), 115.84 (d, J = 21.9 Hz), 97.86, 92.30, 89.57, 81.19, 78.01, 56.10. ^{19}F NMR (471 MHz, Chloroform- d) δ -105.20. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{12}\text{CrFO}_5]^+$, 367.0068, found, 367.0070.

4-(4-chlorobenzoyl)anisole chromium tricarbonyl (1m-Cr)



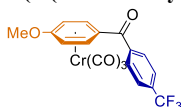
Yellow solid. 52.0 mg, 68% yield. The regioisomeric ratio was > 10:1. R_f = 0.33 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.69 (d, J = 8.5 Hz, 2H), 7.48 (d, J = 8.5 Hz, 2H), 6.15 (d, J = 7.1 Hz, 2H), 5.18 (d, J = 7.2 Hz, 2H), 3.79 (s, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 230.29, 191.66, 144.16, 138.61, 135.01, 129.82, 128.91, 95.97, 90.94, 76.58, 55.92. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{12}\text{ClCrO}_5]^+$, 382.9773, found, 382.9772.

4-(4-bromobenzoyl)anisole chromium tricarbonyl (1n-Cr)



Yellow solid. 52.4 mg, 61% yield. The regioisomeric ratio was > 10:1. R_f = 0.33 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.64 (d, J = 8.2 Hz, 2H), 7.61 (d, J = 8.2 Hz, 2H), 6.15 (d, J = 6.8 Hz, 2H), 5.18 (d, J = 6.8 Hz, 2H), 3.79 (s, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 230.27, 191.81, 144.17, 135.46, 131.88, 129.90, 127.09, 95.94, 90.83, 76.58, 55.93. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{12}\text{BrCrO}_5]^+$, 426.9268, found, 426.9271.

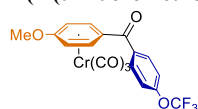
4-(4-(trifluoromethyl)benzoyl)anisole chromium tricarbonyl (1o-Cr)



Orange solid. 63.4 mg, 76% yield. The regioisomeric ratio was > 10:1. R_f = 0.33 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.81 (d, J = 7.9 Hz, 2H), 7.76 (d, J = 8.1 Hz, 2H), 6.14 (d, J = 6.9 Hz, 2H), 5.19 (d, J = 7.0 Hz, 2H), 3.79 (s, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 230.09, 192.01, 144.27, 140.03, 133.51 (q, J = 33.5, 32.3 Hz), 128.45, 125.65, 123.54 (q, J = 273.2, 271.8 Hz), 95.77, 90.02, 76.66, 55.95. ^{19}F NMR (471 MHz, Chloroform- d) δ -63.03. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{12}\text{CrF}_3\text{O}_5]^+$, 417.0036, found,

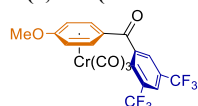
417.0039.

4-(4-(trifluoromethoxy)benzoyl)anisole chromium tricarbonyl (1p-Cr)



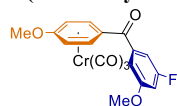
Orange solid. 66.2 mg, 77% yield. The regioisomeric ratio was > 10:1. $R_f = 0.35$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.80 (d, $J = 8.3$ Hz, 2H), 7.33 (d, $J = 8.2$ Hz, 2H), 6.16 (d, $J = 6.9$ Hz, 2H), 5.19 (d, $J = 6.9$ Hz, 2H), 3.79 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 230.26, 191.45, 151.91, 144.18, 135.03, 130.35, 120.59, 120.29 (q, $J = 258.6$ Hz), 95.94, 90.82, 76.61, 55.93. $^{19}\text{F NMR}$ (471 MHz, Chloroform-*d*) δ -57.61. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{12}\text{CrF}_3\text{O}_6]^+$, 432.9986, found, 432.9992.

4-(3,5-bis(trifluoromethyl)benzoyl)anisole chromium tricarbonyl (s3-Cr)



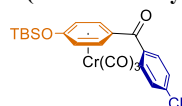
Orange solid. 67.9 mg, 70% yield. The regioisomeric ratio was > 10:1. $R_f = 0.40$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 8.16 (s, 2H), 8.09 (s, 1H), 6.09 (d, $J = 5.5$ Hz, 2H), 5.22 (d, $J = 5.4$ Hz, 2H), 3.81 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 229.69, 190.21, 144.23, 138.61, 132.34 (q, $J = 34.1$ Hz), 128.26, 125.42, 122.73 (q, $J = 273.4$ Hz), 95.35, 89.00, 76.70, 56.02. $^{19}\text{F NMR}$ (471 MHz, Chloroform-*d*) δ -62.85. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{19}\text{H}_{11}\text{CrF}_6\text{O}_5]^+$, 484.9910, found, 484.9917.

4-(3-methoxy-5-fluorobenzoyl)anisole chromium tricarbonyl (1v-Cr)



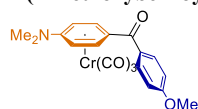
Orange solid. 58.3 mg, 74% yield. The regioisomeric ratio was > 10:1. $R_f = 0.35$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 3:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.02 (s, 1H), 6.98 (d, $J = 8.0$ Hz, 1H), 6.82 (d, $J = 10.2$ Hz, 1H), 6.19 (d, $J = 6.9$ Hz, 2H), 5.19 (d, $J = 6.9$ Hz, 2H), 3.86 (s, 3H), 3.79 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 230.26, 191.54 (d, $J = 2.7$ Hz), 164.06, 162.09, 161.08 (d, $J = 10.9$ Hz), 144.23, 139.00 (d, $J = 8.4$ Hz), 109.47 (d, $J = 2.8$ Hz), 107.67 (d, $J = 23.5$ Hz), 105.54 (d, $J = 24.8$ Hz), 95.95, 90.41, 76.61, 55.93. $^{19}\text{F NMR}$ (471 MHz, Chloroform-*d*) δ -109.80. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{14}\text{CrFO}_6]^+$, 397.0174, found, 397.0173.

4-(4-chlorobenzoyl)(((tert-butyldimethylsilyl)oxy)benzene) chromium tricarbonyl (1q-Cr)



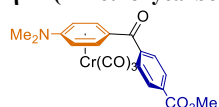
Orange solid. 45.0 mg, 47% yield. The regioisomeric ratio was > 10:1. $R_f = 0.70$ (PE/EA = 9:1 v/v). Elution with PE/EA = 25:1 to 10:1 v/v. $^1\text{H NMR}$ (600 MHz, Chloroform-*d*) δ 7.68 (d, $J = 8.5$ Hz, 2H), 7.47 (d, $J = 8.5$ Hz, 2H), 6.11 (d, $J = 7.1$ Hz, 2H), 5.10 (d, $J = 7.1$ Hz, 2H), 0.98 (s, 9H), 0.31 (s, 6H). $^{13}\text{C NMR}$ (151 MHz, Chloroform-*d*) δ 230.58, 191.81, 140.88, 138.51, 135.09, 129.80, 128.87, 96.23, 90.59, 81.68, 25.28, 18.04, -4.43. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{22}\text{H}_{24}\text{ClCrO}_5\text{Si}]^+$, 483.0481, found, 483.0483.

4-(4-methoxybenzoyl)nn-dimethylaniline chromium tricarbonyl (1r-Cr)



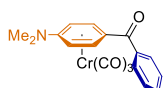
Yellow solid. 62.6 mg, 80% yield. The regioisomeric ratio was > 10:1. (1.1 equiv **2**) $R_f = 0.45$ (PE/EA = 2:1 v/v). Elution with PE/EA = 4:1 to 2:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 7.82 (d, $J = 8.6$ Hz, 2H), 6.97 (d, $J = 8.6$ Hz, 2H), 6.24 (d, $J = 7.2$ Hz, 2H), 4.87 (d, $J = 7.2$ Hz, 2H), 3.88 (s, 3H), 2.98 (s, 6H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 232.27, 191.89, 162.88, 136.12, 130.99, 129.53, 113.71, 98.25, 90.75, 73.03, 55.47, 39.80. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{19}\text{H}_{18}\text{CrNO}_5]^+$, 392.0585, found, 392.0585.

η^6 -4-(4-methoxycarbonyl)nn-dimethylaniline chromium tricarbonyl (1s-Cr)



Red solid. 70.2 mg, 84% yield. The regioisomeric ratio was > 10:1. $R_f = 0.40$ (PE/EA = 2:1 v/v). Elution with PE/EA = 4:1 to 2:1 v/v. (1.1 equiv **2**) $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 8.14 (d, $J = 8.4$ Hz, 2H), 7.76 (d, $J = 8.4$ Hz, 2H), 6.18 (d, $J = 7.5$ Hz, 2H), 4.86 (d, $J = 7.6$ Hz, 2H), 3.96 (s, 3H), 2.99 (s, 6H). $^{13}\text{C NMR}$ (126 MHz, Chloroform-*d*) δ 231.69, 193.09, 166.23, 141.15, 136.37, 132.66, 129.67, 127.99, 97.68, 88.41, 73.08, 52.42, 39.82. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{20}\text{H}_{18}\text{CrNO}_6]^+$, 420.0534, found, 420.0538.

4-(benzoyl)nn-dimethylaniline chromium tricarbonyl (1j-Cr)



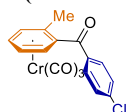
Orange solid. 51.8 mg, 72% yield. The regioisomeric ratio was > 10:1. R_f = 0.35 (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 3:1 v/v. (1.1 equiv) **^1H NMR** (500 MHz, Chloroform-*d*) δ 7.74 (d, J = 7.1 Hz, 2H), 7.56 (t, J = 7.4 Hz, 1H), 7.48 (t, J = 7.5 Hz, 2H), 6.23 (d, J = 7.5 Hz, 2H), 4.86 (d, J = 7.5 Hz, 2H), 2.99 (s, 6H). **^{13}C NMR** (126 MHz, Chloroform-*d*) δ 231.98, 193.58, 137.23, 136.22, 131.78, 128.44, 128.29, 98.02, 89.48, 73.08, 39.82. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{16}\text{CrNO}_4]^+$, 362.0479, found, 362.0476.

2-(2-(trifluoromethyl)benzoyl)toluene chromium tricarbonyl (1u-Cr)



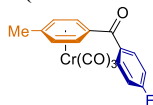
Orange solid. 67.9 mg, 85% yield. The regioisomeric ratio was > 10:1. R_f = 0.55 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform-*d*) δ 7.75 (d, J = 7.7 Hz, 1H), 7.68 (t, J = 7.3 Hz, 1H), 7.65–7.58 (m, 2H), 5.69 (t, J = 6.2 Hz, 1H), 5.45 (d, J = 6.5 Hz, 1H), 5.11 (d, J = 6.3 Hz, 1H), 5.03 (t, J = 6.4 Hz, 1H), 2.53 (s, 3H). **^{13}C NMR** (126 MHz, Chloroform-*d*) δ 230.63, 195.35, 137.41 (q, J = 2.2 Hz), 131.95, 130.25, 128.69, 127.40 (q, J = 32.1 Hz), 126.61 (q, J = 4.6 Hz), 123.59 (q, J = 274.0 Hz), 112.51, 98.85, 96.03, 94.95 (q, J = 1.7 Hz), 92.01, 86.95, 21.11. **^{19}F NMR** (471 MHz, Chloroform-*d*) δ -57.58. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{12}\text{CrF}_3\text{O}_4]^+$, 401.0087, found, 401.0083.

2-(4-chlorobenzoyl)toluene chromium tricarbonyl (1x-Cr)



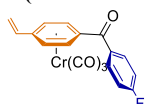
Orange solid. 47.8 mg, 65% yield. The regioisomeric ratio was > 10:1. R_f = 0.40 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (600 MHz, Chloroform-*d*) δ 7.84 (d, J = 8.1 Hz, 2H), 7.47 (d, J = 8.2 Hz, 2H), 5.61–5.54 (m, 2H), 5.17–5.12 (m, 2H), 2.25 (s, 3H). **^{13}C NMR** (151 MHz, Chloroform-*d*) δ 231.27, 192.70, 140.11, 135.24, 130.91, 129.07, 108.98, 103.93, 95.69, 94.72, 90.95, 86.56, 20.00. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{12}\text{ClCrO}_4]^+$, 366.9824, found, 366.9826.

4-(4-fluorobenzoyl)toluene chromium tricarbonyl (1t-Cr)



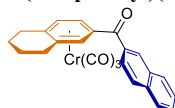
Orange solid. 53.3 mg, 76% yield. The regioisomeric ratio was > 10:1. R_f = 0.35 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform-*d*) δ 7.81 (dd, J = 8.7, 5.3 Hz, 2H), 7.18 (t, J = 8.6 Hz, 2H), 6.05 (d, J = 6.7 Hz, 2H), 5.18 (d, J = 6.7 Hz, 2H), 2.31 (s, 3H). **^{13}C NMR** (126 MHz, Chloroform-*d*) δ 230.67, 191.97, 165.22 (d, J = 254.2 Hz), 132.83 (d, J = 3.2 Hz), 131.20 (d, J = 9.0 Hz), 115.80 (d, J = 22.0 Hz), 110.94, 96.24, 94.23, 90.45, 20.73. **^{19}F NMR** (471 MHz, Chloroform-*d*) δ -105.76. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{12}\text{CrFO}_4]^+$, 351.0119, found, 351.0117.

4-(4-fluorobenzoyl)styrene chromium tricarbonyl (1ac-Cr)



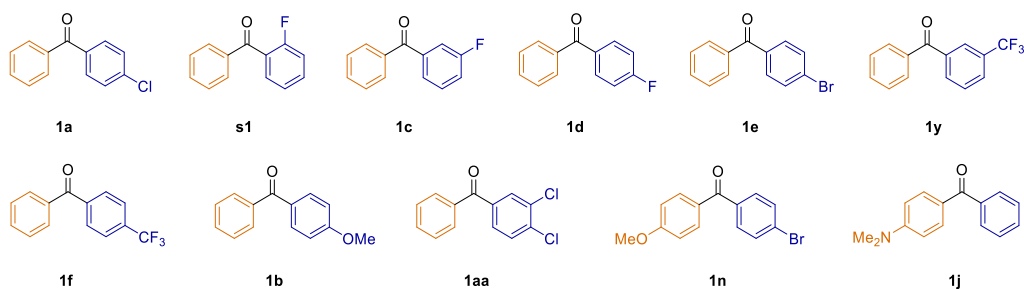
Red solid. 47.7 mg, 66% yield. The regioisomeric ratio was > 10:1. R_f = 0.50 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform-*d*) δ 7.83 (dd, J = 8.6, 5.5 Hz, 2H), 7.19 (t, J = 8.6 Hz, 2H), 6.38 (dd, J = 17.5, 10.8 Hz, 1H), 6.08 (d, J = 6.7 Hz, 2H), 5.79 (d, J = 17.4 Hz, 1H), 5.49 (d, J = 10.8 Hz, 1H), 5.44 (d, J = 6.7 Hz, 2H). **^{13}C NMR** (126 MHz, Chloroform-*d*) δ 230.62, 192.14, 165.39 (d, J = 254.5 Hz), 133.19, 132.87 (d, J = 3.2 Hz), 131.39 (d, J = 9.1 Hz), 118.53, 115.94 (d, J = 22.0 Hz), 107.29, 95.53, 95.28, 87.91. **^{19}F NMR** (471 MHz, Chloroform-*d*) δ -105.55. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{12}\text{CrFO}_4]^+$, 363.0119, found, 363.0125.

2-(2-naphthoyl)(5,6,7,8-tetrahydronaphthalene) chromium tricarbonyl (1w-Cr)

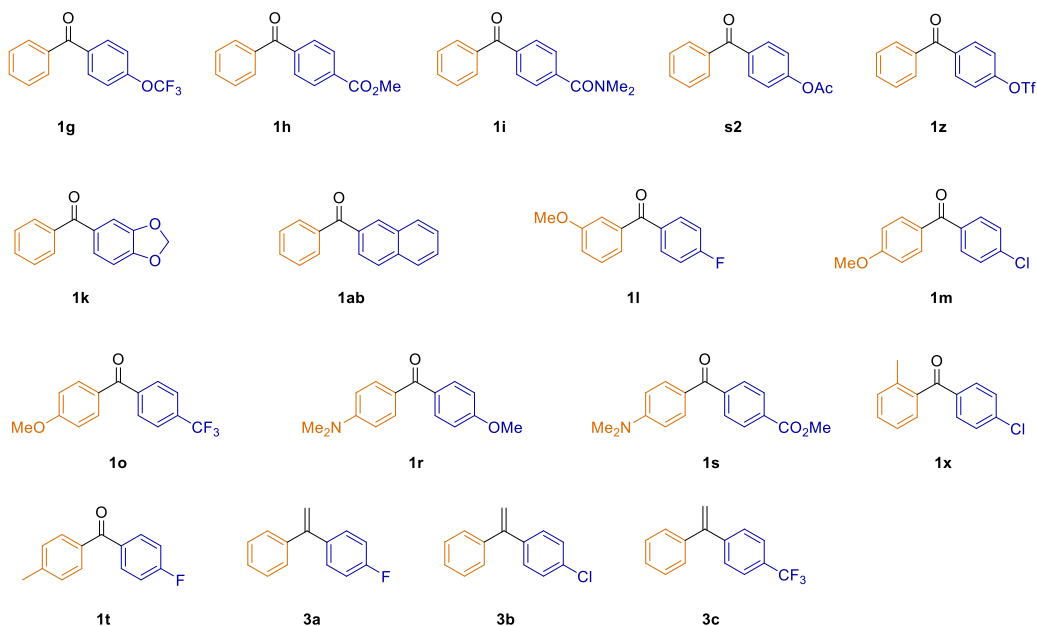


Orange solid. 71.7 mg, 85% yield. The regioisomeric ratio was > 10:1. R_f = 0.45 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. **^1H NMR** (500 MHz, Chloroform-*d*) δ 8.41 (s, 1H), 7.97 (dd, J = 16.0, 8.3 Hz, 2H), 7.91 (d, J = 8.0 Hz, 1H), 7.83 (d, J = 8.5 Hz, 1H), 7.60 (dt, J = 19.8, 7.2 Hz, 2H), 6.03 (d, J = 6.8 Hz, 1H), 5.93 (s, 1H), 5.27 (d, J = 6.8 Hz, 1H), 2.77 (s, 2H), 2.63 (tq, J = 16.4, 7.9, 5.1 Hz, 2H), 1.87–1.70 (m, 4H). **^{13}C NMR** (126 MHz, Chloroform-*d*) δ 231.68, 194.09, 135.11, 133.82, 132.18, 129.93, 129.31, 128.62, 128.35, 127.82, 127.05, 124.99, 111.78, 106.91, 97.64, 95.92, 93.98, 91.39, 28.24, 28.10, 22.00, 21.85. **HRMS-APCI** (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{24}\text{H}_{19}\text{CrO}_4]^+$, 423.0683, found, 423.0685.

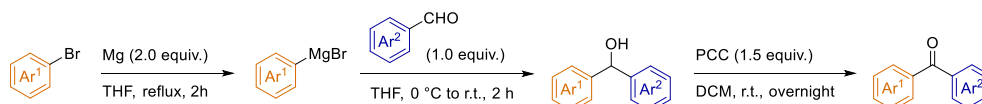
2.3 The preparation of substrates



The following known substrates were commercially available and used without further purifications: **1a**, **s1**, **1c**, **1d**, **1e**, **1y**, **1f**, **1b**, **1aa**, **1n**, **1j**.



The following known substrates were prepared according to the literature procedures: **1g**⁵, **1h**⁵, **1i**⁵, **s2**⁶, **1z**⁷, **1k**⁵, **1ab**⁵, **1l**⁸, **1m**⁹, **1o**⁹, **1r**⁹, **1s**¹⁰, **1x**¹¹, **1t**¹¹, **3a**¹², **3b**¹², **3c**¹².

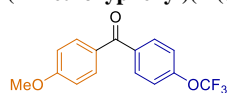


Step i. Under argon atmosphere, magnesium turnings (20.0 mmol, 2.0 equiv) and dry THF (15 mL) were charged in an oven-dried 100 mL round bottom flask equipped with a magnetic stir bar. In a separate oven-dried flask, a 1M solution of the bromoarene (10.0 mmol, 1.0 equiv) in dry THF was prepared, and 20% of this solution was added to the flask containing Mg turnings and heated, followed by addition of 20 μ L 1,2-dibromoethane. Once the reaction was initiated with a change of color or gas evolution, the remaining 80% of the bromoarene solution was added dropwise. The solution was refluxed for additional 2 h. Upon completion of the reaction, the Grignard reagent solution was cooled to room temperature and used for next step without titration.

Step ii. A dry 250 mL round bottom flask equipped with a magnetic stir bar, was charged with benzaldehyde (10.0 mmol, 1.0 equiv) and THF (25 mL) under argon. Under vigorous stirring, the prepared Grignard reagent was slowly added under 0 $^{\circ}$ C. The reaction mixture was stirred under argon for 2.0 h at room temperature and quenched with saturated aqueous NH_4Cl (20 mL). The aqueous layer was extracted with diethyl ether (3 \times 20 mL). The combined organic extract was washed with brine, then dried over anhydrous Na_2SO_4 , filtered, and concentrated to give crude product diarylmethanol directly used in the next step.

Step iii. A 100 mL round bottom flask equipped with a magnetic stir bar, was charged with diarylmethanol, celite (5.0 g) and DCM (30 mL). Under vigorous stirring, PCC (15.0 mmol) was slowly added at room temperature, and the reaction mixture was stirred at room temperature overnight. Then the reaction mixture was filtered, the filter residue was washed with DCM (30 mL \times 3), and the filtrate was concentrated and purified by silica gel column chromatography to afford the desired diaryl ketone.

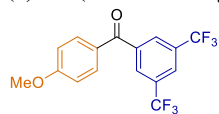
(4-methoxyphenyl)(4-(trifluoromethoxy)phenyl)methanone (**1p**)



White solid. 86% yield. R_f = 0.50 (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.81 (d, J = 8.5 Hz, 4H), 7.31 (d, J = 8.0 Hz, 2H), 7.01–6.95 (m, 2H), 3.90 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, Chloroform- d) δ 193.98, 163.46, 151.75, 136.60, 132.49,

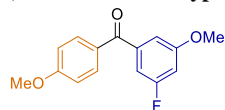
131.59, 129.68, 120.34 (q, $J = 258.4$ Hz), 120.22, 113.71, 55.52. ^{19}F NMR (471 MHz, Chloroform- d) δ -57.61. HRMS-ESI (m/z): $[\text{M}+\text{H}]^+$ calcd. for $[\text{C}_{15}\text{H}_{12}\text{F}_3\text{O}_3]^+$, 297.0733, found, 297.0738.

(3,5-bis(trifluoromethyl)phenyl)(4-methoxyphenyl)methanone (s3)



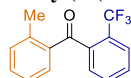
Pale yellow oil. 87% yield. $R_f = 0.55$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 8.19 (s, 2H), 8.07 (s, 1H), 7.84–7.77 (m, 2H), 7.05–6.99 (m, 2H), 3.92 (s, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 192.24, 164.13, 140.15, 132.57, 131.89 (q, $J = 33.7$ Hz), 129.52 (q, $J = 3.2$ Hz), 128.52, 125.12 (h, $J = 3.6$ Hz), 122.95 (q, $J = 273.0$ Hz), 114.15, 55.61. ^{19}F NMR (471 MHz, Chloroform- d) δ -62.90. GCMS-EI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_{10}\text{F}_6\text{O}_2]^+$, 348.1, found, 348.0.

(3-fluoro-5-methoxyphenyl)(4-methoxyphenyl)methanone (1v)



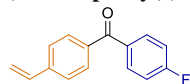
White solid. 92% yield. $R_f = 0.40$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.83 (d, $J = 8.9$ Hz, 2H), 7.11–7.06 (m, 1H), 7.01 (ddd, $J = 8.5, 2.2, 1.3$ Hz, 1H), 6.97 (d, $J = 8.9$ Hz, 2H), 6.81 (dt, $J = 10.2, 2.3$ Hz, 1H), 3.90 (s, 3H), 3.85 (s, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 193.98 (d, $J = 2.5$ Hz), 163.51, 162.99 (d, $J = 247.1$ Hz), 160.76 (d, $J = 10.8$ Hz), 140.69 (d, $J = 8.3$ Hz), 132.54, 129.50, 113.68, 110.70 (d, $J = 2.7$ Hz), 109.06 (d, $J = 23.1$ Hz), 105.28 (d, $J = 25.0$ Hz), 55.81, 55.51. ^{19}F NMR (471 MHz,) δ -110.83. HRMS-ESI (m/z): $[\text{M}+\text{H}]^+$ calcd. for $[\text{C}_{15}\text{H}_{14}\text{FO}_3]^+$, 261.0921, found, 261.0932.

***o*-tolyl(2-(trifluoromethyl)phenyl)methanone (1u)**



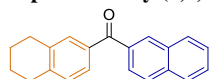
White solid. 93% yield. $R_f = 0.55$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.80–7.75 (m, 1H), 7.63–7.56 (m, 2H), 7.43 (t, $J = 7.5$ Hz, 1H), 7.40–7.35 (m, 1H), 7.32 (d, $J = 7.6$ Hz, 1H), 7.27 (d, $J = 7.8$ Hz, 1H), 7.18 (t, $J = 7.5$ Hz, 1H), 2.62 (s, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 197.38, 140.35, 139.95 (q, $J = 2.0$ Hz), 136.05, 132.58, 132.43, 131.98, 131.40 (q, $J = 1.2$ Hz), 130.00, 128.81, 128.16 (q, $J = 32.3$ Hz), 126.75 (q, $J = 4.9$ Hz), 125.34, 123.61 (q, $J = 274.0$ Hz), 21.41. ^{19}F NMR (471 MHz, Chloroform- d) δ -58.05. GCMS-EI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{15}\text{H}_{11}\text{F}_3\text{O}]^+$, 264.1, found, 264.1.

(4-fluorophenyl)(4-vinylphenyl)methanone (1ac)

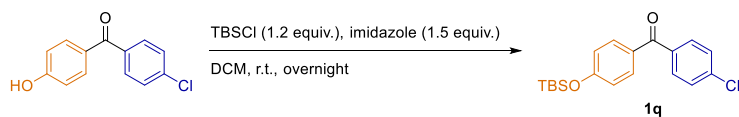


White solid. 86% yield. $R_f = 0.55$ (PE/EA = 9:1 v/v). Elution with PE/EA = 20:1 to 5:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.87–7.80 (m, 2H), 7.75 (d, $J = 8.1$ Hz, 2H), 7.51 (d, $J = 8.2$ Hz, 2H), 7.16 (t, $J = 8.6$ Hz, 2H), 6.78 (dd, $J = 17.6, 10.9$ Hz, 1H), 5.90 (d, $J = 17.6$ Hz, 1H), 5.42 (d, $J = 10.9$ Hz, 1H). ^{13}C NMR (126 MHz, Chloroform- d) δ 194.72, 165.32 (d, $J = 253.8$ Hz), 141.61, 136.53, 135.91, 133.92 (d, $J = 2.8$ Hz), 132.53 (d, $J = 9.2$ Hz), 130.36, 126.09, 116.69, 115.43 (d, $J = 21.9$ Hz). ^{19}F NMR (471 MHz, Chloroform- d) δ -106.10. GCMS-EI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{15}\text{H}_{11}\text{FO}]^+$, 226.1, found, 226.1.

naphthalen-2-yl(5,6,7,8-tetrahydronaphthalen-2-yl)methanone (1w)

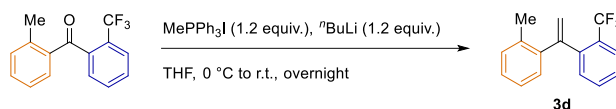


White solid. 84% yield. $R_f = 0.55$ (PE/EA = 9:1 v/v, elution with PE/EA = 20:1 to 5:1 v/v). ^1H NMR (500 MHz, Chloroform- d) δ 8.26 (s, 1H), 7.96–7.88 (m, 4H), 7.64–7.52 (m, 4H), 7.19 (d, $J = 7.9$ Hz, 1H), 2.88–2.81 (m, 4H), 1.85 (hept, $J = 5.2$ Hz, 4H). ^{13}C NMR (126 MHz, Chloroform- d) δ 196.78, 142.52, 137.37, 135.31, 135.18, 135.12, 132.26, 131.49, 131.01, 129.34, 129.00, 128.13, 128.10, 127.79, 127.39, 126.68, 125.89, 29.65, 29.37, 22.98, 22.87. HRMS-ESI (m/z): $[\text{M}+\text{H}]^+$ calcd. for $[\text{C}_{21}\text{H}_{19}\text{O}]^+$, 287.1430, found, 287.1438.



A dry 100 mL round bottom flask equipped with a magnetic stir bar, was charged with (4-chlorophenyl)(4-hydroxyphenyl)methanone (10.0 mmol, 1.0 equiv), TBSCl (12.0 mmol) and DCM (20 mL). Under vigorous stirring, imidazole (15.0 mmol) was slowly added under 0 °C. The reaction mixture was stirred overnight at room temperature and quenched with saturated aqueous NH_4Cl (20 mL). The aqueous layer was extracted with EtOAc (3 \times 20 mL). The combined organic extract was washed with brine, then dried over anhydrous Na_2SO_4 , filtered, and concentrated to give crude product. The crude product was purified by silica gel column chromatography to afford the desired diaryl ketone **1q**. Colorless oil. 96% yield. $R_f = 0.40$ (PE/EA = 20:1 v/v). Elution with PE/EA = 30:1 to 10:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.73 (d, $J = 8.6$ Hz, 2H), 7.71 (d, $J =$

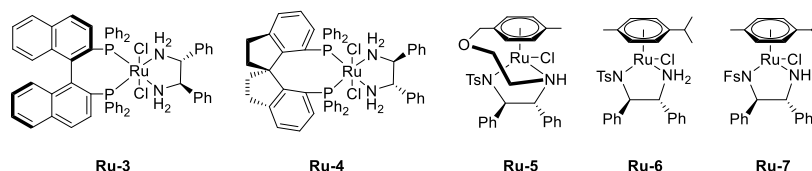
8.4 Hz, 2H), 7.45 (d, J = 8.4 Hz, 2H), 6.90 (d, J = 8.6 Hz, 2H), 1.00 (s, 9H), 0.25 (s, 6H). ^{13}C NMR (126 MHz, Chloroform- d) δ 194.31, 160.12, 138.24, 136.47, 132.32, 131.14, 130.29, 128.46, 119.79, 25.54, 18.20, -4.39. HRMS-ESI (m/z): $[\text{M}+\text{H}]^+$ calcd. for $[\text{C}_{19}\text{H}_{24}\text{ClO}_2\text{Si}]^+$, 347.1229, found, 347.1232.



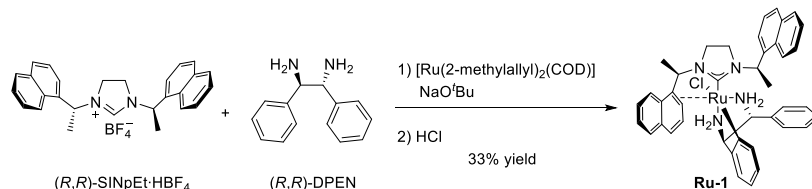
A solution of $t\text{BuLi}$ in hexanes (C = 1.6 M, 12.0 mmol, 1.2 equiv.) was added dropwise, at 0 °C, to a solution of MePPh_3I (12.0 mmol, 1.2 equiv.) in dry THF (30 mL). The reaction mixture was stirred at 0 °C for 30 minutes. A solution of the *o*-tolyl(2-(trifluoromethyl)phenyl)methanone (10.0 mmol, 1.0 equiv.) in dry THF (10 mL) was added dropwise at 0 °C, and the reaction mixture was stirred at room temperature overnight. It was then diluted with brine and extracted with EtOAc (3×40 mL). The combined organic extracts were washed with brine and dried with Na_2SO_4 . The solvent was removed under reduced pressure and the residue was purified by flash chromatography on silica to yield the desired product. Yellow solid. 65% yield. R_f = 0.80 (PE). Elution with PE. ^1H NMR (600 MHz, Chloroform- d) δ 7.68 (d, J = 7.9 Hz, 1H), 7.46 (t, J = 7.5 Hz, 1H), 7.38 (t, J = 7.6 Hz, 1H), 7.27 (d, J = 7.7 Hz, 1H), 7.21–7.15 (m, 2H), 7.16–7.07 (m, 2H), 5.54 (d, J = 18.4 Hz, 2H), 2.23 (s, 3H). ^{13}C NMR (151 MHz, Chloroform- d) δ 146.69, 141.87, 141.04, 135.58, 131.75, 131.35, 130.82, 130.01, 127.91 (q, J = 30.0, 29.6 Hz), 127.48, 127.29, 126.57 (q, J = 5.6 Hz), 125.52, 124.18 (q, J = 273.6 Hz), 120.79, 20.83. ^{19}F NMR (565 MHz, Chloroform- d) δ -57.13. GCMS-EI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_{13}\text{F}]^+$, 262.1, found, 262.1.

3 Asymmetric Hydrogenation of 1,1-Diarylethenes and Benzophenones

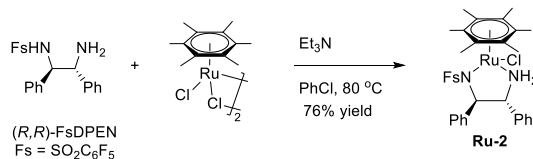
3.1 Synthesis of ruthenium-catalysts



The following known ruthenium-catalysts were commercially available and used without further purifications: **Ru-3**, **Ru-5**, **Ru-6**, **Ru-7**. **Ru-4**¹³ was prepared according to the literature.



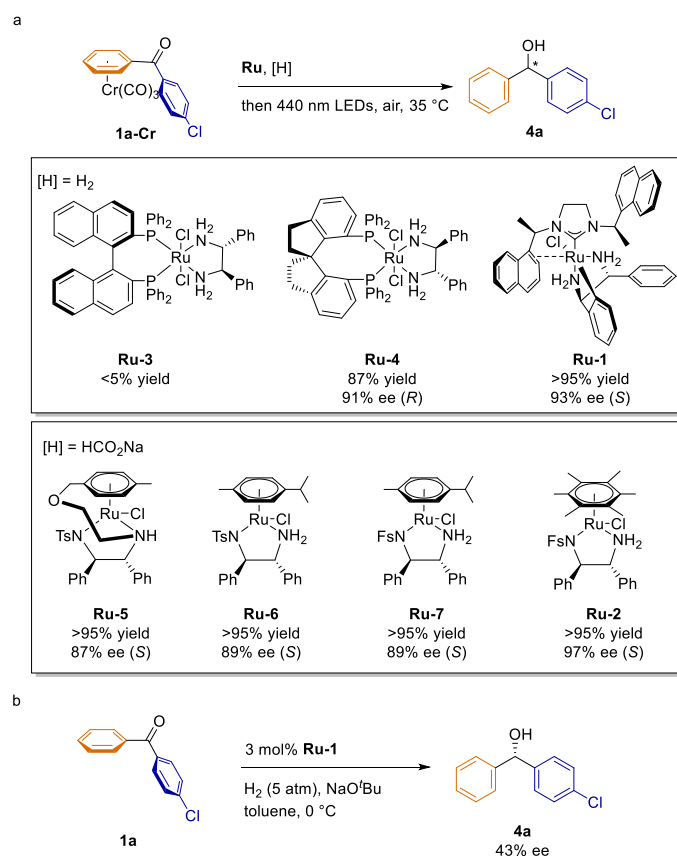
Ru-1 was prepared according to the literature.¹⁴ A typical procedure is described as follow. In a glove box, to an over-dried tube was added $[\text{Ru}(2\text{-methylallyl})_2(\text{COD})]$ (524 mg, 1.5 mmol), $(R,R)\text{-SINpEt-HBF}_4$ (717 mg, 1.5 mmol), $(R,R)\text{-1,2-diphenylethylenediamine}$ (318 mg, 1.5 mmol), and NaO'Bu (173 mg, 1.8 mmol). The mixture was suspended in *n*-hexane (60 mL) and stirred at 40 °C for 4 days. HCl (4M in dioxane) (2.7 mmol) was then added to the reaction mixture at 0 °C. After stirring for 30 min at 0 °C, the mixture was concentrated and purified by column chromatography on silica gel (elution with PE/EA = 20:1 to 2:1 v/v) to yield the complexes **[Ru-1]** (360 mg, 0.49 mmol) in 33% yield. **Ru-1**^{*} was prepared by the same method from $(S,S)\text{-SINpEt-HBF}_4$ and $(S,S)\text{-1,2-diphenylethylenediamine}$. ^1H NMR (500 MHz, Chloroform- d) δ 8.38 (d, J = 8.2 Hz, 1H), 7.79 (d, J = 7.3 Hz, 1H), 7.71 (d, J = 8.0 Hz, 1H), 7.63 (d, J = 8.1 Hz, 1H), 7.52 (d, J = 7.1 Hz, 1H), 7.44 (t, J = 7.7 Hz, 1H), 7.40–7.35 (m, 1H), 7.33 (t, J = 7.3 Hz, 1H), 7.29–7.27 (m, 1H), 7.15–7.14 (m, 2H), 7.13–7.09 (m, 1H), 7.09–7.05 (m, 3H), 7.04–7.00 (m, 3H), 6.56 (d, J = 18.7, 1H), 6.56 (d, J = 8.0 Hz, 1H), 6.15 (dd, J = 9.0, 5.7 Hz, 1H), 5.46 (q, J = 6.4 Hz, 1H), 4.06–3.92 (m, 2H), 3.79 (td, J = 9.4, 2.9 Hz, 1H), 3.75 (d, J = 5.7 Hz, 1H), 3.72–3.65 (m, 2H), 3.58 (d, J = 7.9 Hz, 1H), 2.89–2.74 (m, 2H), 1.63 (d, J = 7.0 Hz, 3H), 1.26 (t, J = 7.1 Hz, 2H), 1.24–1.17 (m, 4H), 1.07 (d, J = 9.4 Hz, 1H), 0.13 (d, J = 9.3 Hz, 1H).



To a premixed solution of $[\text{RuCl}_2(\eta^6\text{-C}_6\text{Me}_6)]_2$ (334.0 mg, 0.5 mmol) and $(R,R)\text{-FsDPEN}$ (442.4 mg, 1.0 mmol) in chlorobenzene (15 mL), Et_3N (404.8 mg, 4.00 mmol) was added and refluxed for 2 h. After evaporation of the resulting reddish orange solution, the residual mixture was dissolved in CH_2Cl_2 (15 mL), washed with water (2 mL \times 3) and dried over Na_2SO_4 . The solvent was removed in vacuum and the residue was purified by column chromatography on silica gel (elution with PE/EA = 1:1 v/v) to afford an orange powder of **Ru-2** (562.4 mg, 0.76 mmol) in 76% yield. ^1H NMR (500 MHz, Chloroform- d) δ 7.20–7.12 (m, 3H), 6.90–6.83 (m, 7H), 4.15 (d, J = 11.0 Hz, 1H), 3.85 (dd, J = 13.6, 9.9 Hz, 1H), 3.75–3.63 (m,

2H). ^{13}C NMR (126 MHz, CDCl_3) δ 139.11, 138.29, 128.78, 128.72, 127.88, 127.27, 126.96, 126.85, 91.66, 71.59, 67.99, 16.07. ^{19}F NMR (471 MHz, CDCl_3) δ -134.33 (d, J = 22.6 Hz), -153.58 (t, J = 20.8 Hz), -163.20 (t, J = 20.8 Hz). HRMS-ESI (m/z): $[\text{M}-\text{Cl}]^+$ calcd. for $[\text{C}_{32}\text{H}_{32}\text{F}_5\text{N}_2\text{O}_2\text{RuS}]^+$, 705.1143, found, 705.1165.

3.2 Asymmetric hydrogenation of **1a-Cr** with different ruthenium-catalysts



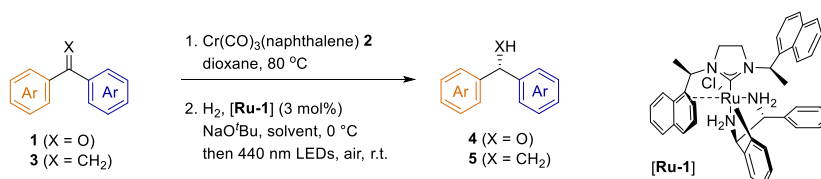
Supplementary Fig. 4 a) Asymmetric hydrogenation of **1a-Cr** with different ruthenium-catalysts. b) Asymmetric hydrogenation of **1a** with **Ru-1**.

General procedure:

$[\text{H}] = \text{H}_2$: To a 5 ml glass tube **1a-Cr** or **1a** (0.1 mmol), **Ru** (3 mol%), NaO'Bu (15 mol%), and toluene (1.0 mL) were added in a glovebox. The glass tube was taken out of the glovebox and placed in a pre-cooled (0 $^\circ\text{C}$) stainless steel autoclave, then the autoclave was pressurized and depressurized with hydrogen gas five times before the indicated pressure (5 atm) was set. The reaction mixture was stirred at 0 $^\circ\text{C}$ for 36 h. After the autoclave was carefully depressurized, the tube was irradiated under 440 nm LEDs (20 W) for 2 hours at room temperature in air atmosphere. Yields were determined by ^1H NMR using 1,1,2,2-tetrachloroethane as the internal standard, and enantiomeric excess was determined by SFC analysis using chiral column.

$[\text{H}] = \text{HCO}_2\text{Na}$: To a 4 ml glass vial, **1a-Cr** (0.1 mmol), **Ru** (7 mol%), HCO_2Na (10.0 equiv.), and DMF/ H_2O (10:1, 0.5 ml/0.05 ml) were added in a glovebox. The reaction mixture was stirred at 35 $^\circ\text{C}$ for 36 h. Then the vial was irradiated under 440 nm LEDs (20 W) for 2 hours at room temperature in air atmosphere. Yields were determined by ^1H NMR using 1,1,2,2-tetrachloroethane as the internal standard, and enantiomeric excess was determined by SFC analysis using chiral column.

3.3 General procedure for asymmetric hydrogenation of benzophenones and 1,1-diarylethenes

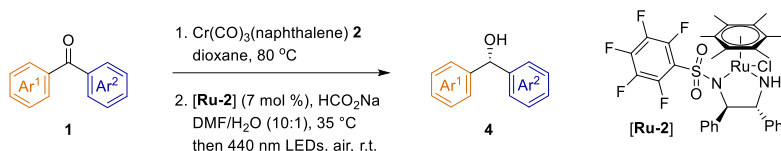


General procedure A:

(1) In a glovebox, a solution of $\text{Cr}(\text{CO})_3(\text{naphthalene})$ **2** (0.3 mmol, 1.5 equiv.), **1** or **3** (0.2 mmol, 1.0 equiv.) in 1,4-dioxane (2 mL) in a 4 ml glass vial was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at

80 °C in the dark for 24 h. After cooling to room temperature, the solution was evaporated under reduced pressure. The residue was purified by column chromatography eluting with EtOAc/Petroleum ether (impurities that cannot be separated by column chromatography can be removed by recrystallization from diethyl ether and pentane).

(2) To a 5 ml glass tube, **Ru-1** (3 mol%), NaO^tBu (15 mol%), **1-Cr** or **3-Cr** (1.0 equiv.), and toluene (hexane for 1,1-diarylethenes) (0.10 M) were added in a glove box. The glass tube was taken out of the glovebox and placed in a pre-cooled (0 °C) stainless steel autoclave, then the autoclave and pressurized and depressurized with hydrogen gas five times before the indicated pressure (5 atm) (50 atm for 1,1-diarylethenes) was set. The reaction mixture was stirred at 0 °C for 36 h. After the autoclave was carefully depressurized, the tube was irradiated under 440 nm LEDs (20 W) for 2 hours at room temperature in air atmosphere. Then the product was purified by flash column chromatography on silica gel. The enantiomeric excess of the product was determined by SFC or HPLC analysis using chiral column (when our chiral column cannot separate the product, the enantiomeric excess was measured before the chromium is removed).

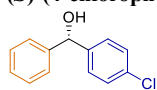


General procedure B:

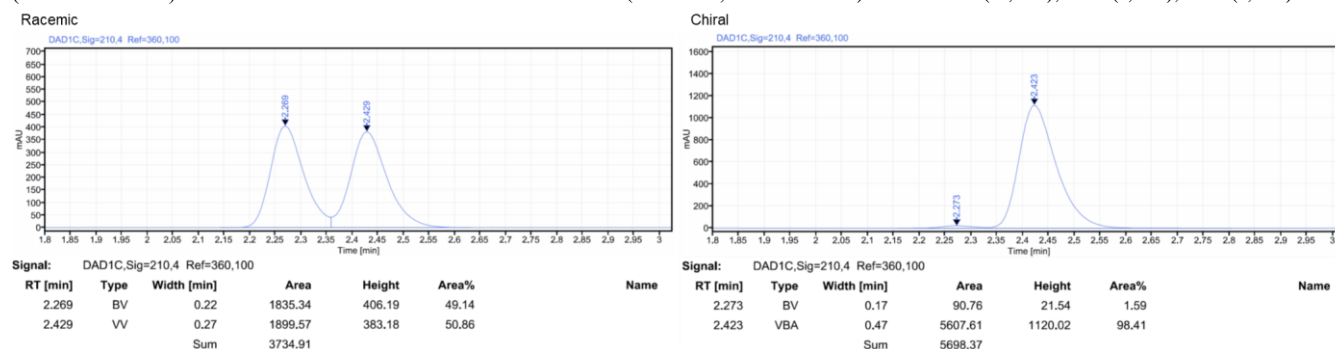
(1) In a glovebox, a solution of Cr(CO)₃(naphthalene) **2** (0.3 mmol, 1.5 equiv.), **1** (0.2 mmol, 1.0 equiv.) in 1,4-dioxane (2 mL) in a 4 ml glass vial was stirred at room temperature for 2 min, then the sealed reaction vial was taken out of the glovebox and the reaction mixture was stirred at 80 °C in the dark for 24 h. After cooling to room temperature, the solution was evaporated under reduced pressure. The red residue was purified by column chromatography eluting with EtOAc/Petroleum ether (impurities that cannot be separated by column chromatography can be removed by recrystallization from diethyl ether and pentane).

(2) To a 4ml glass vial, **Ru-2** (7 mol%), HCO₂Na (10 equiv.), **1-Cr** (1.0 equiv.), and DMF/H₂O (10:1, 0.18 M) were added in a glovebox. The reaction mixture was stirred at 35 °C for 36 h. Then the vial was irradiated under 440 nm LEDs (20 W) for 2 hours at room temperature in air atmosphere. Then the product was purified by flash column chromatography on silica gel. The enantiomeric excess of the product was determined by SFC or HPLC analysis using chiral column (when our chiral column cannot separate the product, the enantiomeric excess was measured before the chromium is removed).

(S)-(4-chlorophenyl)(phenyl)methanol (**4a**)



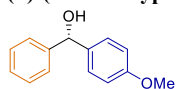
Known compound.¹⁵ **Procedure A**, white solid, 34.0 mg, 78% yield, 93% ee. **Procedure B**, white solid, 32.3 mg, 74% yield, 96% ee. R_f = 0.50 (PE/EA = 6:1 v/v). Elution with PE/EA = 10:1 to 5:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.50–7.09 (m, 9H), 5.66 (s, 1H), 2.75 (s, 1H).



Supplementary Fig. 5 SFC chromatograms of 4a. AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, t_{minor} = 2.3 min, t_{major} = 2.4 min.

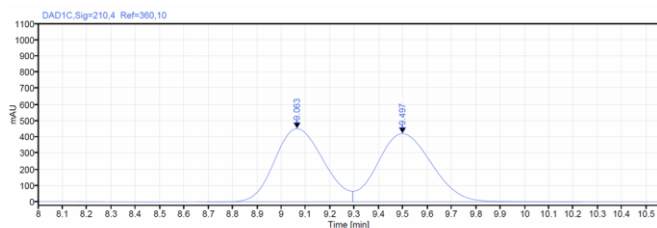
[α]_D²⁰ +19.8 (c 1.02 in CHCl₃) 93% ee (S) (lit.¹⁶ [α]_D²⁰ +8.0 (c 1.51 in CHCl₃) 48% ee (S)).

(S)-(4-methoxyphenyl)(phenyl)methanol (**4b**)



Known compound.¹⁷ **Procedure A**, step (1) with 0.22 mmol **2**, 24.8 mg, white solid, 58% yield, 90% ee. R_f = 0.55 (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 5:1 v/v. ¹H NMR (600 MHz, Chloroform-*d*) δ 7.35 (d, *J* = 7.4 Hz, 2H), 7.31 (t, *J* = 7.6 Hz, 2H), 7.28–7.22 (m, 3H), 6.84 (d, *J* = 8.7 Hz, 2H), 5.76 (s, 1H), 3.76 (s, 3H), 2.32 (s, 1H).

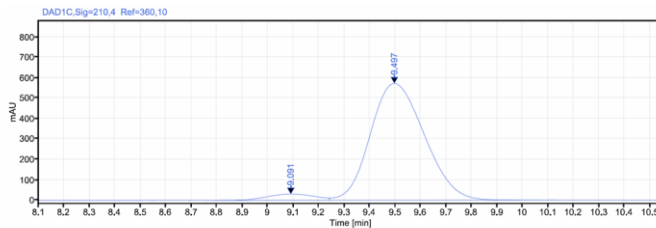
Racemic



Signal: DAD1C, Sig=210.4 Ref=360,10

RT [min]	Type	Width [min]	Area	Height	Area%
9.063	BV	0.60	6272.49	451.81	49.04
9.497	VV	0.94	6518.44	421.21	50.96
Sum			12790.93		

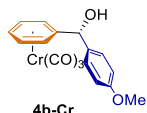
Chiral



Signal: DAD1C, Sig=210.4 Ref=360,10

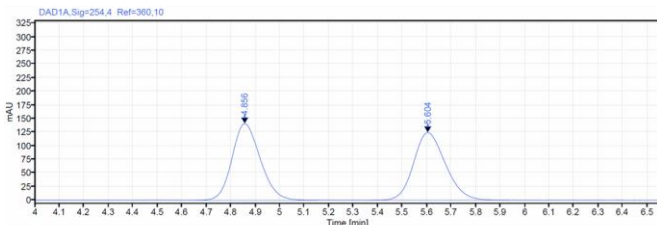
Name	RT [min]	Type	Width [min]	Area	Height	Area%	Name
	9.091	BV	0.45	403.86	30.95	4.56	
	9.497	VV	0.75	8445.58	572.96	95.44	
	Sum			8849.44			

Supplementary Fig. 6 SFC chromatograms of **4b**. OJ-3 column, MeOH/CO₂ = 10:90, 0.6 mL/min, 210 nm, *t*_{minor} = 9.1 min, *t*_{major} = 9.5 min.



4b-Cr, 91% ee. (**1b-Cr** with **Ru-1**)

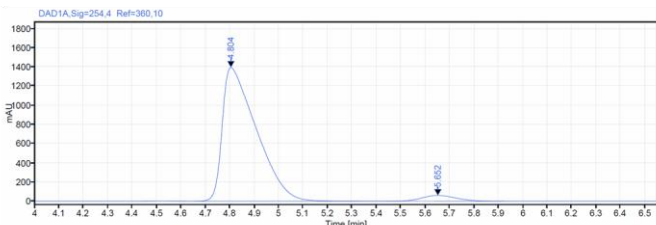
Racemic



Signal: DAD1A, Sig=254.4 Ref=360,10

RT [min]	Type	Width [min]	Area	Height	Area%
4.856	BV	0.50	1133.54	140.89	50.05
5.604	BV	0.57	1131.23	124.55	49.95
Sum			2264.77		

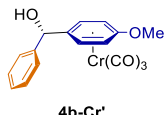
Chiral



Signal: DAD1A, Sig=254.4 Ref=360,10

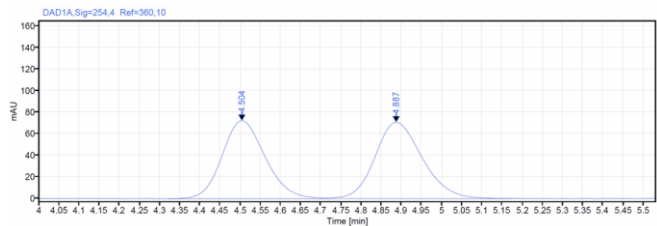
Name	RT [min]	Type	Width [min]	Area	Height	Area%	Name
	4.804	VV	0.71	13504.38	1396.81	95.82	
	5.652	VV	0.52	589.75	60.16	4.18	
	Sum			14094.13			

Supplementary Fig. 7 SFC chromatograms of **4b-Cr**. OD-3 column, MeOH/CO₂ = 10:90, 1.5 mL/min, 254 nm, *t*_{minor} = 5.6 min, *t*_{major} = 4.8 min.



4b-Cr', 93% ee. (**1b-Cr'** with **Ru-1'**)

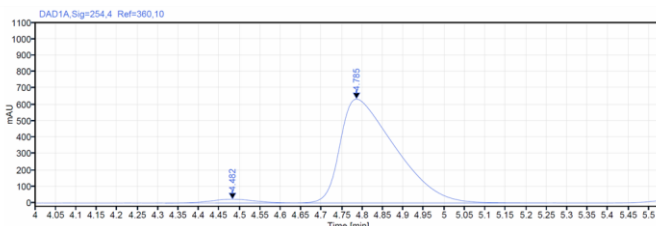
Racemic



Signal: DAD1A, Sig=254.4 Ref=360,10

RT [min]	Type	Width [min]	Area	Height	Area%
4.504	VV	0.38	539.26	72.54	48.13
4.887	VV	0.47	581.10	70.96	51.87
Sum			1120.36		

Chiral

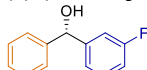


Signal: DAD1A, Sig=254.4 Ref=360,10

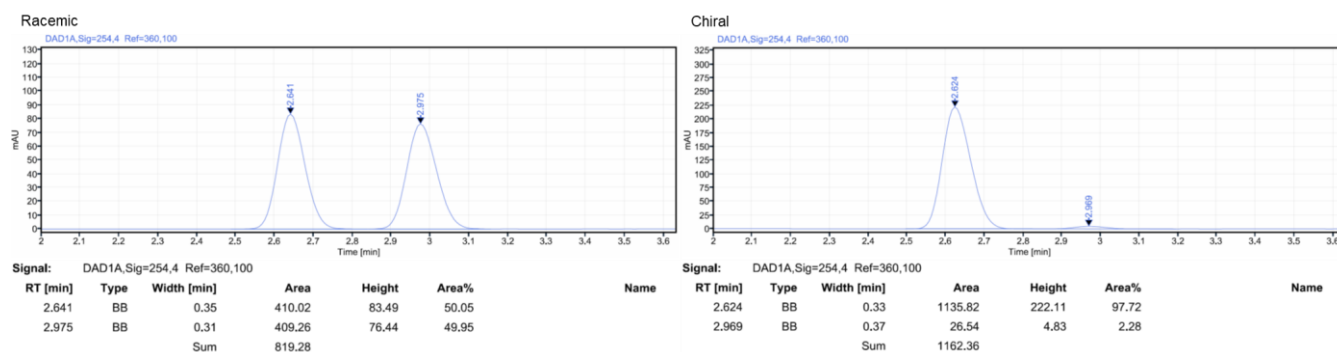
Name	RT [min]	Type	Width [min]	Area	Height	Area%	Name
	4.482	BV	0.31	177.56	23.25	2.91	
	4.785	VV	0.62	5930.63	634.52	97.09	
	Sum			6108.19			

Supplementary Fig. 8 SFC chromatograms of **4b-Cr'**. OD-3 column, MeOH/CO₂ = 10:90, 1.5 mL/min, 254 nm, *t*_{minor} = 4.5 min, *t*_{major} = 4.8 min.

(S)-(3-fluorophenyl)(phenyl)methanol (4c)

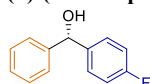


Known compound.¹⁸ **Procedure A**, white solid, 32.3 mg, 80% yield, 82% ee. **Procedure B**, white solid, 30.6 mg, 76% yield, 95% ee. *R*_f = 0.45 (PE/EA = 6:1 v/v). Elution with PE/EA = 10:1 to 5:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.36–7.30 (m, 4H), 7.30–7.24 (m, 2H), 7.15–7.07 (m, 2H), 6.98–6.90 (m, 1H), 5.77 (s, 1H), 2.40 (s, 1H).

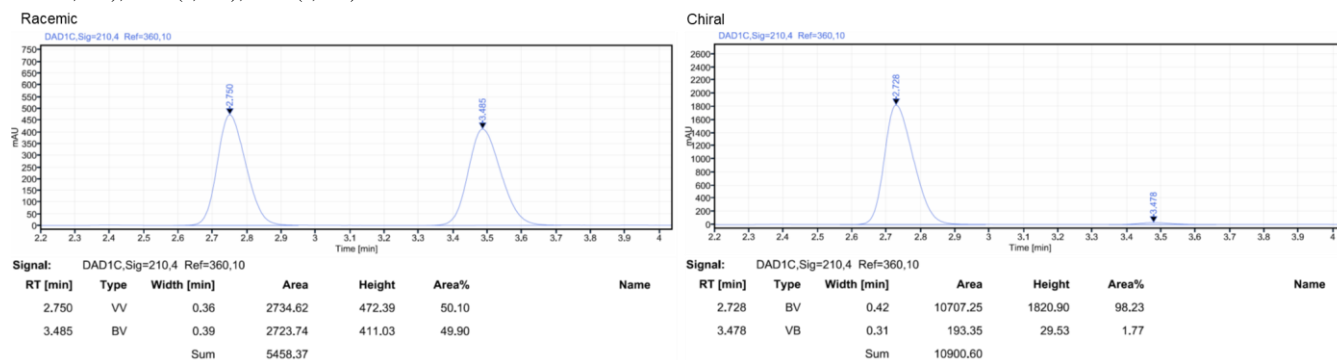


Supplementary Fig. 9 SFC chromatograms of **4c-Cr**. OD-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 254 nm, $t_{\text{minor}} = 3.0$ min, $t_{\text{major}} = 2.6$ min.

(S)-(4-fluorophenyl)(phenyl)methanol (4d)

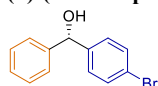


Known compound.¹⁵ **Procedure A**, white solid, 31.0 mg, 77% yield, 96% ee. **Procedure B**, white solid, 31.1 mg, 77% yield, 96% ee. $R_f = 0.45$ (PE/EA = 6:1 v/v). Elution with PE/EA = 10:1 to 5:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.39–7.31 (m, 6H), 7.31–7.24 (m, 1H), 7.01 (t, $J = 8.7$ Hz, 2H), 5.82 (s, 1H), 2.23 (s, 1H).

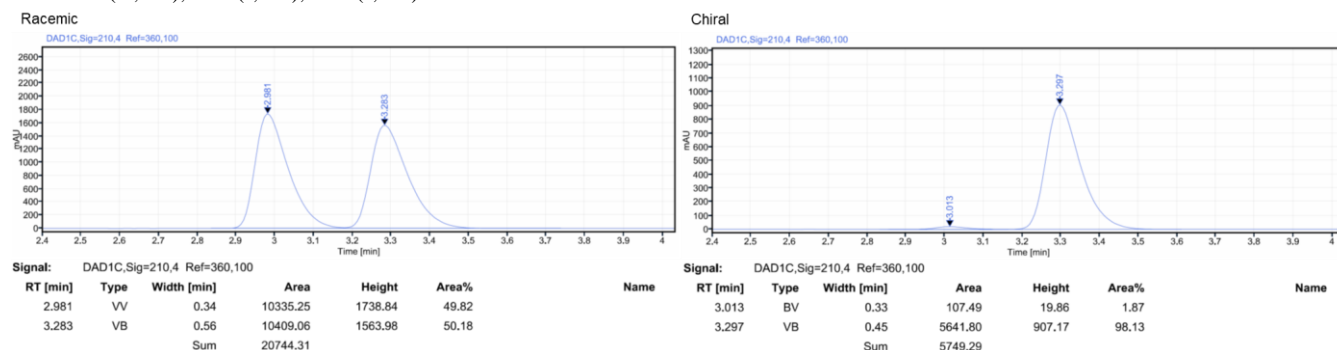


Supplementary Fig. 10 SFC chromatograms of **4d-Cr**. AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 3.5$ min, $t_{\text{major}} = 2.7$ min.

(S)-(4-bromophenyl)(phenyl)methanol (4e)

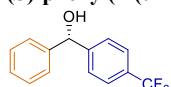


Known compound.¹⁵ **Procedure A**, white solid, 38.3 mg, 73% yield, 92% ee. **Procedure B**, white solid, 36.7 mg, 70% yield, 96% ee. $R_f = 0.45$ (PE/EA = 6:1 v/v). Elution with PE/EA = 10:1 to 5:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.45 (d, $J = 8.5$ Hz, 2H), 7.34 (d, $J = 4.4$ Hz, 4H), 7.30–7.24 (m, 3H), 5.79 (s, 1H), 2.24 (s, 1H).



Supplementary Fig. 11 SFC chromatograms of **4e**. OJ-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 3.0$ min, $t_{\text{major}} = 3.3$ min. $[\alpha]_D^{20} +20.0$ 92% ee (c 1.02 in CHCl₃) (*S*) (lit.¹⁹ $[\alpha]_D^{25} +19.0$ (c 1.00 in CHCl₃) 91% ee (*S*)).

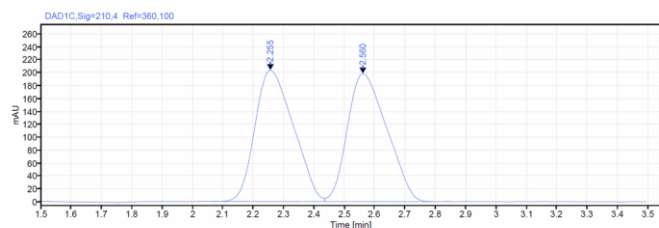
(S)-phenyl(4-(trifluoromethyl)phenyl)methanol (4f)



Known compound.¹⁵ **Procedure A**, white solid, 39.3 mg, 78% yield, 90% ee. **Procedure B**, white solid, 37.7 mg, 75% yield, 99% ee. $R_f = 0.50$

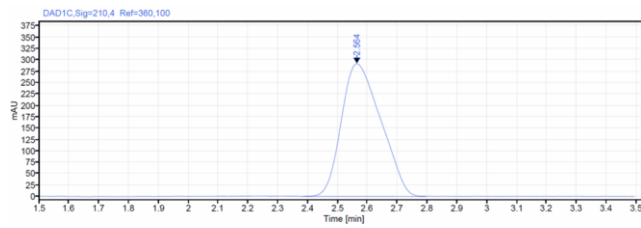
(PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 5:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.59 (d, *J* = 8.2 Hz, 2H), 7.52 (d, *J* = 8.1 Hz, 2H), 7.36 (d, *J* = 4.3 Hz, 4H), 7.34–7.26 (m, 1H), 5.89 (d, *J* = 2.7 Hz, 1H), 2.28 (d, *J* = 3.1 Hz, 1H).

Racemic



RT [min]	Type	Width [min]	Area	Height	Area%
2.255	BV	0.37	1777.26	202.93	50.02
2.560	VB	0.37	1776.03	197.86	49.98
Sum			3553.29		

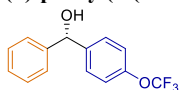
Chiral



Name	RT [min]	Type	Width [min]	Area	Height	Area%	Name
	2.564	BV	0.42	2682.12	292.17	100.00	
Sum				2682.12			

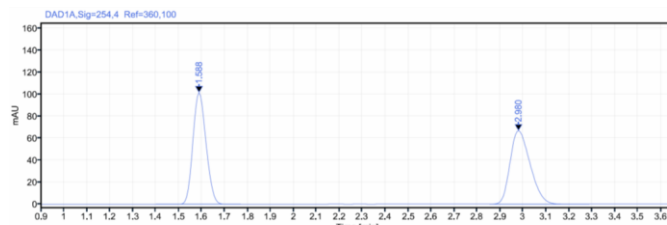
Supplementary Fig. 12 SFC chromatograms of **4f**. OJ-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 210 nm, *t*_{minor} = 2.3 min, *t*_{major} = 2.6 min.

(*S*)-phenyl(4-(trifluoromethoxy)phenyl)methanol (**4g**)



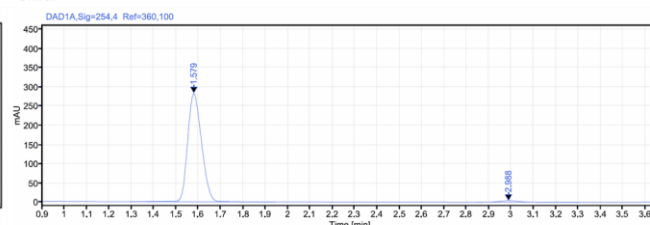
Known compound.²⁰ **Procedure A**, white solid, 42.1 mg, 78% yield, 91% ee. **Procedure B**, white solid, 41.8 mg, 78% yield, 95% ee. *R*_f = 0.45 (PE/EA = 6:1 v/v). Elution with PE/EA = 10:1 to 5:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.40 (d, *J* = 8.4 Hz, 2H), 7.35 (d, *J* = 4.4 Hz, 4H), 7.33–7.25 (m, 1H), 7.17 (d, *J* = 7.6 Hz, 2H), 5.83 (d, *J* = 3.4 Hz, 1H), 2.32 (d, *J* = 3.5 Hz, 1H).

Racemic



RT [min]	Type	Width [min]	Area	Height	Area%	Name
1.588	BB	0.37	402.58	101.75	49.94	
2.980	BB	0.43	403.50	66.91	50.06	
Sum			806.08			

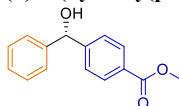
Chiral



RT [min]	Type	Width [min]	Area	Height	Area%	Name
1.579	BB	0.65	1224.66	283.81	97.93	
2.988	BB	0.46	25.92	4.31	2.07	
Sum			1250.58			

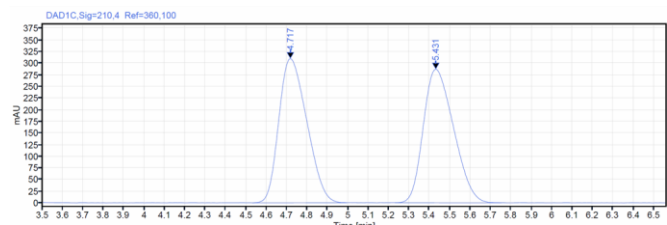
Supplementary Fig. 13 SFC chromatograms of **4g**-Cr. OJ-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 254 nm, *t*_{minor} = 3.0 min, *t*_{major} = 1.6 min.

(*S*)-4-(hydroxy(phenyl)methyl)benzoate (**4h**)



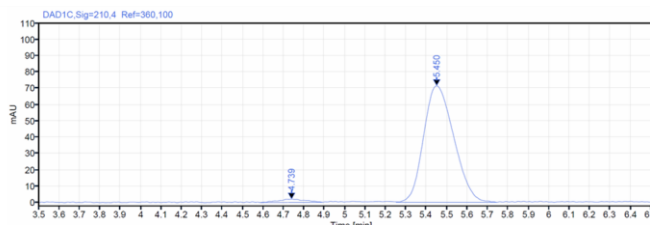
Known compound.²¹ **Procedure A**, white solid, 29.1 mg, 60% yield, 87% ee. **Procedure B**, white solid, 28.2 mg, 58% yield, 95% ee. *R*_f = 0.50 (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.97 (d, *J* = 8.4 Hz, 2H), 7.44 (d, *J* = 8.3 Hz, 2H), 7.37–7.28 (m, 4H), 7.28–7.23 (m, 1H), 5.84 (s, 1H), 3.87 (s, 3H), 2.68 (s, 1H).

Racemic



RT [min]	Type	Width [min]	Area	Height	Area%	Name
4.717	BV	0.46	2959.23	310.10	49.96	
5.431	VV	0.52	2964.24	287.27	50.04	
Sum			5923.47			

Chiral

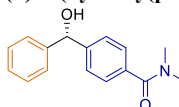


Name	RT [min]	Type	Width [min]	Area	Height	Area%	Name
	4.739	BB	0.30	18.61	2.00	2.47	
	5.450	VB	0.49	734.30	71.46	97.53	
Sum				752.91			

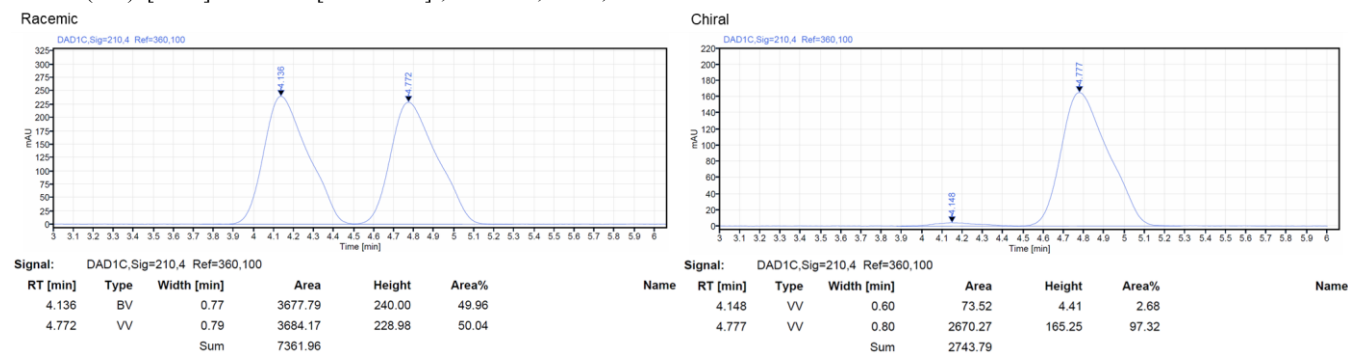
Supplementary Fig. 14 SFC chromatograms of **4h**. OJ-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, *t*_{minor} = 4.7 min, *t*_{major} = 5.5 min.

[α]_D²⁰ +17.6 (c 1.62 in CHCl₃) 87% ee (*S*) (lit.²¹ [α]_D²³ +32.0 (c 1.00 in CHCl₃) 92% ee (*S*)).

(*S*)-4-(hydroxy(phenyl)methyl)-*N,N*-dimethylbenzamide (**4i**)

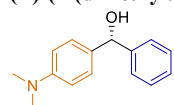


Procedure A, white solid, 19.4 mg, 38% yield, 83% ee. **Procedure B**, white solid, 17.9 mg, 35% yield, 94% ee. $R_f = 0.50$ (EA). Elution with EA. $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.39 (d, $J = 8.1$ Hz, 2H), 7.37–7.30 (m, 6H), 7.26 (t, $J = 6.9$ Hz, 1H), 5.82 (s, 1H), 3.01 (d, $J = 62.8$ Hz, 6H), 2.82 (s, 1H). $^{13}\text{C NMR}$ (126 MHz, Chloroform- d) δ 171.45, 145.36, 143.54, 135.18, 128.50, 127.67, 127.20, 126.59, 126.39, 75.76, 39.57, 35.33. HRMS (m/z): $[\text{M}+\text{H}]^+$ calcd. for $[\text{C}_{16}\text{H}_{18}\text{NO}_2]^+$, 256.1332, found, 256.1336.

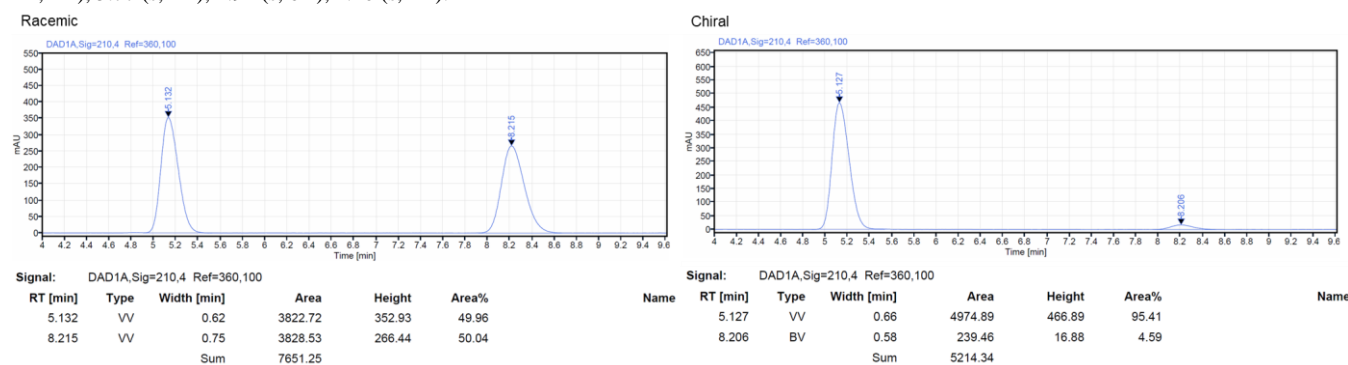


Supplementary Fig. 15 SFC chromatograms of **4i**. OJ-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 4.1$ min, $t_{\text{major}} = 4.8$ min,.

(R)-4-(dimethylamino)phenyl(phenyl)methanol (4j)

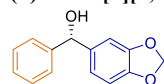


Known compound.²² **Procedure A**, step (1) with 0.22 mmol **2**, white solid, 31.1 mg, 68% yield, 90% ee. $R_f = 0.45$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. $^1\text{H NMR}$ (600 MHz, Chloroform- d) δ 7.38 (d, $J = 7.3$ Hz, 2H), 7.32 (t, $J = 7.7$ Hz, 2H), 7.25–7.19 (m, 3H), 6.69 (d, $J = 8.8$ Hz, 2H), 5.77 (s, 1H), 2.92 (s, 6H), 2.10 (s, 1H).

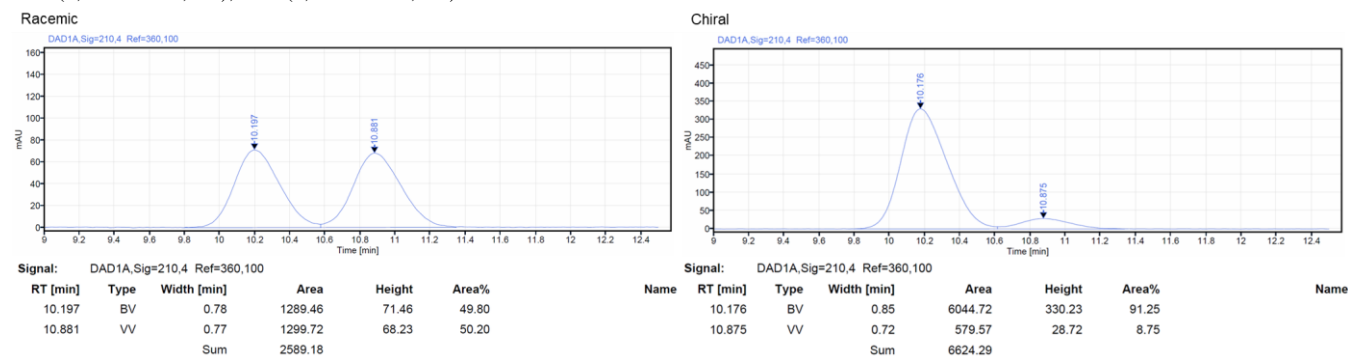


Supplementary Fig. 16 SFC chromatograms of **4j**. IH-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 8.2$ min, $t_{\text{major}} = 5.1$ min.

(S)-benzo[d][1,3]dioxol-5-yl(phenyl)methanol (4k)

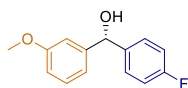


Known compound.¹⁷ **Procedure A**, colorless oil, 36.7 mg, 80% yield, 82% ee. $R_f = 0.45$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.39–7.30 (m, 4H), 7.29–7.22 (m, 1H), 6.87–6.82 (m, 2H), 6.75 (d, $J = 8.4$ Hz, 1H), 5.91 (q, $J = 1.4$ Hz, 2H), 5.75 (d, $J = 2.9$ Hz, 1H), 2.23 (d, $J = 3.3$ Hz, 1H).

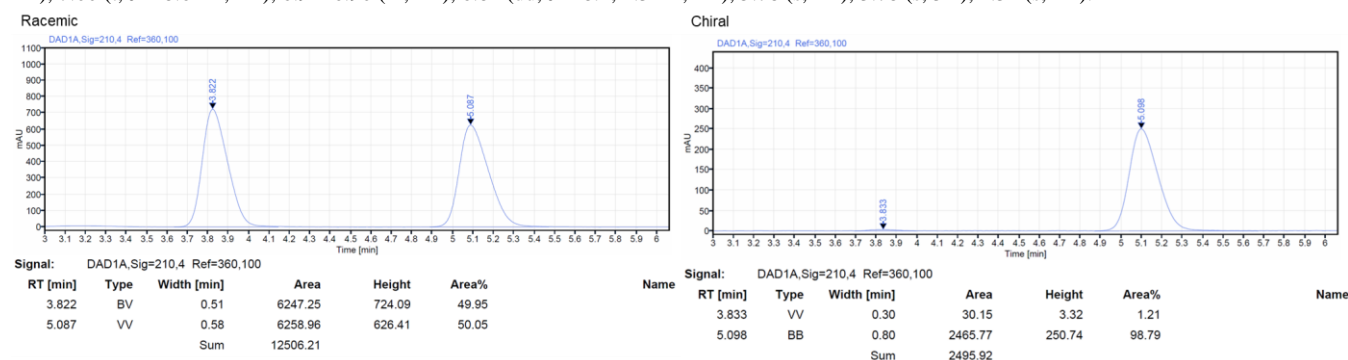


Supplementary Fig. 17 SFC chromatograms of **4k**. OD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 10.9$ min, $t_{\text{major}} = 10.2$ min.

(R)-4-(4-fluorophenyl)(3-methoxyphenyl)methanol (4l)

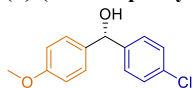


Known compound.²³ **Procedure A**, colorless oil, 35.2 mg, 76% yield, 97% ee. **Procedure B**, colorless oil, 33.8 mg, 73% yield, 91% ee. $R_f = 0.45$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. **¹H NMR** (600 MHz, Chloroform-*d*) δ 7.33 (dd, $J = 8.4, 5.6$ Hz, 2H), 7.25 (t, $J = 8.1$ Hz, 1H), 7.00 (t, $J = 8.6$ Hz, 2H), 6.94–6.90 (m, 2H), 6.81 (dd, $J = 8.2, 2.3$ Hz, 1H), 5.78 (s, 1H), 3.78 (s, 3H), 2.31 (s, 1H).

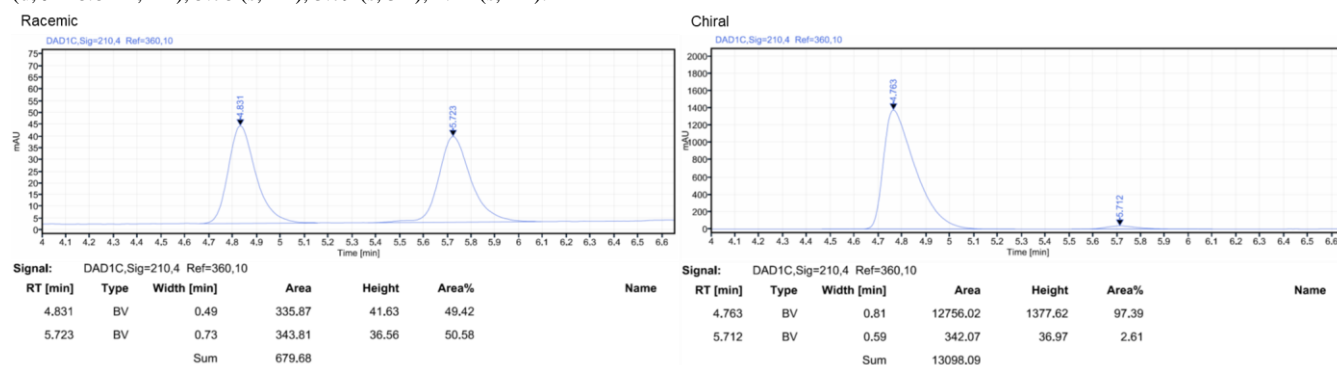


Supplementary Fig. 18 SFC chromatograms of **4l**. OD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 3.8$ min, $t_{\text{major}} = 5.1$ min.

(S)-1-(4-chlorophenyl)-2-(4-methoxyphenyl)ethanol (4m)

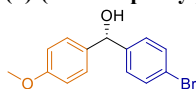


Known compound.¹⁶ **Procedure A**, white solid, 33.0 mg, 66% yield, 94% ee. **Procedure B**, white solid, 32.2 mg, 65% yield, 93% ee. $R_f = 0.45$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. **¹H NMR** (500 MHz, Chloroform-*d*) δ 7.33–7.28 (m, 4H), 7.25 (d, $J = 8.0$ Hz, 2H), 6.87 (d, $J = 8.8$ Hz, 2H), 5.78 (s, 1H), 3.79 (s, 3H), 2.14 (s, 1H).

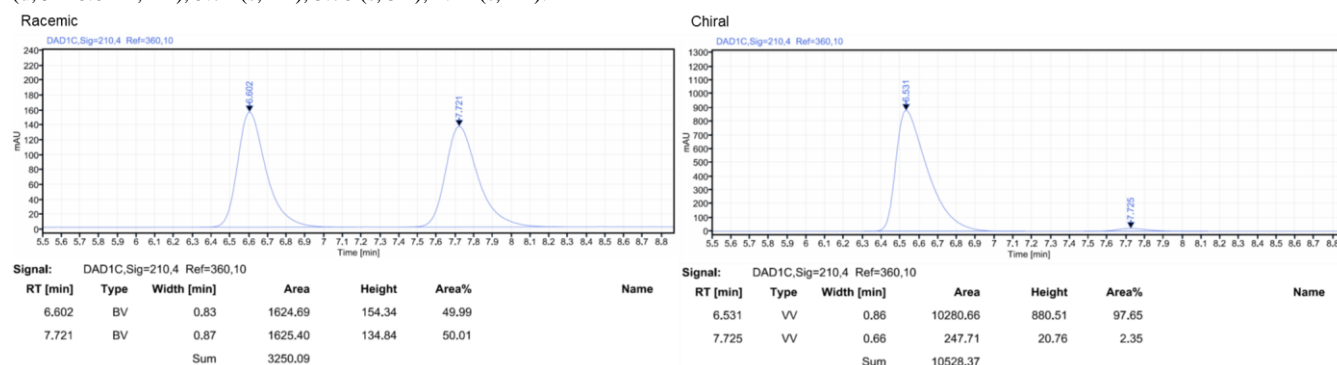


Supplementary Fig. 19 SFC chromatograms of **4m**. AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 5.7$ min, $t_{\text{major}} = 4.8$ min. $[\alpha]_D^{20} +27.8$ (c 1.61 in CHCl₃) 94% ee (*S*). (lit.¹⁶ $[\alpha]_D^{20} +16.6$ (c 0.73 in CHCl₃) 53% ee (*S*)).

(S)-1-(4-bromophenyl)-2-(4-methoxyphenyl)ethanol (4n)

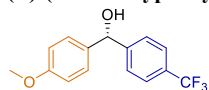


Known compound.²⁴ **Procedure A**, white solid, 34.3 mg, 59% yield, 95% ee. **Procedure B**, white solid, 34.2 mg, 59% yield, 91% ee. $R_f = 0.45$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. **¹H NMR** (500 MHz, Chloroform-*d*) δ 7.44 (d, $J = 8.5$ Hz, 2H), 7.27–7.21 (m, 4H), 6.86 (d, $J = 8.8$ Hz, 2H), 5.74 (s, 1H), 3.78 (s, 3H), 2.22 (s, 1H).



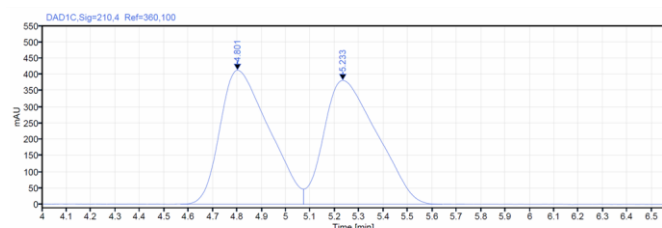
Supplementary Fig. 20 SFC chromatograms of **4n**. AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 7.7$ min, $t_{\text{major}} = 6.5$ min.

(R)-(4-methoxyphenyl)(4-(trifluoromethyl)phenyl)methanol (4o)



Known compound.²⁵ **Procedure A**, white solid, 40.9 mg, 73% yield, 94% ee. **Procedure B**, white solid, 40.5 mg, 72% yield, 96% ee. $R_f = 0.45$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.58 (d, $J = 8.2$ Hz, 2H), 7.48 (d, $J = 8.5$ Hz, 2H), 7.24 (d, $J = 8.5$ Hz, 2H), 6.86 (d, $J = 8.8$ Hz, 2H), 5.82 (s, 1H), 3.78 (s, 3H), 2.35 (s, 1H).

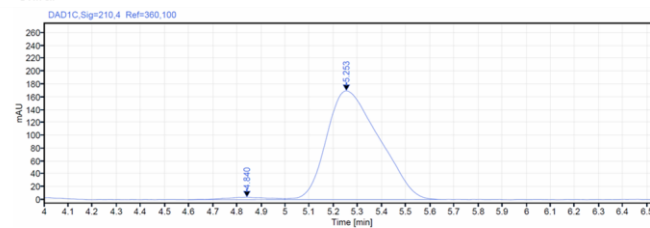
Racemic



Signal: DAD1C, Sig=210.4 Ref=360,100

RT [min]	Type	Width [min]	Area	Height	Area%
4.801	VV	0.50	5746.61	413.28	49.84
5.233	VV	0.57	5782.65	382.15	50.16
Sum			11529.26		

Chiral

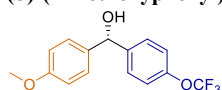


Signal: DAD1C, Sig=210.4 Ref=360,100

Name	RT [min]	Type	Width [min]	Area	Height	Area%
	4.840	BV	0.40	47.11	3.62	1.81
	5.253	VB	0.64	2556.47	169.28	98.19
Sum				2603.58		

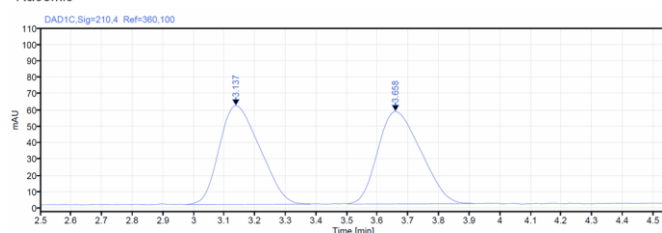
Supplementary Fig. 21 SFC chromatograms of **4o**. OJ-3 column, MeOH/CO₂ = 5:95, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 4.8$ min, $t_{\text{major}} = 5.3$ min.

(S)-(4-methoxyphenyl)(4-(trifluoromethoxy)phenyl)methanol (4p)



Procedure A, white solid, 45.1 mg, 76% yield, 93% ee. **Procedure B**, white solid, 44.7 mg, 75% yield, 95% ee. $R_f = 0.45$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.39 (d, $J = 8.5$ Hz, 2H), 7.26 (d, $J = 8.6$ Hz, 2H), 7.17 (d, $J = 8.1$ Hz, 2H), 6.87 (d, $J = 8.7$ Hz, 2H), 5.79 (s, 1H), 3.79 (s, 3H), 2.26 (s, 1H). $^{13}\text{C NMR}$ (126 MHz, Chloroform- d) δ 159.28, 148.37, 142.59, 135.68, 127.91, 127.74, 120.87, 120.45 (q, $J = 257.0$ Hz), 114.02, 75.10, 55.26. $^{19}\text{F NMR}$ (565 MHz, Chloroform- d) δ -57.85. **HRMS-APCI** (m/z): $[\text{M-OH}]^+$ calcd. for $[\text{C}_{15}\text{H}_{12}\text{F}_3\text{O}_2]^+$, 281.0784, found, 281.0788.

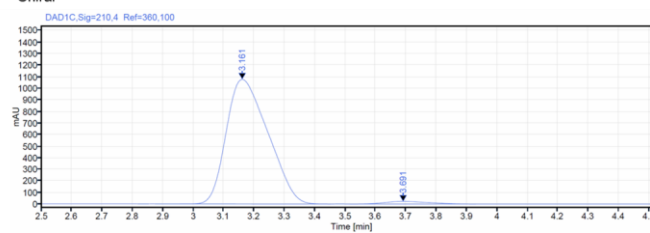
Racemic



Signal: DAD1C, Sig=210.4 Ref=360,100

RT [min]	Type	Width [min]	Area	Height	Area%
3.137	BV	0.41	555.34	60.38	50.19
3.658	VB	0.41	551.04	56.59	49.81
Sum			1106.38		

Chiral

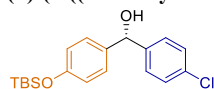


Signal: DAD1C, Sig=210.4 Ref=360,100

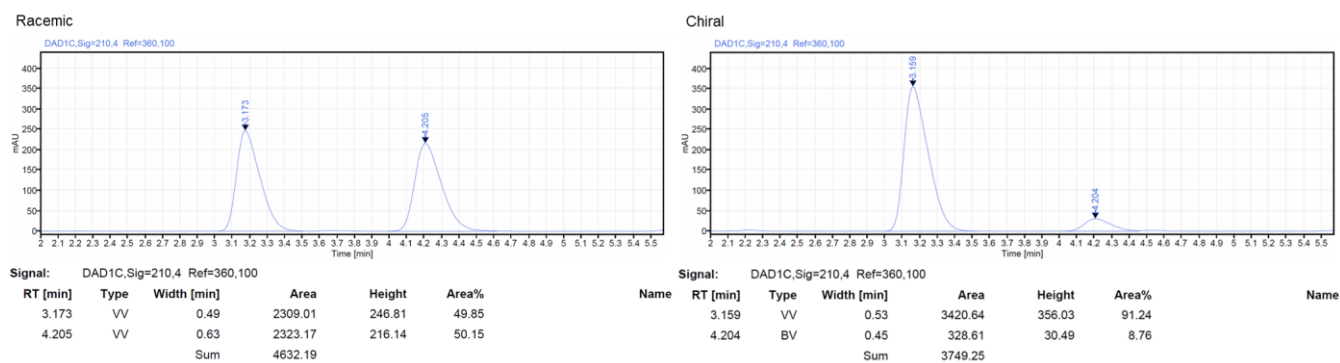
Name	RT [min]	Type	Width [min]	Area	Height	Area%
	3.161	BV	0.50	10256.46	1078.28	97.84
	3.691	BV	0.41	226.84	22.30	2.16
Sum				10483.31		

Supplementary Fig. 22 SFC chromatograms of **4p-Cr**. OJ-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 3.7$ min, $t_{\text{major}} = 3.2$ min.

(S)-4-((tert-butyldimethylsilyloxy)phenyl)(4-chlorophenyl)methanol (4q)

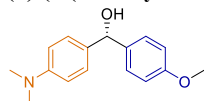


Procedure A, colorless oil, 31.6 mg, 45% yield, 82% ee. $R_f = 0.55$ (PE/EA = 6:1 v/v). Elution with PE/EA = 10:1 to 5:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.29 (s, 4H), 7.17 (d, $J = 8.2$ Hz, 2H), 6.79 (d, $J = 8.6$ Hz, 2H), 5.73 (d, $J = 3.3$ Hz, 1H), 2.25 (d, $J = 3.5$ Hz, 1H), 0.97 (s, 9H), 0.18 (s, 6H). $^{13}\text{C NMR}$ (126 MHz, Chloroform- d) δ 155.35, 142.40, 136.27, 133.06, 128.47, 127.85, 127.78, 120.11, 75.19, 25.62, 18.15, -4.45. **HRMS-ESI** (m/z): $[\text{M-OH}]^+$ calcd. for $[\text{C}_{19}\text{H}_{24}\text{ClOSi}]^+$, 331.1279, found, 331.1284.

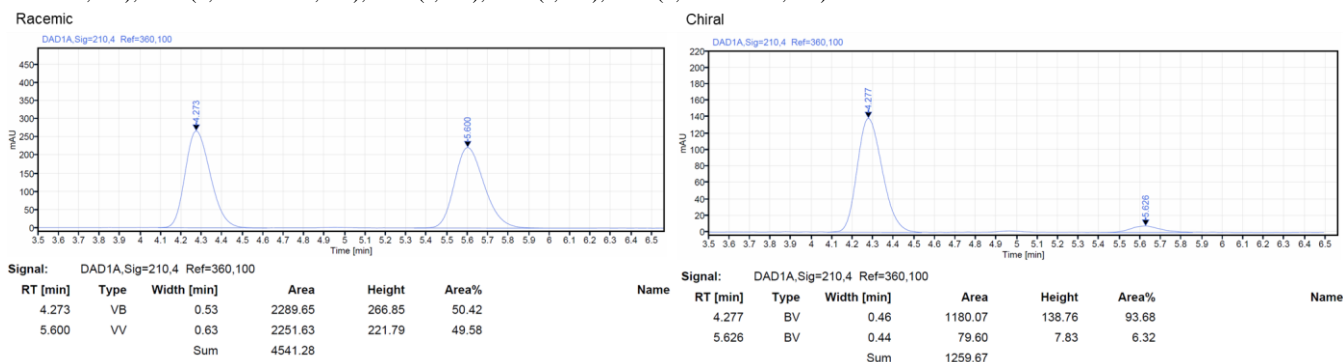


Supplementary Fig. 23 SFC chromatograms of **4q**. AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 254 nm, *t*_{minor} = 4.2 min, *t*_{major} = 3.2 min.

(S)-(4-(dimethylamino)phenyl)(4-methoxyphenyl)methanol (4r)

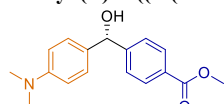


Known compound.²² **Procedure A**, step (1) with 0.22 mmol **2**, white solid, 37.3 mg, 72% yield, 87% ee. *R*_f = 0.25 (PE/EA = 4:1 v/v). Elution with PE/EA = 6:1 to 3:1 v/v. ¹H NMR (600 MHz, Chloroform-*d*) δ 7.29 (d, *J* = 8.4 Hz, 2H), 7.21 (d, *J* = 8.6 Hz, 2H), 6.86 (d, *J* = 8.8 Hz, 2H), 6.70 (d, *J* = 8.8 Hz, 2H), 5.74 (d, *J* = 2.6 Hz, 1H), 3.79 (s, 3H), 2.93 (s, 6H), 2.02 (d, *J* = 3.5 Hz, 1H).

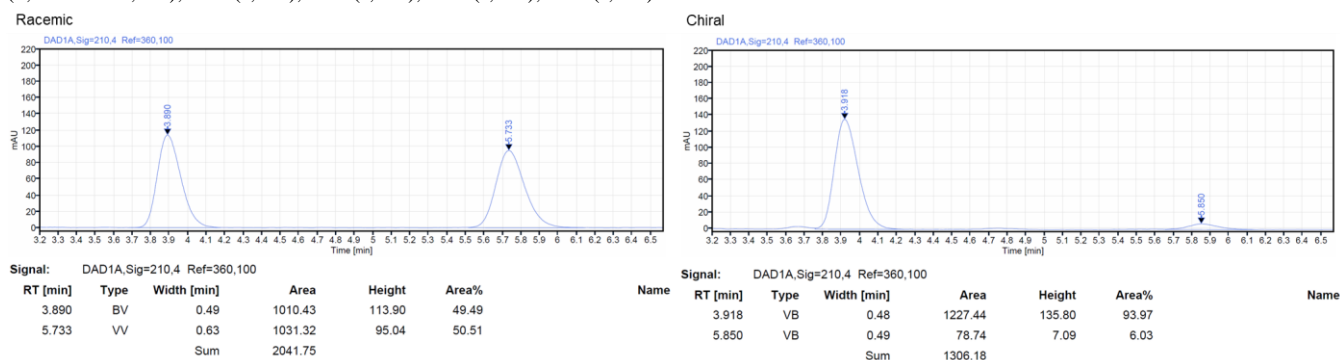


Supplementary Fig. 24 SFC chromatograms of **4r**. IH-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, *t*_{minor} = 5.6 min, *t*_{major} = 4.3 min.

methyl (R)-4-((4-(dimethylamino)phenyl)(hydroxy)methyl)benzoate (4s)

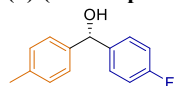


Known compound.²² **Procedure A**, step (1) with 0.22 mmol **2**, pale yellow solid, 44.1 mg, 77% yield, 87% ee. *R*_f = 0.25 (PE/EA = 4:1 v/v). Elution with PE/EA = 6:1 to 3:1 v/v. ¹H NMR (600 MHz, Chloroform-*d*) δ 7.99 (d, *J* = 8.1 Hz, 2H), 7.47 (d, *J* = 8.1 Hz, 2H), 7.18 (d, *J* = 8.5 Hz, 2H), 6.68 (d, *J* = 8.5 Hz, 2H), 5.80 (s, 1H), 3.89 (s, 3H), 2.93 (s, 6H), 2.20 (s, 1H).



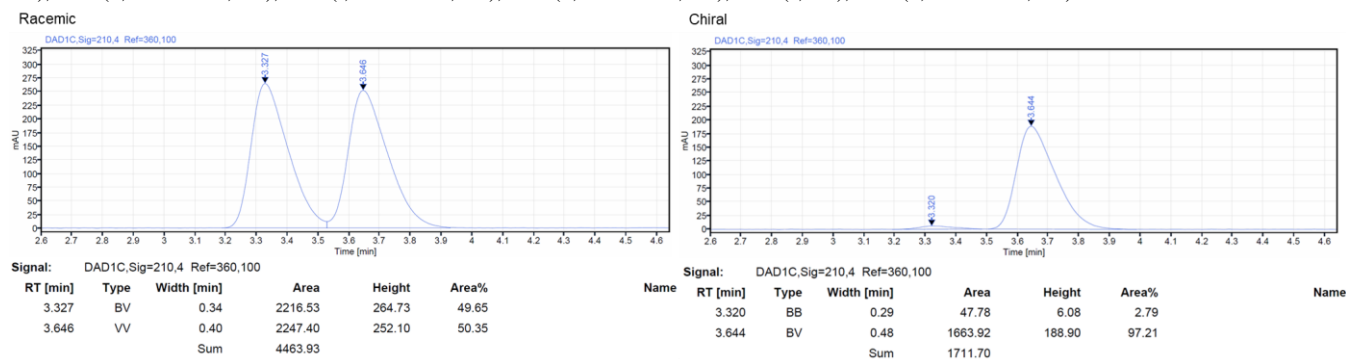
Supplementary Fig. 25 SFC chromatograms of **4s**. IH-3 column, MeOH/CO₂ = 15:85, 1.0 mL/min, 210 nm, *t*_{minor} = 5.8 min, *t*_{major} = 3.9 min.

(S)-(4-fluorophenyl)(p-tolyl)methanol (4t)



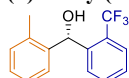
Known compound.²⁶ **Procedure A**, 32.8 mg, white solid, 76% yield, 87% ee. **Procedure B**, 31.5 mg, white solid, 73% yield, 94% ee. *R*_f = 0.55 (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.33 (dd, *J* = 8.3, 5.4 Hz, 2H), 7.23 (d, *J* = 8.1 Hz,

2H), 7.14 (d, $J = 7.6$ Hz, 2H), 7.00 (t, $J = 8.7$ Hz, 2H), 5.78 (d, $J = 3.4$ Hz, 1H), 2.33 (s, 3H), 2.21 (d, $J = 3.5$ Hz, 1H).

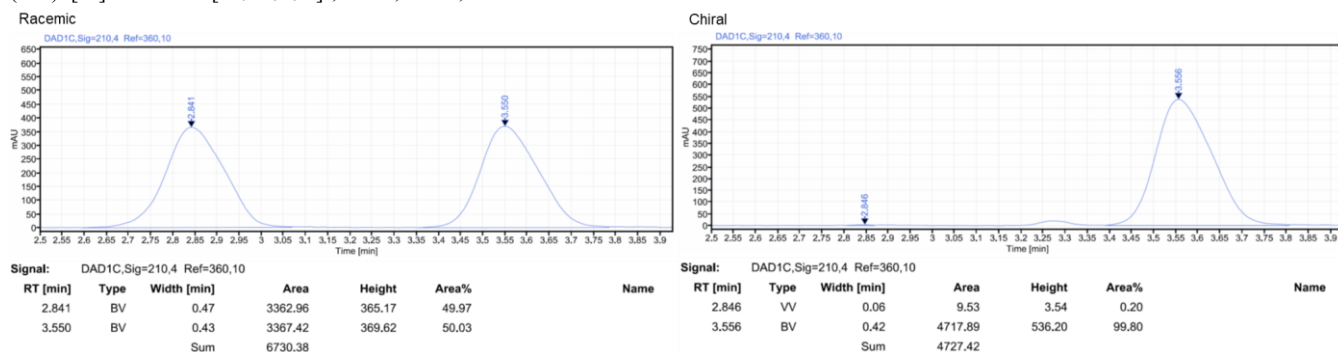


Supplementary Fig. 26 SFC chromatograms of **4t**. AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 3.3$ min, $t_{\text{major}} = 3.6$ min.

(S)-o-tolyl(2-(trifluoromethyl)phenyl)methanol (4u)

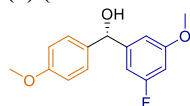


Procedure A, step (2) at room temperature, colorless oil, 44.1 mg, 83% yield, 99% ee. $R_f = 0.55$ (PE/EA = 6:1 v/v). Elution with PE/EA = 10:1 to 6:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.69 (d, $J = 7.3$ Hz, 1H), 7.53 – 7.44 (m, 2H), 7.44 – 7.36 (m, 2H), 7.27 – 7.18 (m, 2H), 7.17 – 7.12 (m, 1H), 6.39 (d, $J = 2.5$ Hz, 1H), 2.27 (d, $J = 3.5$ Hz, 1H), 2.14 (s, 3H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 141.24, 140.14, 135.38, 132.27, 130.59, 129.21, 128.11 (q, $J = 30.3$ Hz), 127.89, 127.71, 126.18, 126.10 (q, $J = 5.9$ Hz), 125.87, 124.53 (q, $J = 274.1$ Hz), 68.60, 19.11. GCMS-EI (m/z): [M]⁺ calcd. for [C₁₅H₁₃F₃O]⁺, 266.1, found, 266.1.

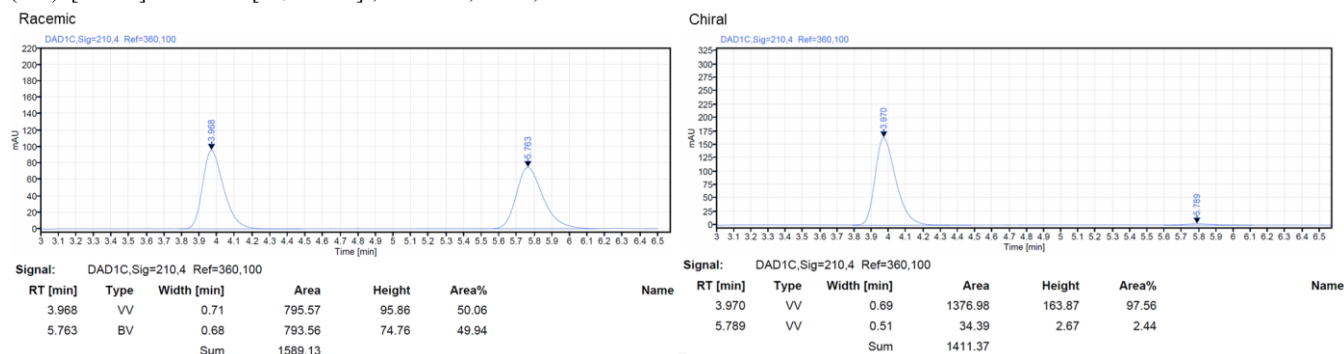


Supplementary Fig. 27 SFC chromatograms of **4u**. OD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 2.8$ min, $t_{\text{major}} = 3.6$ min.

(S)-(3-fluoro-5-methoxyphenyl)(4-methoxyphenyl)methanol (4v)

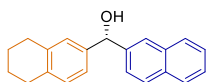


Procedure A, colorless oil, 38.7 mg, 74% yield, 82% ee. **Procedure B**, colorless oil, 36.5 mg, 70% yield, 95% ee. $R_f = 0.40$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.25 (d, $J = 8.5$ Hz, 2H), 6.86 (d, $J = 8.8$ Hz, 2H), 6.75 – 6.71 (m, 1H), 6.70 – 6.65 (m, 1H), 6.49 (dt, $J = 10.5, 2.3$ Hz, 1H), 5.71 (d, $J = 3.1$ Hz, 1H), 3.78 (s, 3H), 3.76 (s, 3H), 2.27 (d, $J = 3.4$ Hz, 1H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 163.58 (d, $J = 245.1$ Hz), 160.86 (d, $J = 11.3$ Hz), 159.25, 147.14 (d, $J = 8.8$ Hz), 135.50, 127.91, 113.98, 107.77 (d, $J = 2.6$ Hz), 105.52 (d, $J = 22.6$ Hz), 100.34 (d, $J = 25.3$ Hz), 75.31 (d, $J = 2.3$ Hz), 55.50, 55.26. ¹⁹F NMR (471 MHz, CDCl₃) δ -111.37. HRMS-APCI (m/z): [M-OH]⁺ calcd. for [C₁₅H₁₄FO₂]⁺, 245.0972, found, 245.0987.

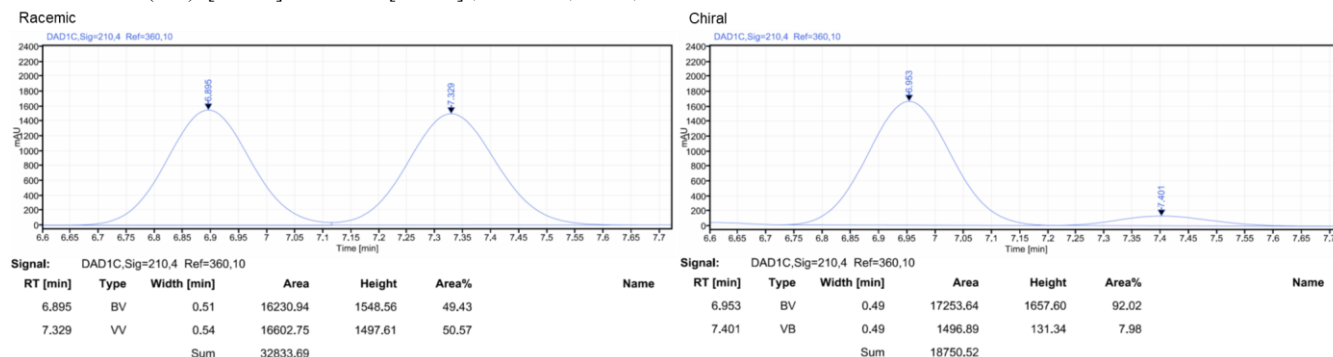


Supplementary Fig. 28 SFC chromatograms of **4v**. AD-3 column, MeOH/CO₂ = 10:90, 1.5 mL/min, 210 nm, $t_{\text{minor}} = 5.8$ min, $t_{\text{major}} = 4.0$ min.

(S)-naphthalen-2-yl(5,6,7,8-tetrahydronaphthalen-2-yl)methanol (4w)

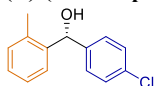


Procedure A, 48.9 mg, white solid, 85% yield, 84% ee. $R_f = 0.55$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.90 (s, 1H), 7.85–7.82 (m, 1H), 7.81–7.76 (m, 2H), 7.49–7.40 (m, 3H), 7.12–7.08 (m, 2H), 7.02 (d, $J = 8.4$ Hz, 1H), 5.92 (d, $J = 3.3$ Hz, 1H), 2.73 (d, $J = 3.2$ Hz, 4H), 2.28 (d, $J = 3.5$ Hz, 1H), 1.76 (dt, $J = 6.7, 3.5$ Hz, 4H). $^{13}\text{C NMR}$ (126 MHz, Chloroform- d) δ 141.28, 140.84, 137.36, 136.71, 133.25, 132.80, 129.32, 128.19, 128.05, 127.63, 127.38, 126.08, 125.82, 124.78, 124.74, 123.92, 76.27, 29.43, 29.12, 23.14, 23.12. **HRMS-APCI** (m/z): [M-OH] $^+$ calcd. for [C₂₁H₁₉] $^+$, 271.1481, found, 271.1482.

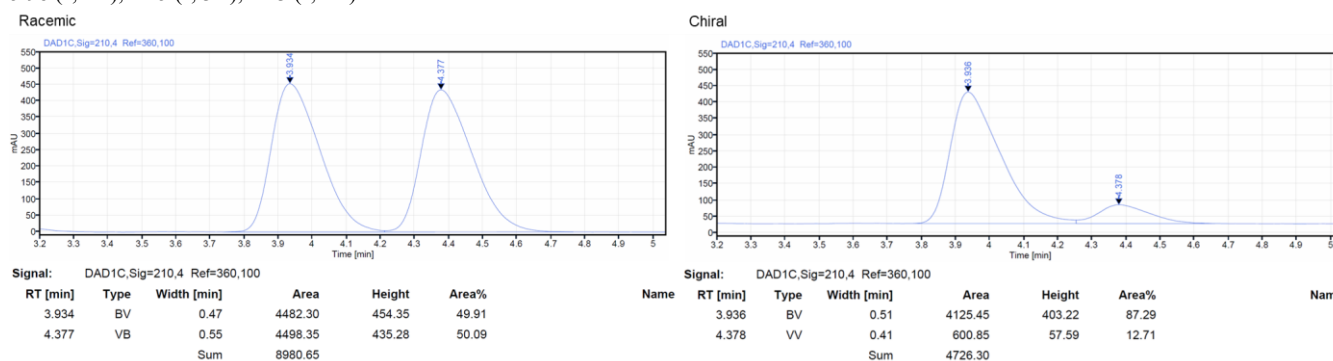


Supplementary Fig. 29 SFC chromatograms of **4w**. IC-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 7.4$ min, $t_{\text{major}} = 6.9$ min.

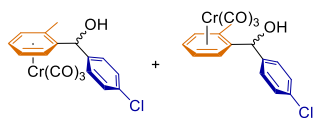
(R)-4-chlorophenyl(o-tolyl)methanol (4x)



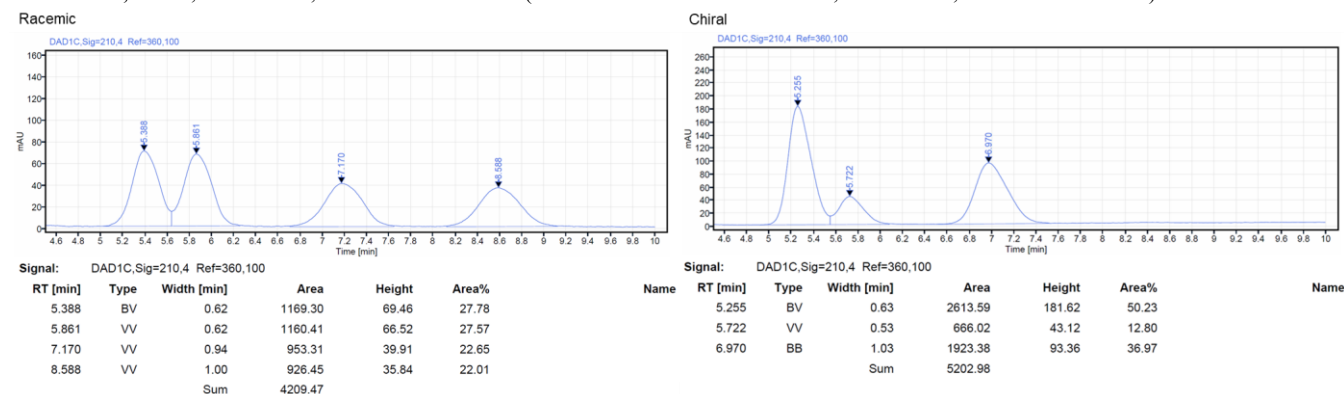
Known compound.²⁷ **Procedure A**, white solid, 29.7 mg, 64% yield, 74% ee. $R_f = 0.55$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.45 (dd, $J = 7.3, 1.6$ Hz, 1H), 7.32–7.25 (m, 4H), 7.22 (qd, $J = 7.2, 1.7$ Hz, 2H), 7.15 (dd, $J = 7.2, 1.5$ Hz, 1H), 5.98 (s, 1H), 2.25 (s, 3H), 2.13 (s, 1H).



Supplementary Fig. 30 SFC chromatograms of **4x**. AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 4.4$ min, $t_{\text{major}} = 3.9$ min.

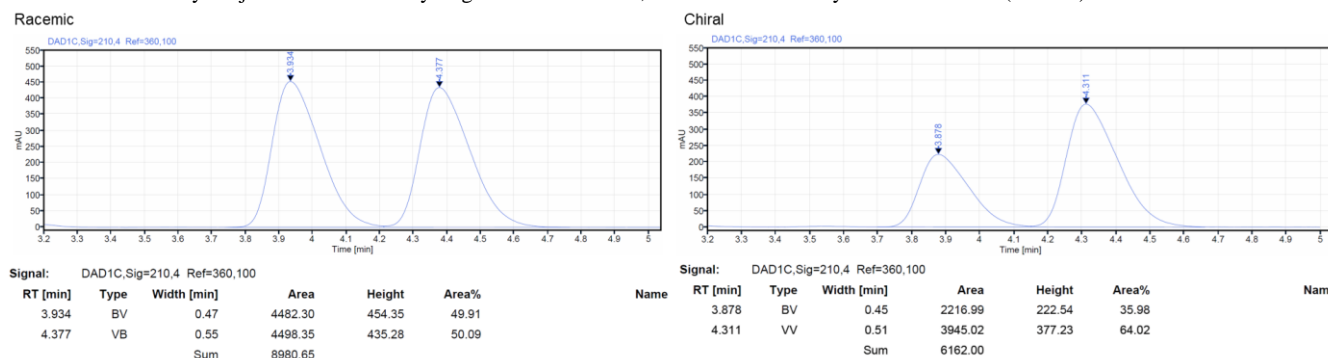


Procedure A, **4x-Cr**, 50.23:49.77, >20:1 dr and 2.9:1 dr. (Racemate from NaBH₄ reduction, 50.43:49.58, 1.2:1 dr and 1:1.2 dr)



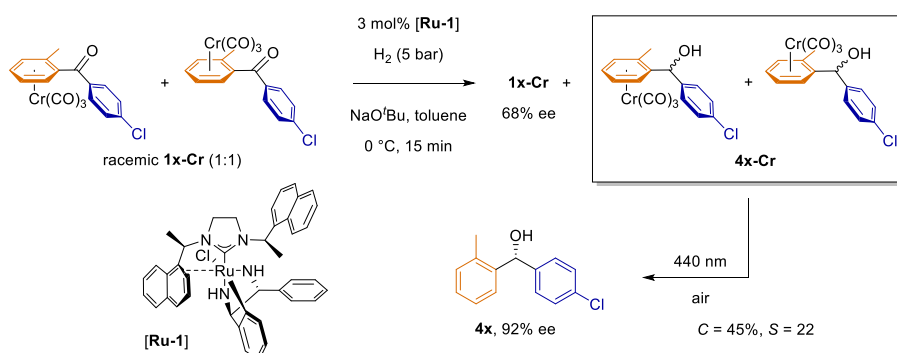
Supplementary Fig. 31 SFC chromatograms of **4x-Cr**. AS-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 8.5$ min, $t_{\text{major}} = 5.3$ min and $t_{\text{minor}} = 5.7$ min, $t_{\text{major}} = 7.0$ min.

When **1x** was directly subjected to the same hydrogenation conditions, the enantioselectivity was much lower (28% ee).

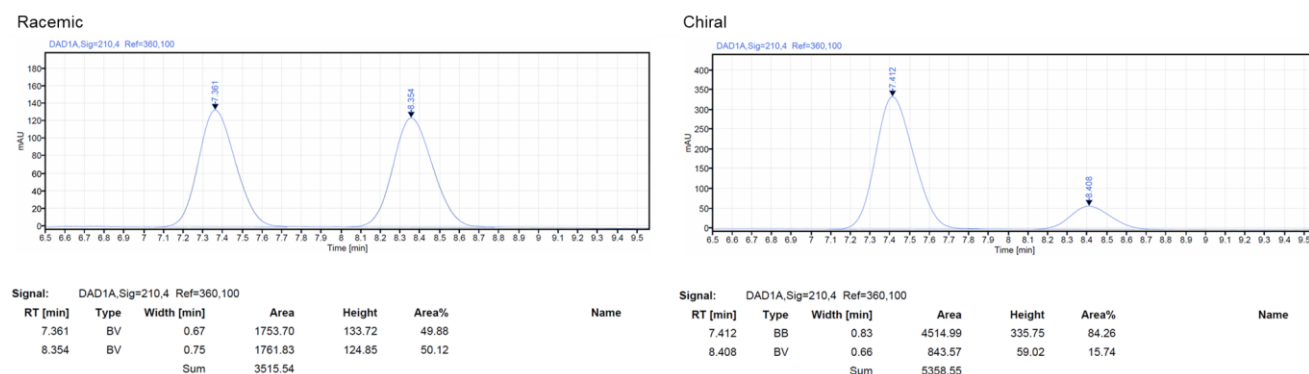


Supplementary Fig. 32 SFC chromatograms of **4x** (directly hydrogenation from **1x**). AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, t_{major} = 3.9 min, t_{minor} = 4.4 min.

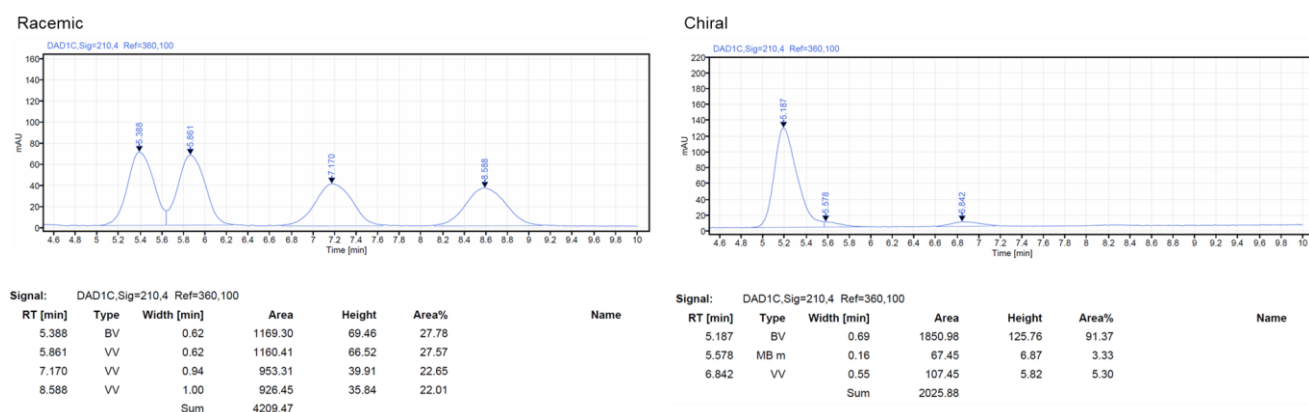
The kinetic resolution of **1x-Cr** can be achieved by shortening the reaction time. After 15 minutes of reaction at 0 °C, the conversion of the reaction was 45%, and two pair of diastereomers (**4x-Cr**, 91.37:8.63) with > 20:1 dr and 1.6:1 dr were obtained respectively. The recovered **1x-Cr** was also obtained with 68% ee. After removing the Cr(CO)₃ unit, we obtained the free alcohol **4x** with 92% ee.



Supplementary Fig. 33 Kinetic resolution of **1x-Cr**. Conversion (*C*). *S* factor (*S*) = $\ln[(1 - C)(1 - ee_s)] / \ln[(1 - C)(1 + ee_s)]$.

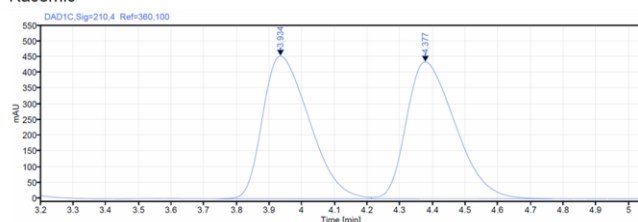


Supplementary Fig. 34 SFC chromatograms of recovered **1x-Cr** (kinetic resolution of **1x-Cr**). OD-3 column, MeOH/CO₂ = 5:95, 1.0 mL/min, 210 nm, t_{major} = 7.4 min, t_{minor} = 8.4 min.

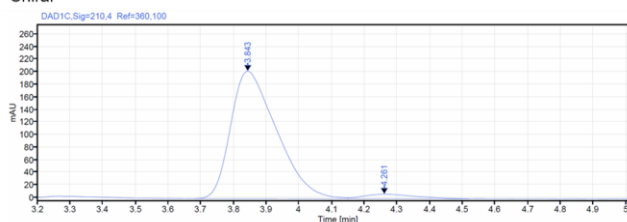


Supplementary Fig. 35 SFC chromatograms of 4x-Cr (kinetic resolution of 1x-Cr). AS-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 8.5$ min, $t_{\text{major}} = 5.2$ min and $t_{\text{minor}} = 5.6$ min, $t_{\text{major}} = 6.8$ min. (Racemate from NaBH₄ reduction, 50.43:49.58, 1.2:1 dr and 1:1.2 dr)

Racemic

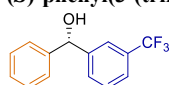


Chiral



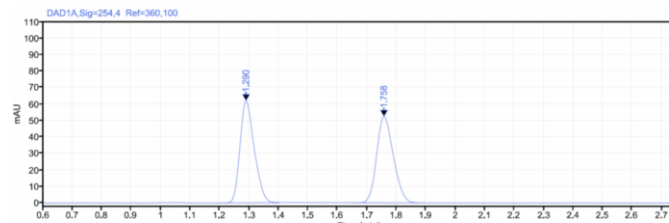
Supplementary Fig. 36 SFC chromatograms of 4x (kinetic resolution of 1x-Cr). AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{major}} = 3.8$ min, $t_{\text{minor}} = 4.3$ min.

(S)-phenyl(3-(trifluoromethyl)phenyl)methanol (4y)

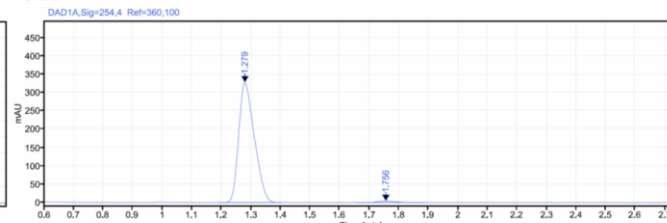


Known compound.²⁸ **Procedure B**, white solid, 34.9 mg, 69% yield, 96% ee. $R_f = 0.55$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.70 (s, 1H), 7.53 (t, $J = 8.0$ Hz, 2H), 7.44 (t, $J = 7.7$ Hz, 1H), 7.36 (d, $J = 4.4$ Hz, 4H), 7.34–7.26 (m, 1H), 5.87 (d, $J = 3.4$ Hz, 1H), 2.34 (d, $J = 3.5$ Hz, 1H).

Racemic

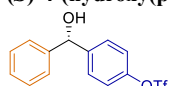


Chiral



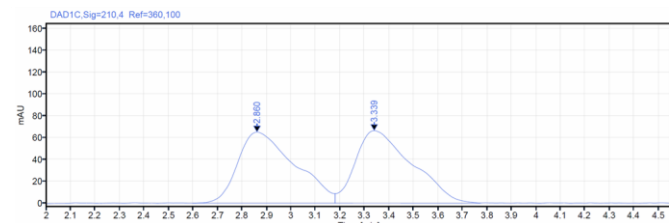
Supplementary Fig. 37 SFC chromatograms of 4y-Cr. OD-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 254 nm, $t_{\text{minor}} = 1.8$ min, $t_{\text{major}} = 1.3$ min.

(S)-4-(hydroxy(phenyl)methyl)phenyl trifluoromethanesulfonate (1z)

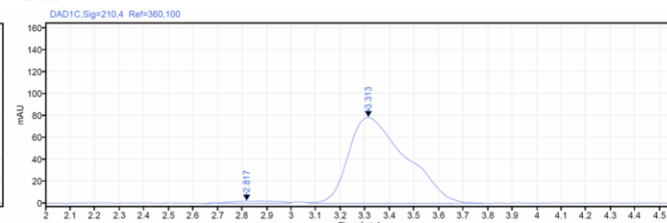


Procedure B, white solid, 51.6 mg, 78% yield, 94% ee. $R_f = 0.45$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.46 (d, $J = 8.4$ Hz, 2H), 7.39–7.27 (m, 5H), 7.23 (d, $J = 8.8$ Hz, 2H), 5.84 (d, $J = 3.2$ Hz, 1H), 2.35 (d, $J = 3.4$ Hz, 1H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 148.68, 144.10, 143.03, 128.79, 128.28, 128.13, 126.59, 121.27, 118.71 (q, $J = 320.7$ Hz), 75.38. ¹⁹F NMR (471 MHz, Chloroform-*d*) δ -72.84. **HRMS** (m/z): [M-OH]⁺ calcd. for [C₁₄H₁₀F₃O₃S]⁺, 315.0297 found, 315.0308.

Racemic

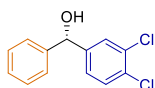


Chiral



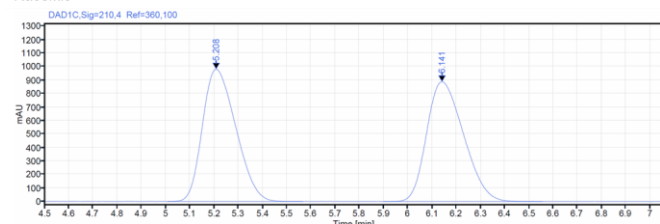
Supplementary Fig. 38 SFC chromatograms of 4z. AD-3 column, MeOH/CO₂ = 5:95, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 2.8$ min, $t_{\text{major}} = 3.3$ min.

(S)-(3,4-dichlorophenyl)(phenyl)methanol (4aa)



Known compound.¹⁶ **Procedure B**, white solid, 40.4 mg, 80% yield, 97% ee. $R_f = 0.50$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.47 (d, $J = 1.8$ Hz, 1H), 7.38–7.26 (m, 6H), 7.15 (dd, $J = 8.3, 1.8$ Hz, 1H), 5.71 (s, 1H), 2.53 (s, 1H).

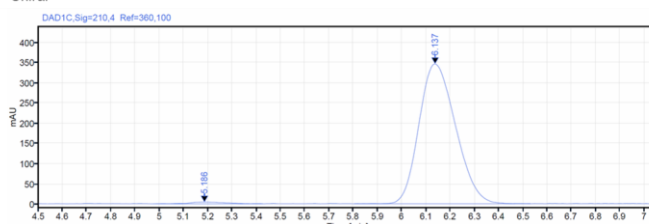
Racemic



Signal: DAD1C, Sig=210.4 Ref=360,100

RT [min]	Type	Width [min]	Area	Height	Area%
5.208	VV	0.56	9386.00	981.96	49.92
6.141	BB	0.66	9414.70	889.01	50.08
Sum			18800.70		

Chiral

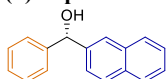


Signal: DAD1C, Sig=210.4 Ref=360,100

Name	RT [min]	Type	Width [min]	Area	Height	Area%	Name
	5.186	BB	0.45	45.52	4.31	1.24	
	6.137	BB	0.73	3626.13	345.94	98.76	
	Sum			3671.65			

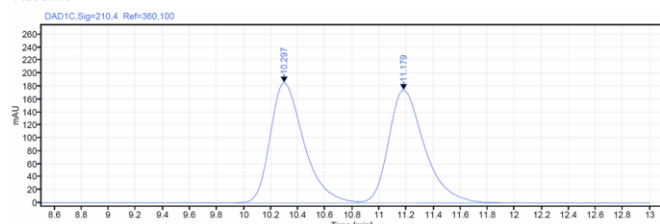
Supplementary Fig. 39 SFC chromatograms of **4aa**. OJ-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 5.2$ min, $t_{\text{major}} = 6.1$ min.

(*S*)-naphthalen-2-yl(phenyl)methanol (**4ab**)



Known compound.¹⁵ **Procedure B**, white solid, 35.1 mg, 75% yield, 92% ee. $R_f = 0.50$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.88 (s, 1H), 7.84–7.76 (m, 3H), 7.49–7.43 (m, 2H), 7.43–7.40 (m, 3H), 7.35–7.31 (m, 2H), 7.28–7.24 (m, 1H), 5.98 (d, $J = 3.2$ Hz, 1H), 2.36 (d, $J = 3.5$ Hz, 1H).

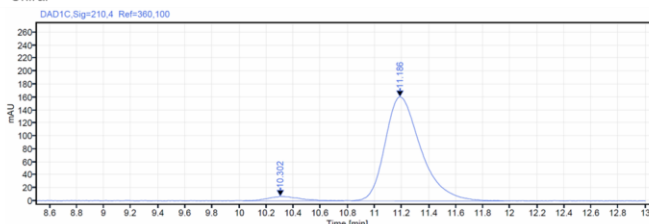
Racemic



Signal: DAD1C, Sig=210.4 Ref=360,100

RT [min]	Type	Width [min]	Area	Height	Area%
10.297	BV	0.92	3205.84	185.69	49.93
11.179	VV	1.11	3215.07	174.15	50.07
Sum			6420.92		

Chiral

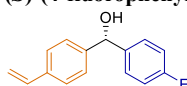


Signal: DAD1C, Sig=210.4 Ref=360,100

Name	RT [min]	Type	Width [min]	Area	Height	Area%	Name
	10.302	VV	0.75	120.47	7.06	3.93	
	11.186	VV	1.13	2946.42	160.50	96.07	
	Sum			3066.89			

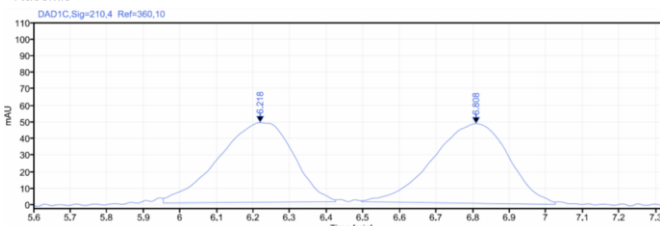
Supplementary Fig. 40 SFC chromatograms of **4ab**. AD-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 10.3$ min, $t_{\text{major}} = 11.2$ min.

(*S*)-(4-fluorophenyl)(4-vinylphenyl)methanol (**4ac**)



Procedure B, white solid, 28.1 mg, 62% yield, 93% ee. $R_f = 0.55$ (PE/EA = 4:1 v/v). Elution with PE/EA = 10:1 to 4:1 v/v. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.38 (d, $J = 8.2$ Hz, 2H), 7.34–7.29 (m, 4H), 7.01 (t, $J = 8.7$ Hz, 2H), 6.69 (dd, $J = 17.6, 10.9$ Hz, 1H), 5.80 (d, $J = 2.4$ Hz, 1H), 5.73 (dd, $J = 17.6, 0.8$ Hz, 1H), 5.24 (dd, $J = 10.9, 0.8$ Hz, 1H), 2.26 (d, $J = 3.2$ Hz, 1H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 162.16 (d, $J = 245.9$ Hz), 143.14, 139.44 (d, $J = 3.1$ Hz), 137.07, 136.31, 128.19 (d, $J = 8.1$ Hz), 126.62, 126.40, 115.30 (d, $J = 21.5$ Hz), 114.09, 75.35. ¹⁹F NMR (471 MHz, CDCl₃) δ -114.95. GCMS-EI (m/z): [M]⁺ calcd. for [C₁₅H₁₃FO]⁺, 228.1, found, 228.1.

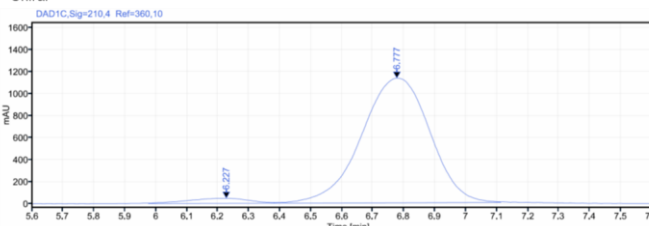
Racemic



Signal: DAD1C, Sig=210.4 Ref=360,10

RT [min]	Type	Width [min]	Area	Height	Area%
6.218	VV	0.47	685.72	48.05	49.71
6.808	VV	0.53	693.68	48.01	50.29
Sum			1379.40		

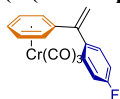
Chiral



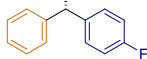
Signal: DAD1C, Sig=210.4 Ref=360,10

Name	RT [min]	Type	Width [min]	Area	Height	Area%	Name
	6.227	VV	0.41	619.94	45.23	3.44	
	6.777	VV	0.73	17384.63	1133.96	96.56	
	Sum			18004.57			

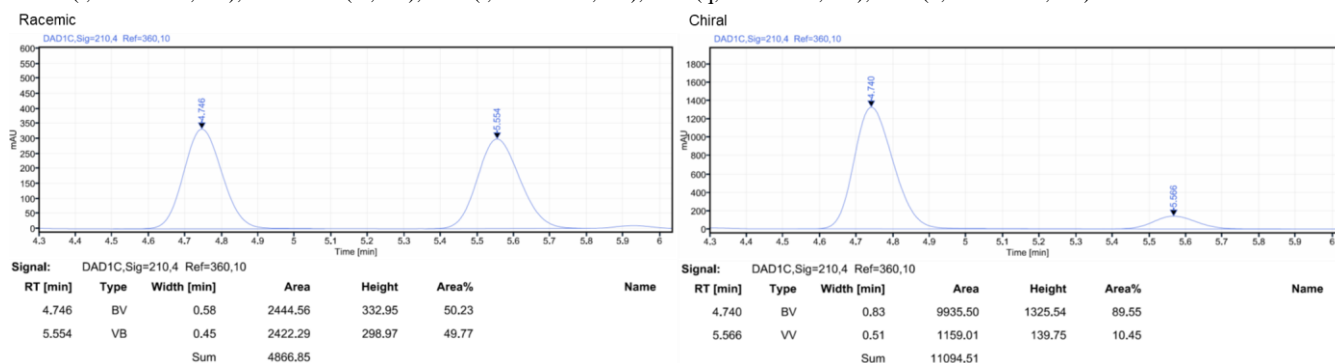
Supplementary Fig. 41 SFC chromatograms of **4ac**. IC-3 column, MeOH/CO₂ = 10:90, 1.0 mL/min, 202 nm, $t_{\text{minor}} = 6.2$ min, $t_{\text{major}} = 6.8$ min.

(1-(4-fluorophenyl)vinyl)benzene chromium tricarbonyl (3a-Cr)

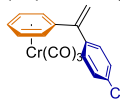
Yellow solid. 54.7 mg, 82% yield. The regioisomeric ratio was > 10:1. R_f = 0.40 (PE/EA = 20:1 v/v). Elution with PE/EA = 30:1 to 20:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.35 (t, J = 6.9 Hz, 2H), 7.07 (t, J = 8.5 Hz, 2H), 5.69 (s, 1H), 5.50–5.25 (m, 6H). ^{13}C NMR (126 MHz, Chloroform- d) δ 232.55, 162.73 (d, J = 247.8 Hz), 144.40, 135.45 (d, J = 3.4 Hz), 130.15 (d, J = 8.1 Hz), 117.31, 115.39 (d, J = 21.6 Hz), 108.45, 93.25, 92.54, 91.28. ^{19}F NMR (471 MHz, Chloroform- d) δ -113.48. HRMS-APCI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{12}\text{CrFO}_3]^+$, 335.0170, found, 335.0166.

(S)-1-fluoro-4-(1-phenylethyl)benzene (5a)

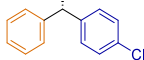
Known compound.¹² **Procedure A**, colorless oil, 32.7 mg, 82% yield, 79% ee. R_f = 0.80 (PE). Elution with PE. ^1H NMR (500 MHz, Chloroform- d) δ 7.28 (t, J = 7.5 Hz, 1H), 7.22–7.14 (m, 3H), 6.96 (t, J = 8.4 Hz, 1H), 4.13 (q, J = 7.1 Hz, 1H), 1.62 (d, J = 7.2 Hz, 2H).



Supplementary Fig. 42 SFC chromatograms of 5a-Cr. OJ-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 210 nm, t_{minor} = 5.6 min, t_{major} = 4.7 min.

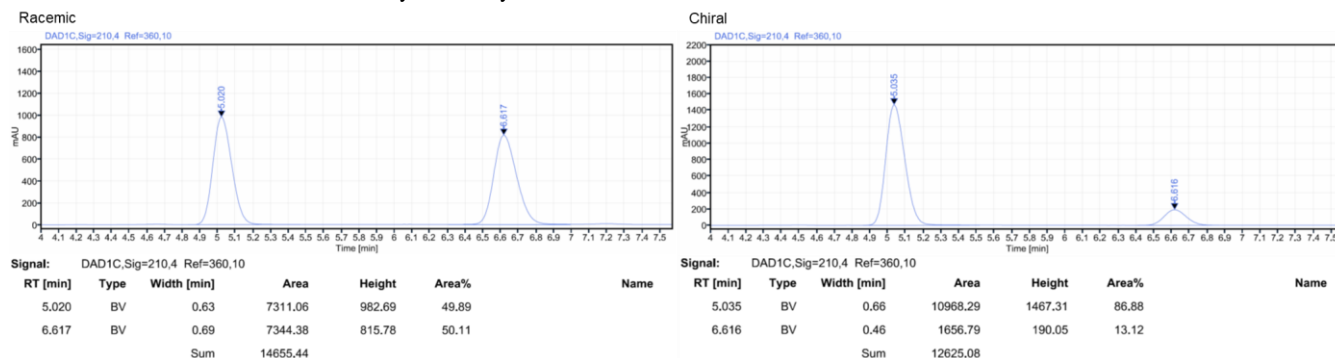
(1-(4-chlorophenyl)vinyl)benzene chromium tricarbonyl (3b-Cr)

Yellow solid. 56.7 mg, 81% yield. The regioisomeric ratio was > 10:1. R_f = 0.40 (PE/EA = 20:1 v/v). Elution with PE/EA = 30:1 to 20:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.35 (d, J = 8.3 Hz, 2H), 7.32 (d, J = 8.1 Hz, 2H), 5.70 (s, 1H), 5.46 (s, 1H), 5.41 (d, J = 6.1 Hz, 2H), 5.37 (d, J = 5.7 Hz, 1H), 5.32 (t, J = 5.8 Hz, 2H). ^{13}C NMR (126 MHz, Chloroform- d) δ 232.48, 144.34, 137.87, 134.33, 129.74, 128.66, 117.59, 108.13, 93.22, 92.53, 91.24. HRMS-APCI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{17}\text{H}_{12}\text{ClCrO}_3]^+$, 350.9875, found, 350.9886.

(S)-1-chloro-4-(1-phenylethyl)benzene (5b)

Known compound.¹² **Procedure A**, colorless oil, 35.0 mg, 81% yield, 73% ee. R_f = 0.80 (PE). Elution with PE. ^1H NMR (500 MHz, Chloroform- d) δ 7.28 (t, J = 7.6 Hz, 1H), 7.26–7.22 (m, 1H), 7.21–7.17 (m, 2H), 7.14 (d, J = 8.4 Hz, 1H), 4.12 (q, J = 7.2 Hz, 1H), 1.61 (d, J = 7.2 Hz, 1H).

The enantiomeric excess was determined by SFC analysis:



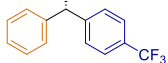
Supplementary Fig. 43 SFC chromatograms of 5b. OJ-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 210 nm, t_{minor} = 6.6 min, t_{major} = 5.0 min.

(1-(4-(trifluoromethyl)phenyl)vinyl)benzene chromium tricarbonyl (3c-Cr)

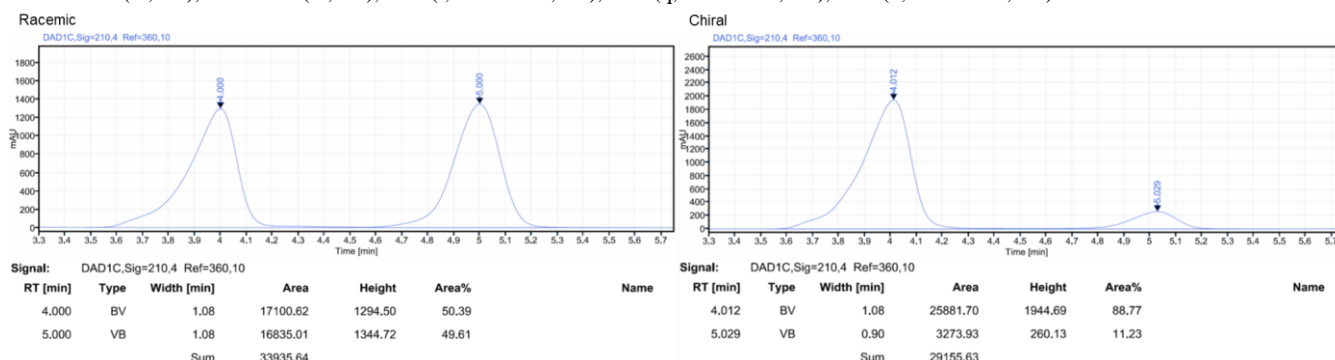


Yellow solid. 62.5 mg, 81% yield. The regioisomeric ratio was > 10:1. R_f = 0.40 (PE/EA = 20:1 v/v). Elution with PE/EA = 30:1 to 20:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.65 (d, J = 7.9 Hz, 2H), 7.51 (d, J = 8.0 Hz, 2H), 5.78 (s, 1H), 5.51 (s, 1H), 5.37 (d, J = 24.5 Hz, 5H). ^{13}C NMR (126 MHz, Chloroform- d) δ 232.46, 144.47, 143.14, 130.56 (q, J = 32.5 Hz), 128.93, 125.56 (q, J = 3.9 Hz), 124.09 (q, J = 272.2 Hz), 118.52, 107.61, 93.13, 92.62, 91.30. ^{19}F NMR (565 MHz, Chloroform- d) δ -62.64. HRMS-APCI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{18}\text{H}_{12}\text{CrF}_3\text{O}_3]^+$, 385.0138, found, 385.0142.

(S)-1-(1-phenylethyl)-4-(trifluoromethyl)benzene (5c)



Known compound.¹² **Procedure A**, colorless oil, 40.7 mg, 81% yield, 77% ee. R_f = 0.80 (PE). Elution with PE. ^1H NMR (500 MHz, Chloroform- d) δ 7.31–7.24 (m, 2H), 7.24–7.12 (m, 5H), 6.95 (t, J = 8.7 Hz, 2H), 4.12 (q, J = 7.2 Hz, 1H), 1.61 (d, J = 7.2 Hz, 3H).



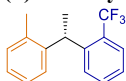
Supplementary Fig. 44 SFC chromatograms of 5c-Cr. OJ-3 column, MeOH/CO₂ = 7:93, 1.0 mL/min, 210 nm, t_{minor} = 5.0 min, t_{major} = 4.0 min.

2-(1-(2-(trifluoromethyl)phenyl)vinyl)toluene chromium tricarbonyl (3d-Cr)



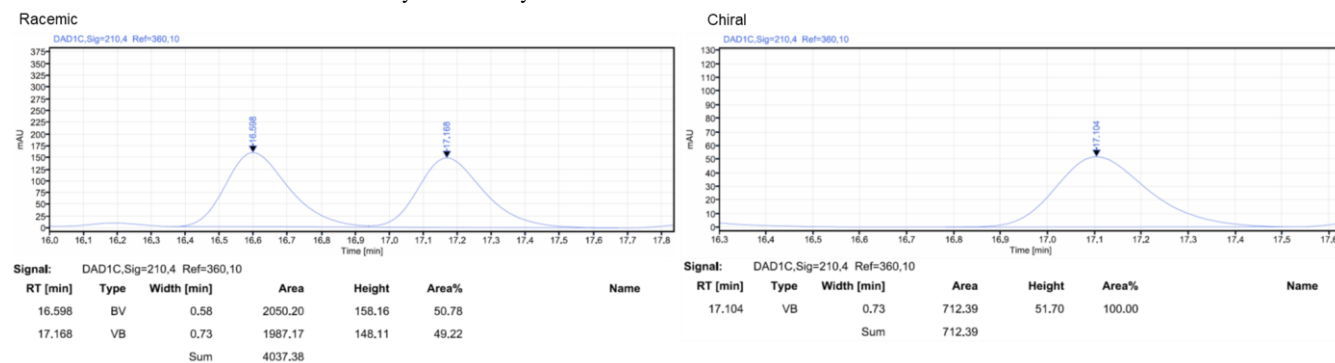
Yellow solid. 67.5 mg, 85% yield. The regioisomeric ratio was > 10:1. R_f = 0.45 (PE/EA = 20:1 v/v). Elution with PE/EA = 30:1 to 20:1 v/v. ^1H NMR (500 MHz, Chloroform- d) δ 7.68 (d, J = 7.7 Hz, 1H), 7.56 (t, J = 7.4 Hz, 1H), 7.43 (t, J = 7.8 Hz, 2H), 5.80 (s, 1H), 5.67 (s, 1H), 5.53–5.38 (m, 2H), 5.11 (t, J = 6.1 Hz, 2H), 2.13 (s, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 232.98, 142.44, 139.93, 132.13, 131.86, 128.20, 127.87 (q, J = 30.2 Hz), 126.57 (q, J = 5.6 Hz), 125.12, 123.92 (q, J = 274.0 Hz), 111.10, 109.36, 98.03, 94.87, 92.05, 88.13, 20.35. ^{19}F NMR (565 MHz, Chloroform- d) δ -57.00. HRMS-APCI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{19}\text{H}_{14}\text{CrF}_3\text{O}_3]^+$, 399.0295, found, 399.0286.

(S)-1-methyl-2-(1-(2-(trifluoromethyl)phenyl)ethyl)benzene (5d)



Procedure A, step (2) at room temperature, colorless oil, 45.1 mg, 85% yield, 99% ee. R_f = 0.80 (PE). Elution with PE. ^1H NMR (500 MHz, Chloroform- d) δ 7.65 (d, J = 7.6 Hz, 1H), 7.42–7.34 (m, 2H), 7.29–7.24 (m, 2H), 7.25–7.22 (m, 1H), 7.16 (td, J = 7.4, 1.2 Hz, 1H), 7.12 (d, J = 7.4 Hz, 2H), 4.68 (q, J = 7.0 Hz, 1H), 2.10 (s, 3H), 1.62 (d, J = 7.0 Hz, 3H). ^{13}C NMR (126 MHz, Chloroform- d) δ 145.39 (q, J = 1.7 Hz), 142.86, 136.67, 131.98, 130.74, 129.04, 127.94 (q, J = 29.6 Hz), 126.63, 126.45, 126.11 (q, J = 6.1 Hz), 125.96, 125.76, 124.79 (q, J = 274.1 Hz), 37.35, 22.57, 19.73. ^{19}F NMR (471 MHz, Chloroform- d) δ -58.95. GCMS-EI (m/z): $[\text{M}]^+$ calcd. for $[\text{C}_{16}\text{H}_{15}\text{F}_3]^+$, 264.1, found, 264.1.

The enantiomeric excess was determined by HPLC analysis:



Supplementary Fig. 45 SFC chromatograms of 5d-Cr. IH-3 + IBN-3 column, MeOH/hexane = 2:98, 1.0 mL/min, 210 nm, $t_{\text{minor}} = 16.6$ min, $t_{\text{major}} = 17.1$ min.

4 Computational Studies

Density Functional Theory (DFT) calculation of the thermodynamic data was conducted with Gaussian 16 C.01²⁹ and ORCA 5.0.3^{30, 31}. Structures were first optimized at PBE0-D3(BJ)/def2SVP³²⁻³⁵ level with SMD³⁶ implicit solvent model using toluene as solvent in Gaussian. Vibrational analysis was done at the optimization level, with the result that no imaginary vibration frequency for substrates and products, and only one imaginary frequency for the transition states were given. Free energy correction of the optimized structures was obtained by directly reading from the output file of vibrational analysis. The transition states were further checked by instinct reaction coordinates calculation. Single point energy was calculated in ORCA, at PWPB95-D3(BJ)/def2-QZVPP^{32, 34, 35, 37} level, still using SMD implicit solvent model and toluene as solvent. The free energies were obtained by directly summing of the single point energy and the free energy correction. (Cartesian coordinates of optimized structure see Supplementary Data 1)

To get structural and energetical understanding of the impact to rotation of aryl-carbonyl C–C bond with or without $\text{Cr}(\text{CO})_3$ ligation, relaxed scan of dihedral angles calculation were taken at the optimization level. The relaxed scan of dihedral angles calculation proceeded with 6° stepsize, and 30 steps. The output electronic energy was directly used. (For detailed calculation data, see Supplementary Data 2)

To give explanation of the enantioselectivity of the hydrogenation, and what role that the coordinated $\text{Cr}(\text{CO})_3$ part play, Independent gradient model based on Hirshfeld partition (IGMH)³⁸ analysis of the transition states were taken. This analysis was taken with Multiwfn³⁹, by using the wavefunction after optimization from the .chk file (transformed from .chk). Molecular structure and IGMH isosurface were drawn with VMD⁴⁰, using the gird data in the .cub file generated by Multiwfn. Atomic charge population and the conceptual DFT calculation were also conducted with Multiwfn.

Supplementary Table 1 Distortion and interaction energies (kcal/mol) at PBE0-D3(BJ)/def2SVP level.

Entry	TS	$\Delta E_{\text{dist-sub}}$	$\Delta E_{\text{dist-cat}}$	ΔE_{dist}	ΔE_{int}
1	TS-3	9.0	3.8	12.8	-13.4
2	TS-4	6.9	2.5	9.4	-12.2

We also used activation strain model^{41, 42} to analyze activation barriers that determine reaction rates. the transition state energy depends upon ΔE_{dist} and ΔE_{int} . By definition, $\Delta E^\ddagger = \Delta E_{\text{dist}} + \Delta E_{\text{int}}$; therefore, since ΔE_{int} and $\Delta E_{\text{dist-cat}}$ are approximately the same, as in **TS-3** and **TS-4**, the distortion energies of the substrate become the determining factor, which is consistent with the results of relaxed scan of dihedral angles calculation.

Supplementary Table 2 Atomic charge population.

Entry	Atom	Hirshfeld Charge	ADCH Charge	CM5 Charge
1	C1 (1a)	0.1520	0.2304	0.1853
2	O1 (1a)	-0.2507	-0.3179	-0.3009
3	C1 (1a-Cr)	0.1591	0.2505	0.1910
4	O1 (1a-Cr)	-0.2382	-0.2961	-0.2894

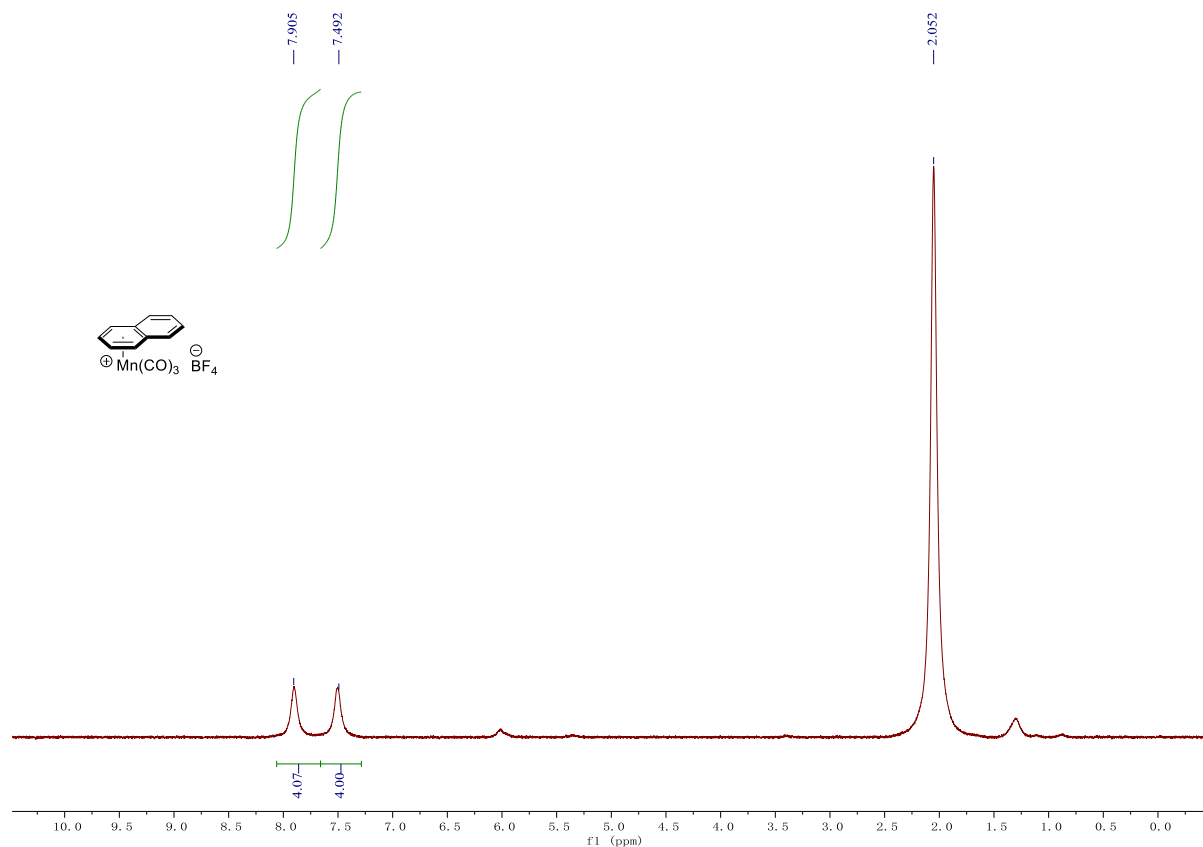
Atomic charge population was a powerful tool for providing quantities information for increased electron-withdrawing effect by $\text{Cr}(\text{CO})_3$ η^6 -coordination, we calculated various atomic charge population including Hirshfeld atomic charge⁴³, ADCH atomic charge⁴⁴ and CM5 atomic charge⁴⁵. And the result shown that the raise of positive atomic charge after $(\text{CrCO})_3$ η^6 -coordination.

Supplementary Table 3 Vertical electron affinity and localized Fukui indexes (f₊) of **1a and **1a-Cr**.**

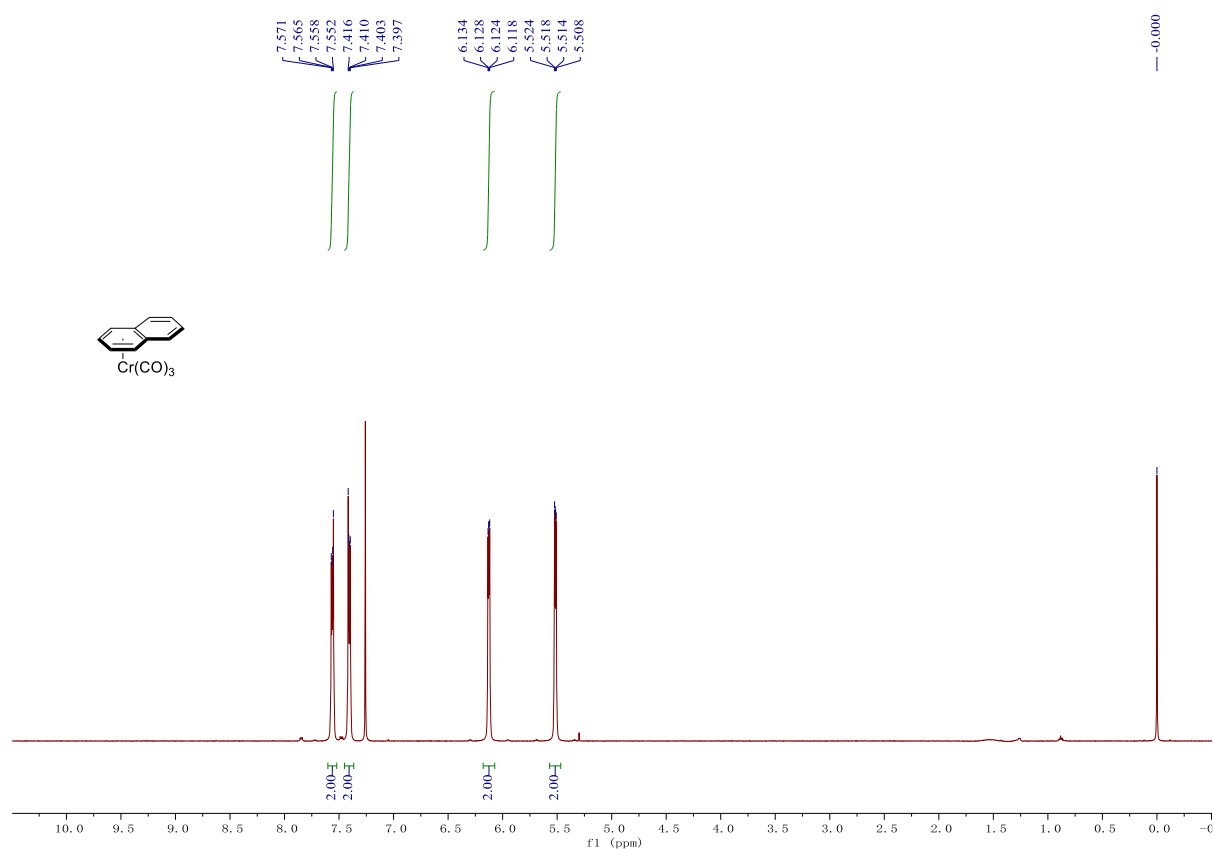
Entry	Compound	VEA (eV)		f ₊
1	1a	0.6520	C1	0.0386
			O1	0.1225
2	1a-Cr	0.9468	C1	0.1019
			O1	0.1154

Despite of the atomic charge, orbital factor (interaction between d orbitals of Cr center and the π system) should be one of another concern too. So, we further performed the conceptual density functional analysis⁴⁶ to provide information from another aspect. Result shown that the raise of vertical electron affinity of **1a-Cr** and C1 localized f₊ to provide information for electron withdrawing effect provided by η^6 -coordination of Cr(CO)₃.

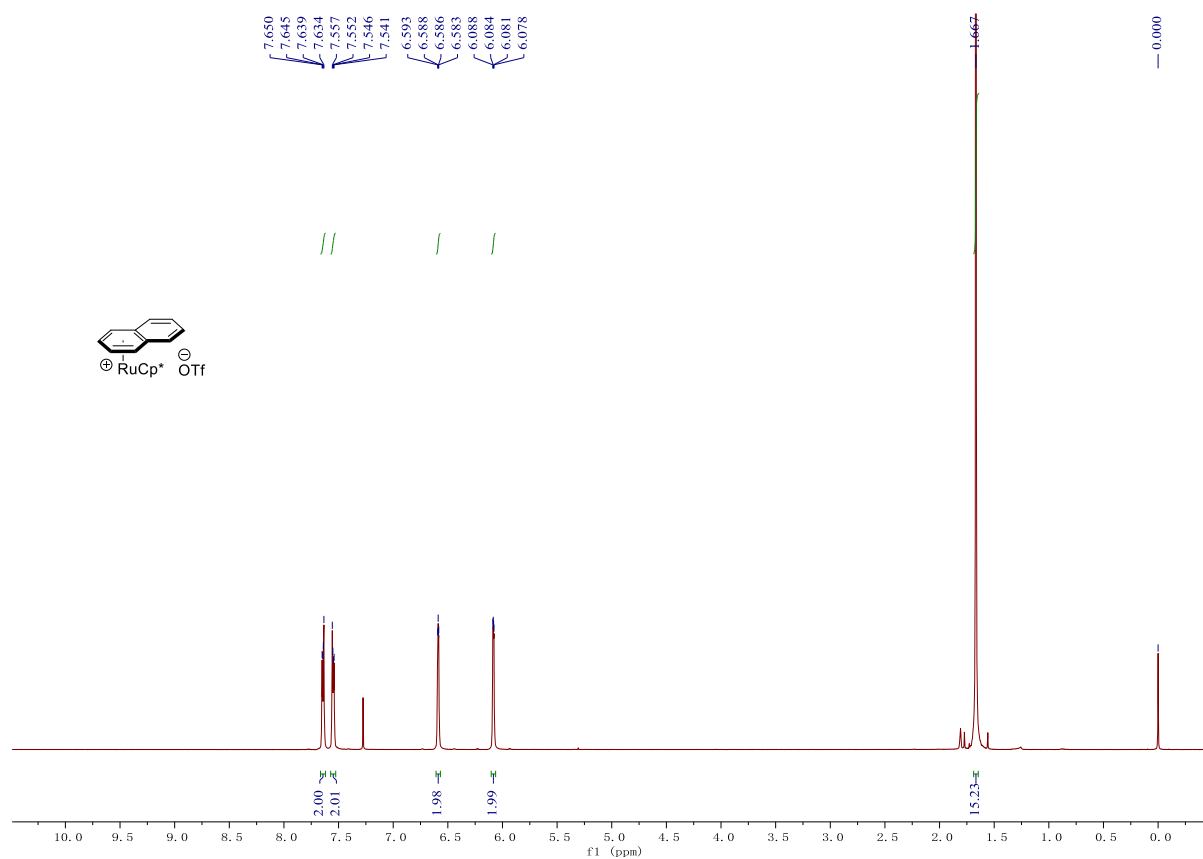
5 NMR Spectra of Compounds



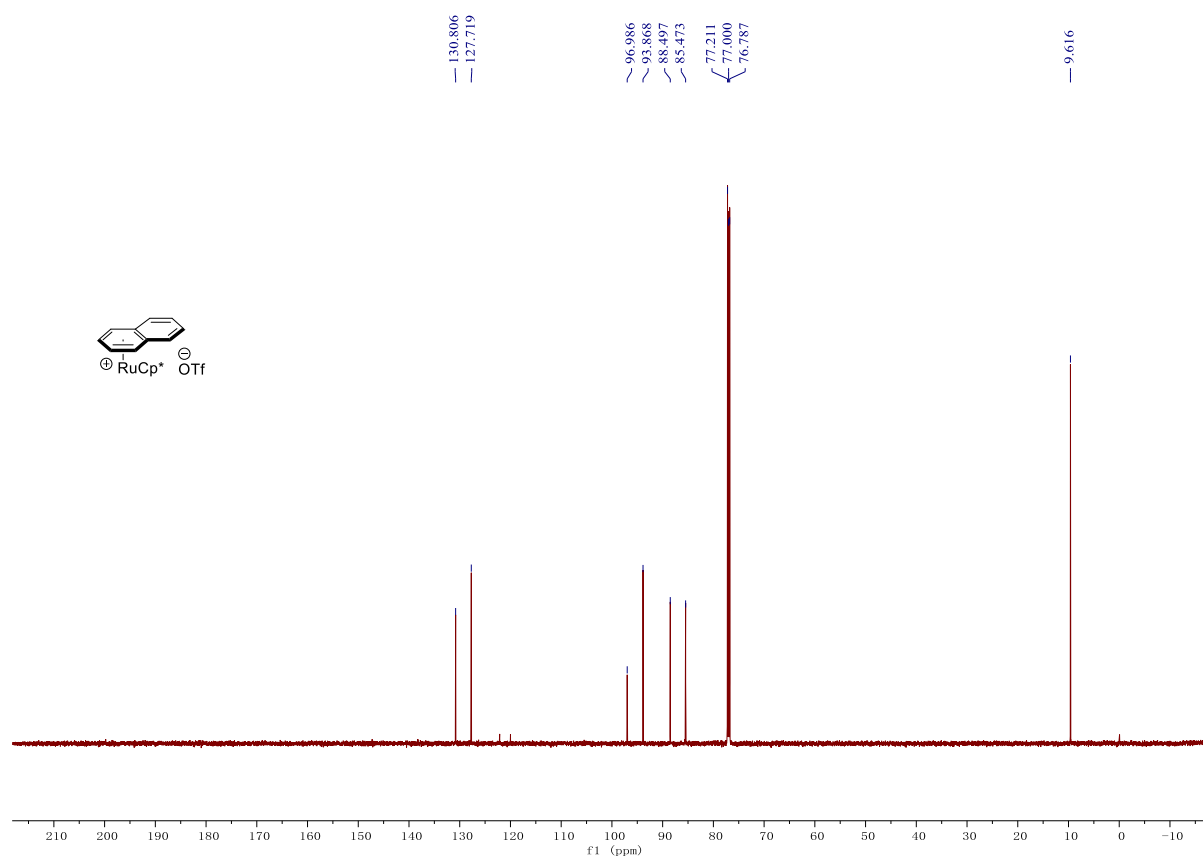
Supplementary Fig. 46 ^1H NMR (500 MHz, Acetone- d_6) of $[\text{Mn}(\text{CO})_3(\text{naphthalene})]\text{BF}_4$.



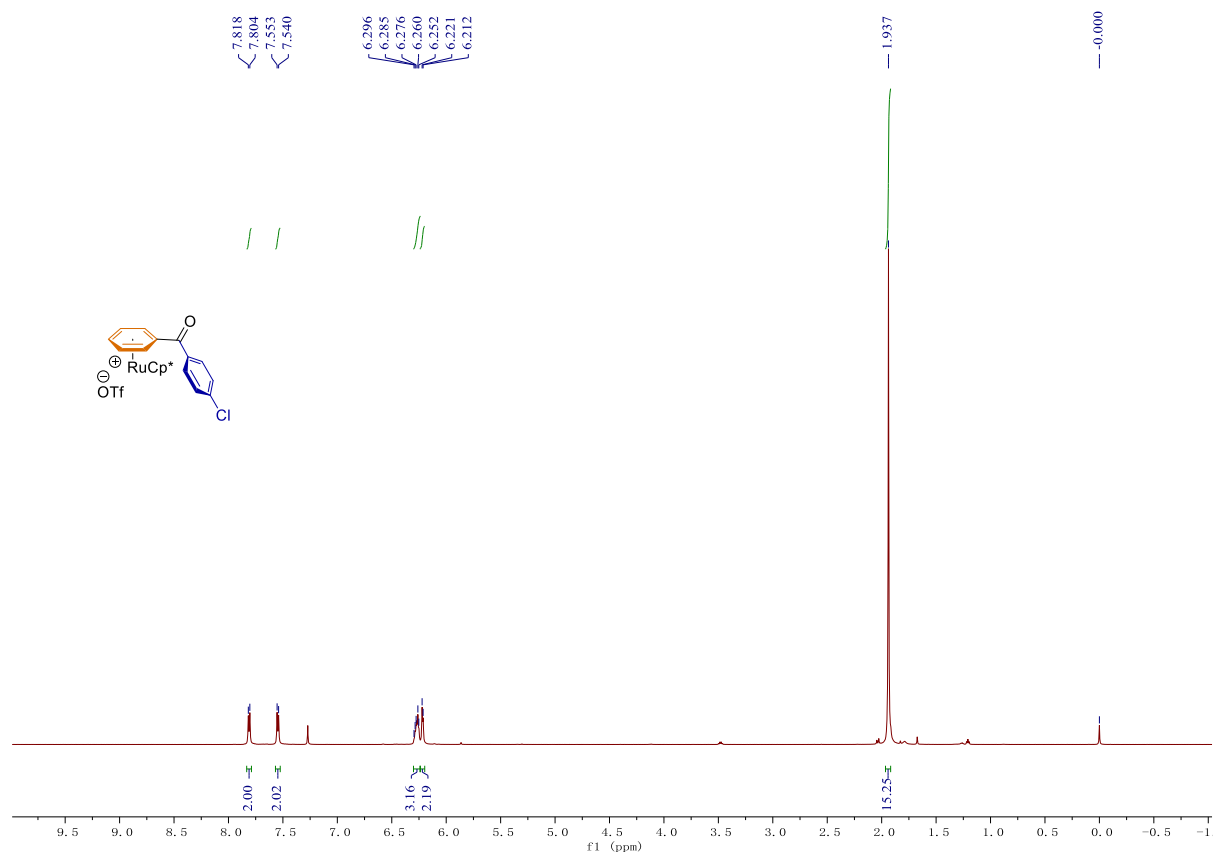
Supplementary Fig. 47 ^1H NMR (500 MHz, Chloroform- d) of $\text{Cr}(\text{CO})_3(\text{naphthalene})$.



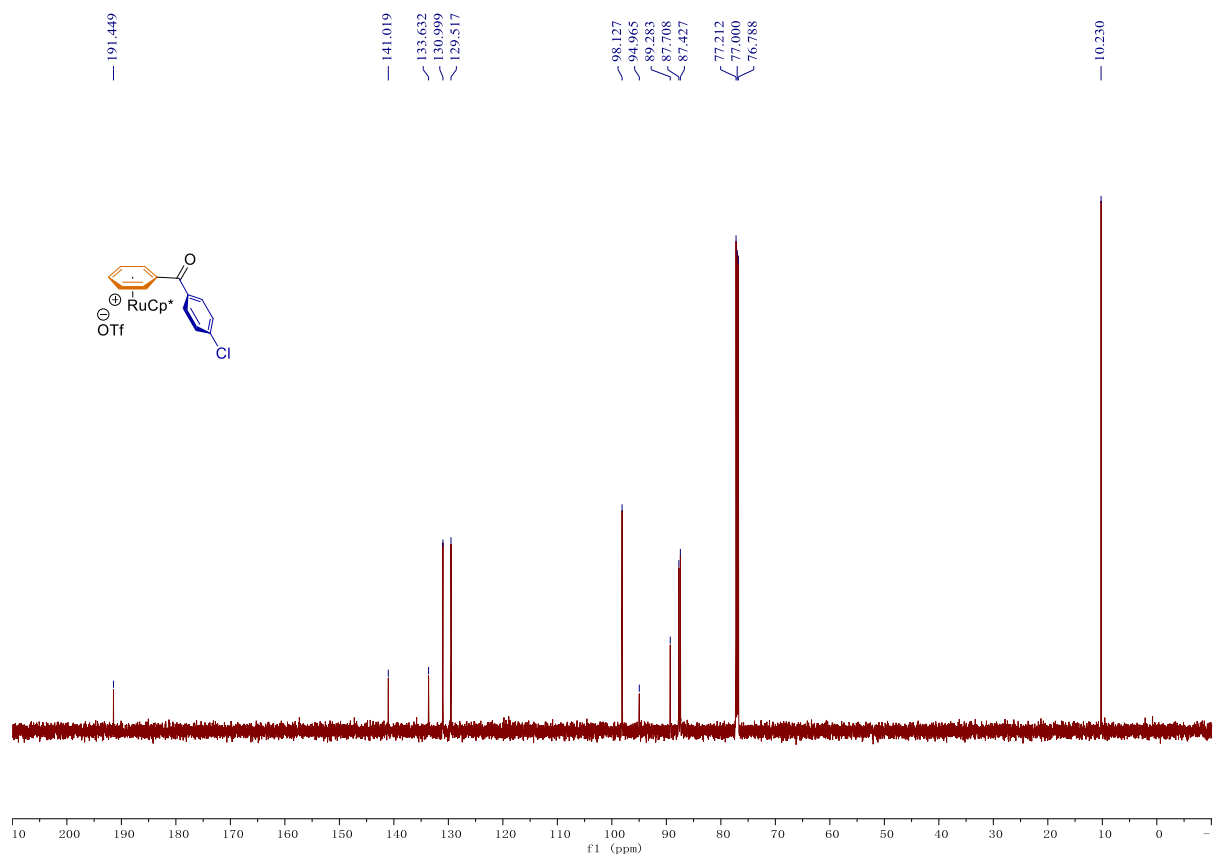
Supplementary Fig. 48 ¹H NMR (600 MHz, Chloroform-*d*) of [Cp*Ru(naphthalene)]OTf.



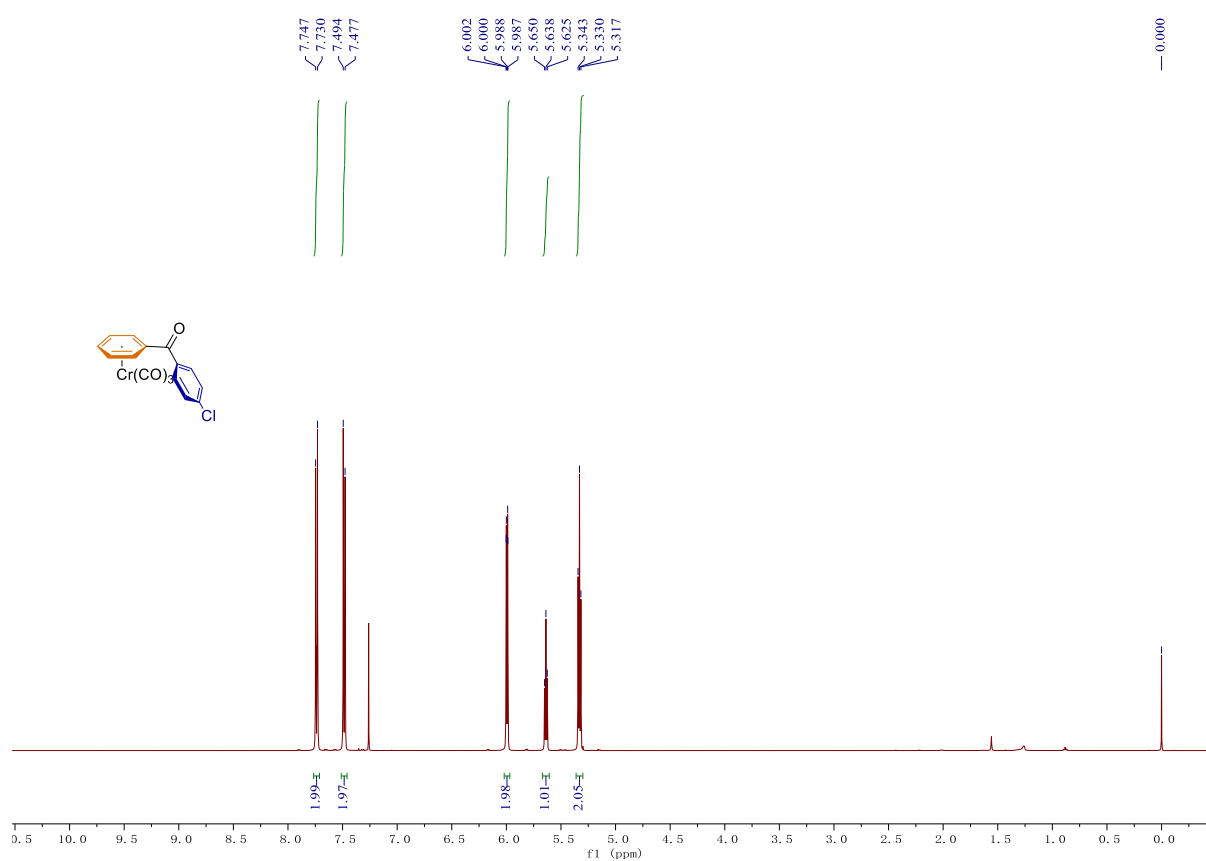
Supplementary Fig. 49 ¹³C NMR (151 MHz, Chloroform-*d*) of [Cp*Ru(naphthalene)]OTf.



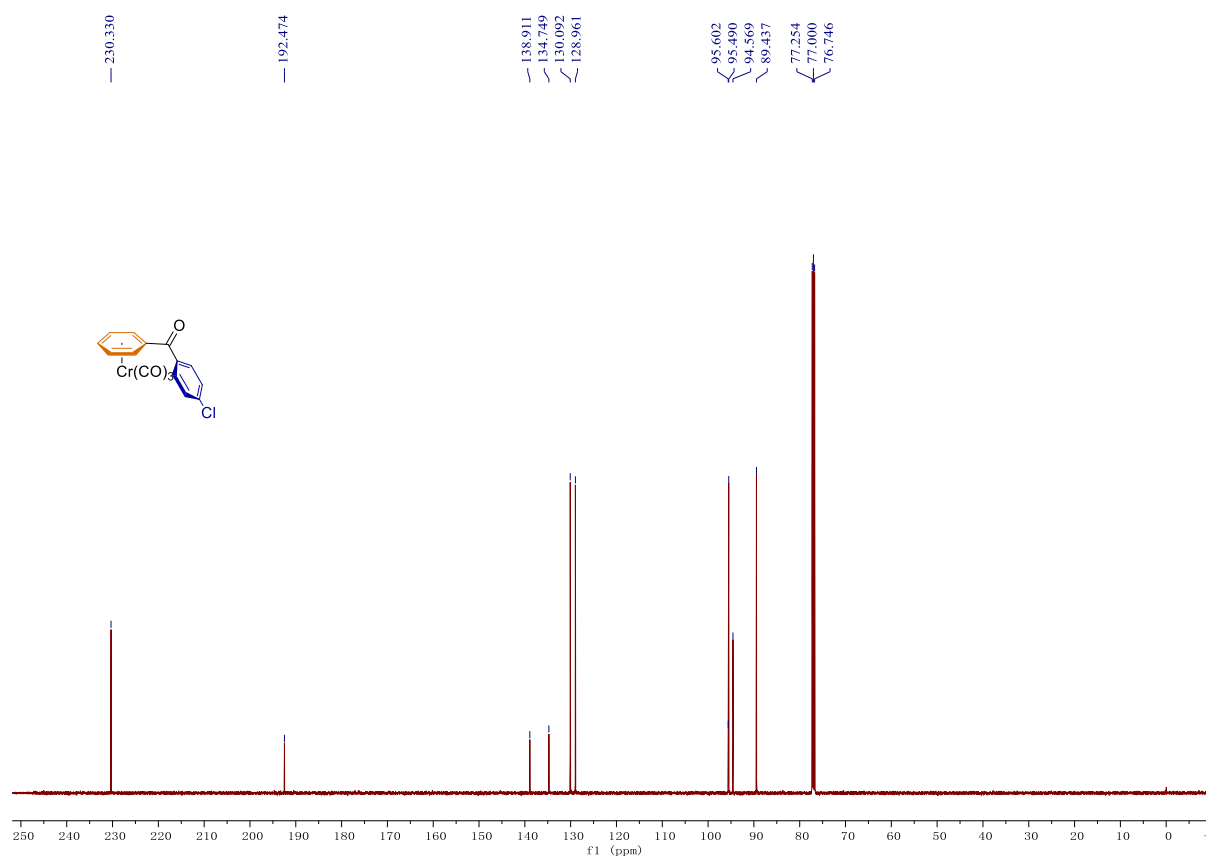
Supplementary Fig. 50 ¹H NMR (600 MHz, Chloroform-*d*) of [Cp*Ru(4-chlorobenzoylbenzene)]OTf (1a-Ru).



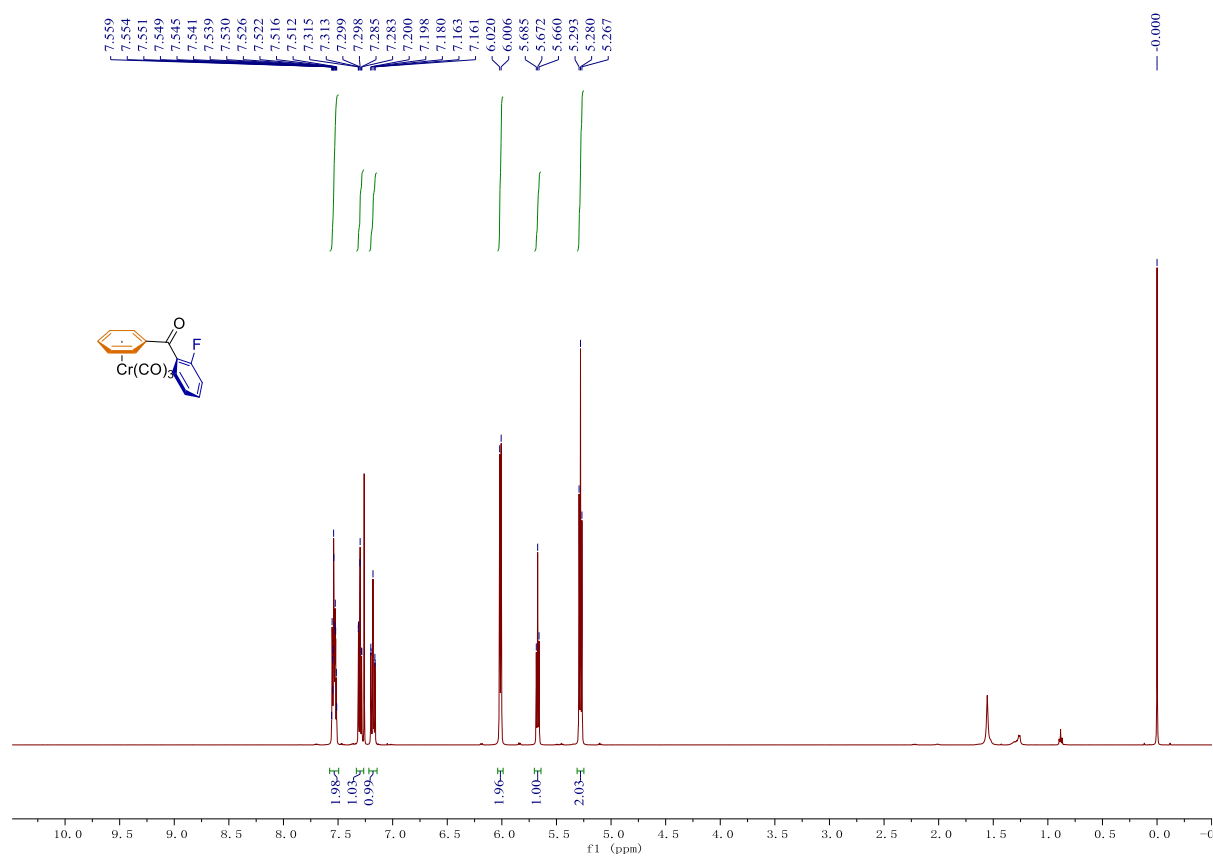
Supplementary Fig. 51 ¹³C NMR (151 MHz, Chloroform-*d*) of [Cp*Ru(4-chlorobenzoylbenzene)]OTf (1a-Ru).



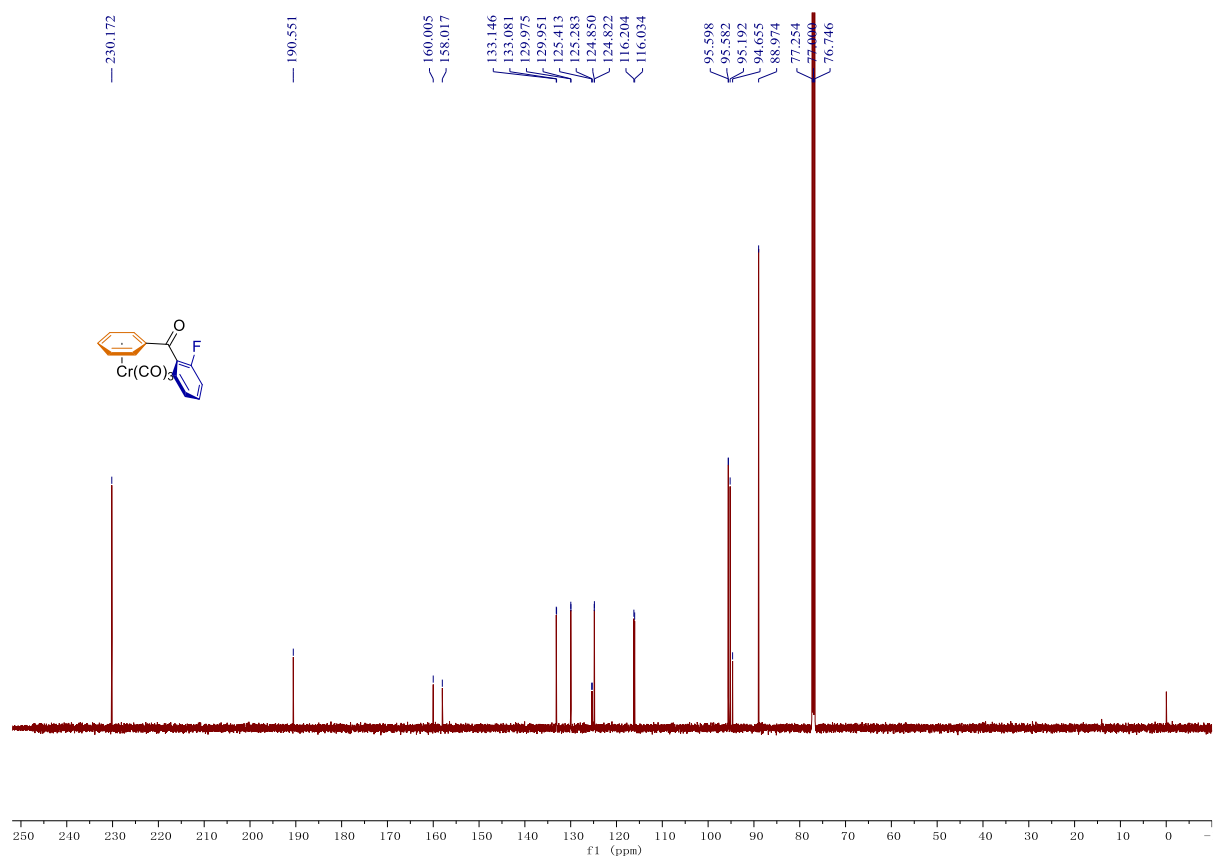
Supplementary Fig. 52 ¹H NMR (500 MHz, Chloroform-*d*) of 4-chlorobenzoylbenzene chromium tricarbonyl (1a-Cr).



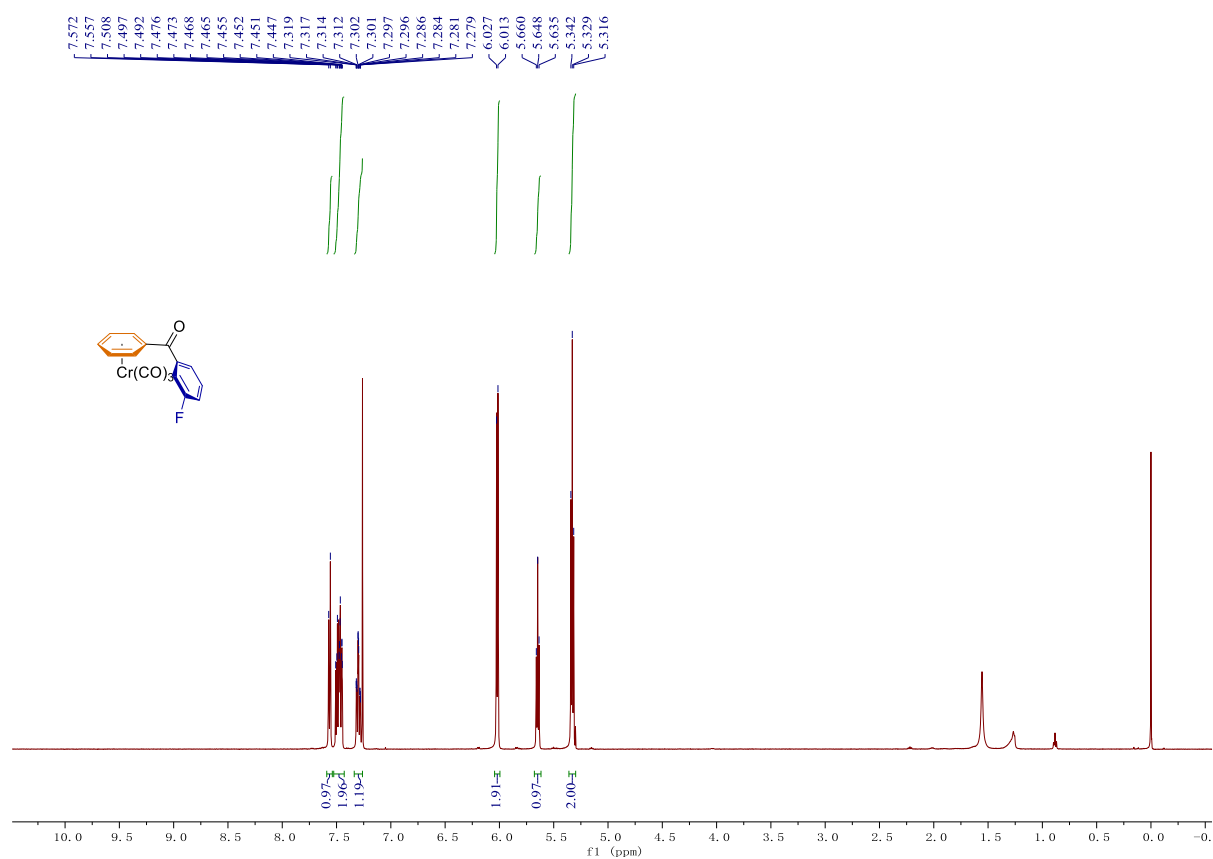
Supplementary Fig. 53 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-chlorobenzoylbenzene chromium tricarbonyl (1a-Cr).



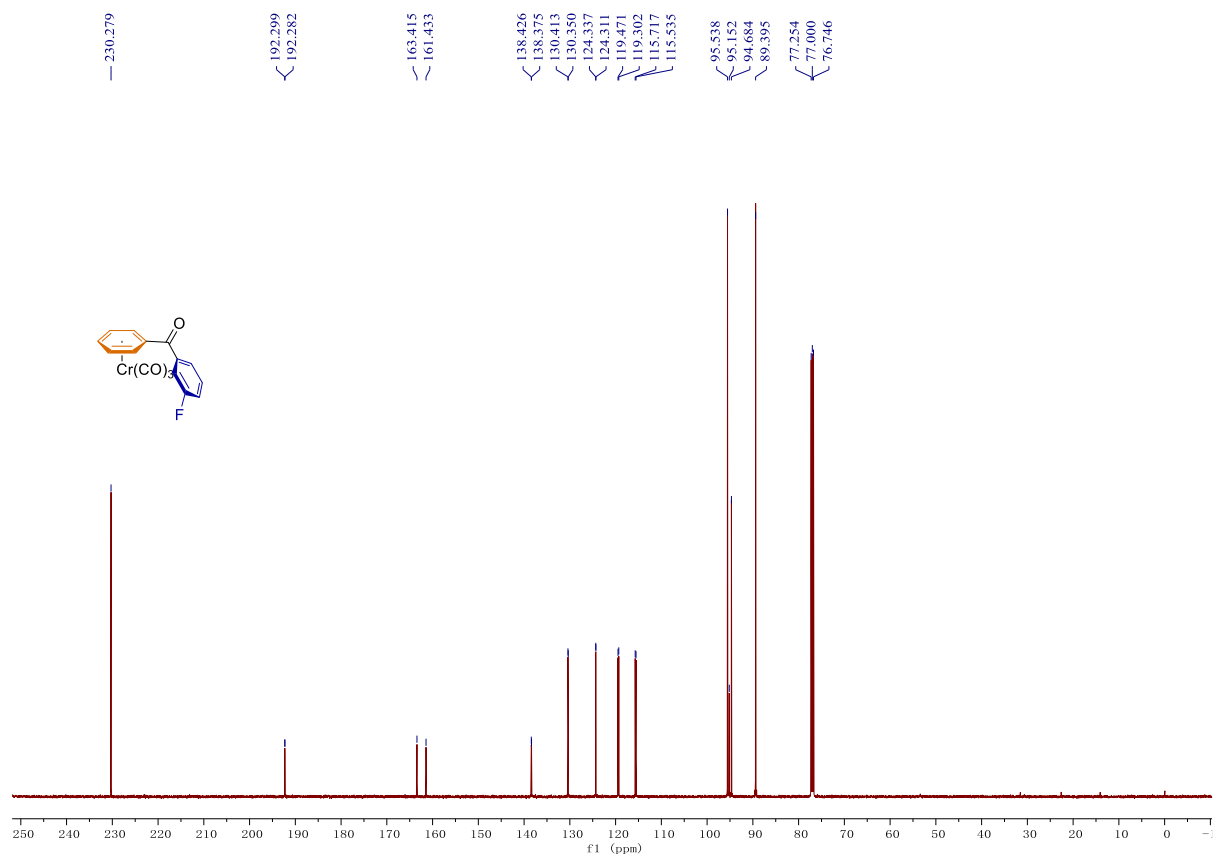
Supplementary Fig. 54 ¹H NMR (500 MHz, Chloroform-*d*) of 2-fluorobenzoylbenzene chromium tricarbonyl (s1-Cr).



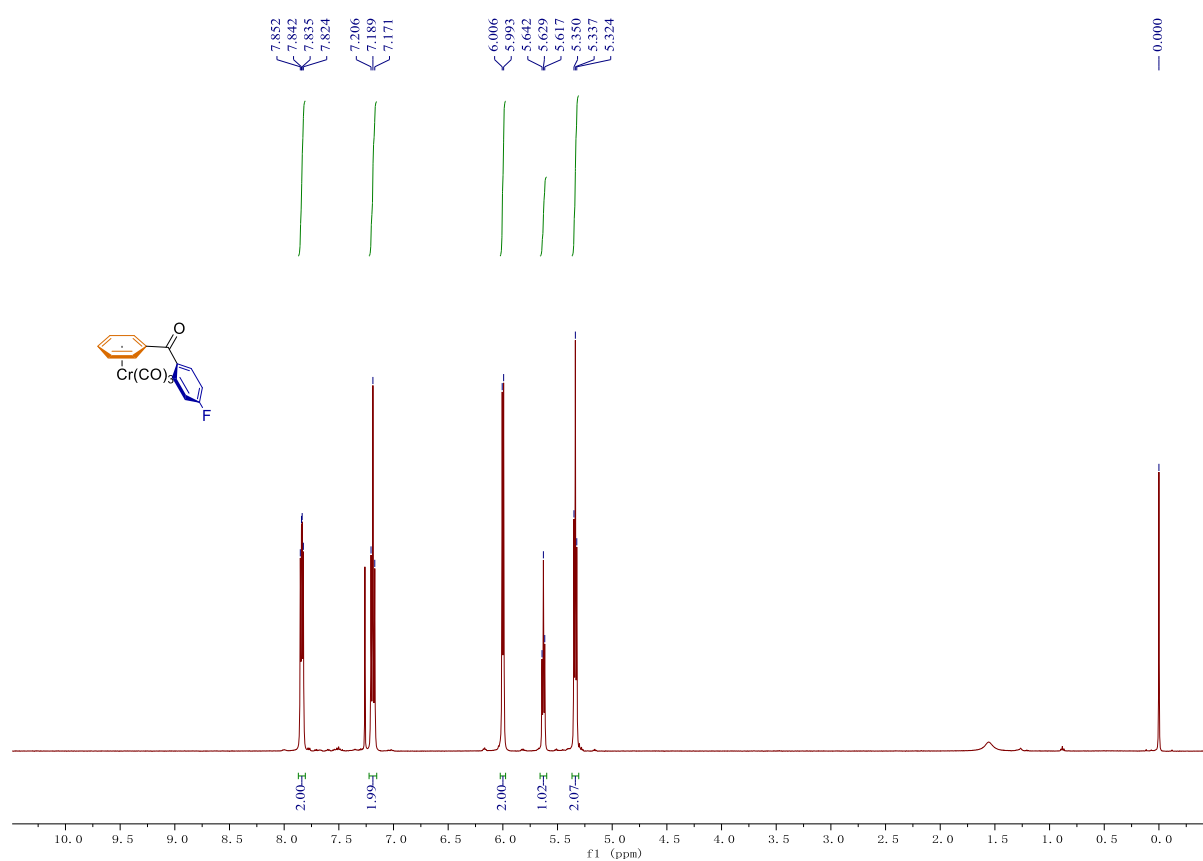
Supplementary Fig. 55 ¹³C NMR (126 MHz, Chloroform-*d*) of 2-fluorobenzoylbenzene chromium tricarbonyl (s1-Cr).



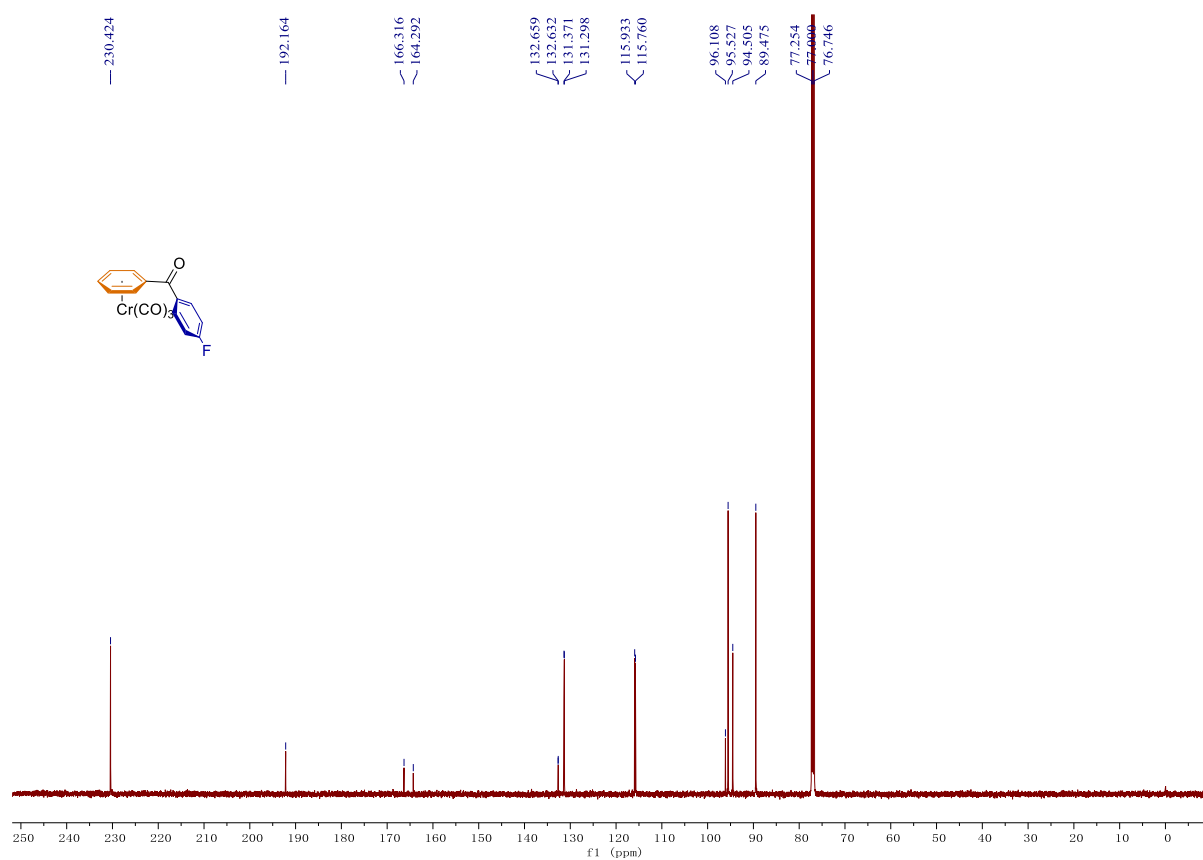
Supplementary Fig. 56 ¹H NMR (500 MHz, Chloroform-*d*) of 3-fluorobenzoylbenzene chromium tricarbonyl (1c-Cr).



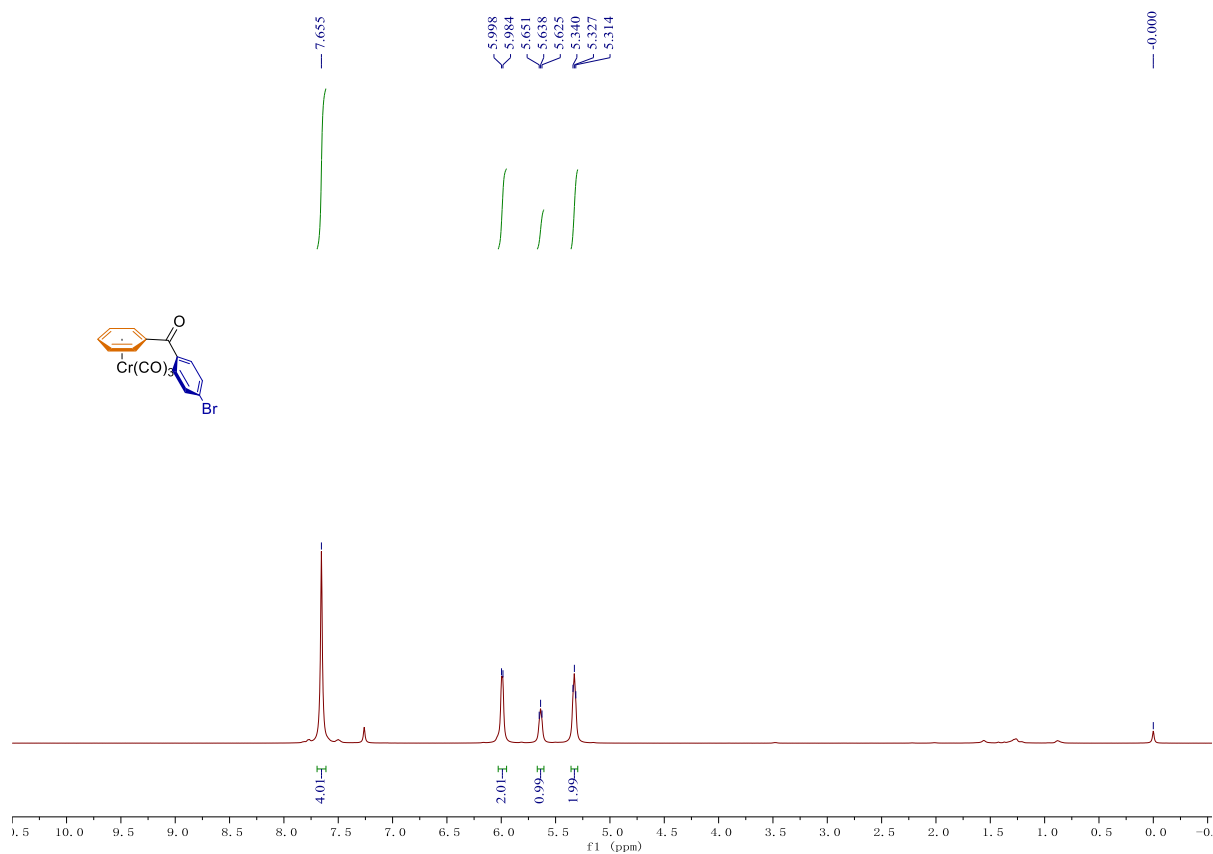
Supplementary Fig. 57 ¹³C NMR (126 MHz, Chloroform-*d*) of 3-fluorobenzoylbenzene chromium tricarbonyl (1c-Cr).



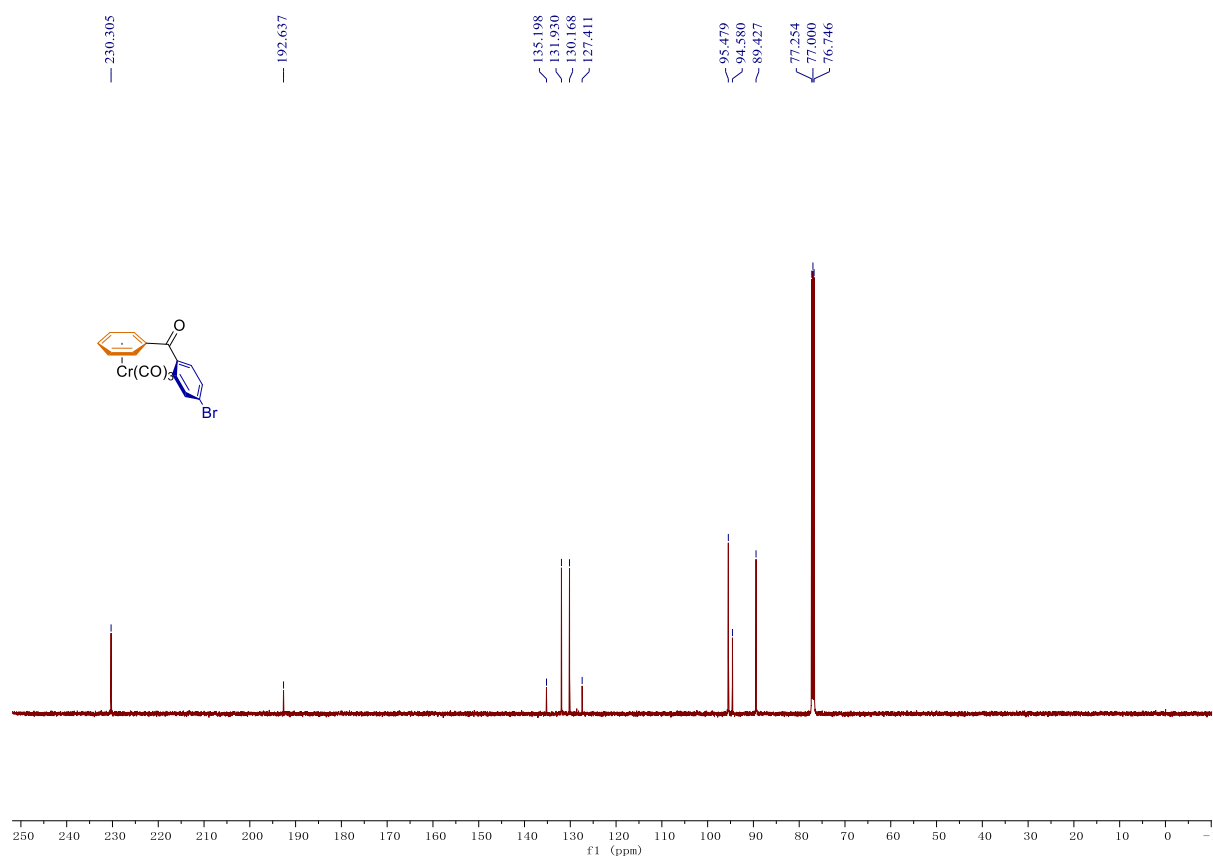
Supplementary Fig. 58 ¹H NMR (500 MHz, Chloroform-*d*) of 4-fluorobenzoylbenzene chromium tricarbonyl (1d-Cr).



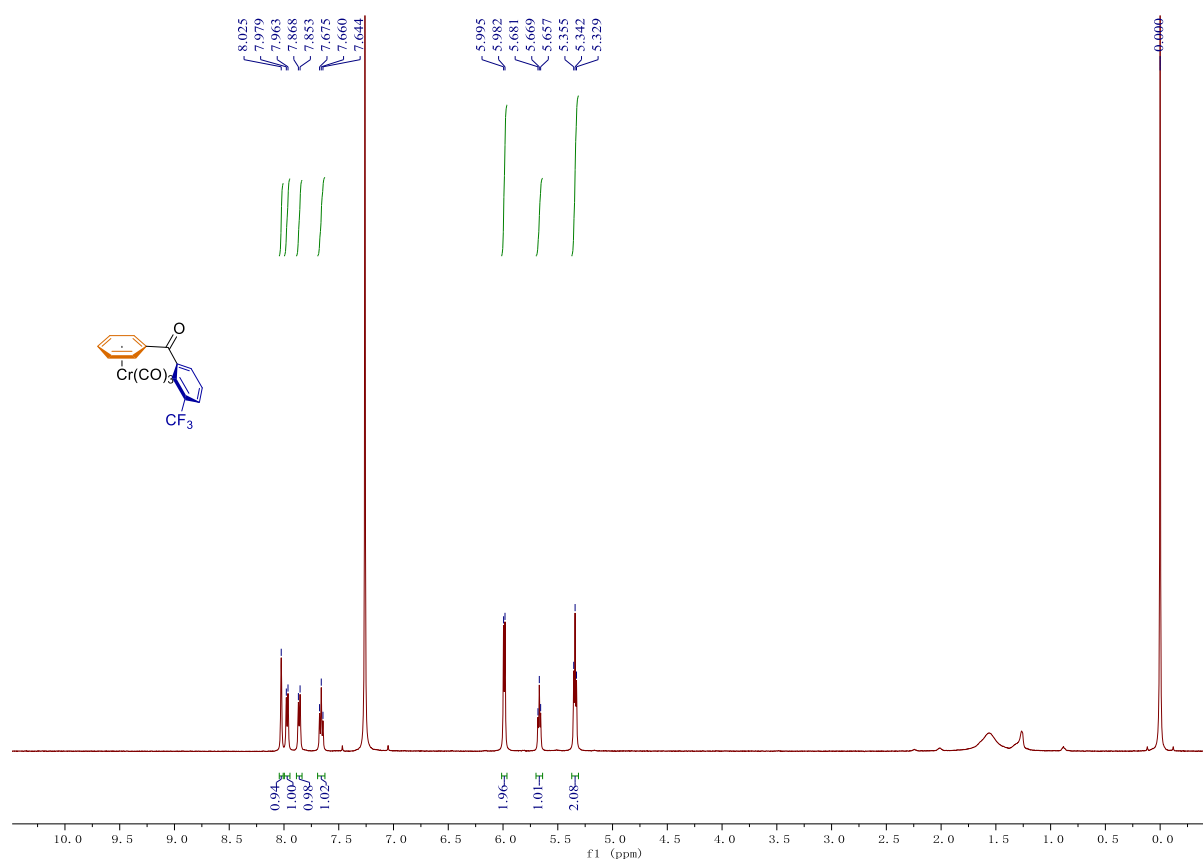
Supplementary Fig. 59 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-fluorobenzoylbenzene chromium tricarbonyl (1d-Cr).



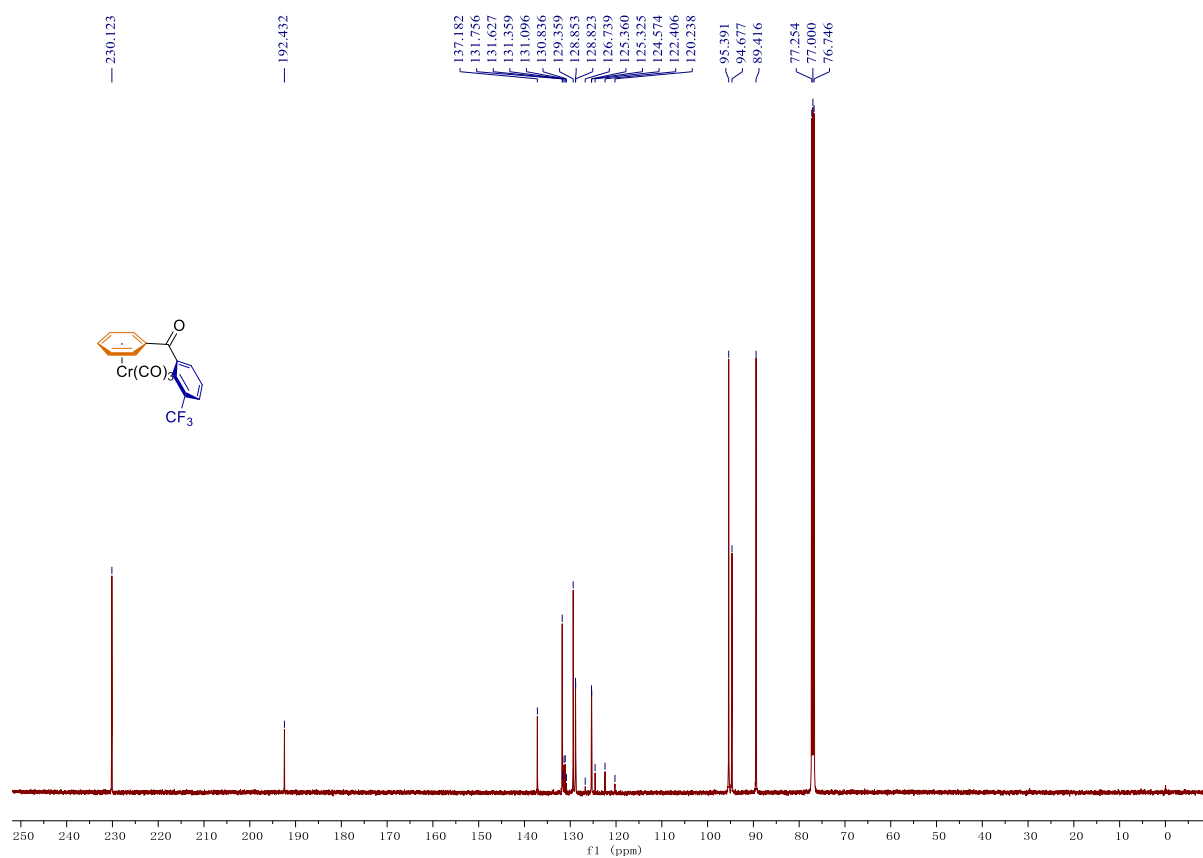
Supplementary Fig. 60 ¹H NMR (500 MHz, Chloroform-*d*) of 4-bromobenzoylbenzene chromium tricarbonyl (1e-Cr).



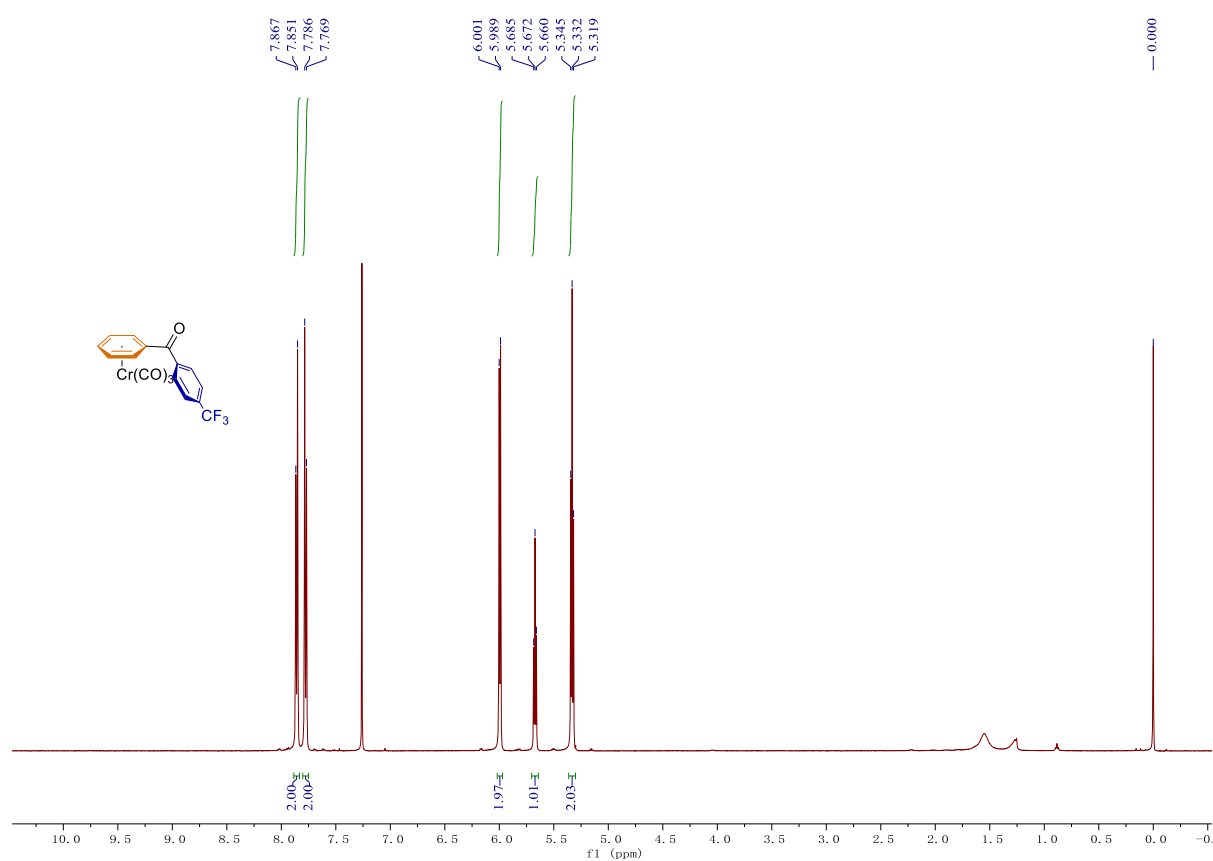
Supplementary Fig. 61 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-bromobenzoylbenzene chromium tricarbonyl (1e-Cr).



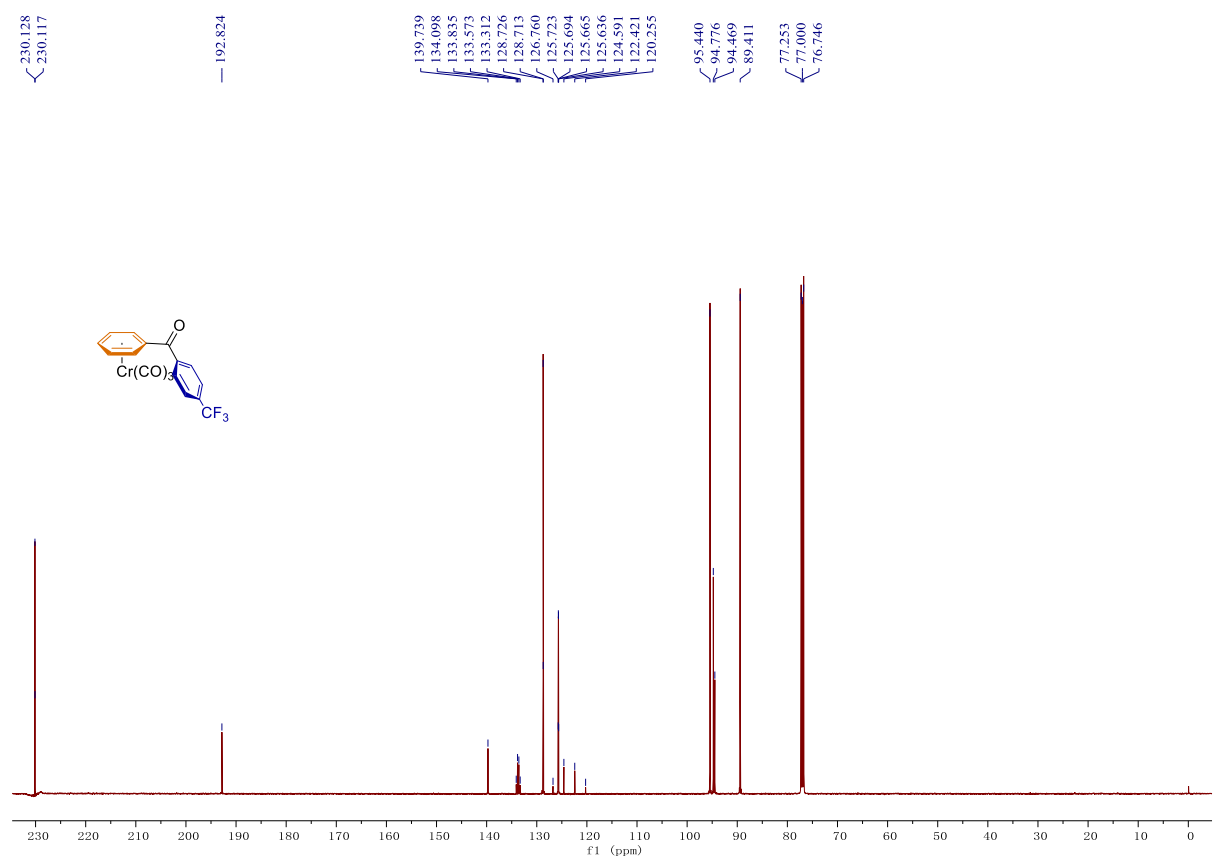
Supplementary Fig. 62 ¹H NMR (500 MHz, Chloroform-*d*) of 3-(trifluoromethyl)benzoylbenzene chromium tricarbonyl (1y-Cr).



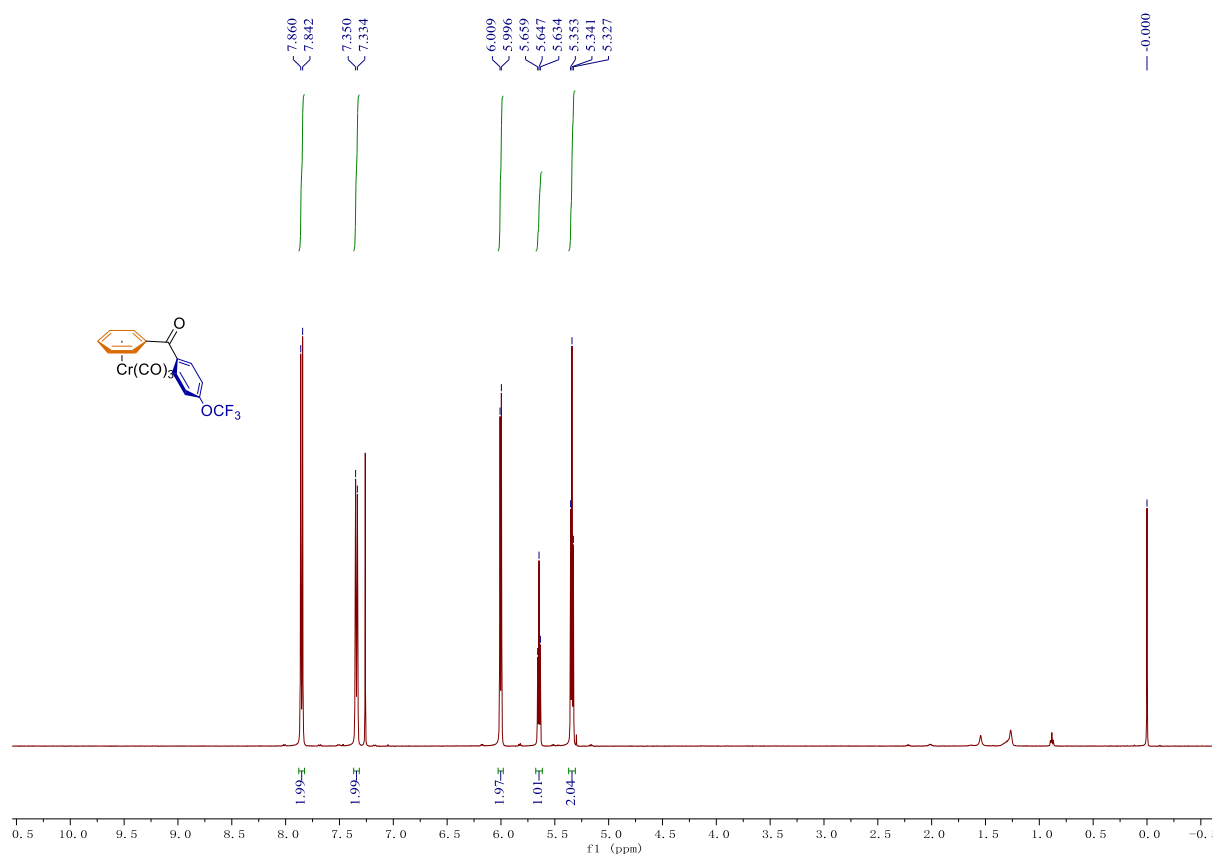
Supplementary Fig. 63 ¹³C NMR (126 MHz, Chloroform-*d*) of 3-(trifluoromethyl)benzoylbenzene chromium tricarbonyl (1y-Cr).



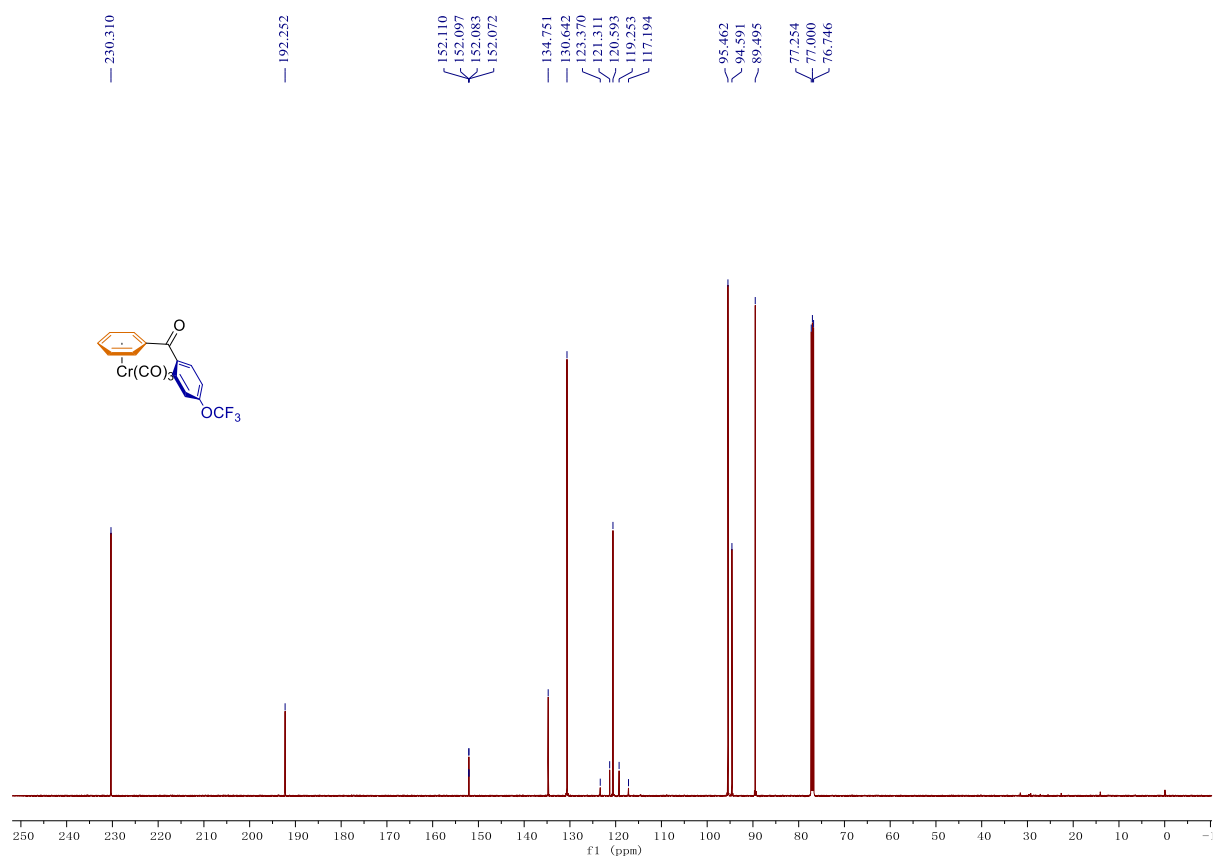
Supplementary Fig. 64 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(trifluoromethyl)benzoylbenzene chromium tricarbonyl (1f-Cr).



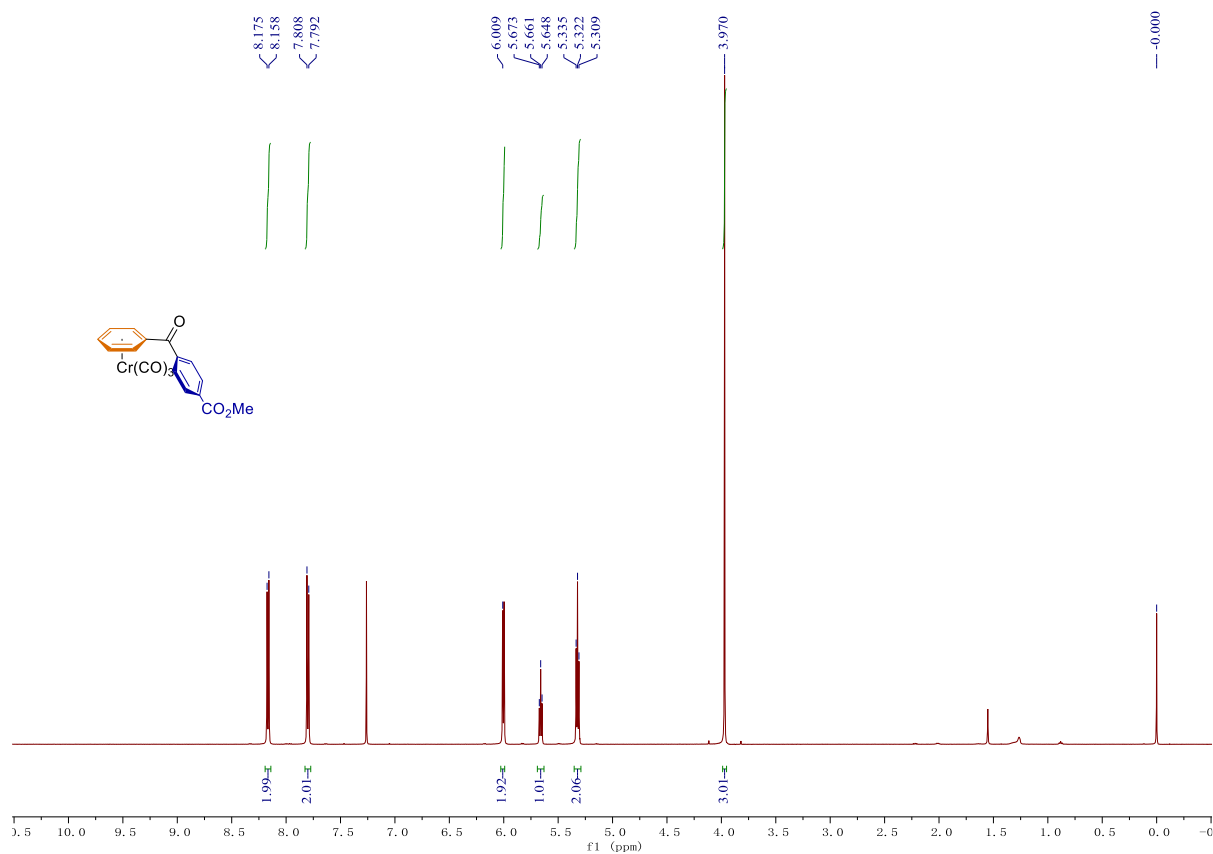
Supplementary Fig. 65 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(trifluoromethyl)benzoylbenzene chromium tricarbonyl (1f-Cr).



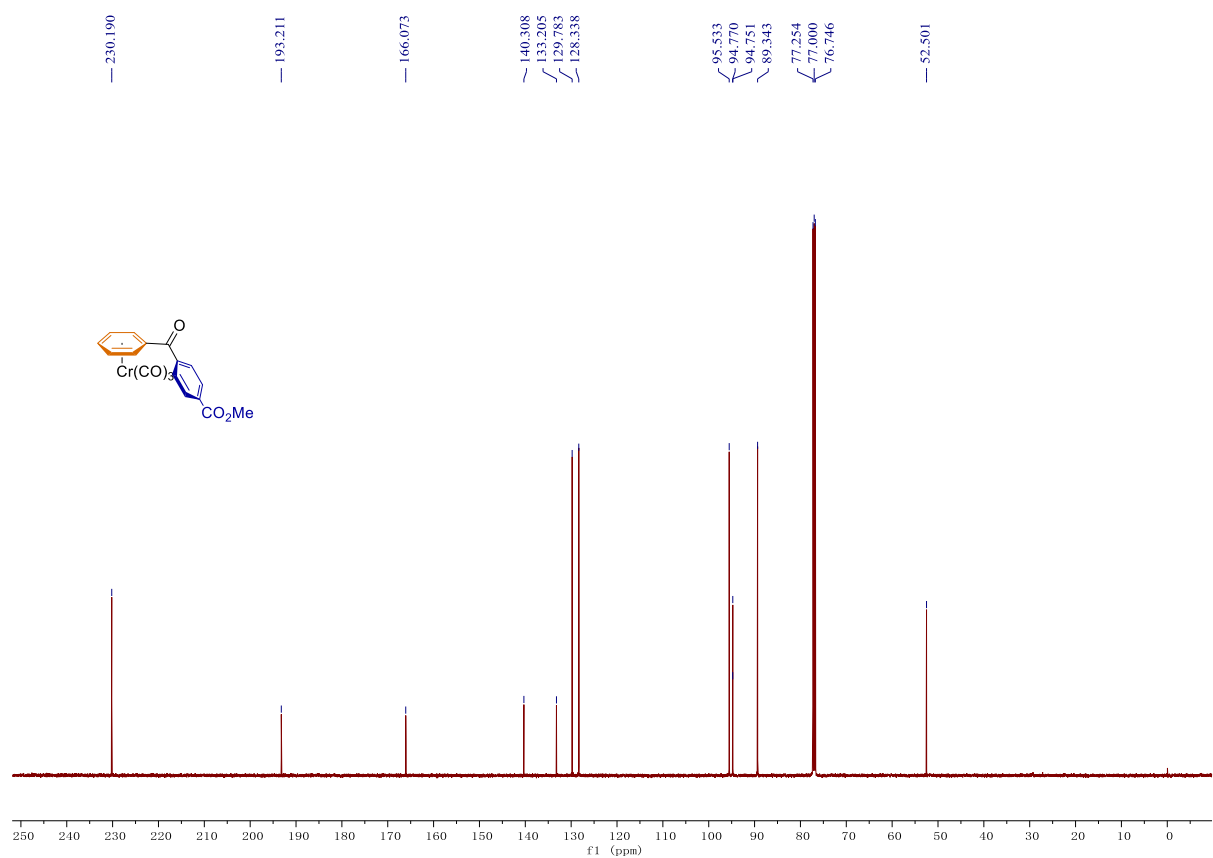
Supplementary Fig. 66 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(trifluoromethoxy)benzoylbenzene chromium tricarbonyl (1g-Cr).



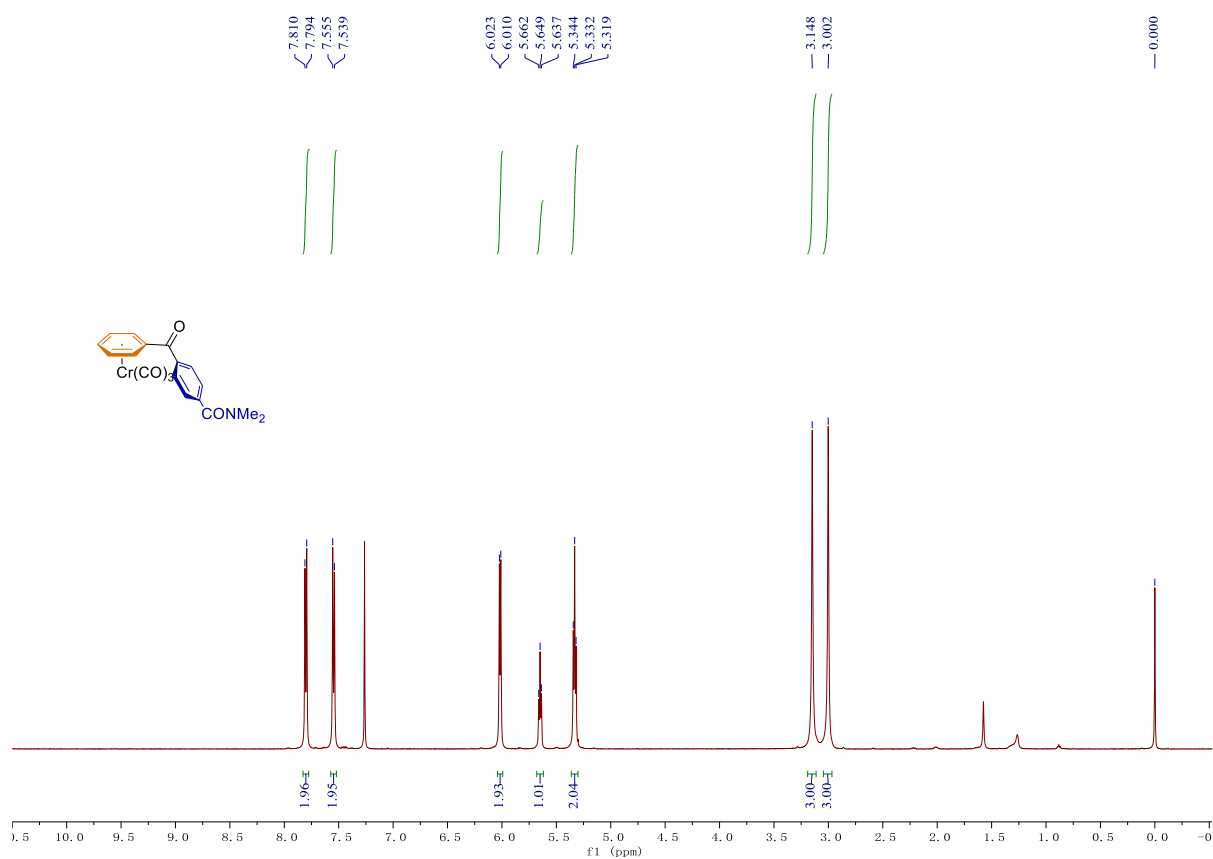
Supplementary Fig. 67 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(trifluoromethoxy)benzoylbenzene chromium tricarbonyl (1g-Cr).



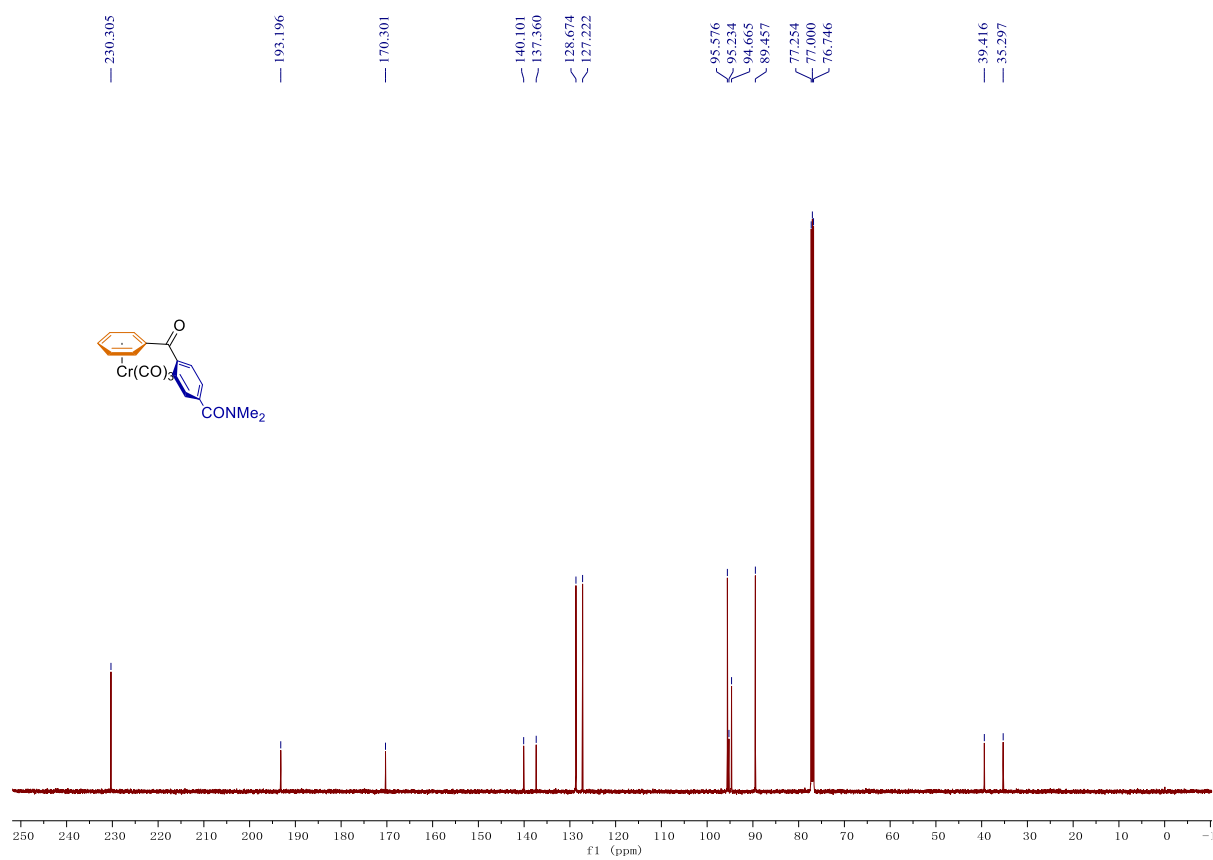
Supplementary Fig. 68 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(methoxycarbonyl)benzoylbenzene chromium tricarbonyl (1h-Cr).



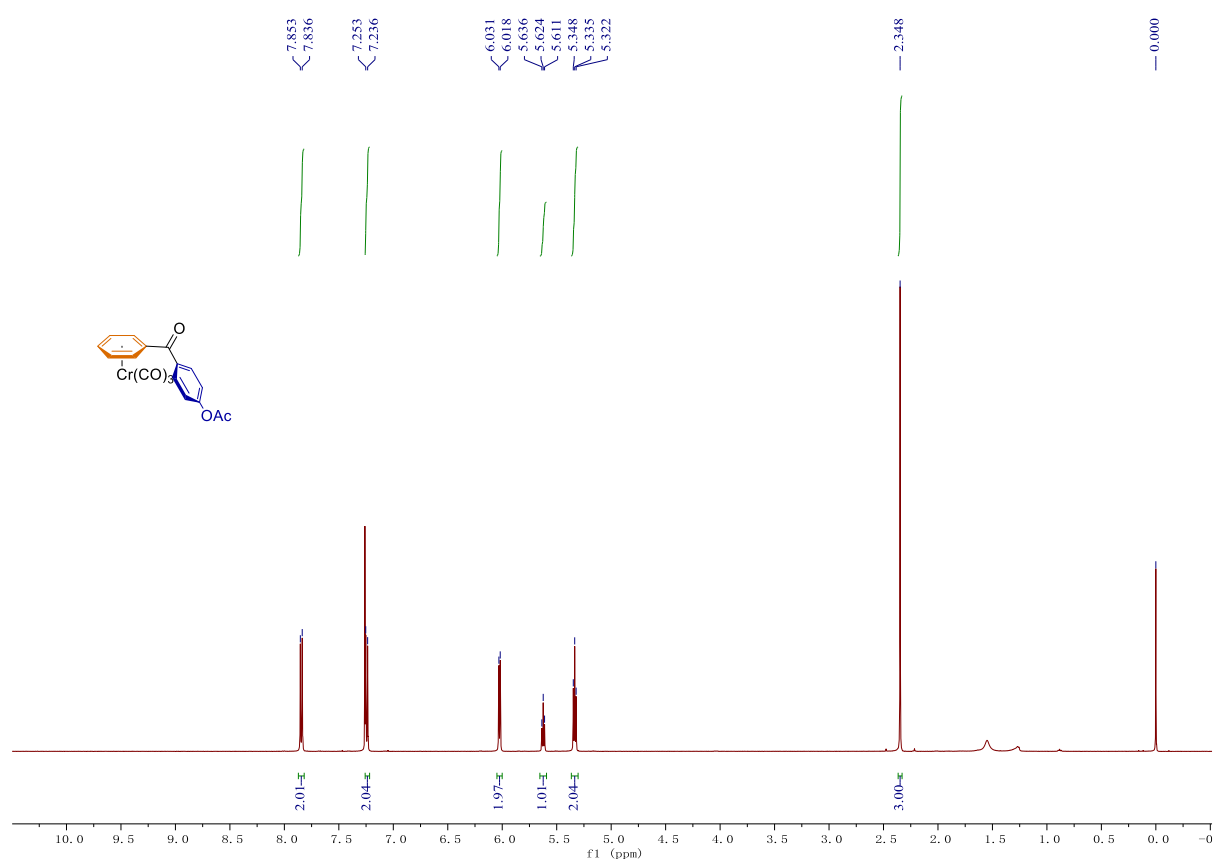
Supplementary Fig. 69 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(methoxycarbonyl)benzoylbenzene chromium tricarbonyl (1h-Cr).



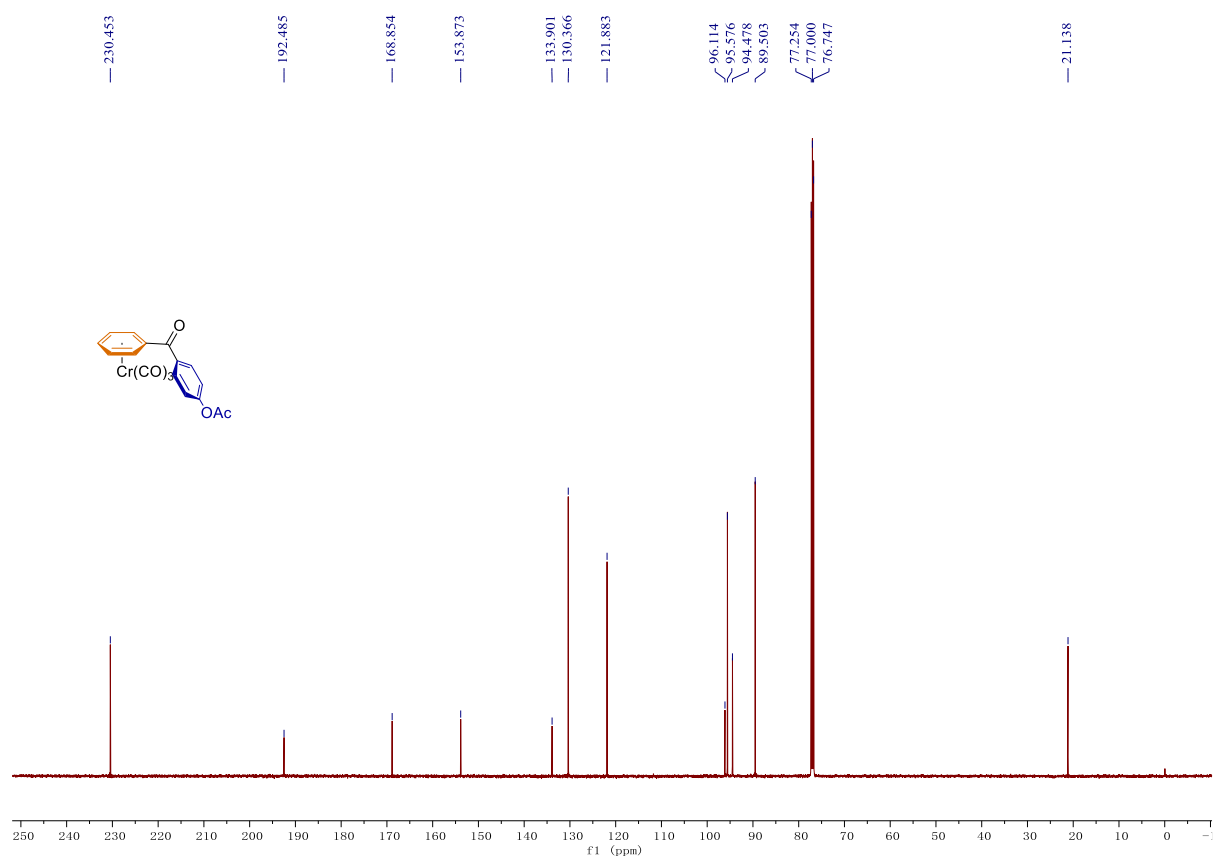
Supplementary Fig. 70 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(dimethylcarbamoyl)benzoylbenzene chromium tricarbonyl (1i-Cr).



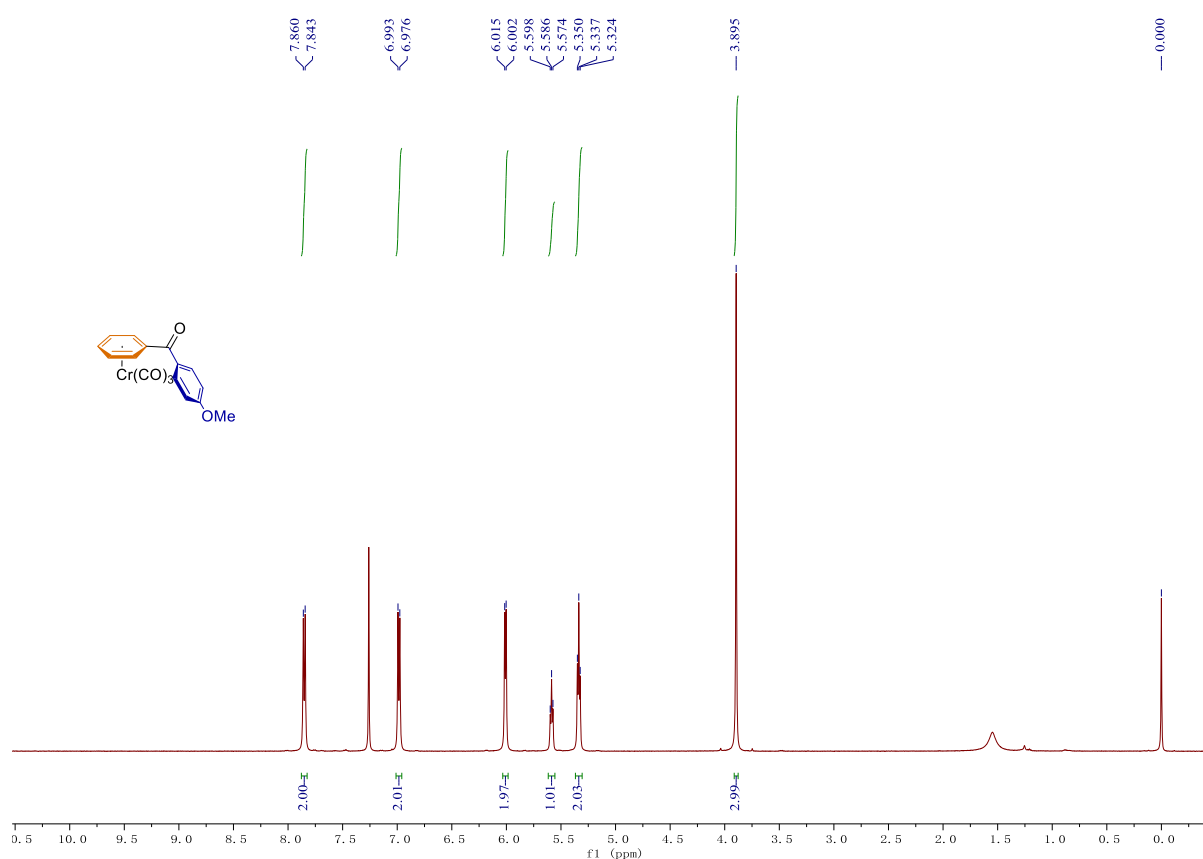
Supplementary Fig. 71 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(dimethylcarbamoyl)benzoylbenzene chromium tricarbonyl (1i-Cr).



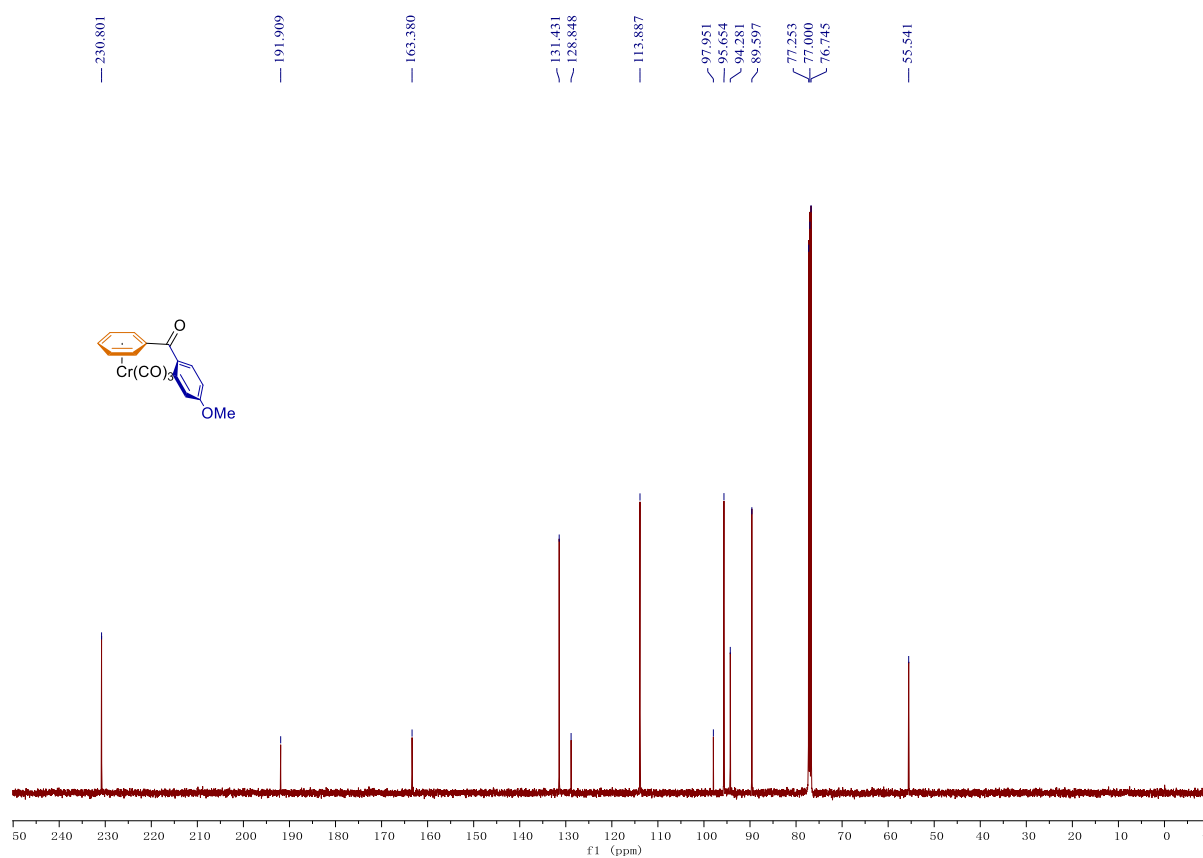
Supplementary Fig. 72 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(acetoxy)benzoylbenzene chromium tricarbonyl (s2-Cr).



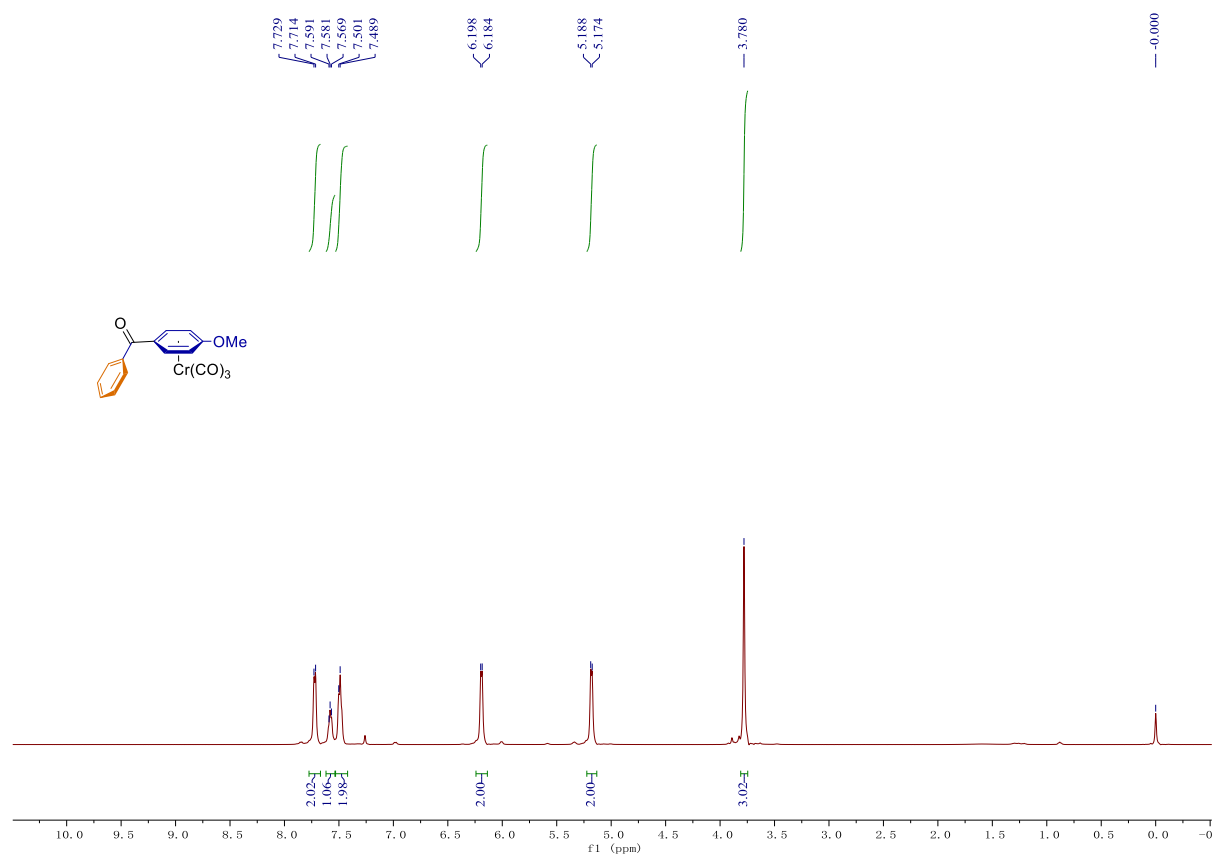
Supplementary Fig. 73 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(acetoxy)benzoylbenzene chromium tricarbonyl (s2-Cr).



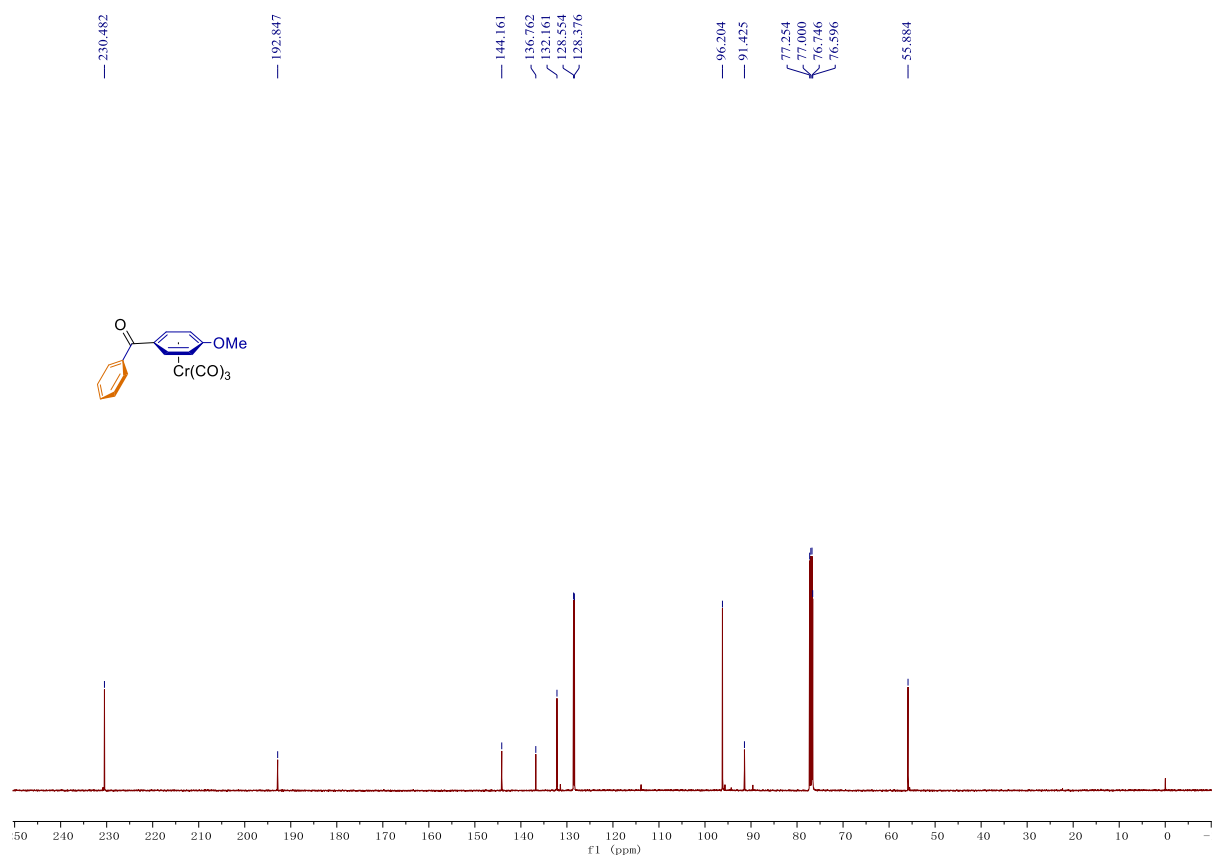
Supplementary Fig. 74 ^1H NMR (500 MHz, Chloroform- d) of 4-(methoxy)benzoylbenzene chromium tricarbonyl (1b-Cr).



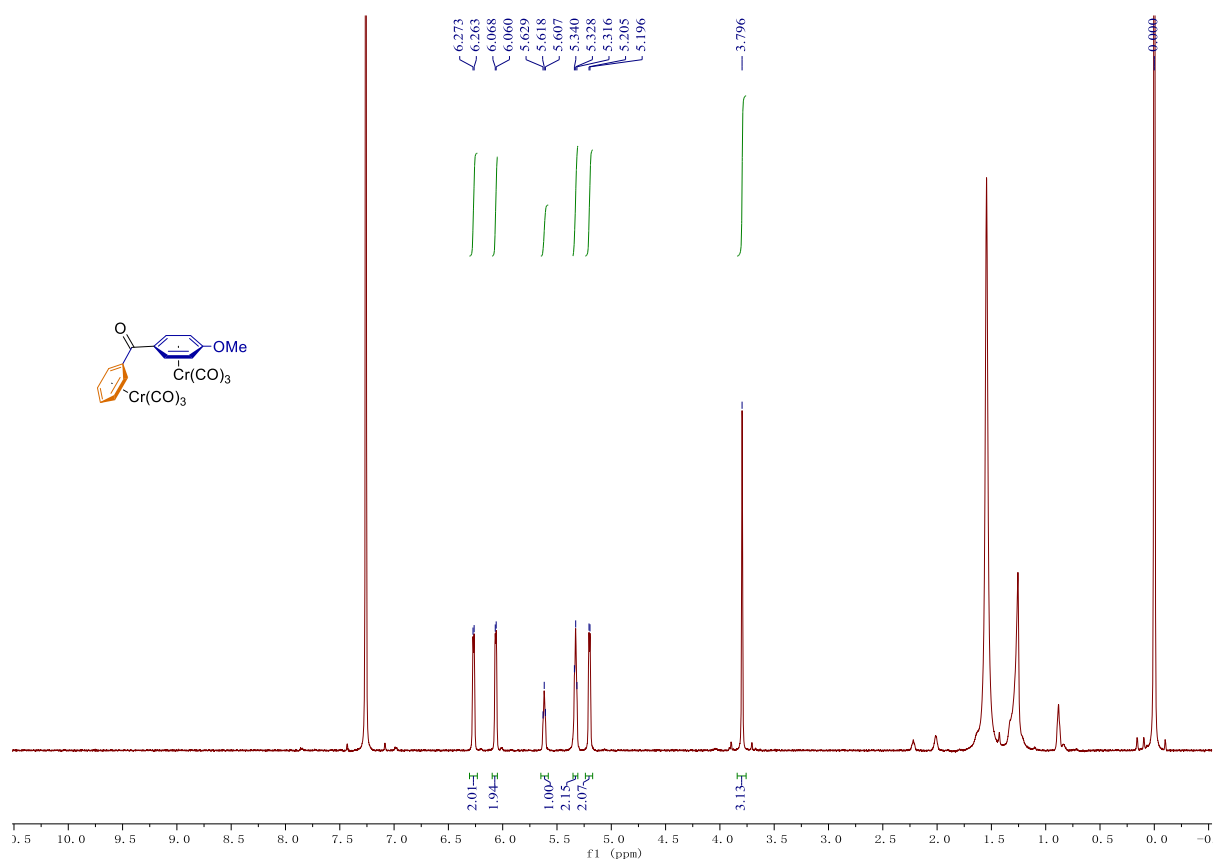
Supplementary Fig. 75 ^{13}C NMR (126 MHz, Chloroform- d) of 4-(methoxy)benzoylbenzene chromium tricarbonyl (1b-Cr).



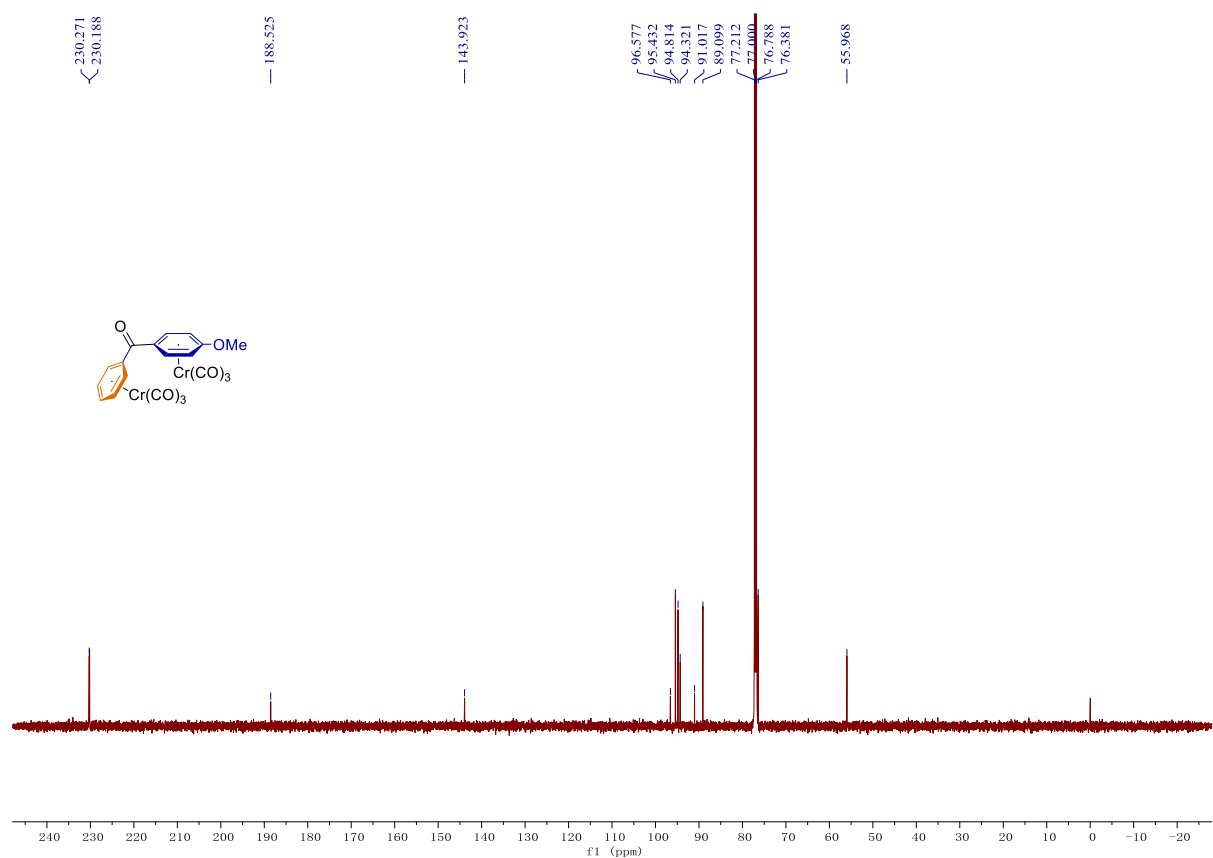
Supplementary Fig. 76 ¹H NMR (500 MHz, Chloroform-*d*) of 4-benzoylanisole chromium tricarbonyl (1b-Cr').



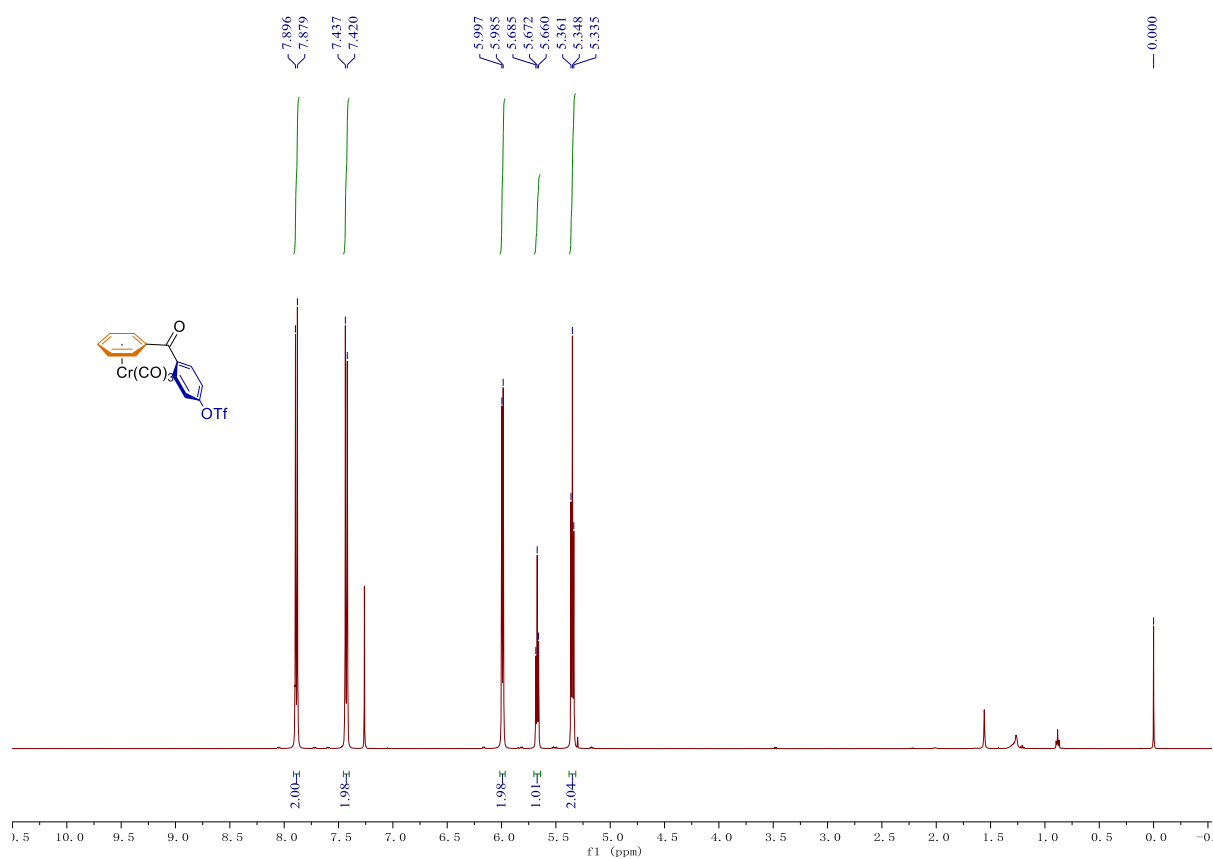
Supplementary Fig. 77 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-benzoylanisole chromium tricarbonyl (1b-Cr').



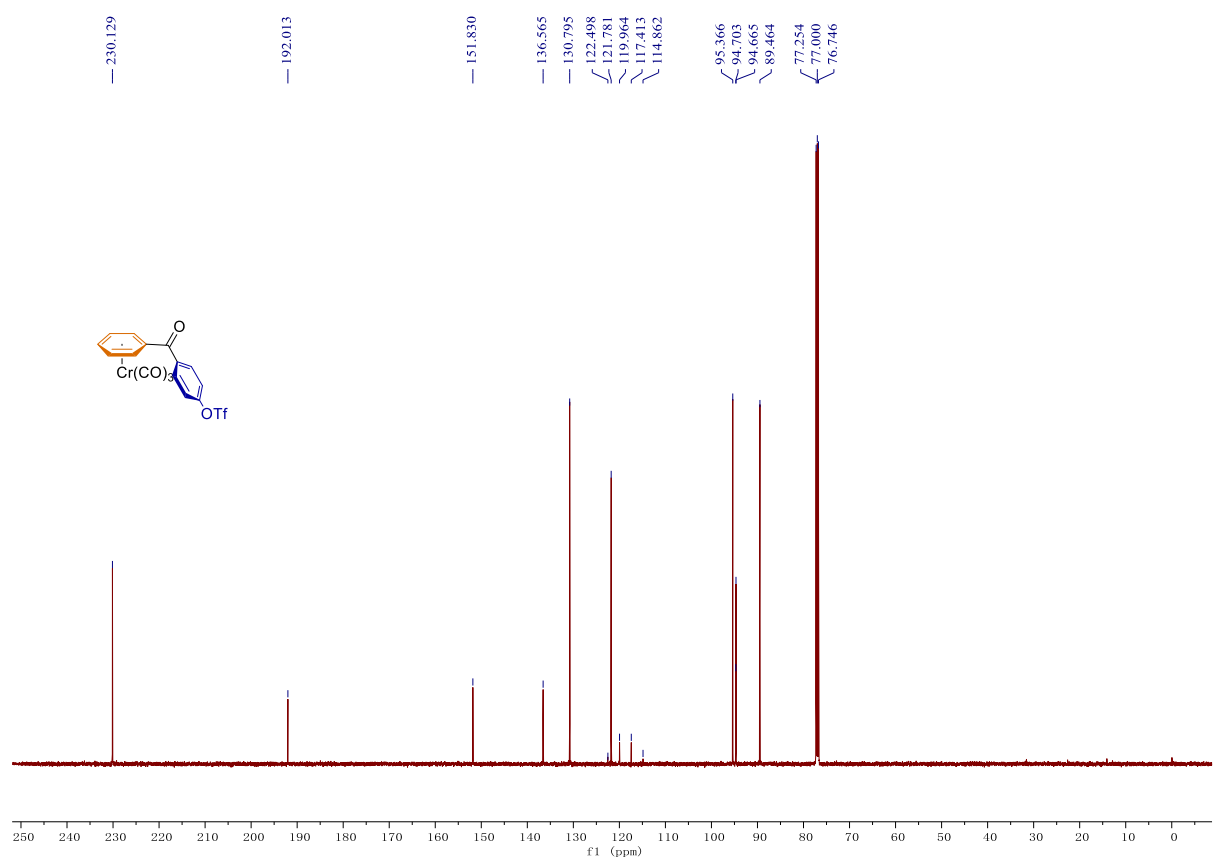
Supplementary Fig. 78 ¹H NMR (600 MHz, Chloroform-*d*) of (4-methoxyphenyl chromium tricarbonyl)(phenyl chromium tricarbonyl)methanone (1b-diCr).



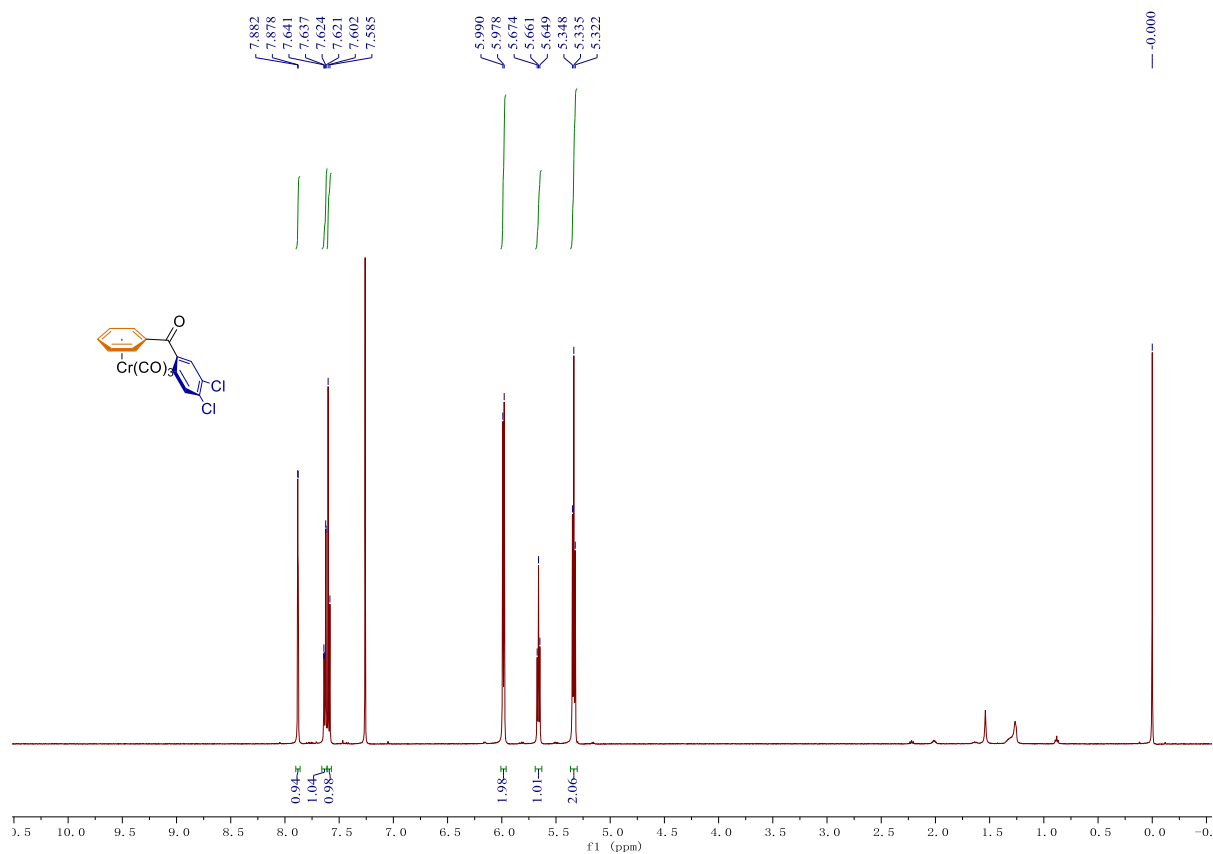
Supplementary Fig. 79 ¹³C NMR (151 MHz, Chloroform-*d*) of (4-methoxyphenyl chromium tricarbonyl)(phenyl chromium tricarbonyl)methanone (1b-diCr).



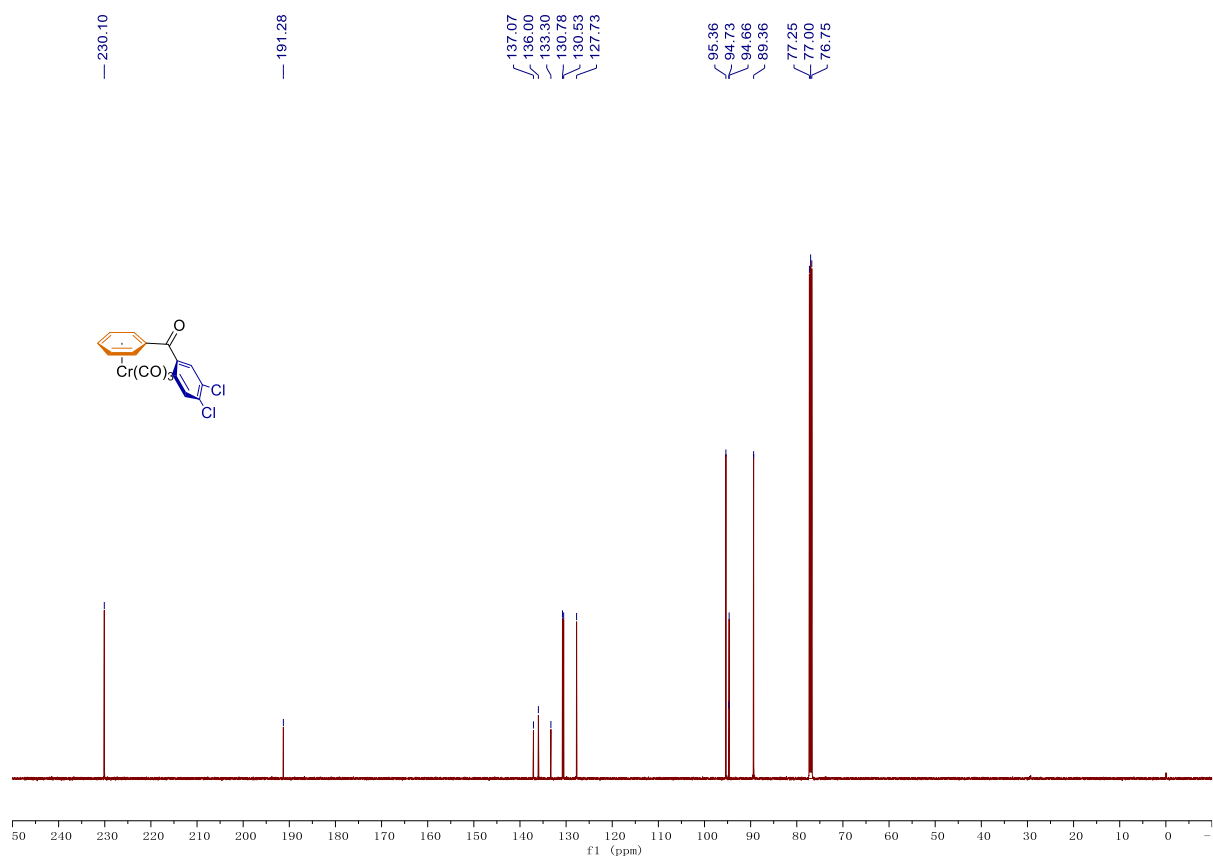
Supplementary Fig. 80 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(((trifluoromethyl)sulfonyl)oxy)benzoylbenzene chromium tricarbonyl (1z-Cr).



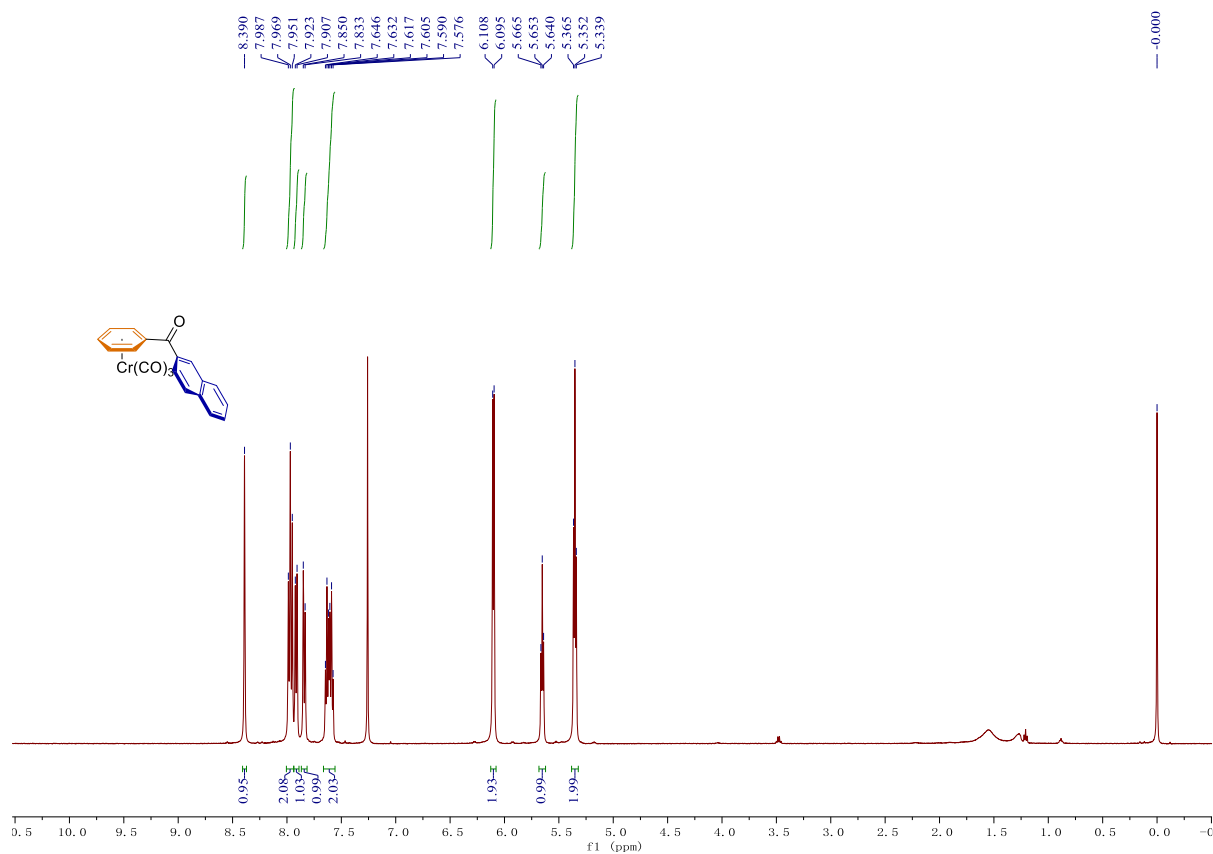
Supplementary Fig. 81 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(((trifluoromethyl)sulfonyl)oxy)benzoylbenzene chromium tricarbonyl (1z-Cr).



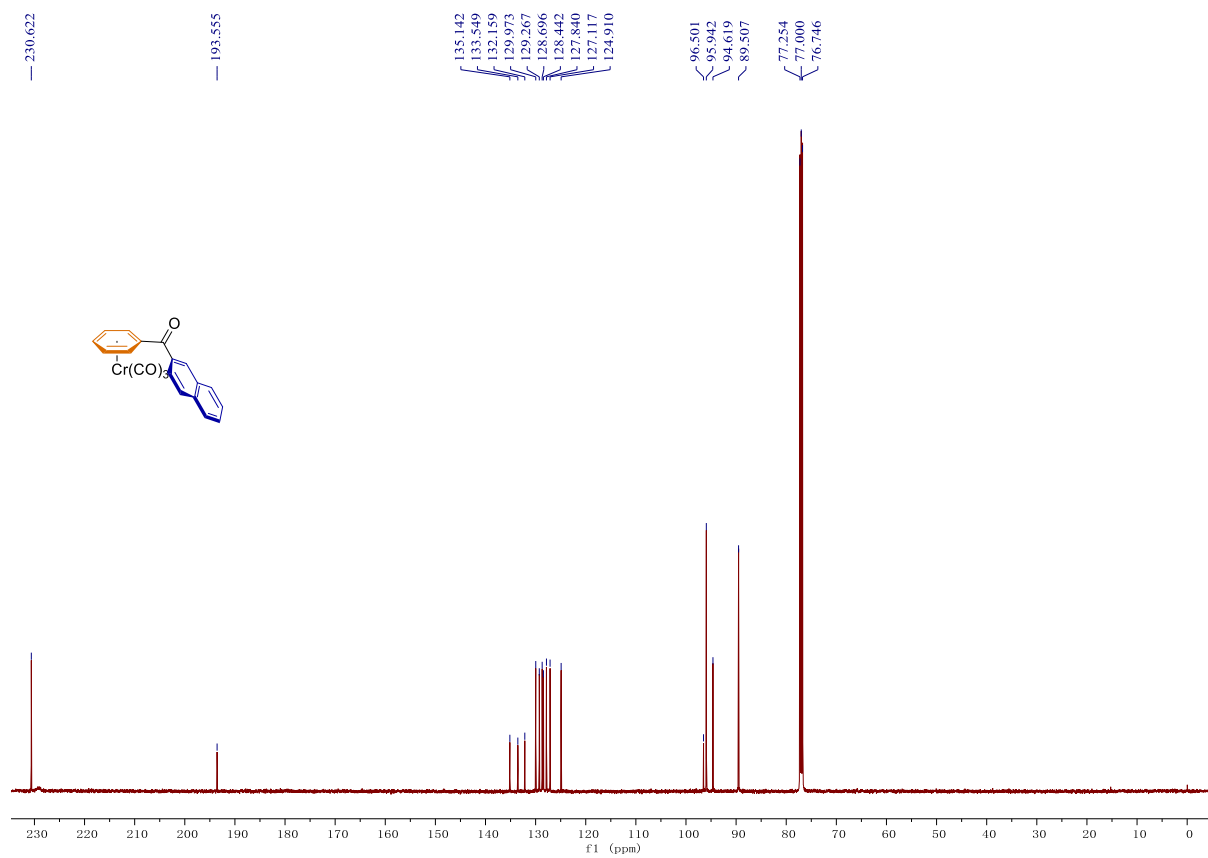
Supplementary Fig. 82 ¹H NMR (500 MHz, Chloroform-*d*) of 3,4-dichlorobenzoylbenzene chromium tricarbonyl (1aa-Cr).



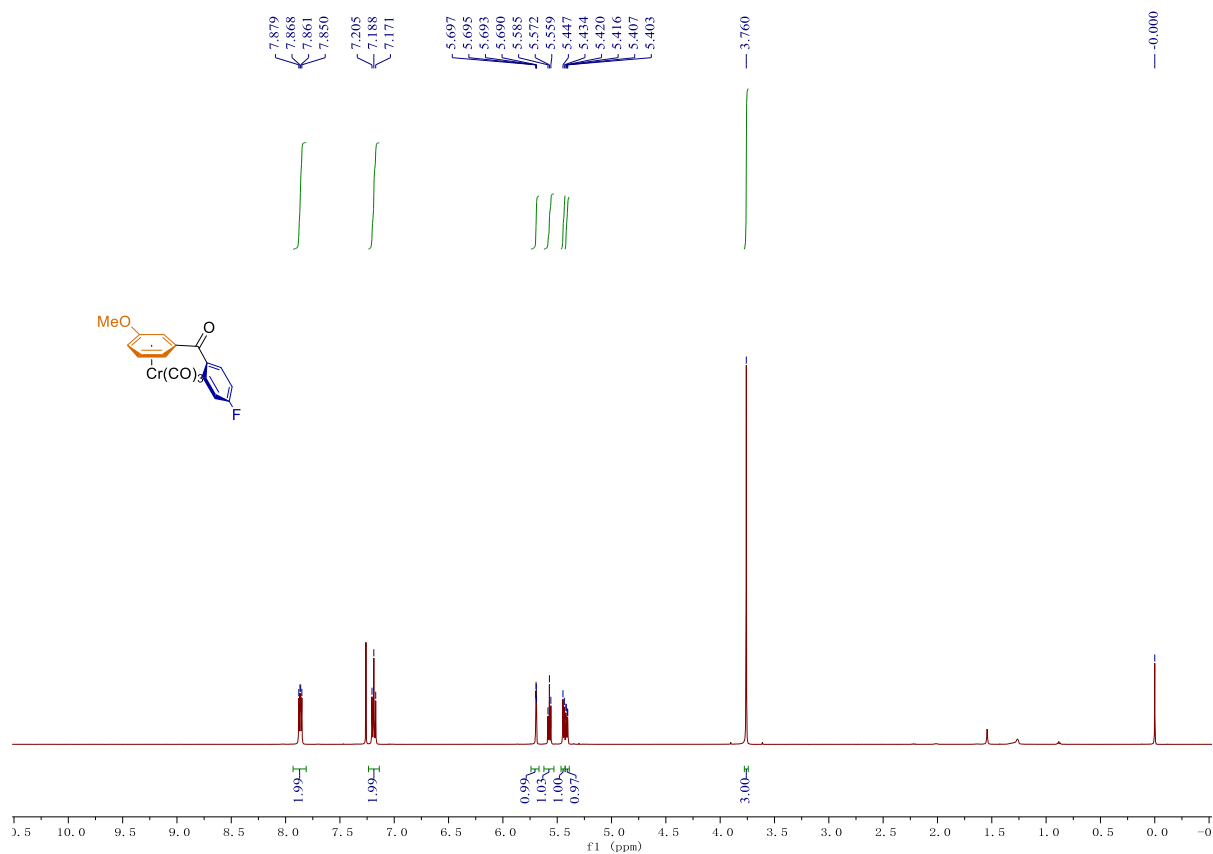
Supplementary Fig. 83 ¹³C NMR (126 MHz, Chloroform-*d*) of 3,4-dichlorobenzoylbenzene chromium tricarbonyl (1aa-Cr).



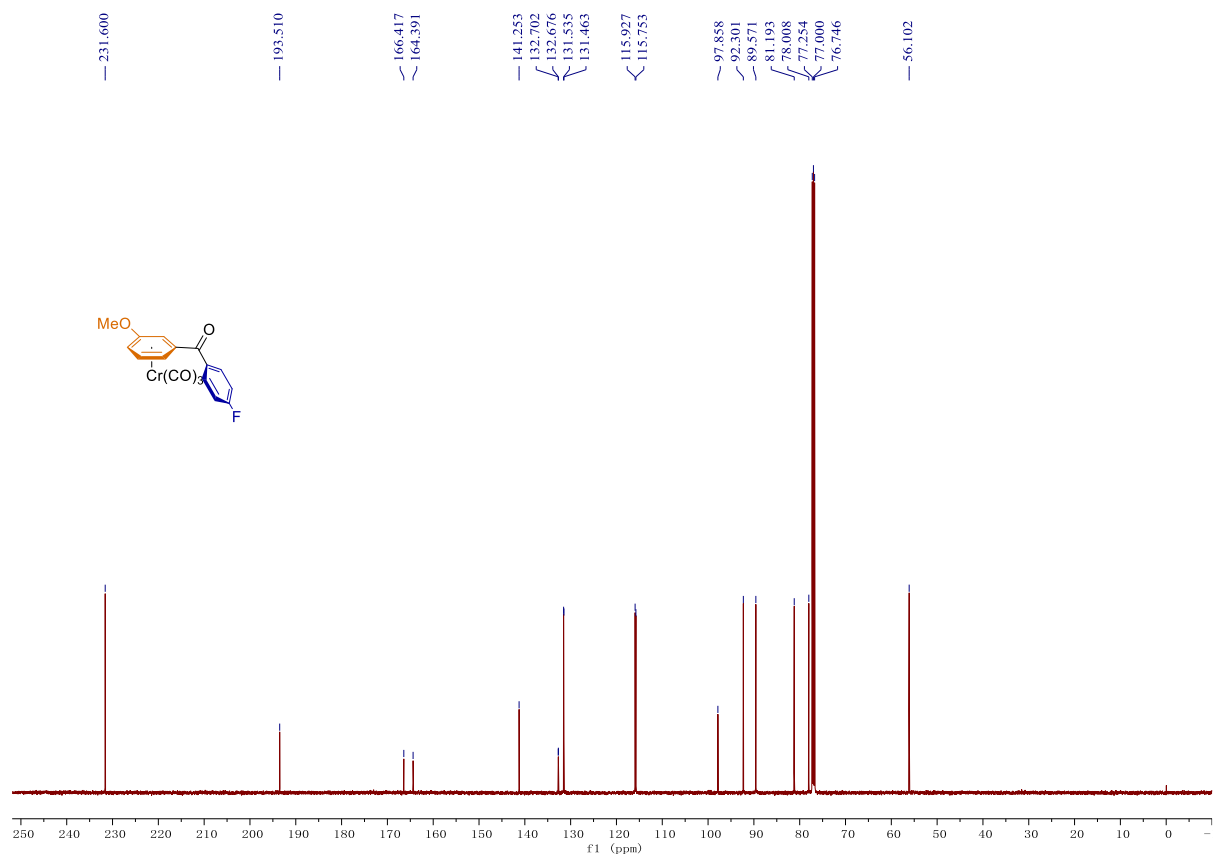
Supplementary Fig. 86 ¹H NMR (500 MHz, Chloroform-*d*) of 2-naphthoylbenzene chromium tricarbonyl (1ab-Cr).



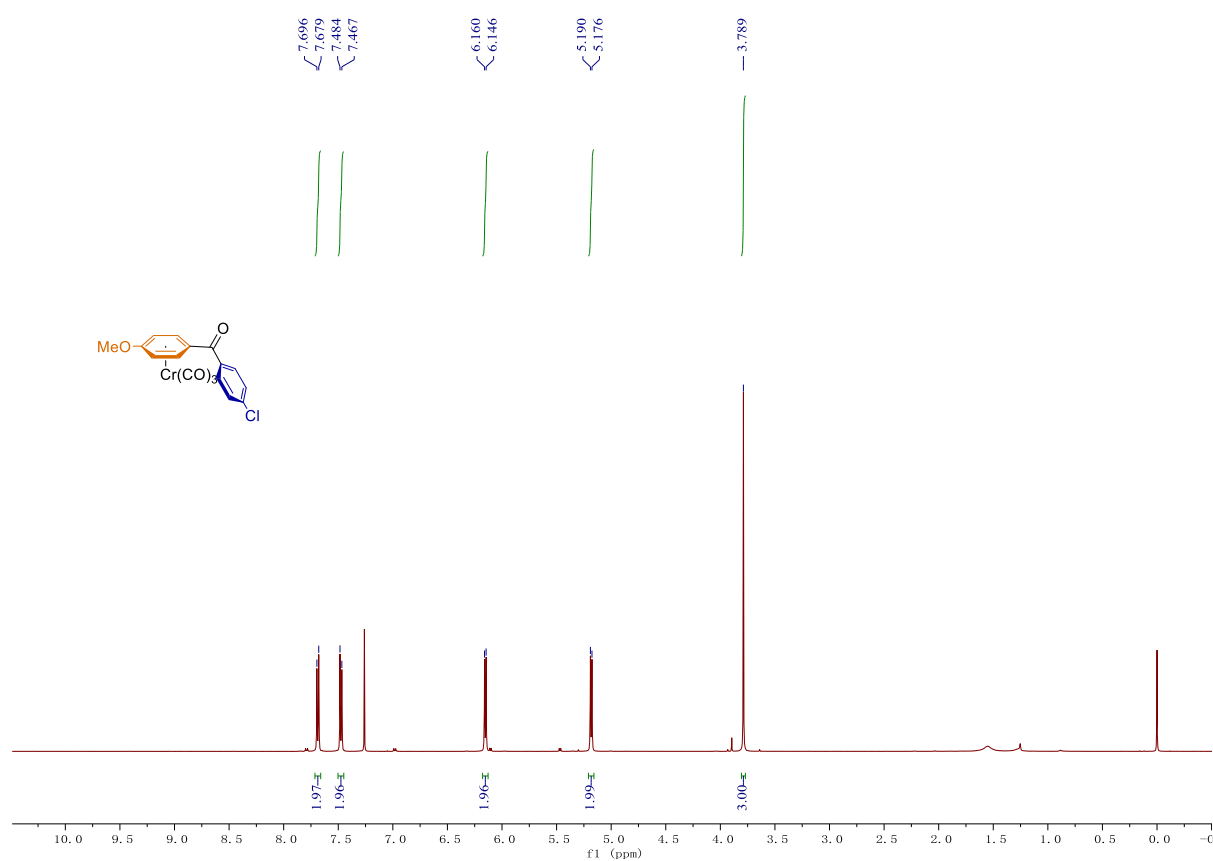
Supplementary Fig. 87 ¹³C NMR (126 MHz, Chloroform-*d*) of 2-naphthoylbenzene chromium tricarbonyl (1ab-Cr).



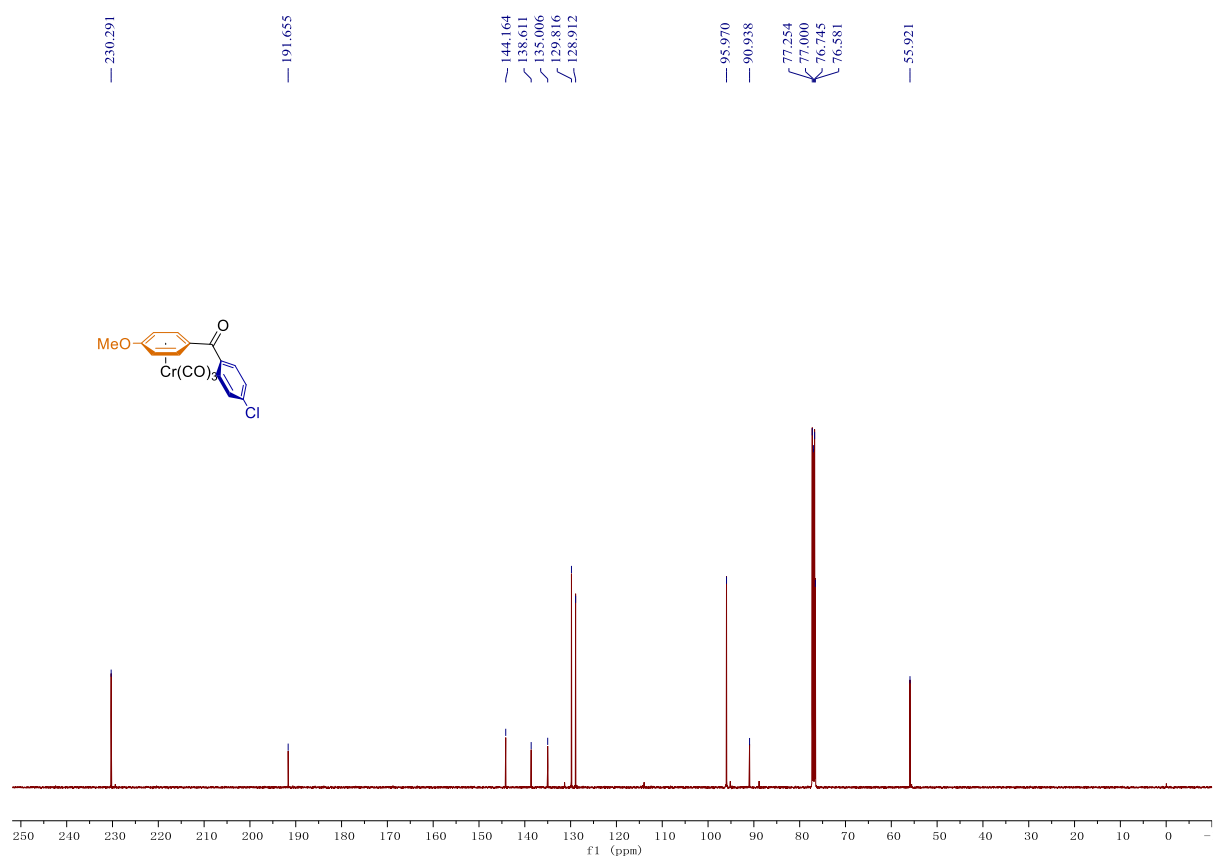
Supplementary Fig. 88 ¹H NMR (500 MHz, Chloroform-*d*) of 3-(4-fluorobenzoyl)anisole chromium tricarbonyl (11-Cr).



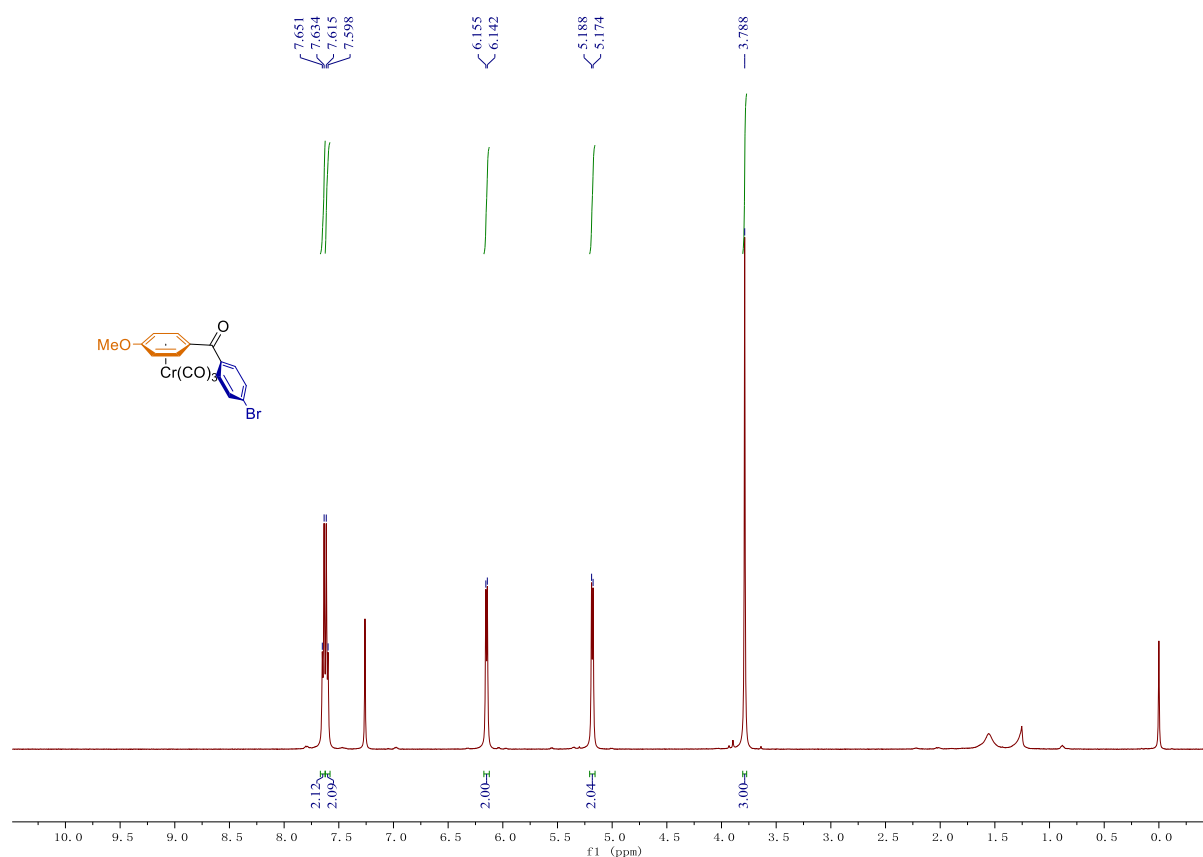
Supplementary Fig. 89 ¹³C NMR (126 MHz, Chloroform-*d*) of 3-(4-fluorobenzoyl)anisole chromium tricarbonyl (11-Cr).



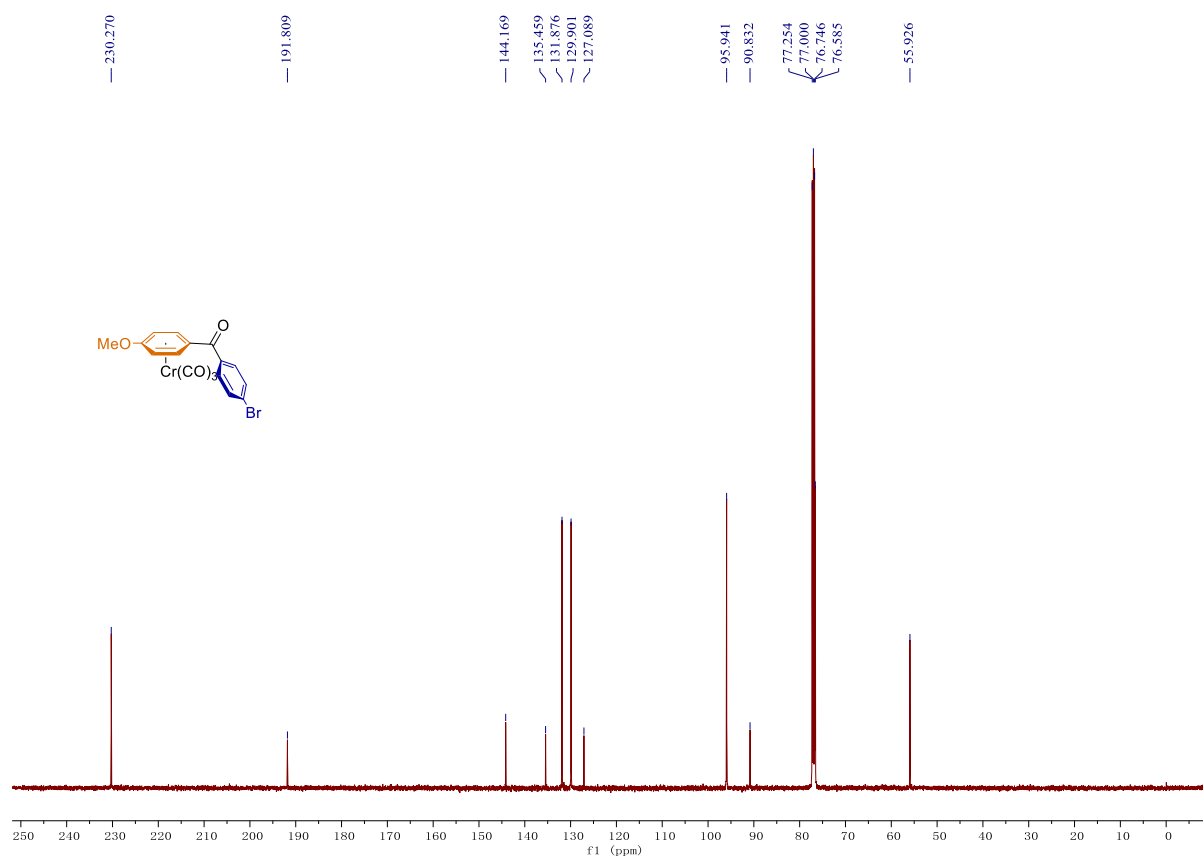
Supplementary Fig. 90 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(4-chlorobenzoyl)anisole chromium tricarbonyl (1m-Cr).



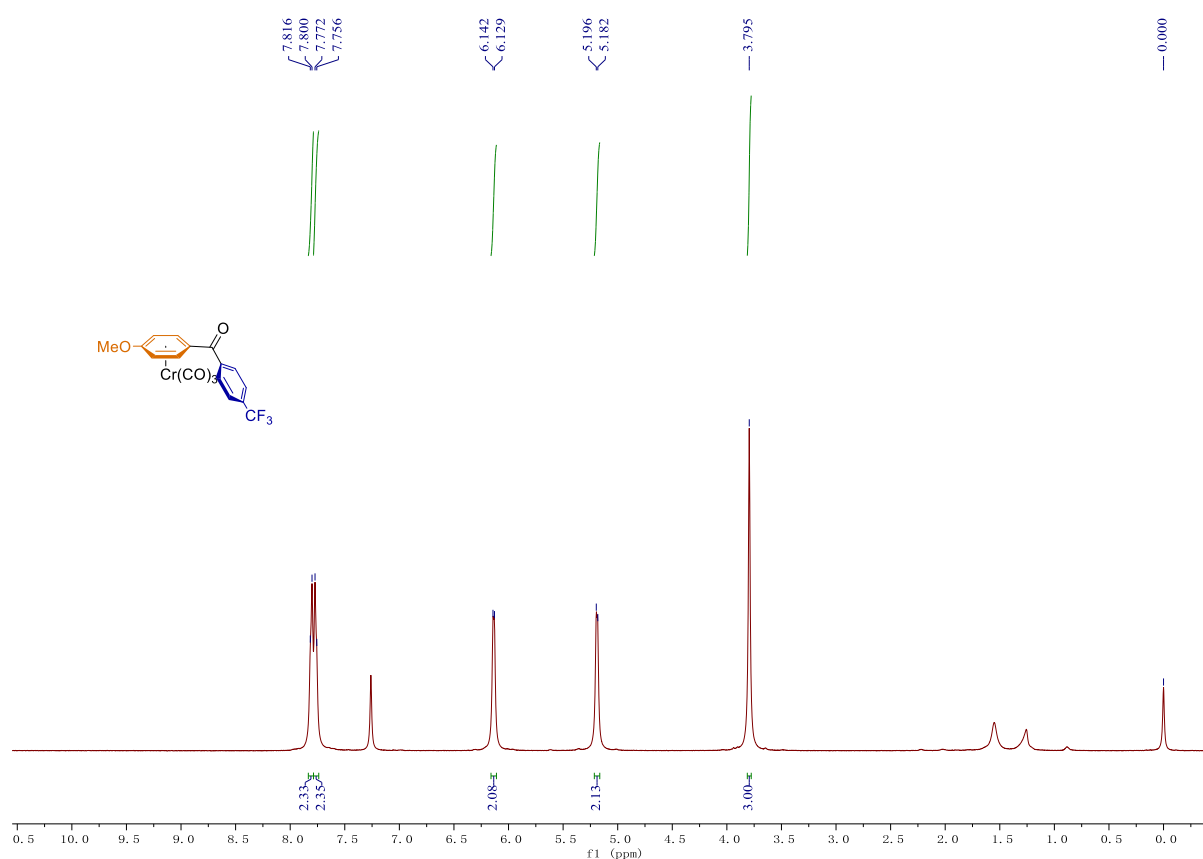
Supplementary Fig. 91 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(4-chlorobenzoyl)anisole chromium tricarbonyl (1m-Cr).



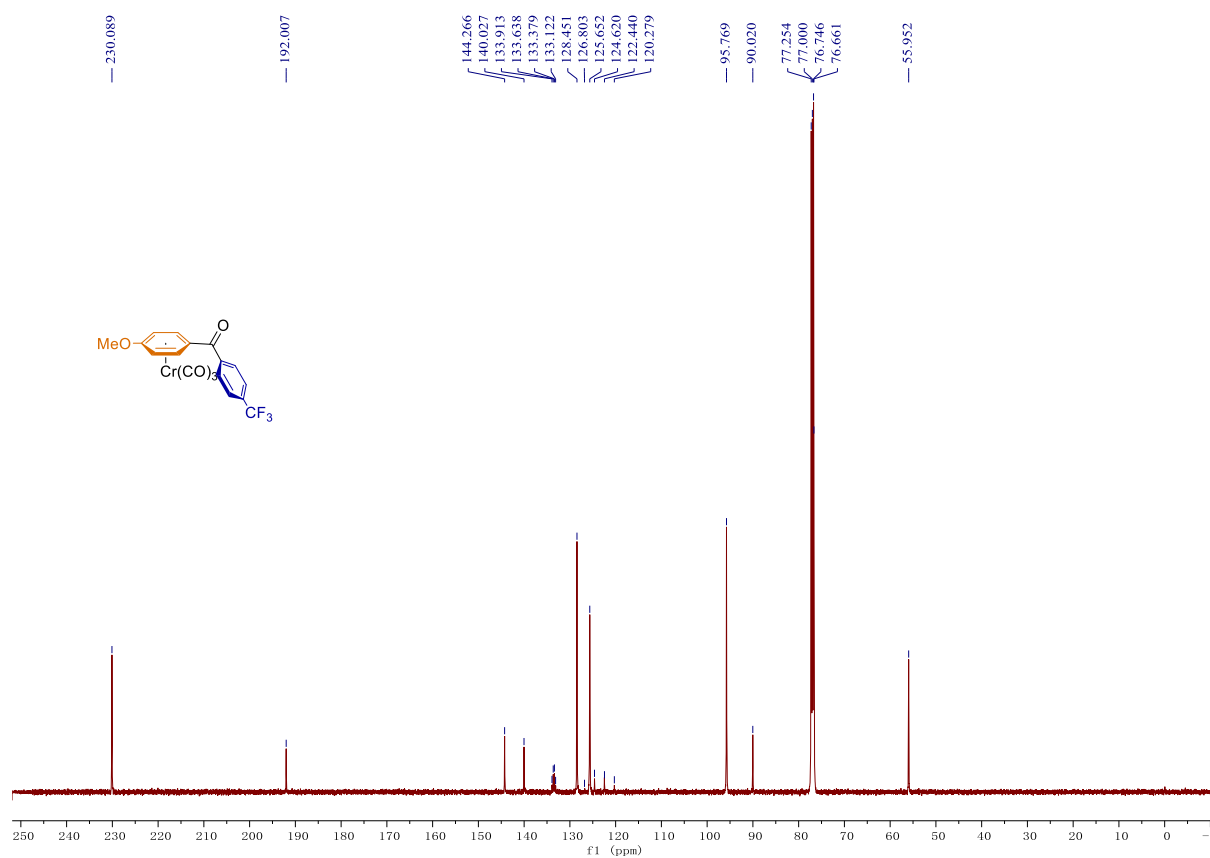
Supplementary Fig. 92 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(4-bromobenzoyl)anisole chromium tricarbonyl (1n-Cr).



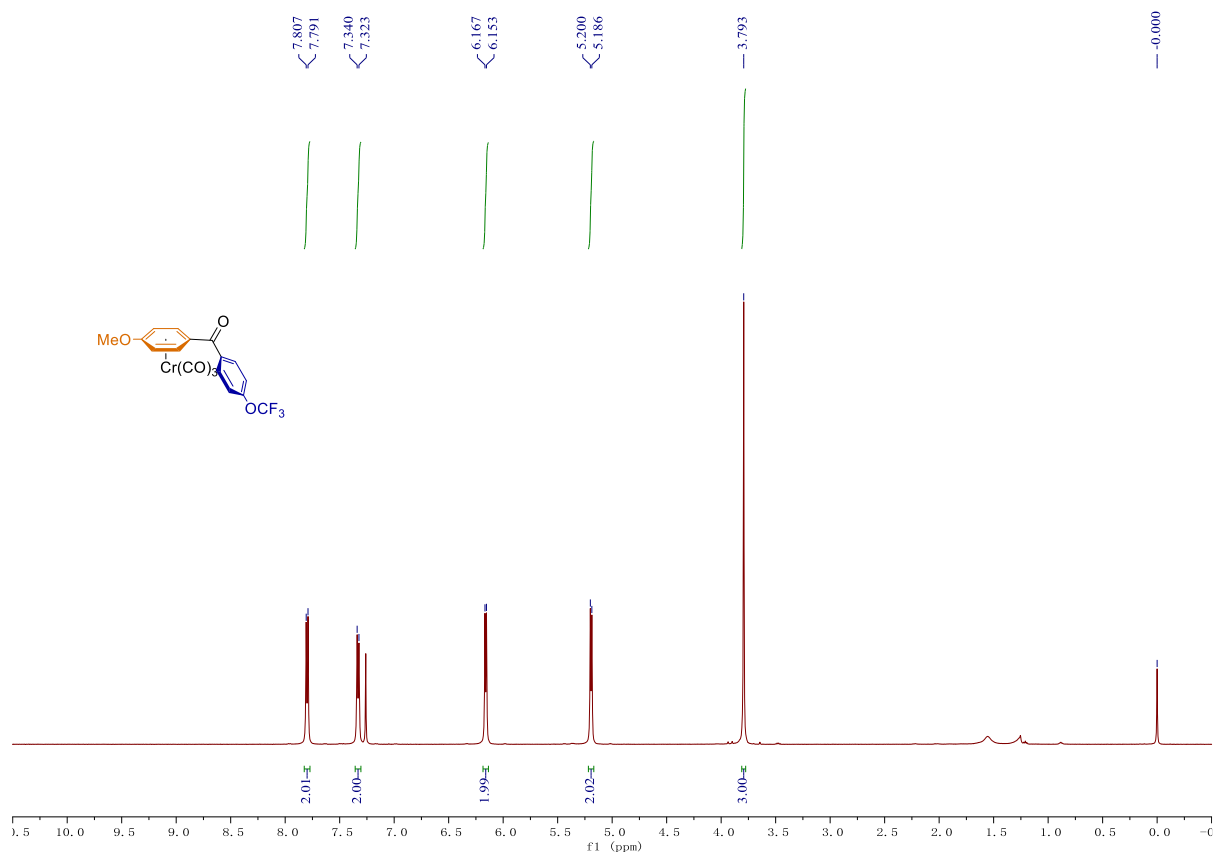
Supplementary Fig. 93 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(4-bromobenzoyl)anisole chromium tricarbonyl (1n-Cr).



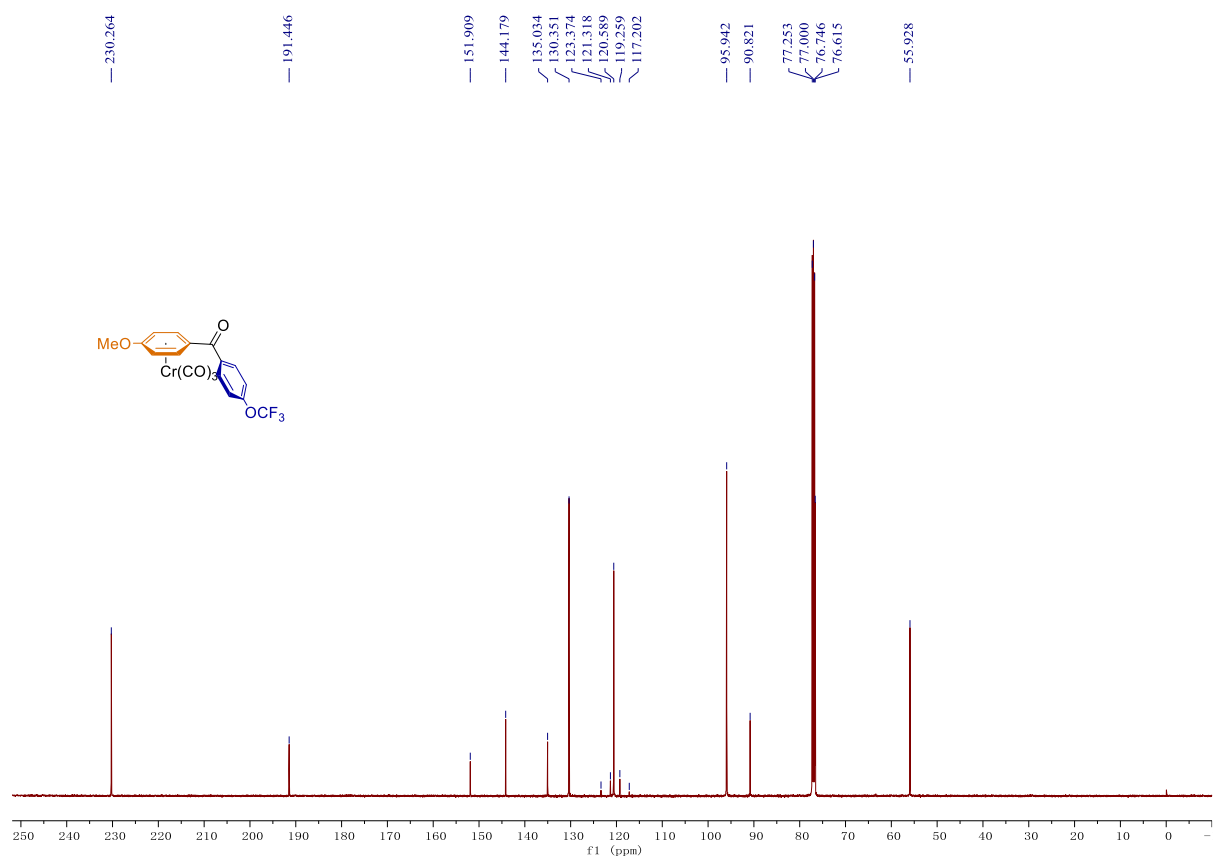
Supplementary Fig. 94 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(4-(trifluoromethyl)benzoyl)anisole chromium tricarbonyl (1o-Cr).



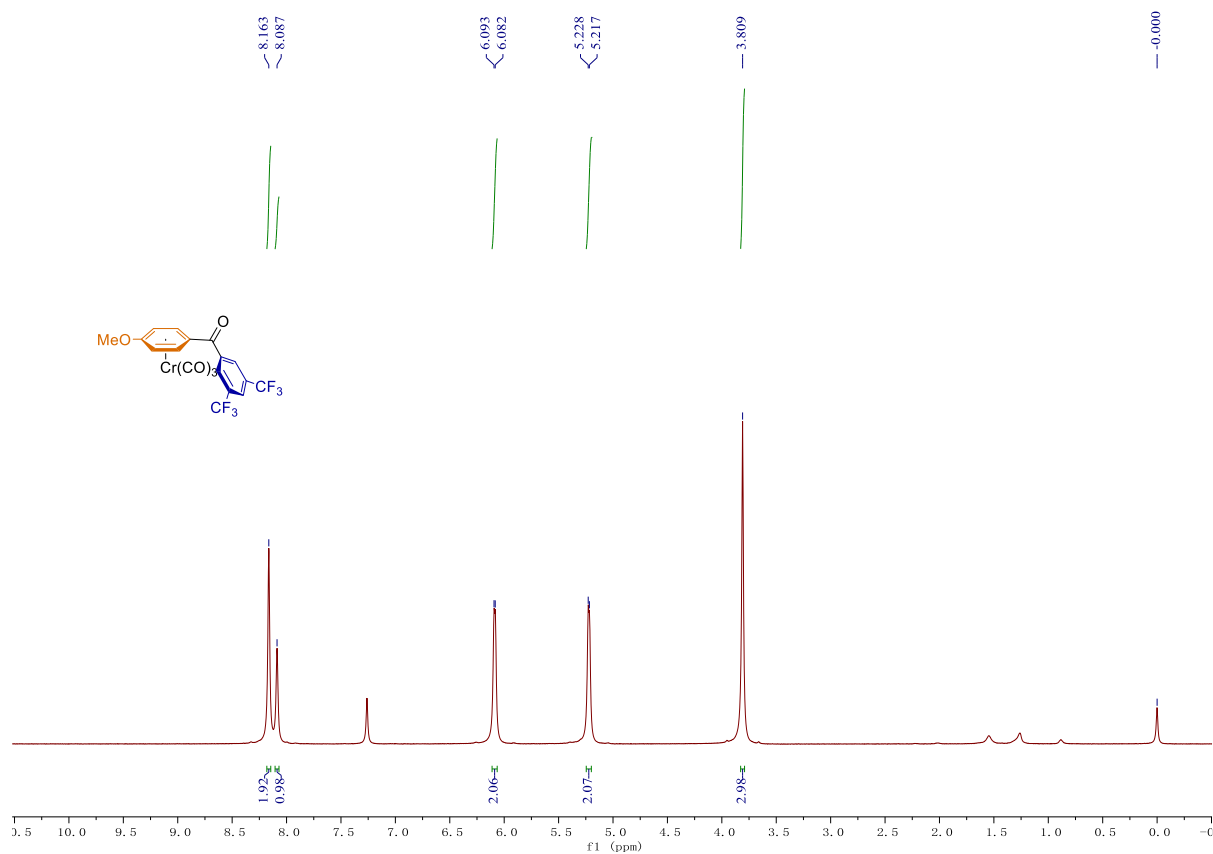
Supplementary Fig. 95 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(4-(trifluoromethyl)benzoyl)anisole chromium tricarbonyl (1o-Cr).



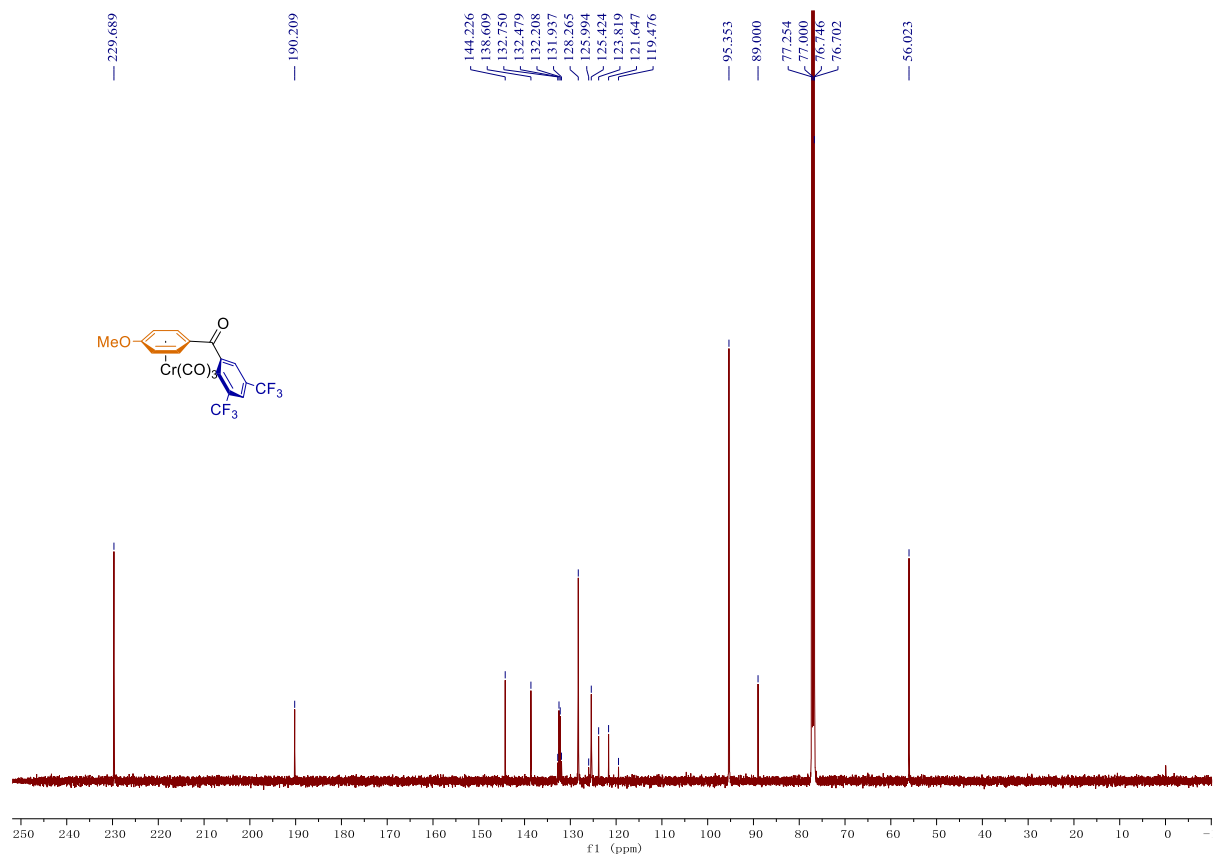
Supplementary Fig. 96 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(4-(trifluoromethoxy)benzoyl)anisole chromium tricarbonyl (1p-Cr).



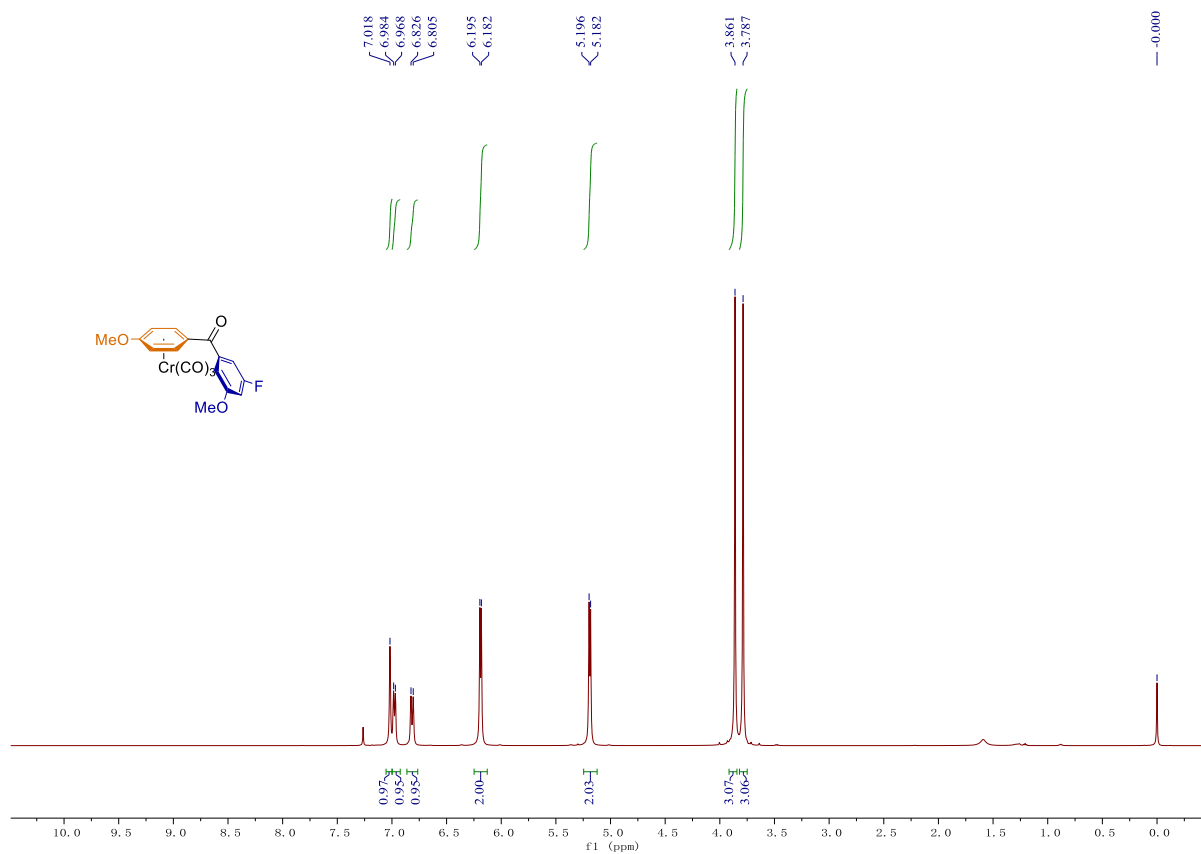
Supplementary Fig. 97 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(4-(trifluoromethoxy)benzoyl)anisole chromium tricarbonyl (1p-Cr).



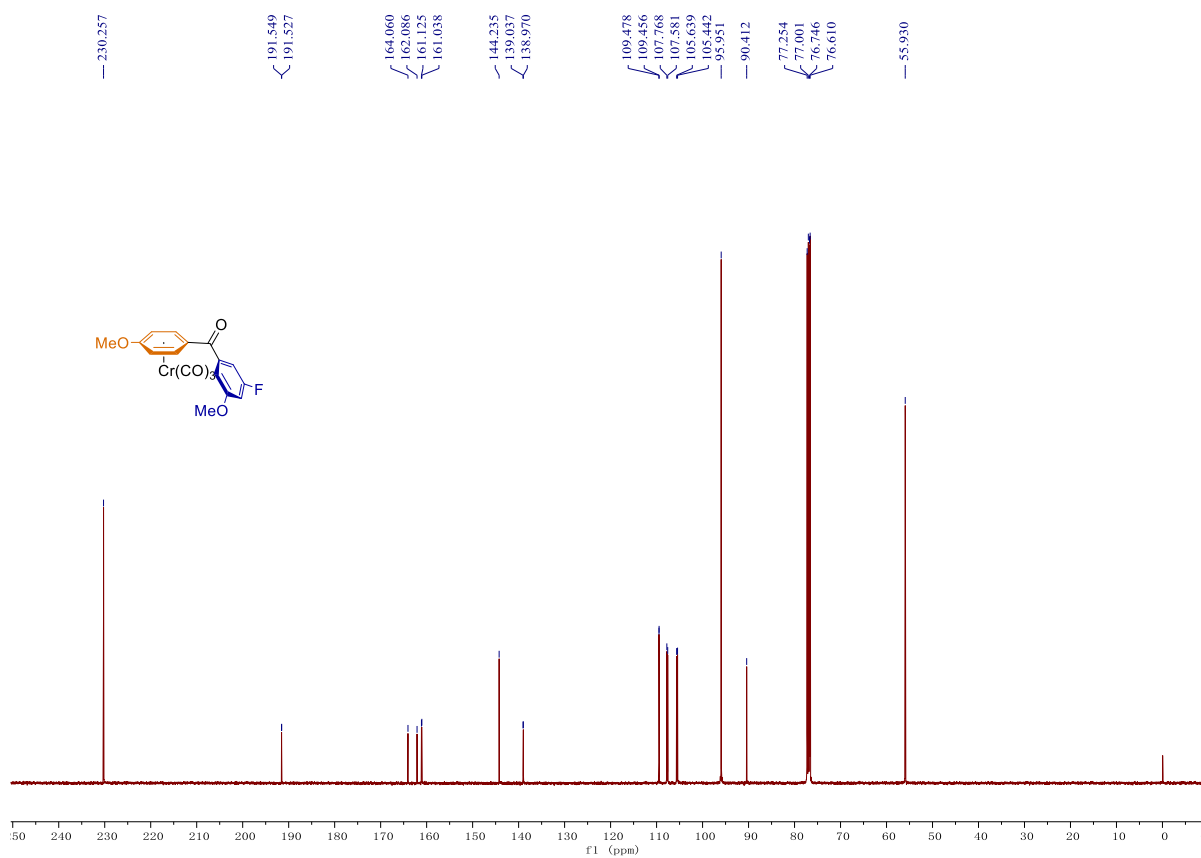
Supplementary Fig. 98 $^1\text{H NMR}$ (500 MHz, $\text{Chloroform-}d$) of 4-(3,5-bis(trifluoromethyl)benzoyl)anisole chromium tricarbonyl (s3-Cr).



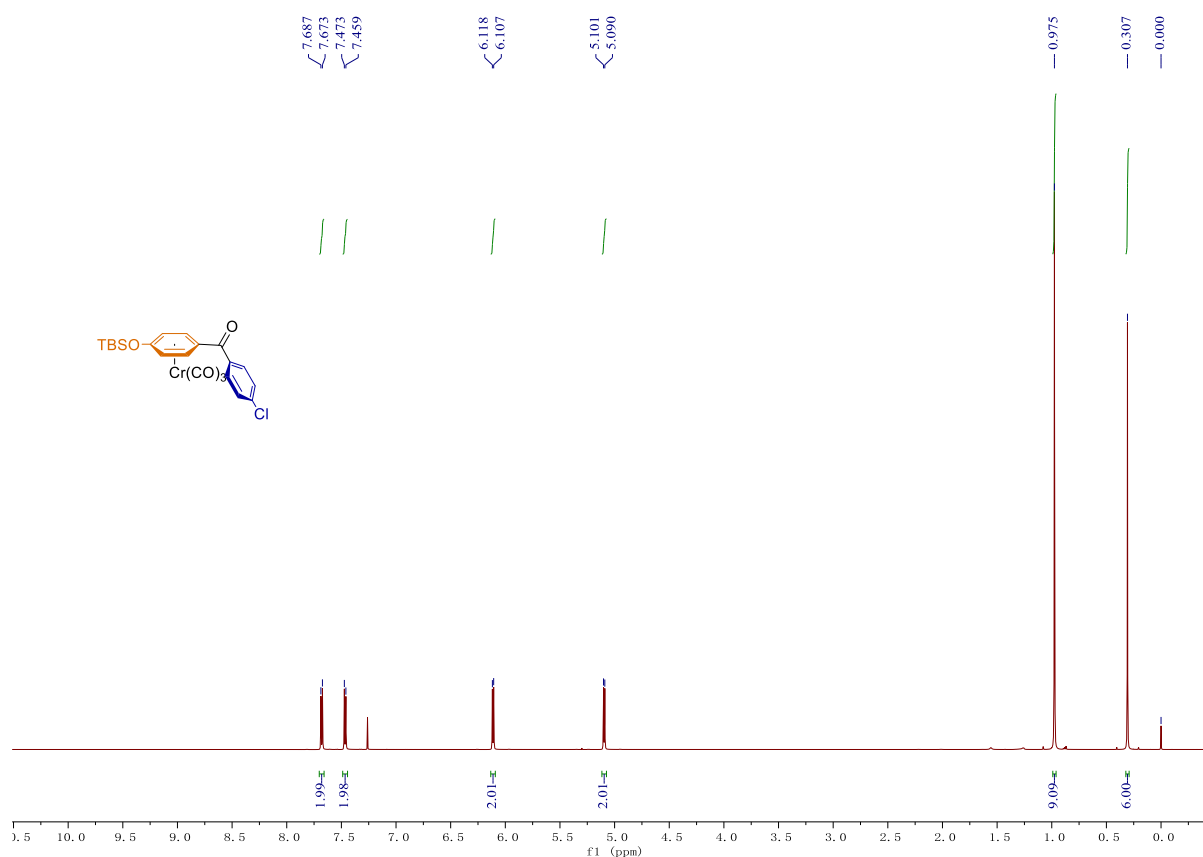
Supplementary Fig. 99 $^{13}\text{C NMR}$ (126 MHz, $\text{Chloroform-}d$) of 4-(3,5-bis(trifluoromethyl)benzoyl)anisole chromium tricarbonyl (s3-Cr).



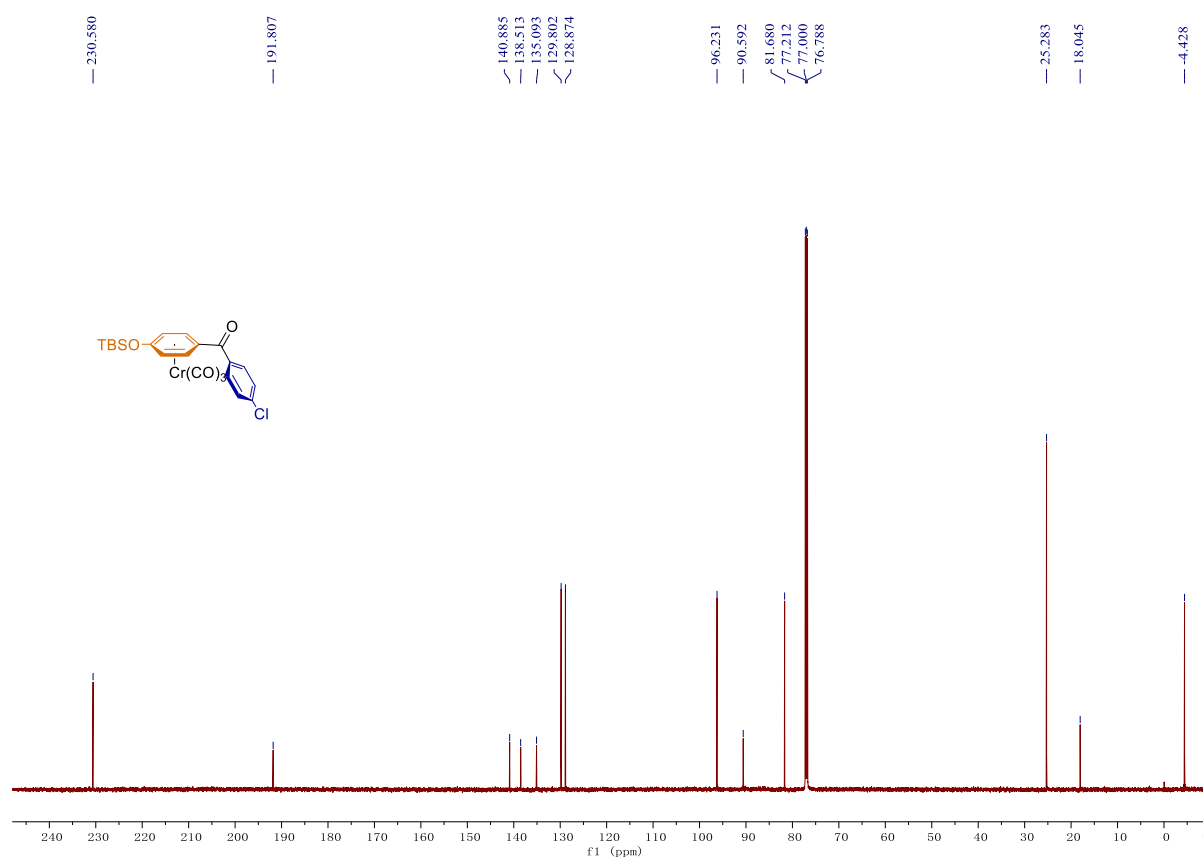
Supplementary Fig. 100 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(3-methoxy-5-fluorobenzoyl)anisole chromium tricarbonyl (1v-Cr).



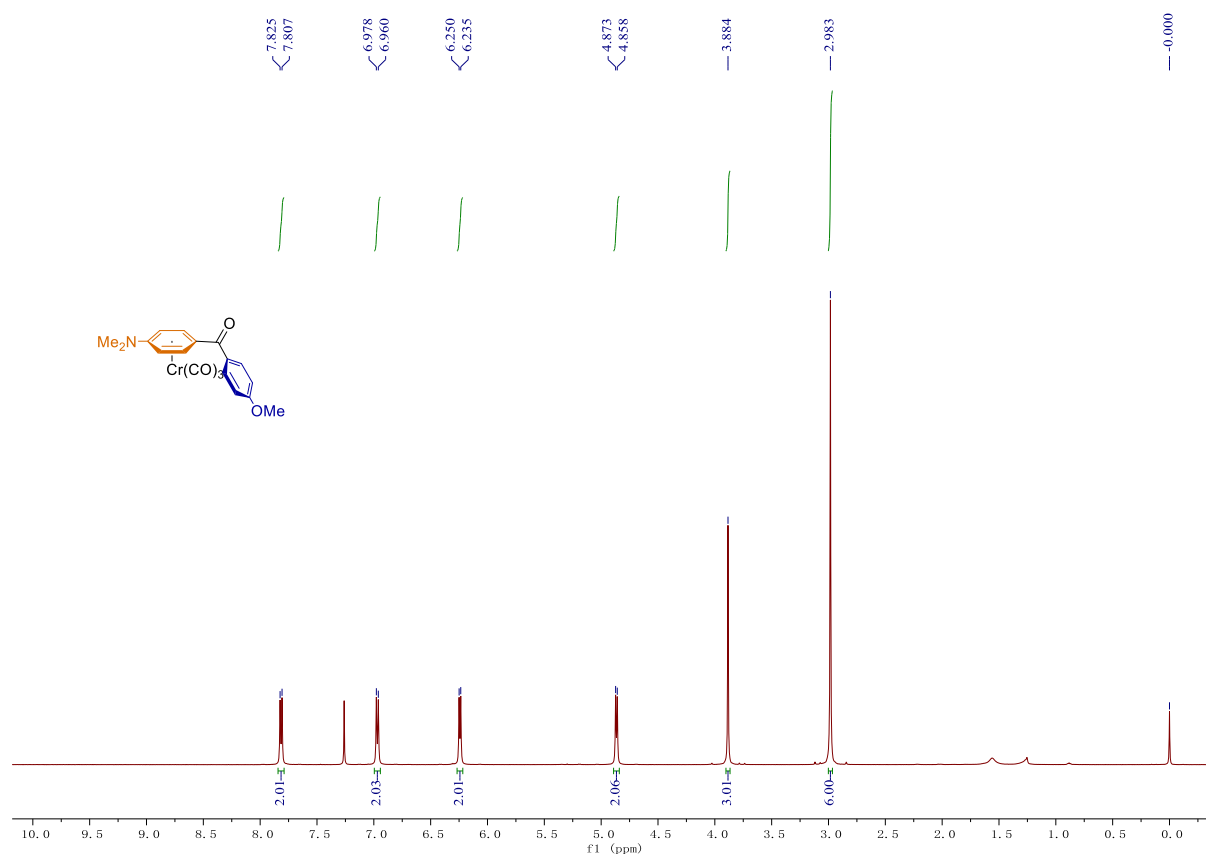
Supplementary Fig. 101 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(3-methoxy-5-fluorobenzoyl)anisole chromium tricarbonyl (1v-Cr).



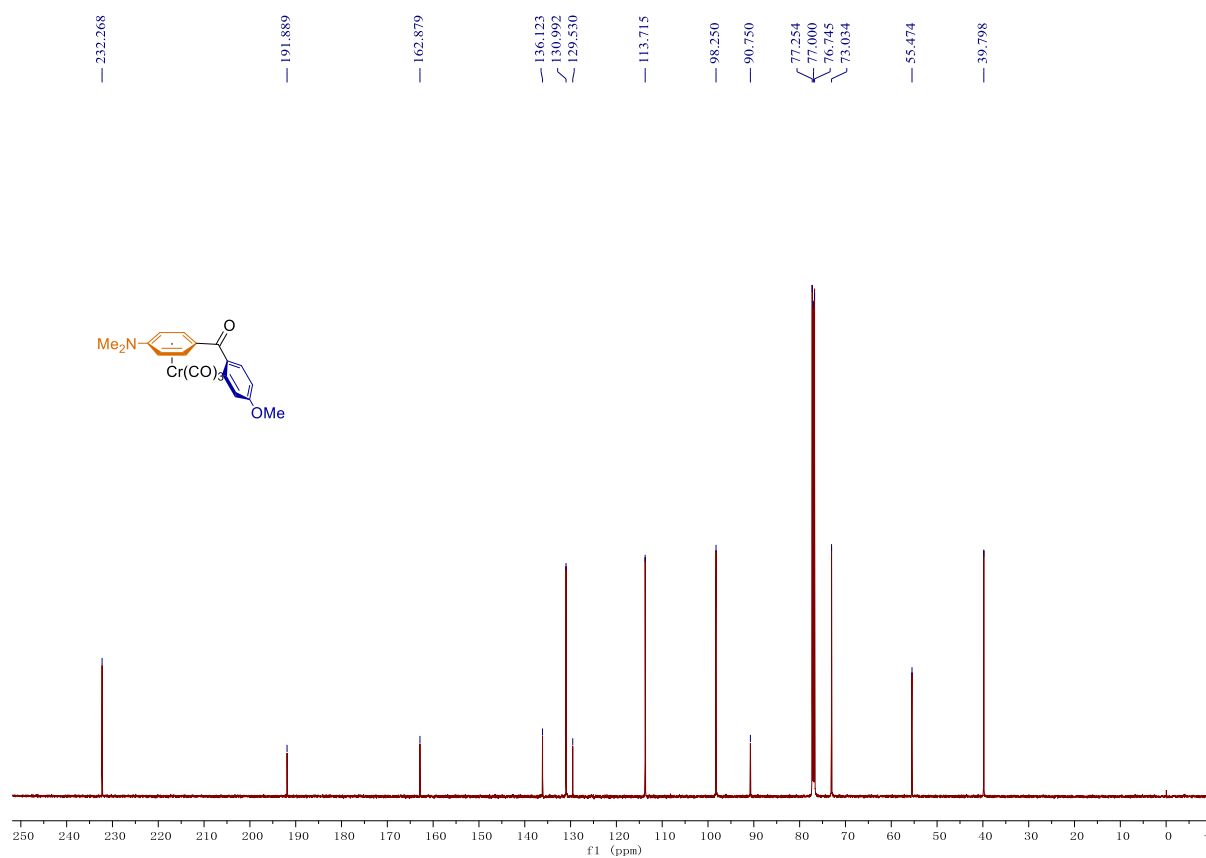
Supplementary Fig. 102 ¹H NMR (600 MHz, Chloroform-*d*) of 4-(4-chlorobenzoyl)((tert-butyldimethylsilyl)oxy)benzene chromium tricarbonyl (1q-Cr).



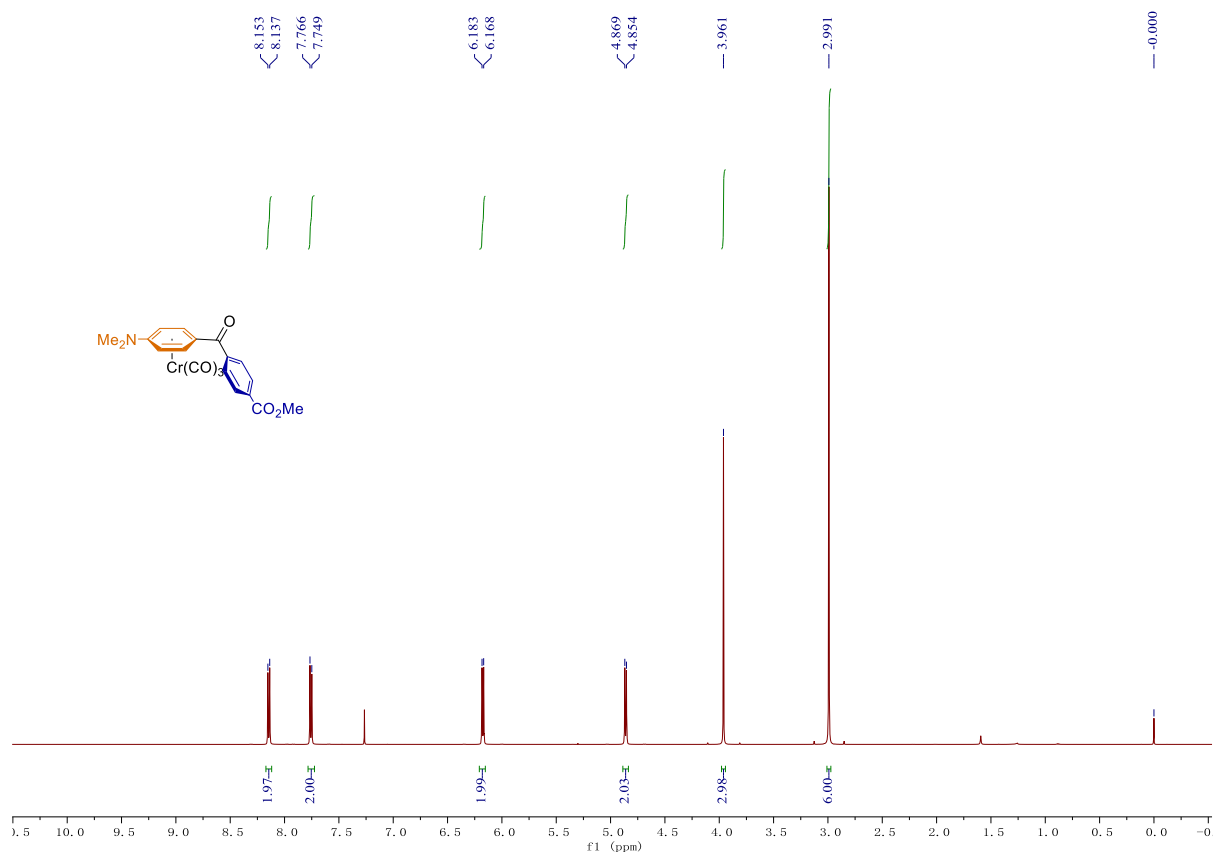
Supplementary Fig. 103 ¹³C NMR (151 MHz, Chloroform-*d*) of 4-(4-chlorobenzoyl)((tert-butyldimethylsilyl)oxy)benzene chromium tricarbonyl (1q-Cr).



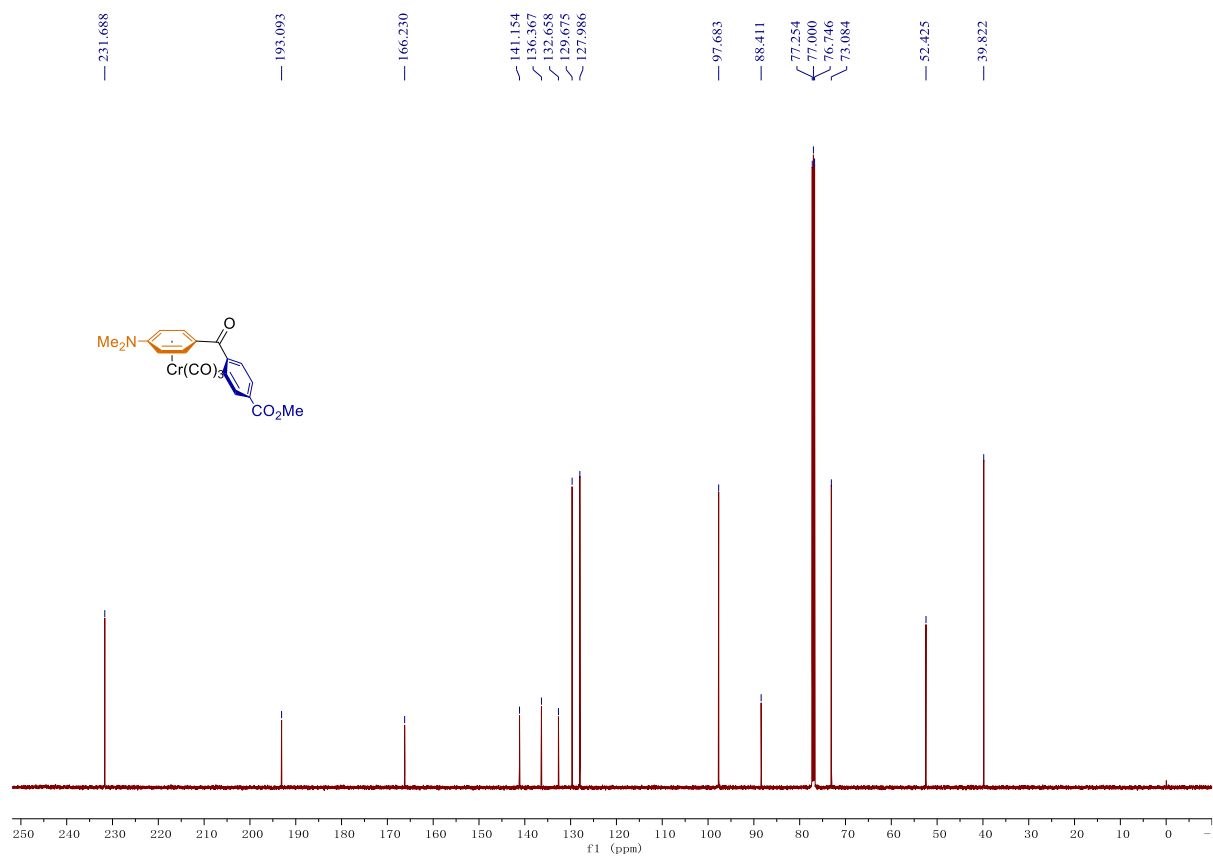
Supplementary Fig. 104 ^1H NMR (500 MHz, CDCl_3) of 4-(4-methoxybenzoyl)nn-dimethylaniline chromium tricarbonyl (1r-Cr).



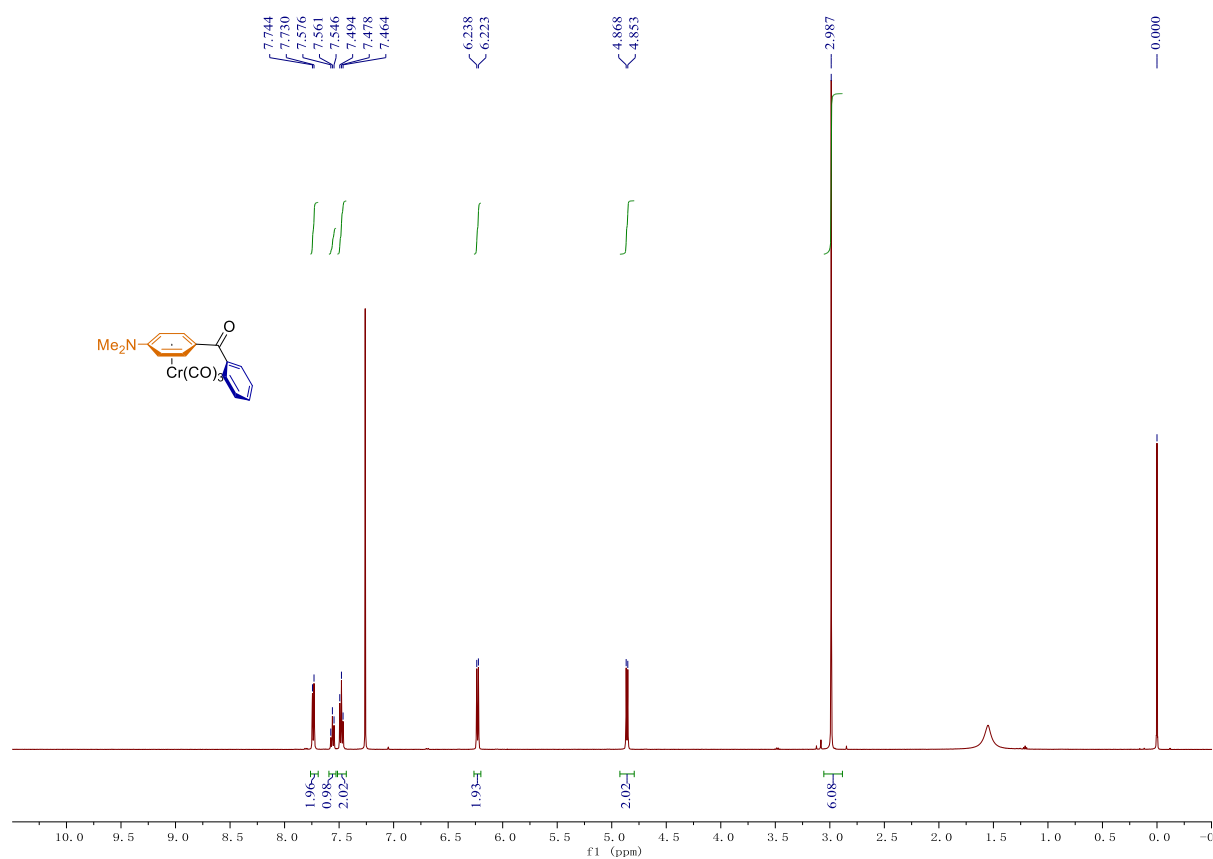
Supplementary Fig. 105 ^{13}C NMR (126 MHz, CDCl_3) of 4-(4-methoxybenzoyl)nn-dimethylaniline chromium tricarbonyl (1r-Cr).



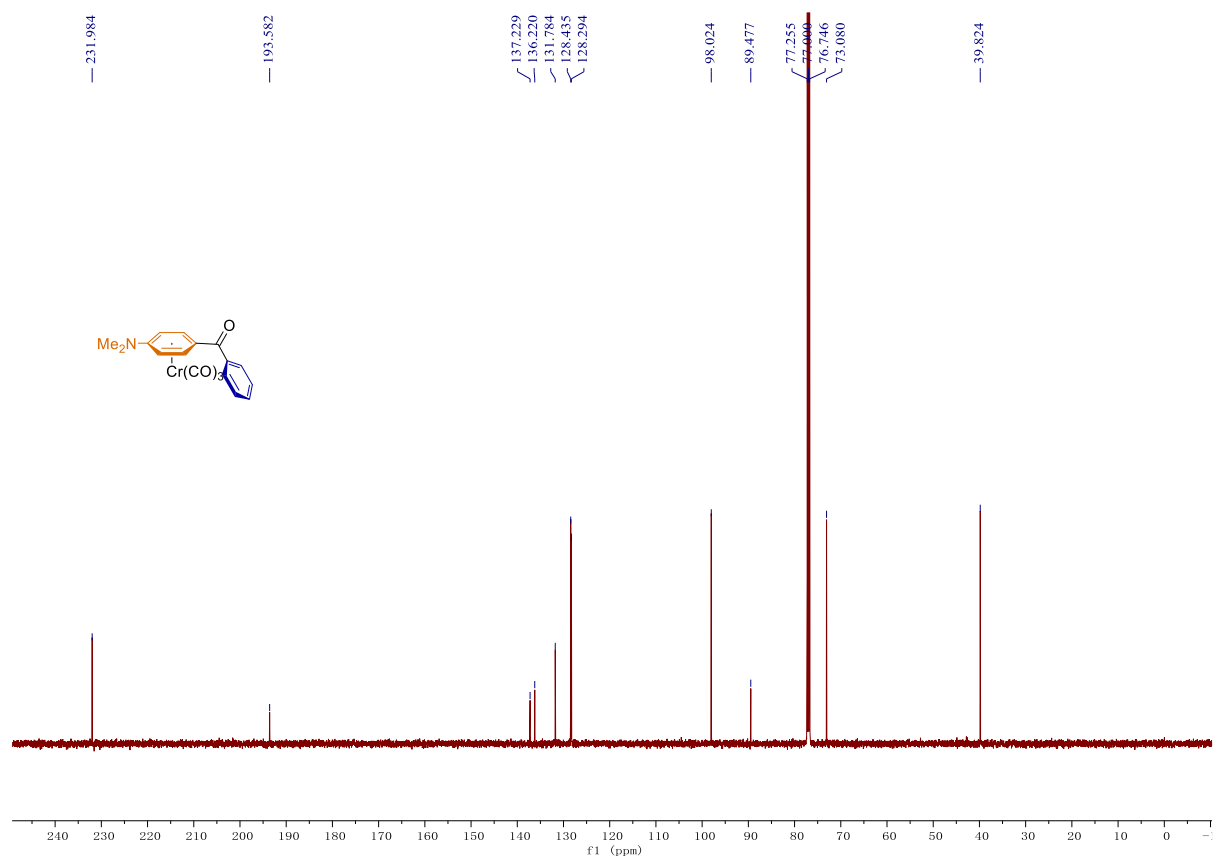
Supplementary Fig. 106 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(4-methoxycarbonyl)nn-dimethylaniline chromium tricarbonyl (1s-Cr).



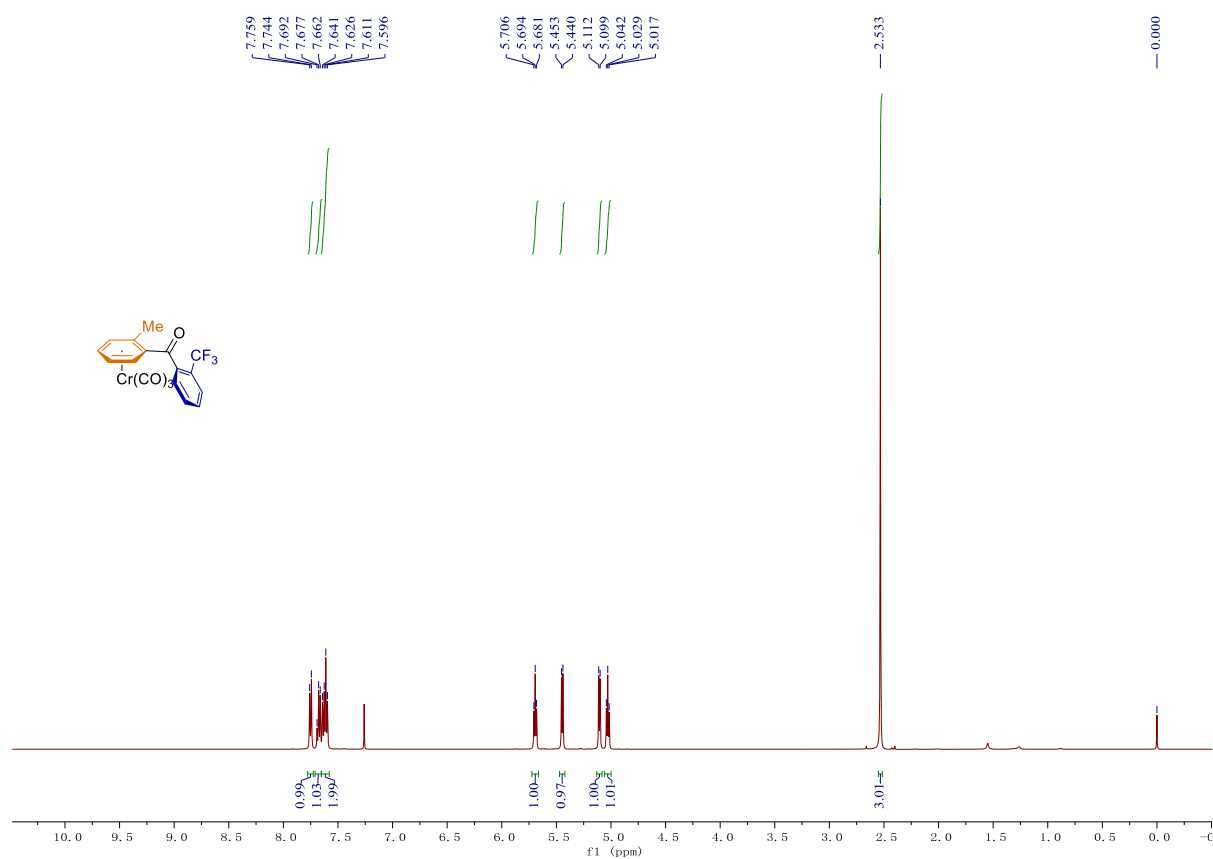
Supplementary Fig. 107 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(4-methoxycarbonyl)nn-dimethylaniline chromium tricarbonyl (1s-Cr).



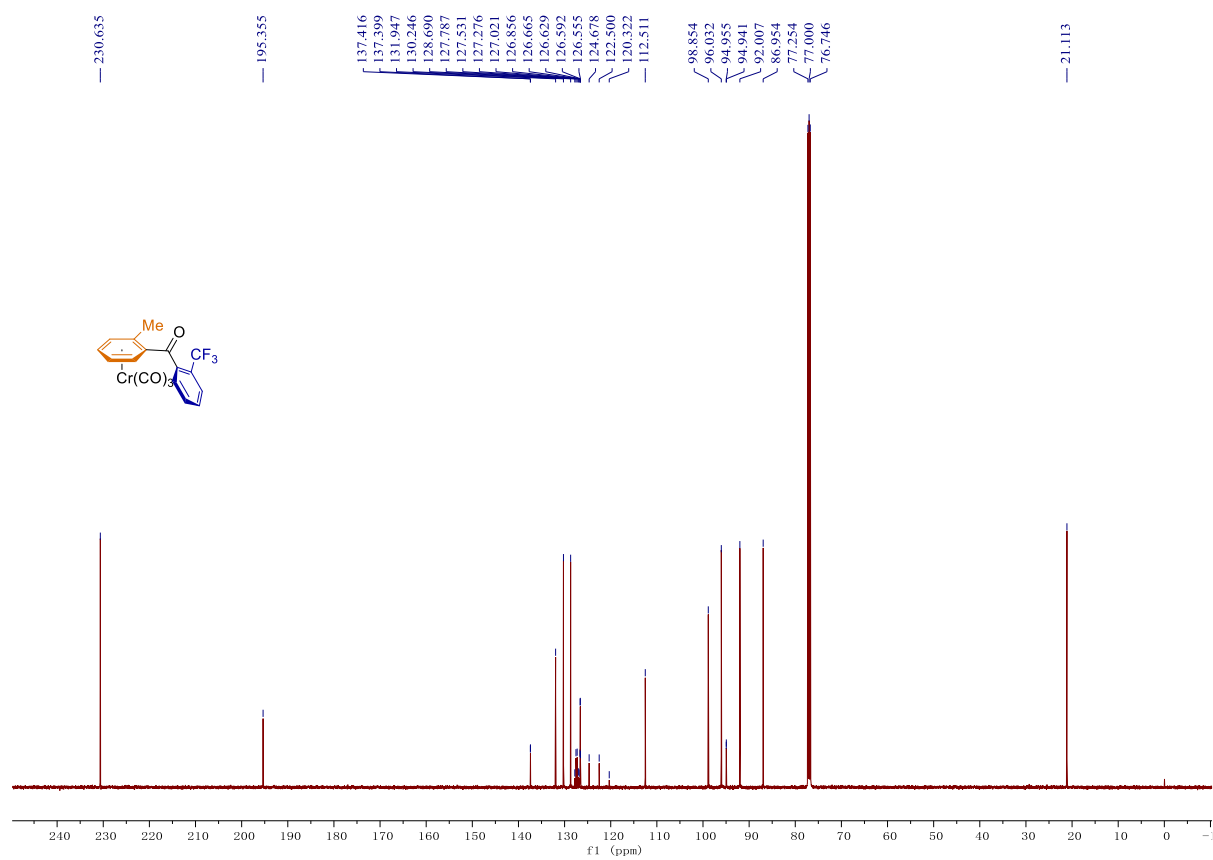
Supplementary Fig. 108 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(benzoyl)nn-dimethylaniline chromium tricarbonyl (1j-Cr).



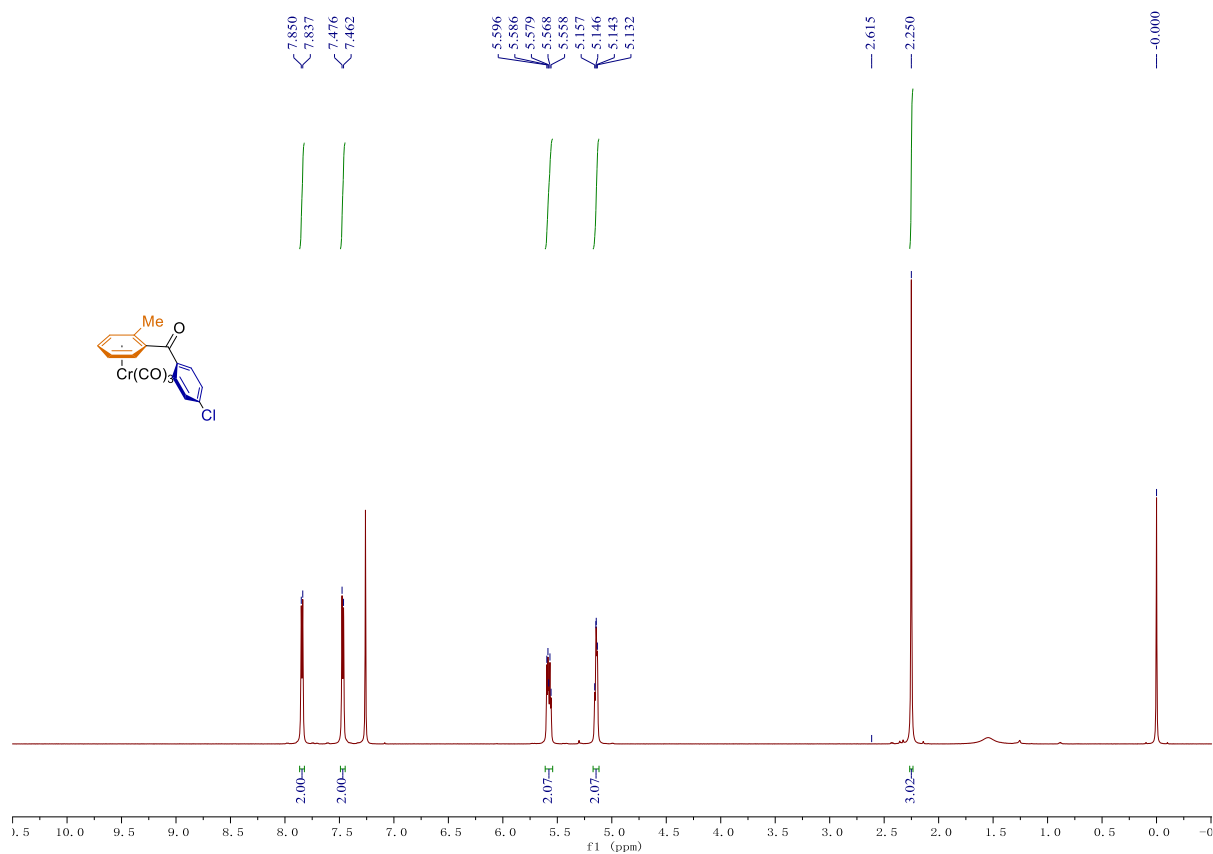
Supplementary Fig. 109 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(benzoyl)nn-dimethylaniline chromium tricarbonyl (1j-Cr).



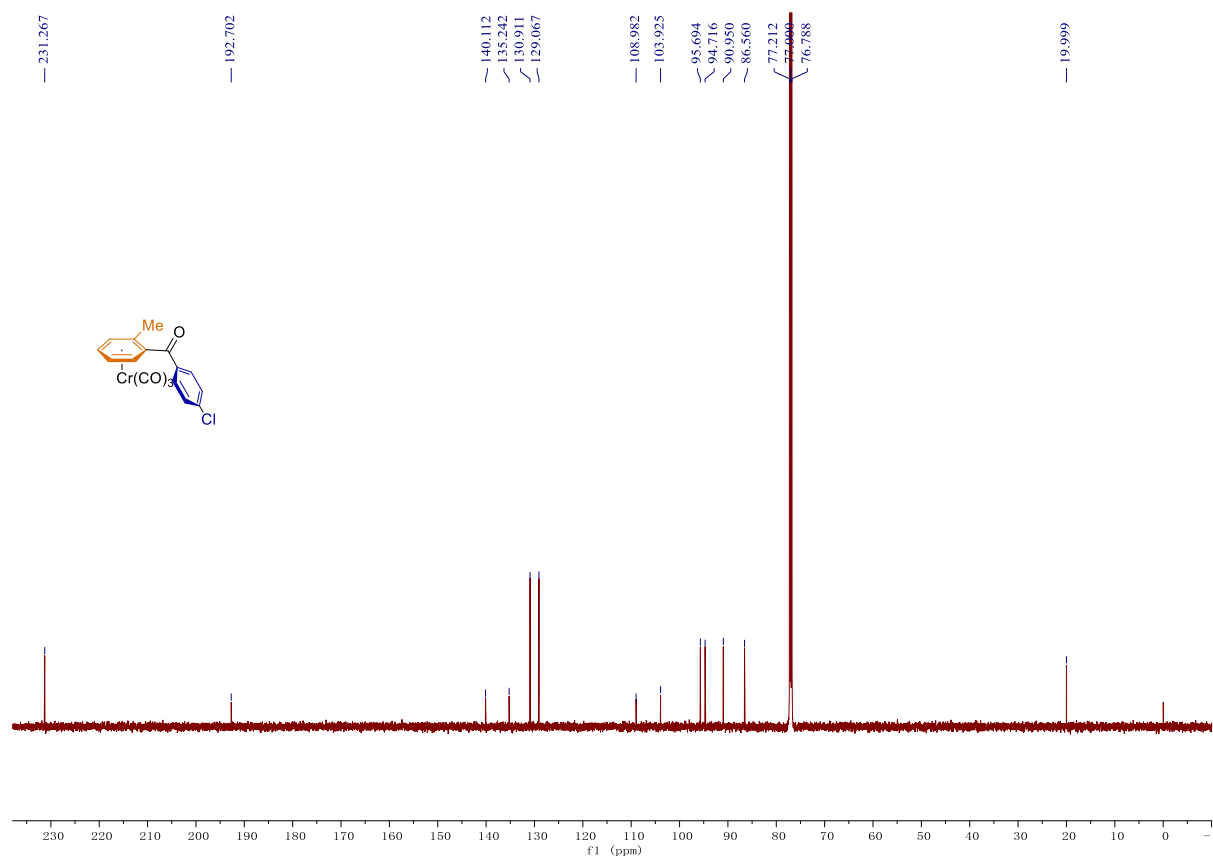
Supplementary Fig. 110 ¹H NMR (500 MHz, Chloroform-*d*) of 2-(2-(trifluoromethyl)benzoyl)toluene chromium tricarbonyl (1u-Cr).



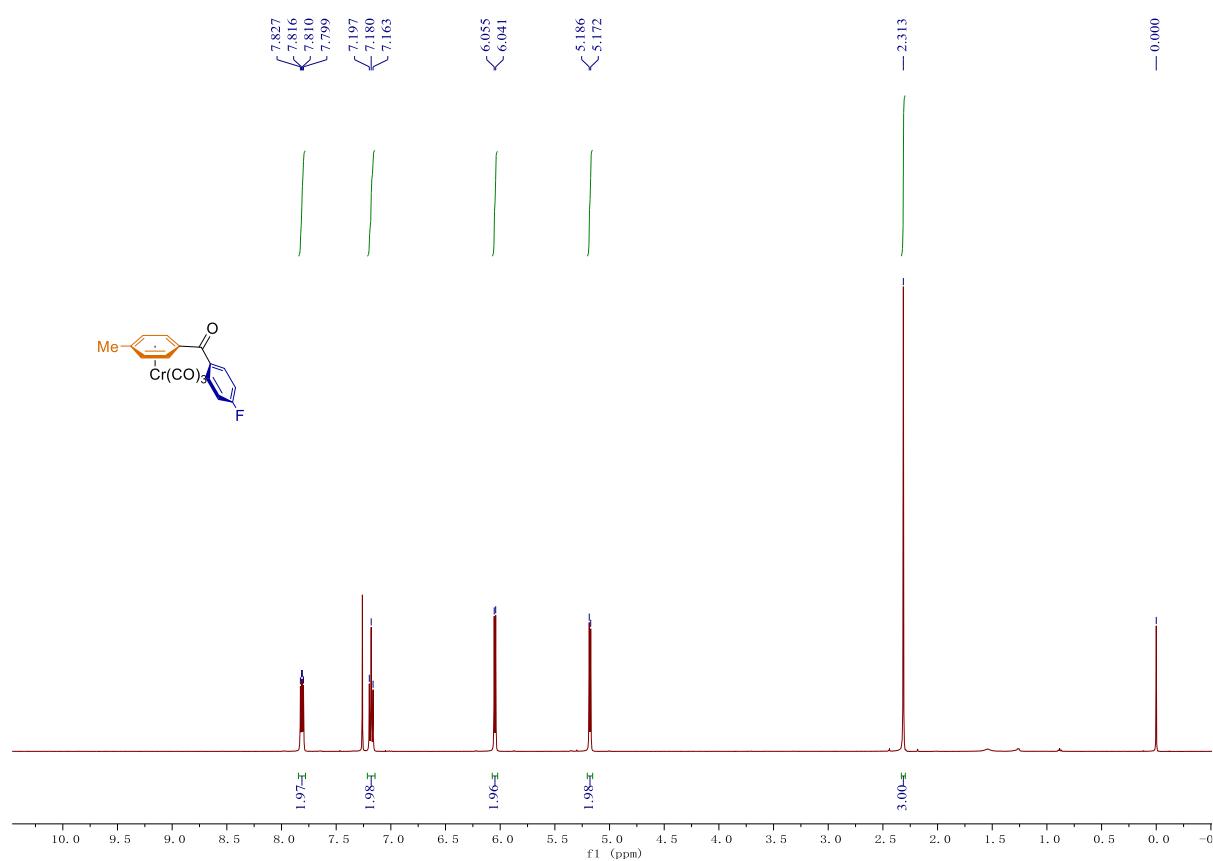
Supplementary Fig. 111 ¹³C NMR (126 MHz, Chloroform-*d*) of 2-(2-(trifluoromethyl)benzoyl)toluene chromium tricarbonyl (1u-Cr).



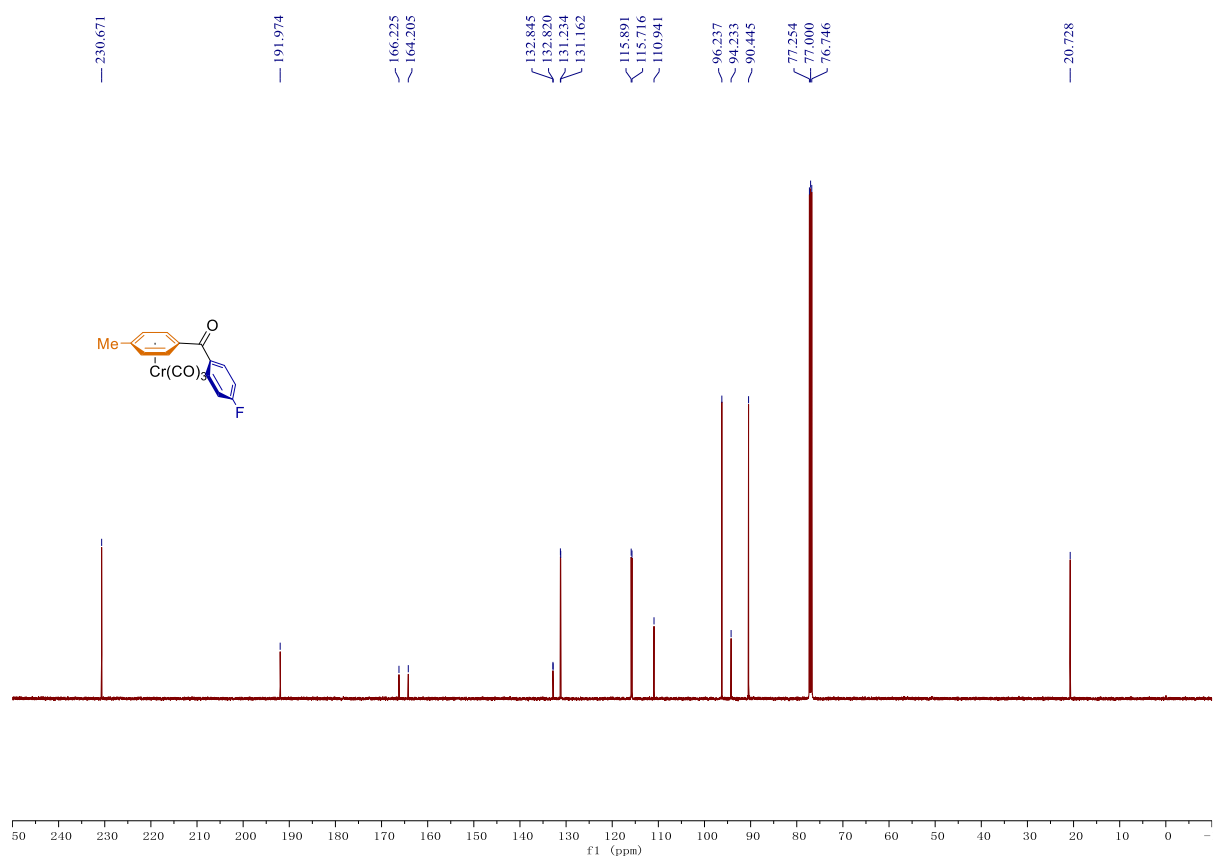
Supplementary Fig. 112 ^1H NMR (600 MHz, Chloroform-*d*) of 2-(4-chlorobenzoyl)toluene chromium tricarbonyl (1x-Cr).



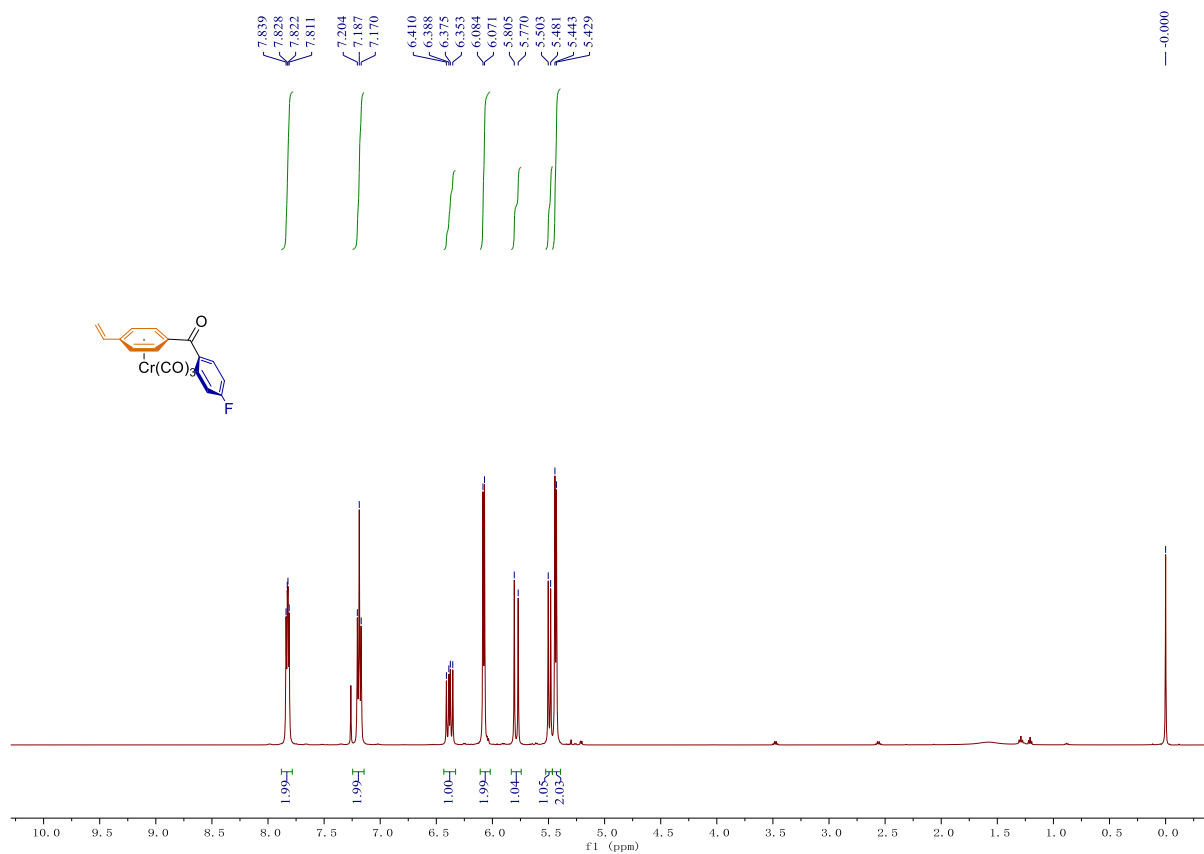
Supplementary Fig. 113 ^{13}C NMR (151 MHz, Chloroform-*d*) of 2-(4-chlorobenzoyl)toluene chromium tricarbonyl (1x-Cr).



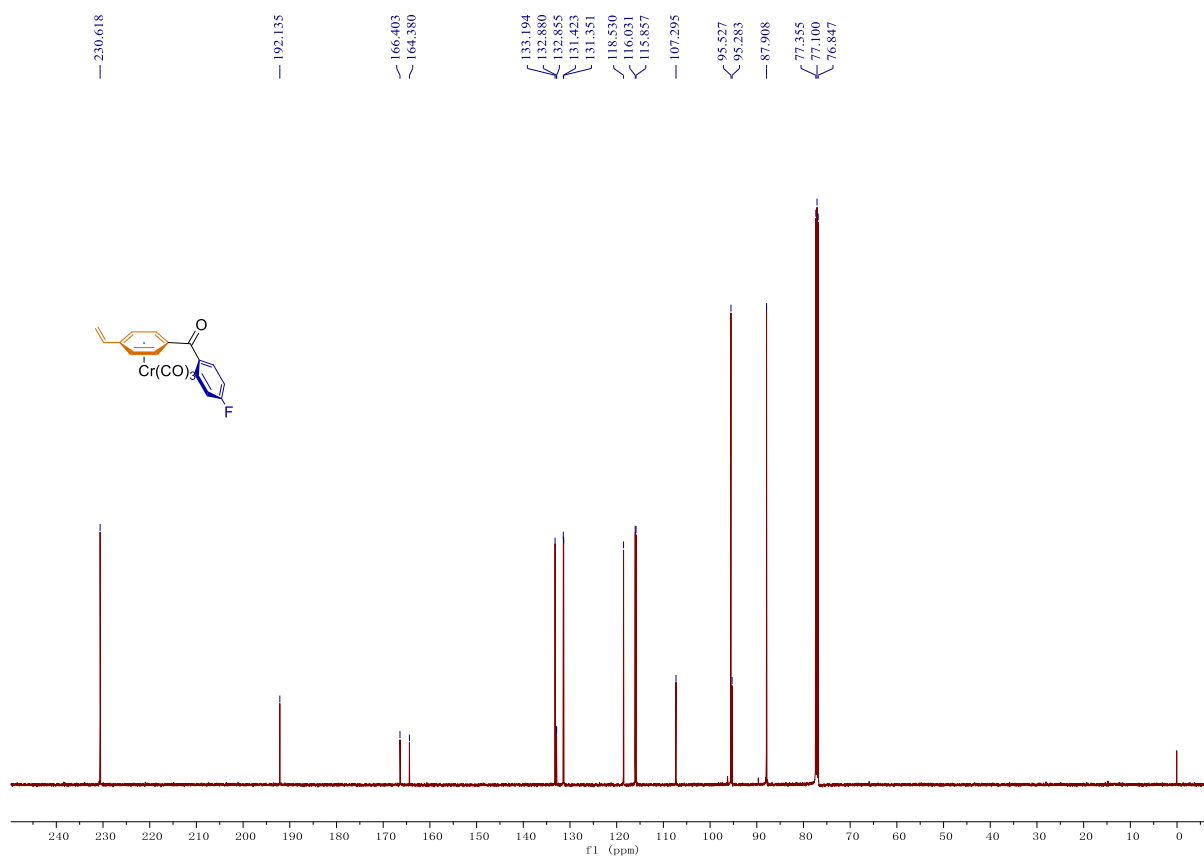
Supplementary Fig. 114 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(4-fluorobenzoyl)toluene chromium tricarbonyl (1t-Cr).



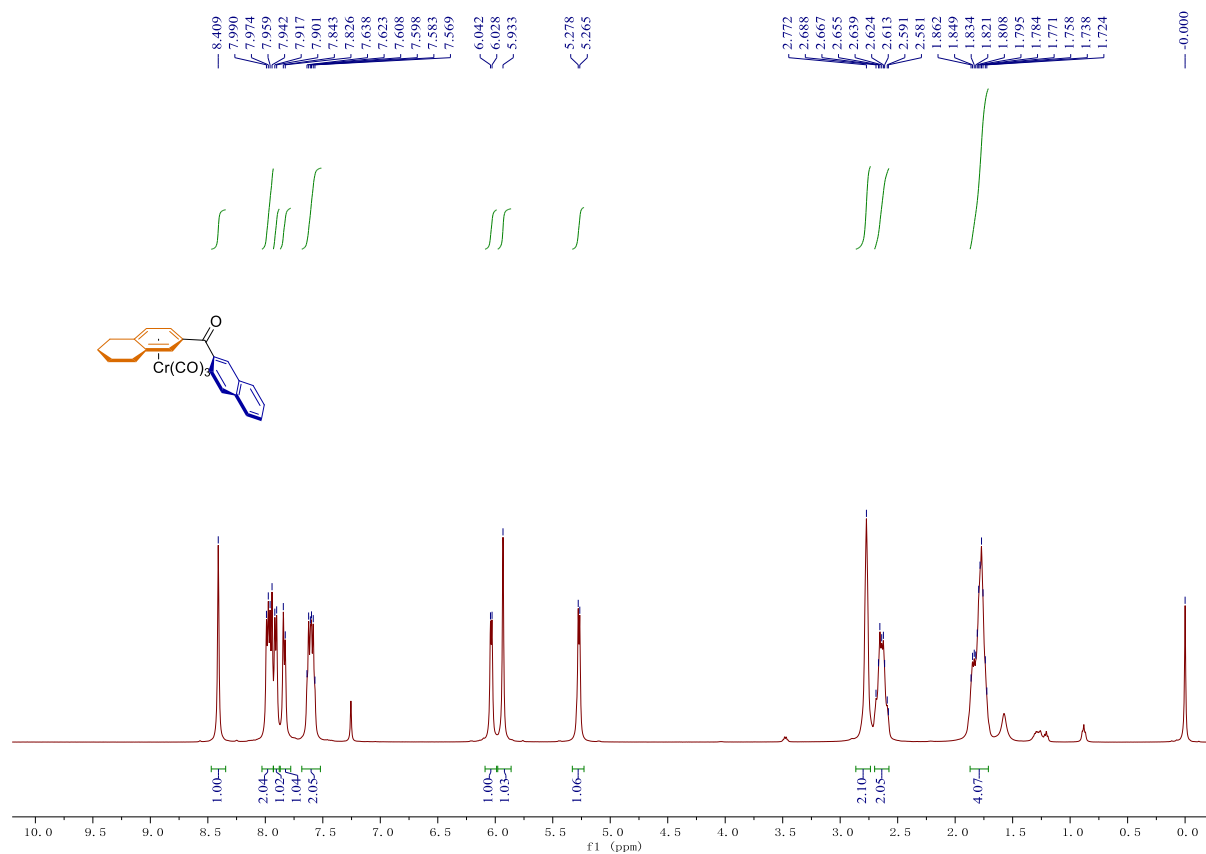
Supplementary Fig. 115 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(4-fluorobenzoyl)toluene chromium tricarbonyl (1t-Cr).



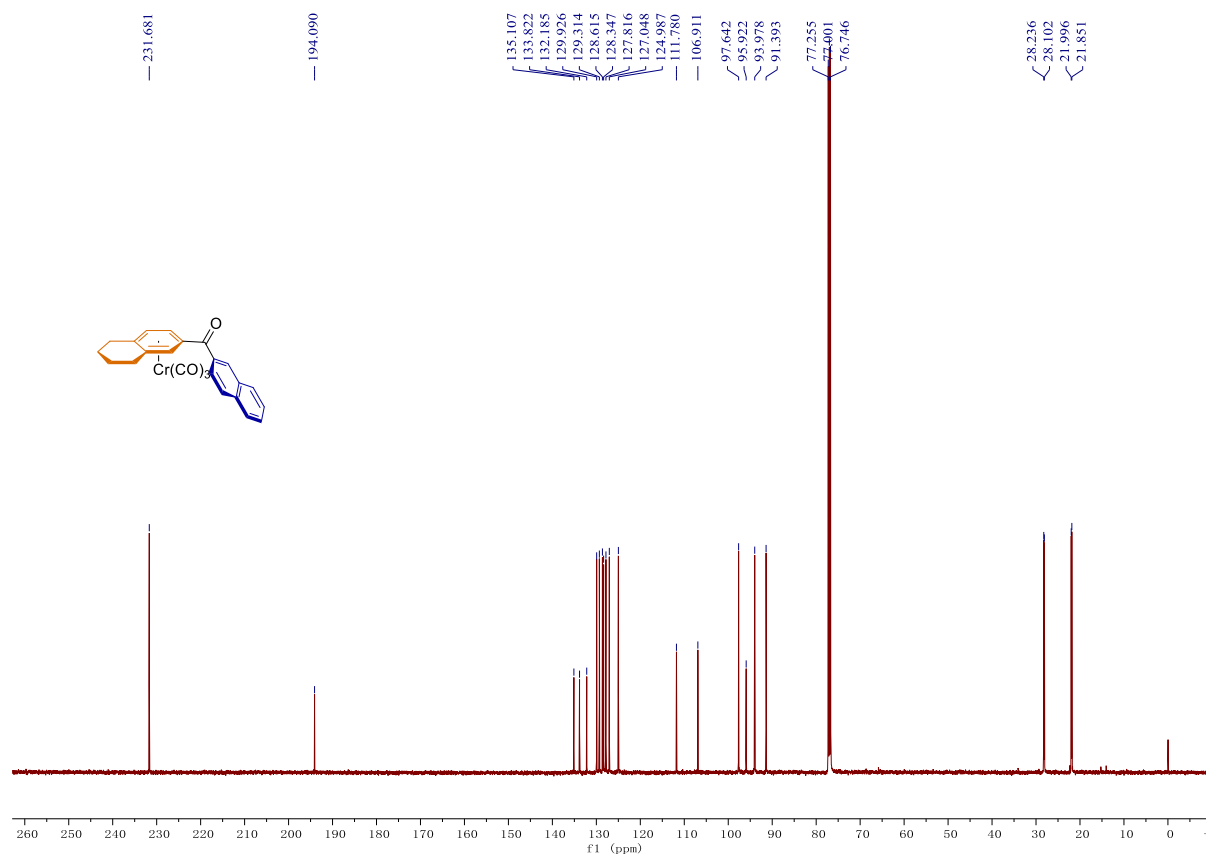
Supplementary Fig. 116 ¹H NMR (500 MHz, Chloroform-*d*) of 4-(4-fluorobenzoyl)styrene chromium tricarbonyl (1ac-Cr).



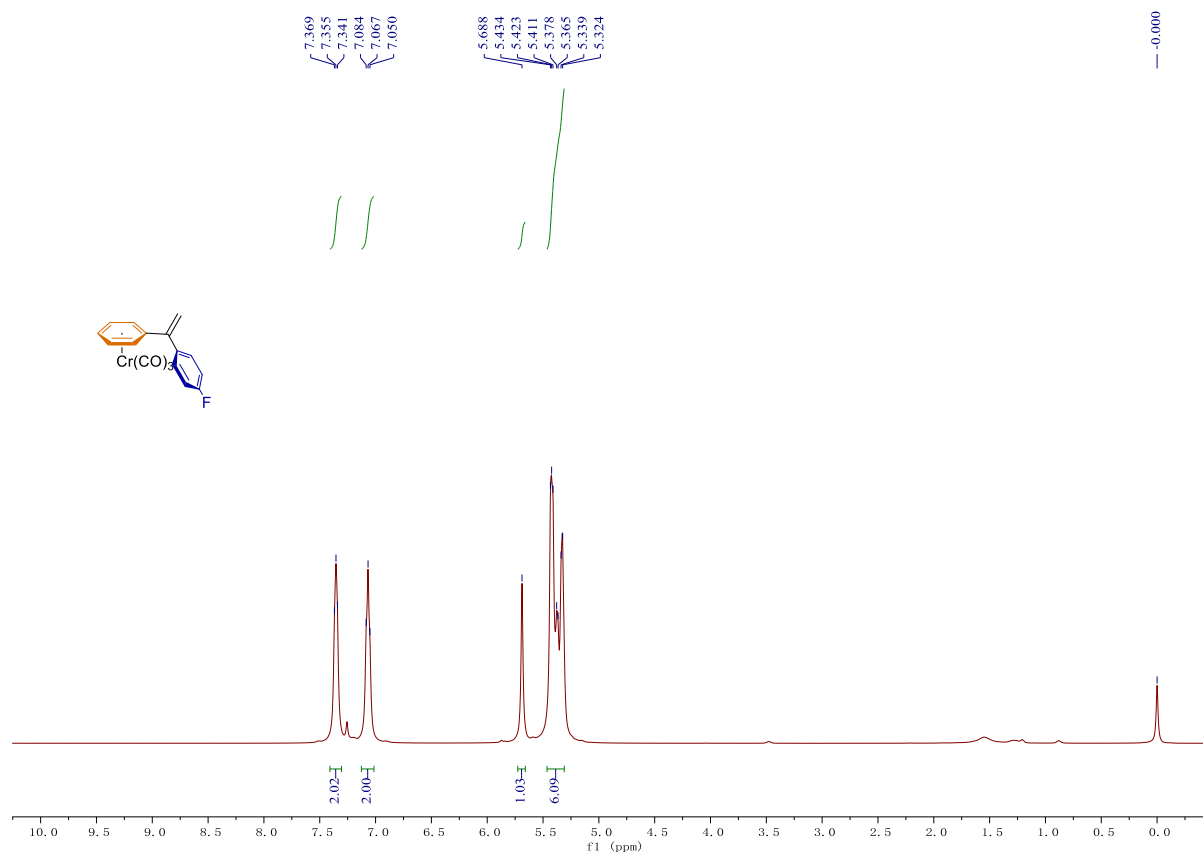
Supplementary Fig. 117 ¹³C NMR (126 MHz, Chloroform-*d*) of 4-(4-fluorobenzoyl)styrene chromium tricarbonyl (1ac-Cr).



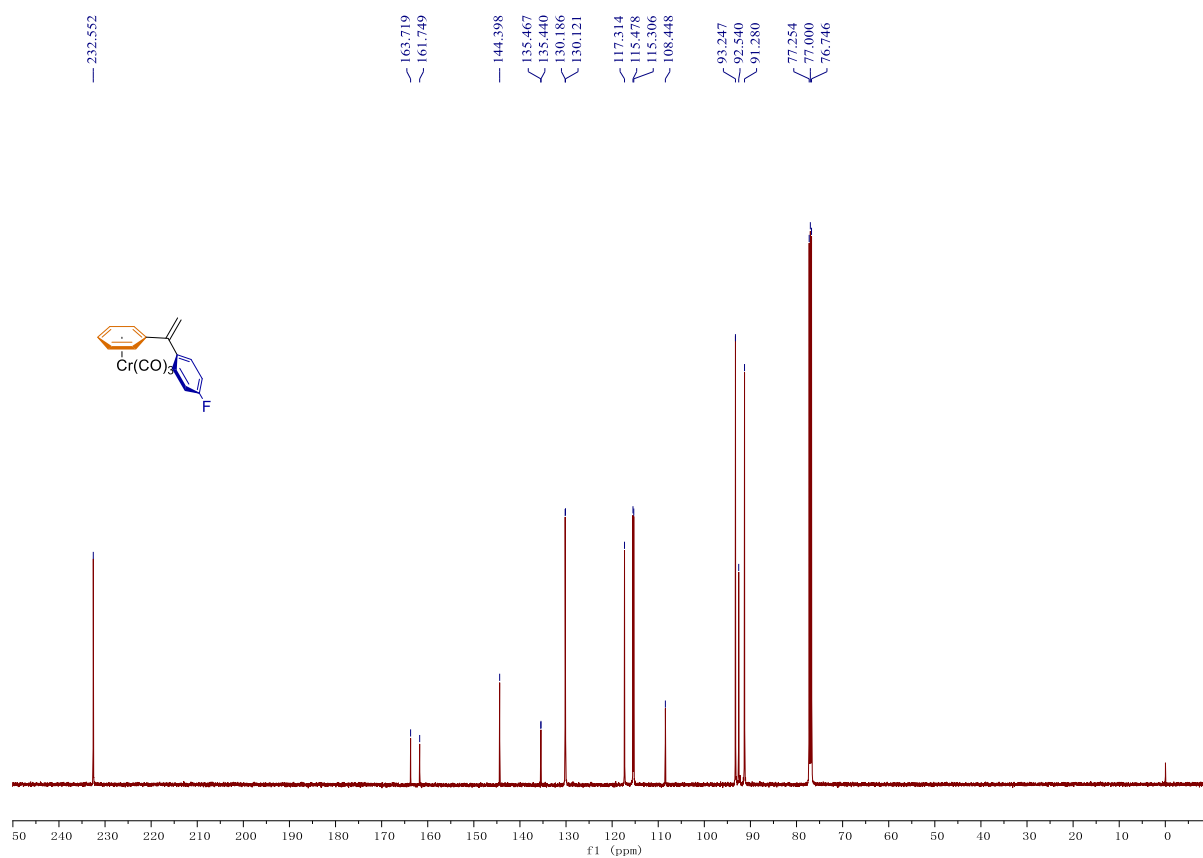
Supplementary Fig. 118 ¹H NMR (500 MHz, Chloroform-*d*) of 2-(2-naphthoyl)(5,6,7,8-tetrahydronaphthalene) chromium tricarbonyl (1w-Cr).



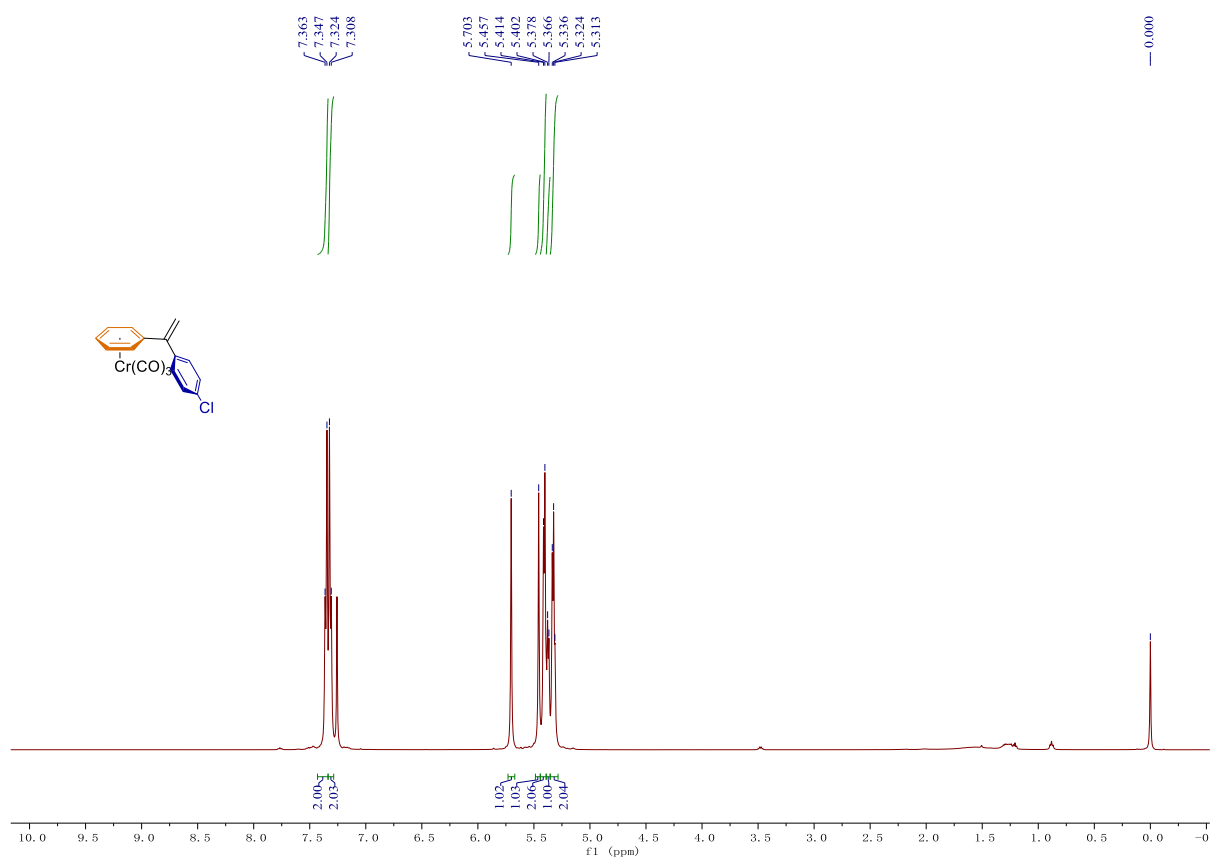
Supplementary Fig. 119 ¹³C NMR (126 MHz, Chloroform-*d*) of 2-(2-naphthoyl)(5,6,7,8-tetrahydronaphthalene) chromium tricarbonyl (1w-Cr).



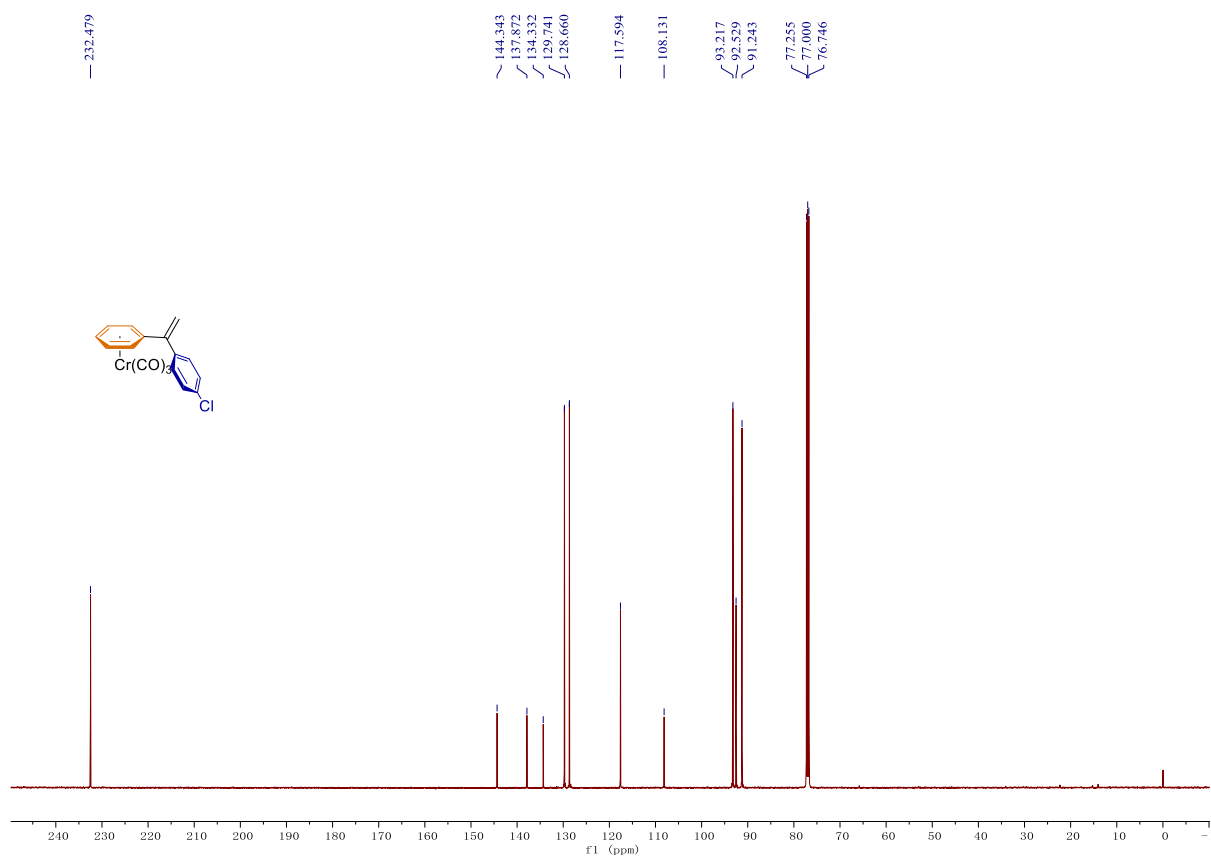
Supplementary Fig. 120 ¹H NMR (500 MHz, Chloroform-*d*) of (1-(4-fluorophenyl)vinyl)benzene chromium tricarbonyl (3a-Cr).



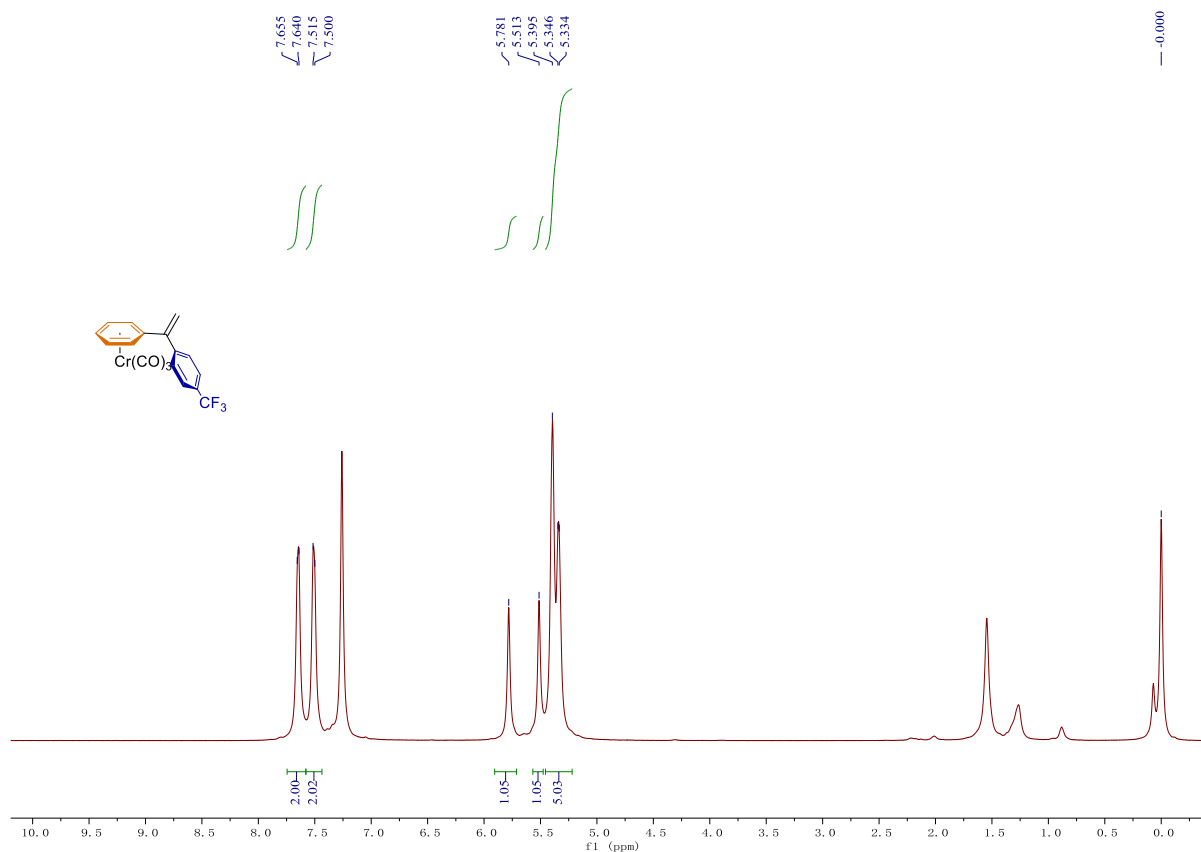
Supplementary Fig. 121 ¹³C NMR (126 MHz, Chloroform-*d*) of (1-(4-fluorophenyl)vinyl)benzene chromium tricarbonyl (3a-Cr).



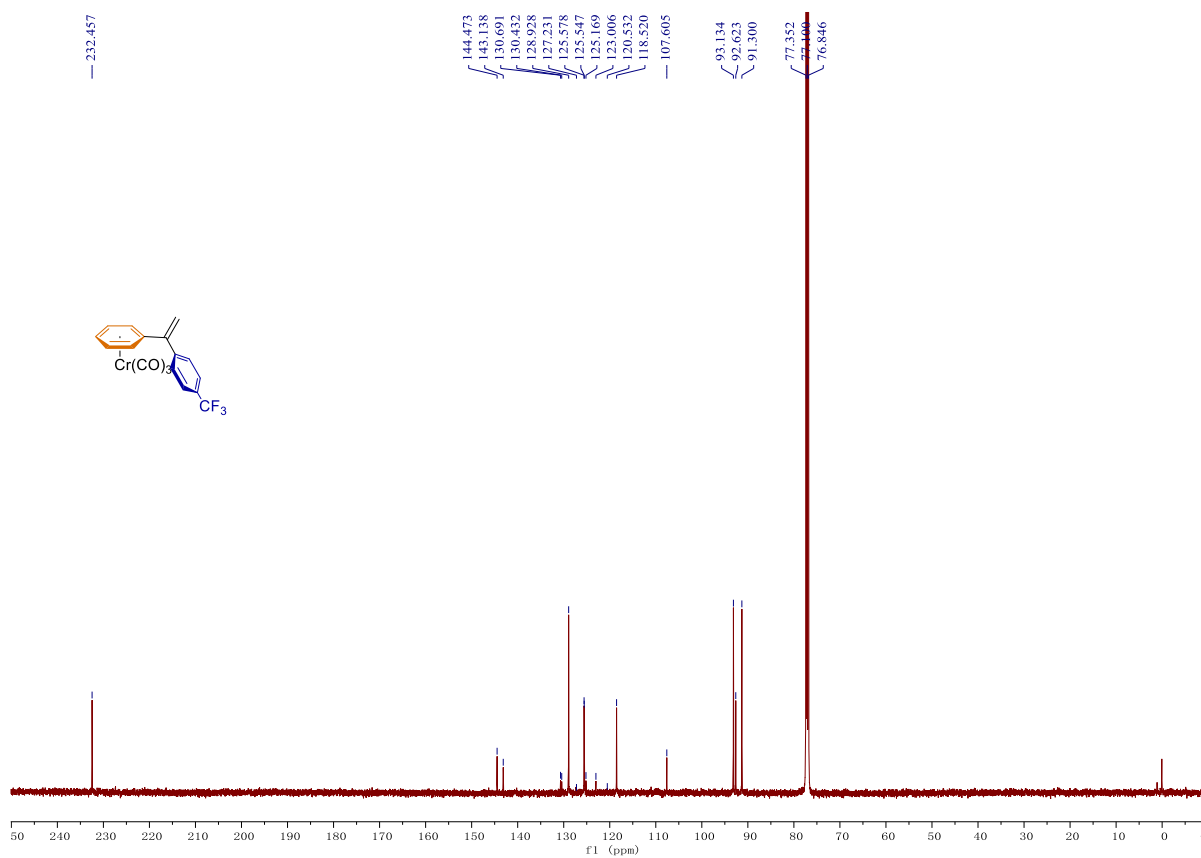
Supplementary Fig. 122 ¹H NMR (500 MHz, Chloroform-*d*) of (1-(4-chlorophenyl)vinyl)benzene chromium tricarbonyl (3b-Cr).



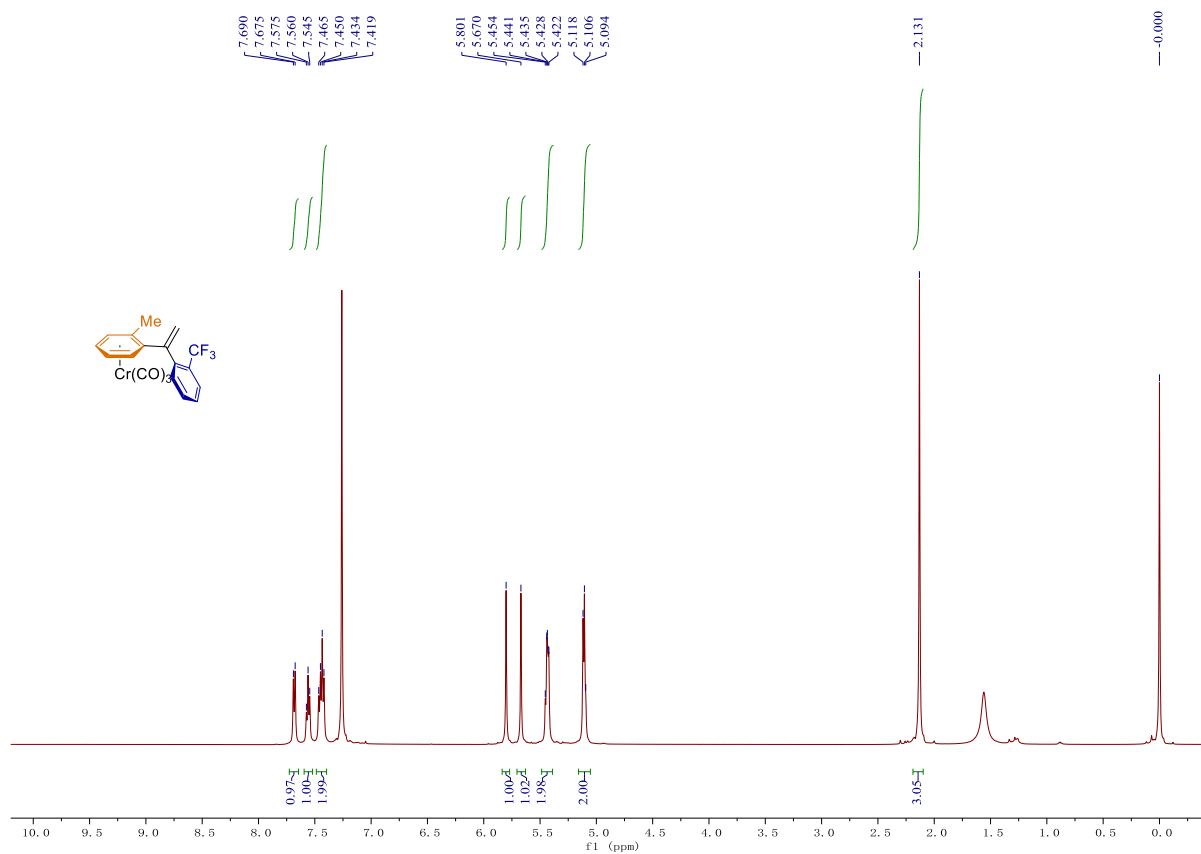
Supplementary Fig. 123 ¹³C NMR (126 MHz, Chloroform-*d*) of (1-(4-chlorophenyl)vinyl)benzene chromium tricarbonyl (3b-Cr).



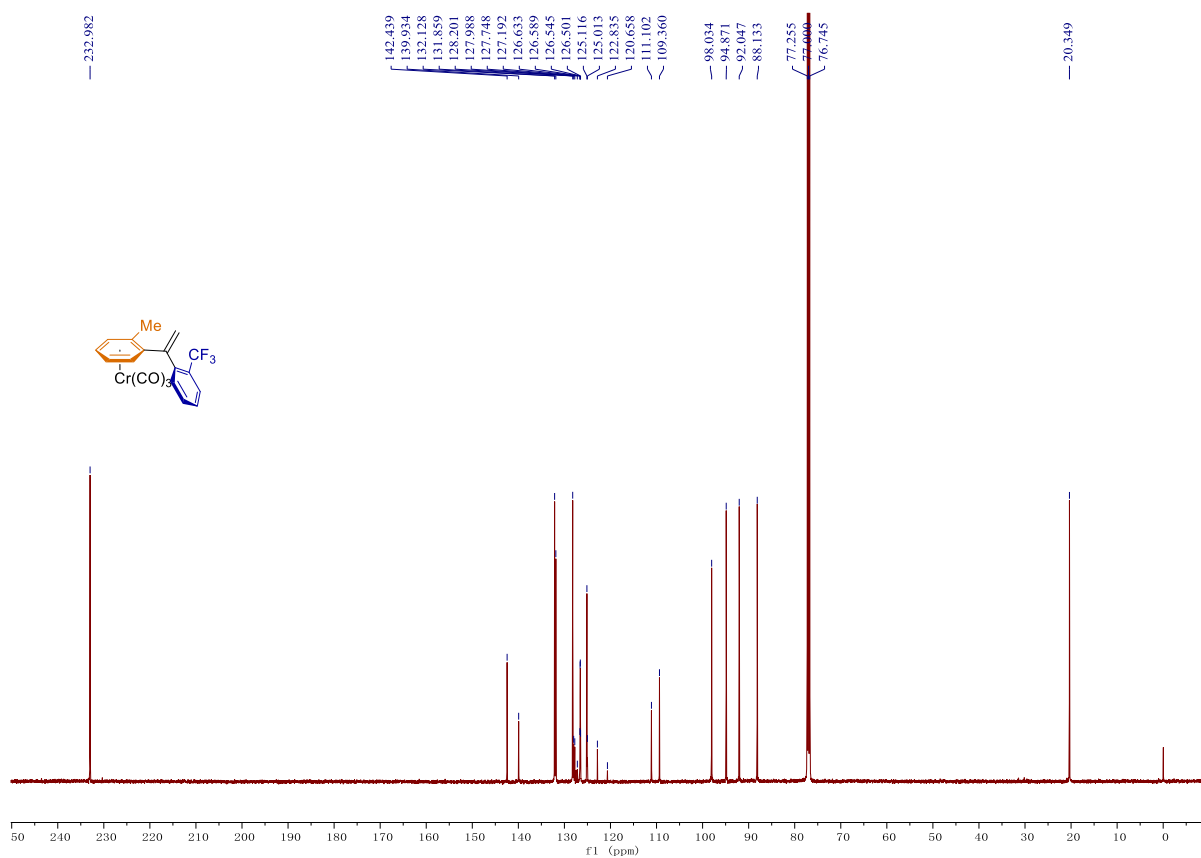
Supplementary Fig. 124 ¹H NMR (500 MHz, Chloroform-*d*) of (1-(4-(trifluoromethyl)phenyl)vinyl)benzene chromium tricarbonyl (3c-Cr).



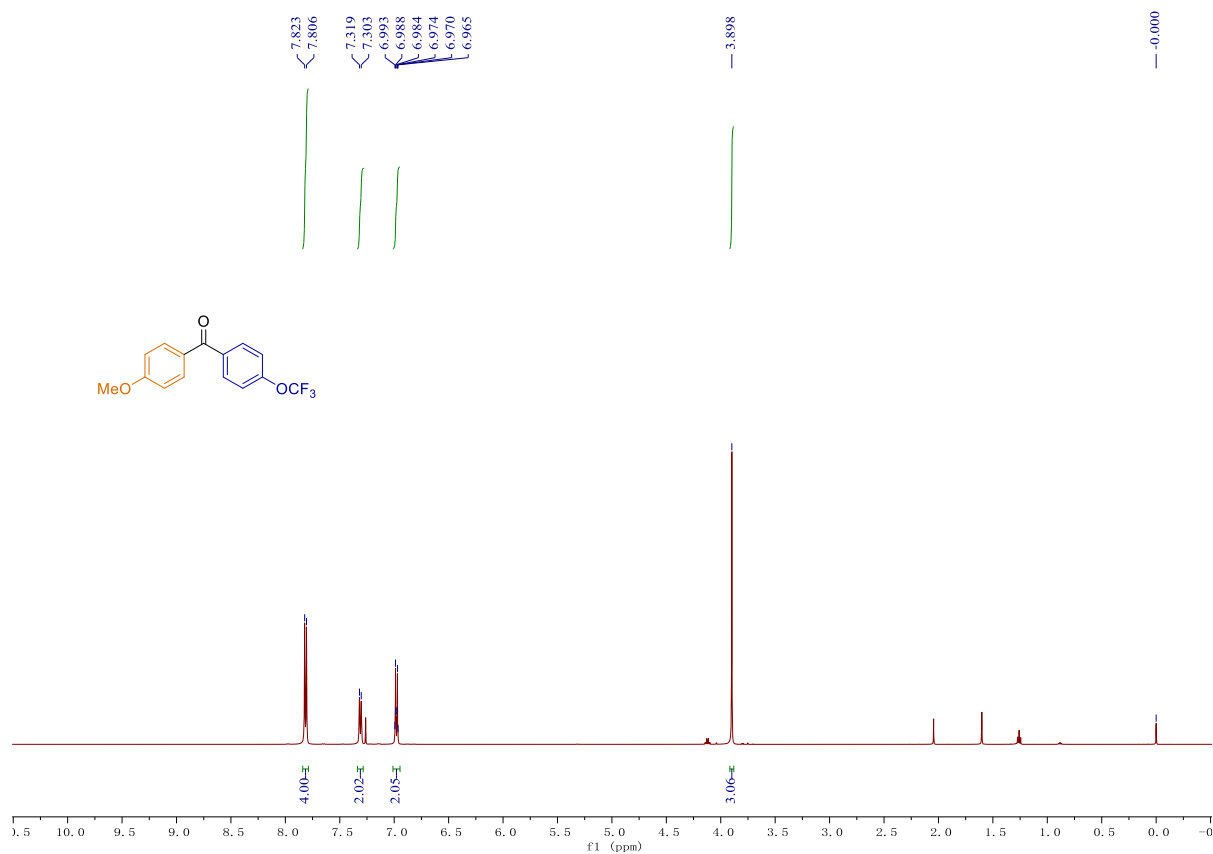
Supplementary Fig. 125 ¹³C NMR (126 MHz, Chloroform-*d*) of (1-(4-(trifluoromethyl)phenyl)vinyl)benzene chromium tricarbonyl (3c-Cr).



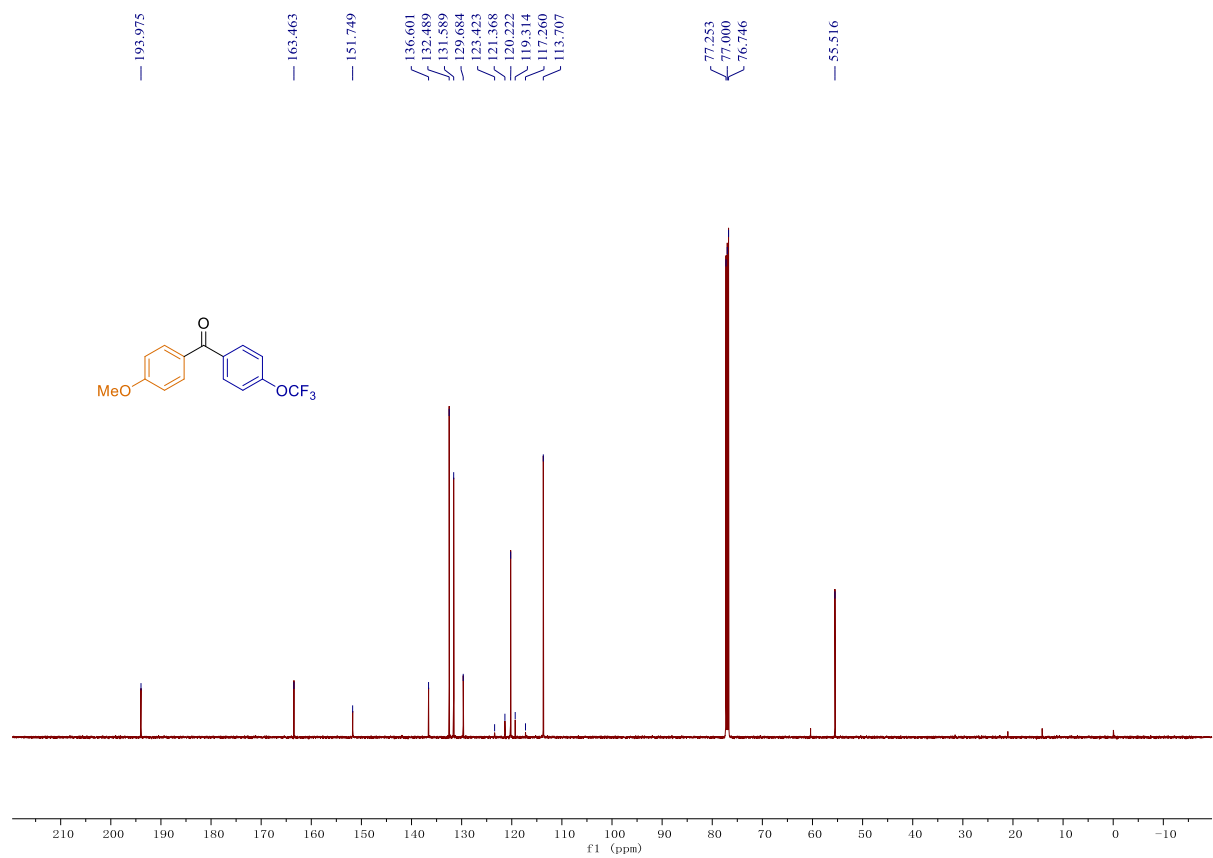
Supplementary Fig. 126 ¹H NMR (500 MHz, Chloroform-*d*) of 2-(1-(2-(trifluoromethyl)phenyl)vinyl)toluene chromium tricarbonyl (3d-Cr).



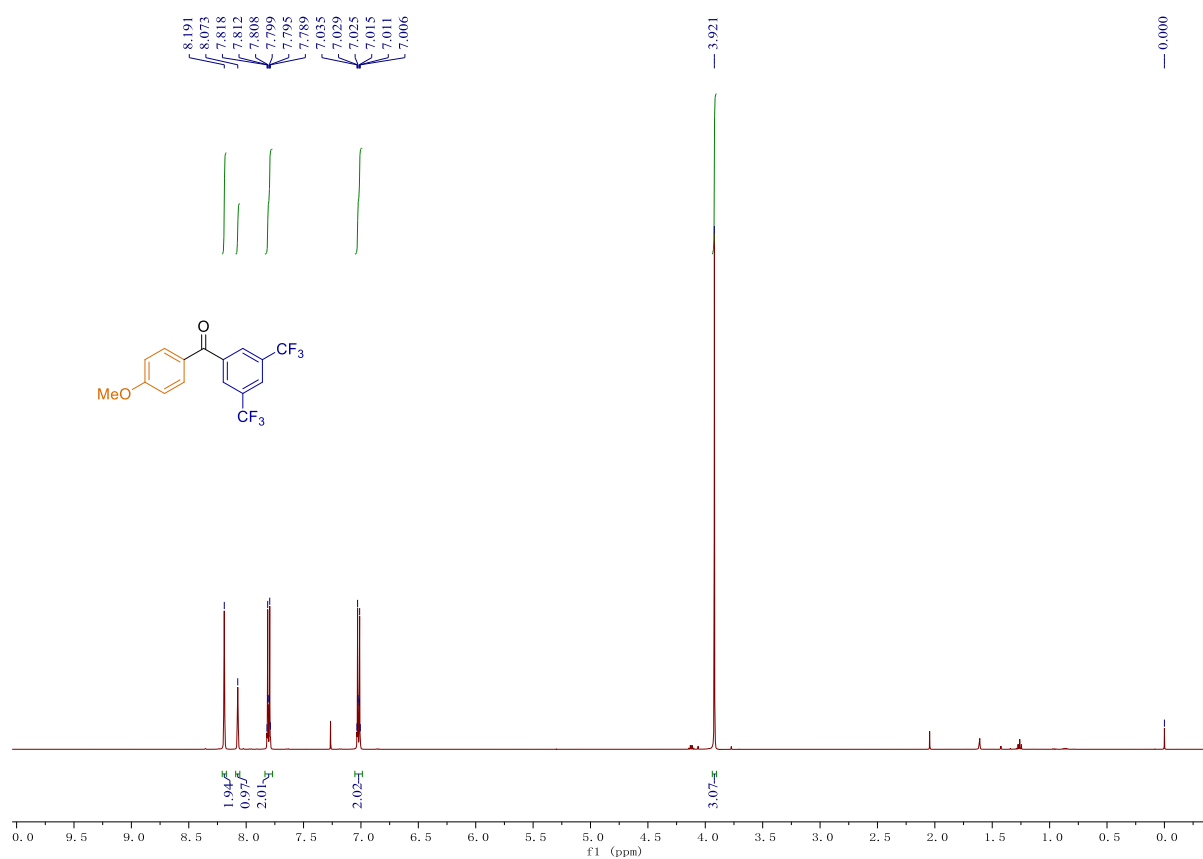
Supplementary Fig. 127 ¹³C NMR (126 MHz, Chloroform-*d*) of 2-(1-(2-(trifluoromethyl)phenyl)vinyl)toluene chromium tricarbonyl (3d-Cr).



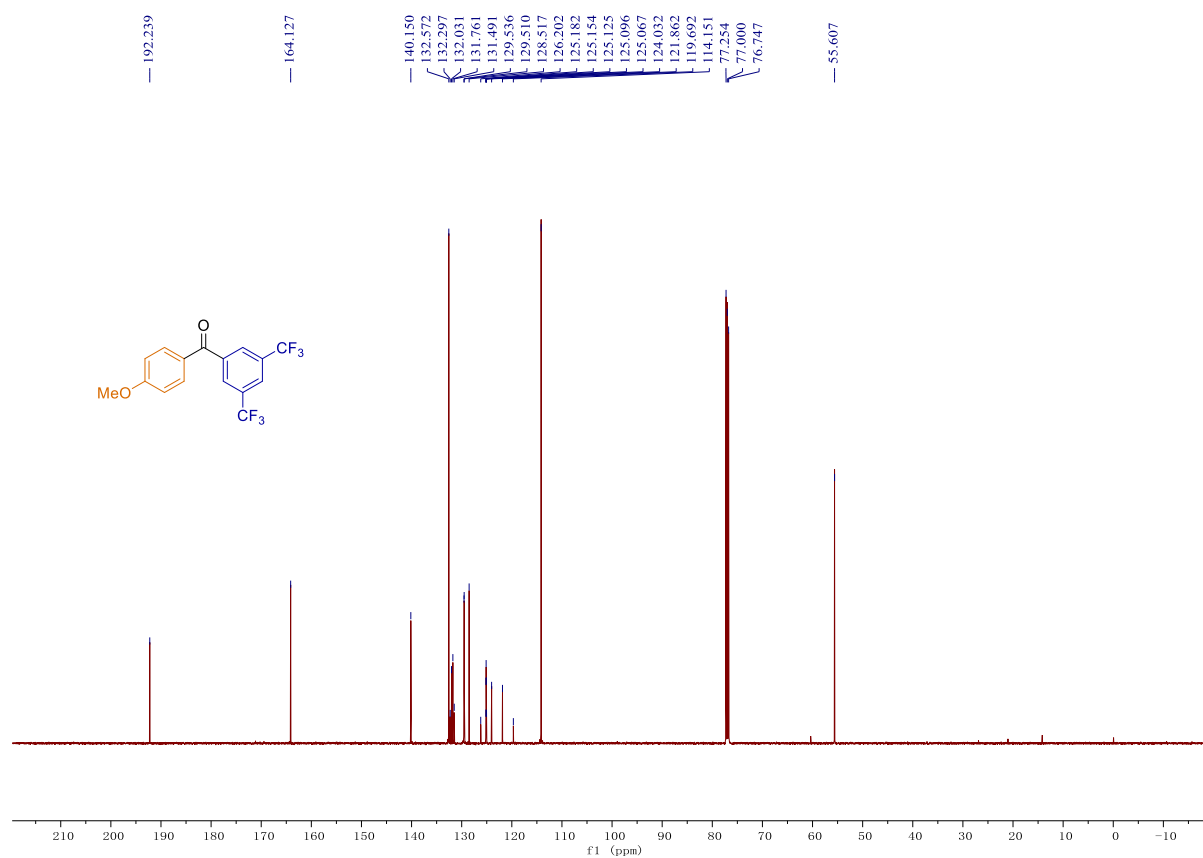
Supplementary Fig. 128 ¹H NMR (500 MHz, Chloroform-*d*) of (4-methoxyphenyl)(4-(trifluoromethoxy)phenyl)methanone (1p).



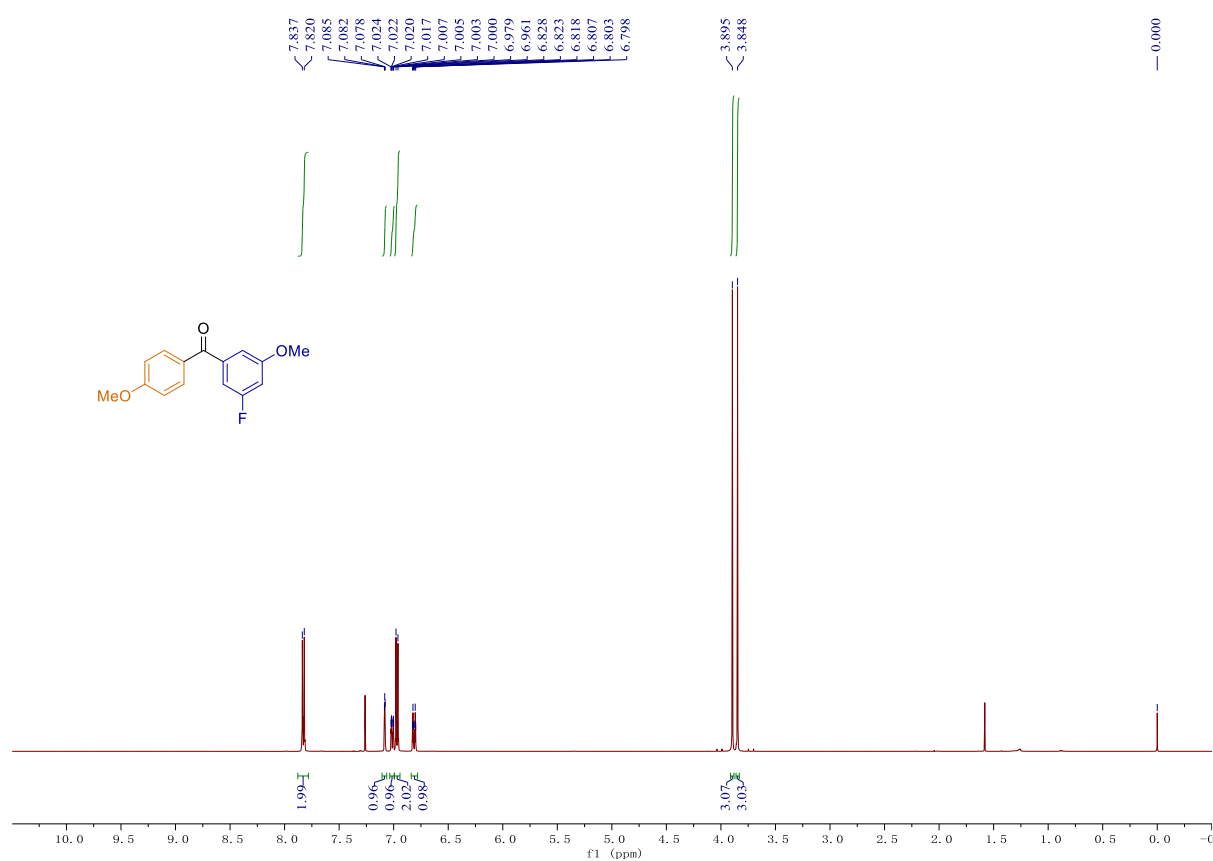
Supplementary Fig. 129 ¹³C NMR (126 MHz, Chloroform-*d*) of (4-methoxyphenyl)(4-(trifluoromethoxy)phenyl)methanone (1p).



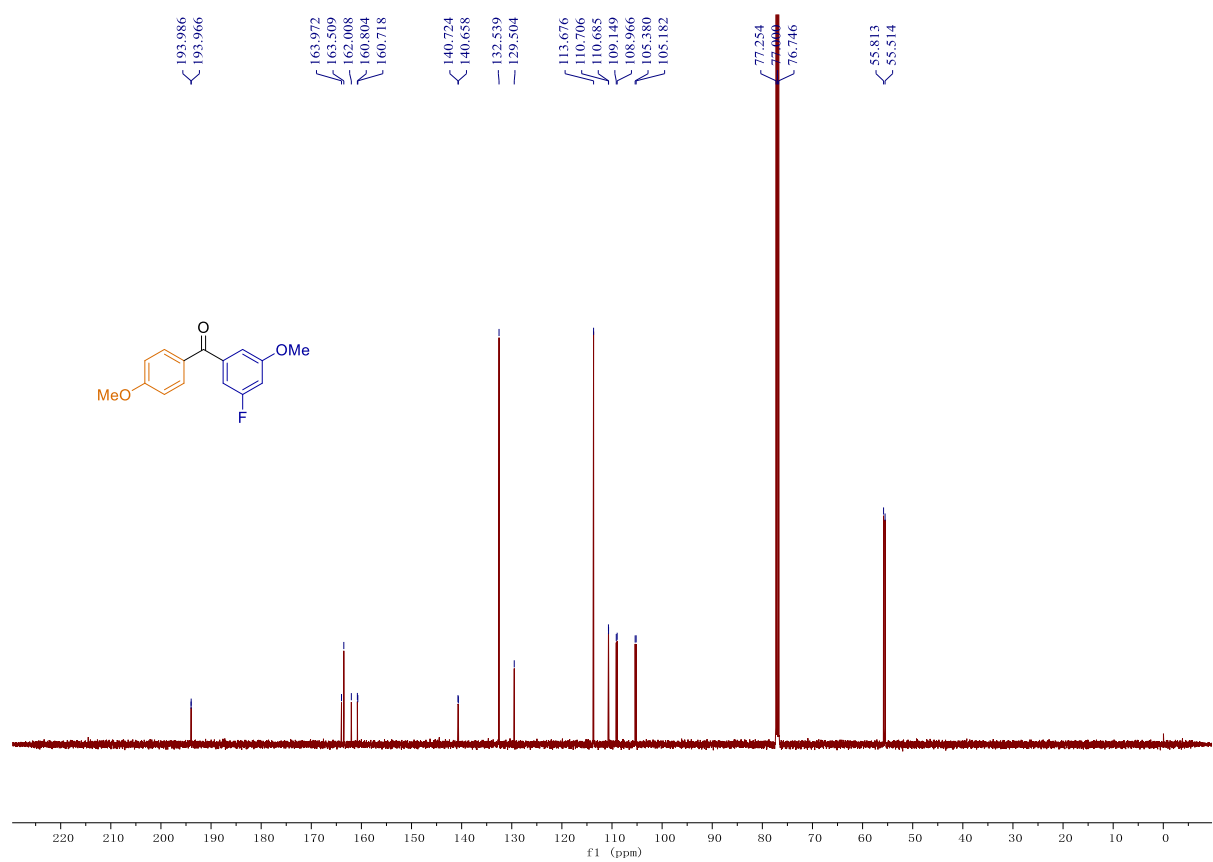
Supplementary Fig. 130 ¹H NMR (500 MHz, Chloroform-*d*) of (3,5-bis(trifluoromethyl)phenyl)(4-methoxyphenyl)methanone (s3).



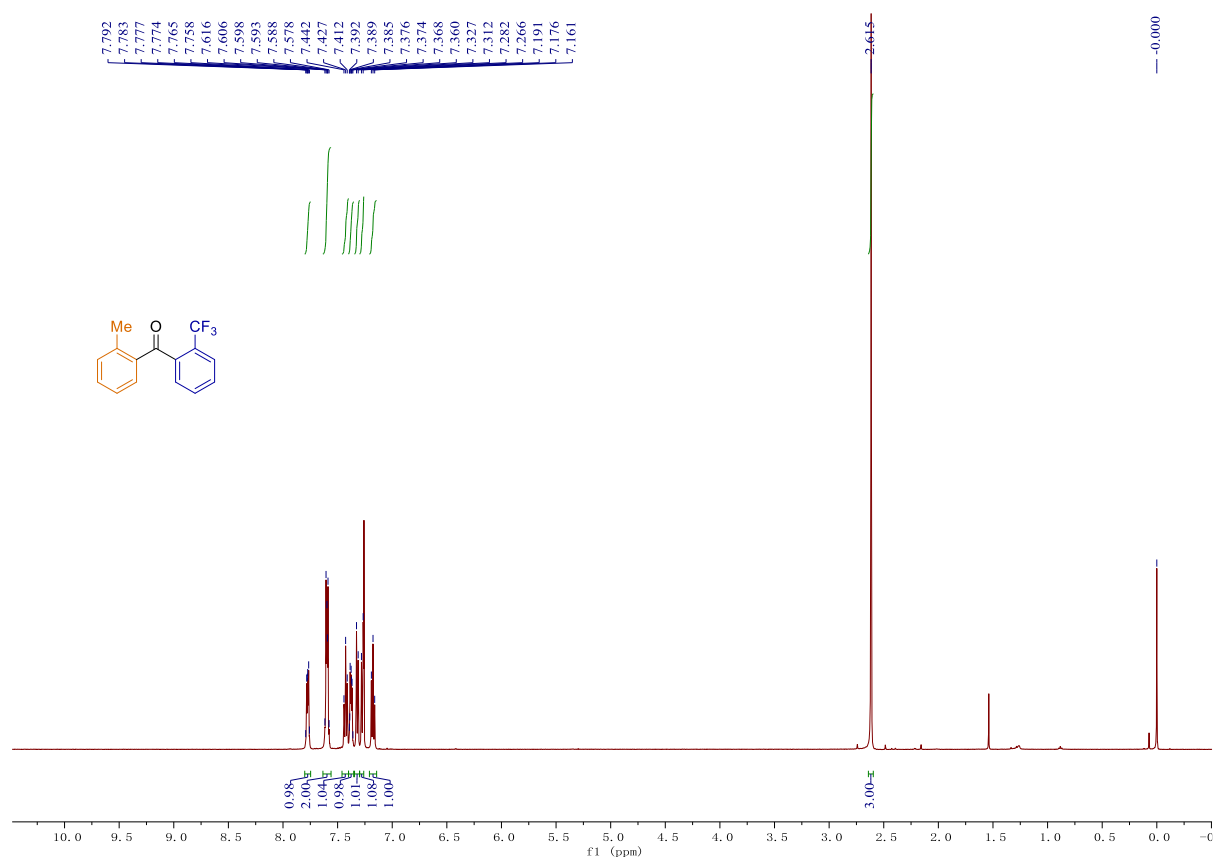
Supplementary Fig. 131 ¹³C NMR (126 MHz, Chloroform-*d*) of (3,5-bis(trifluoromethyl)phenyl)(4-methoxyphenyl)methanone (s3).



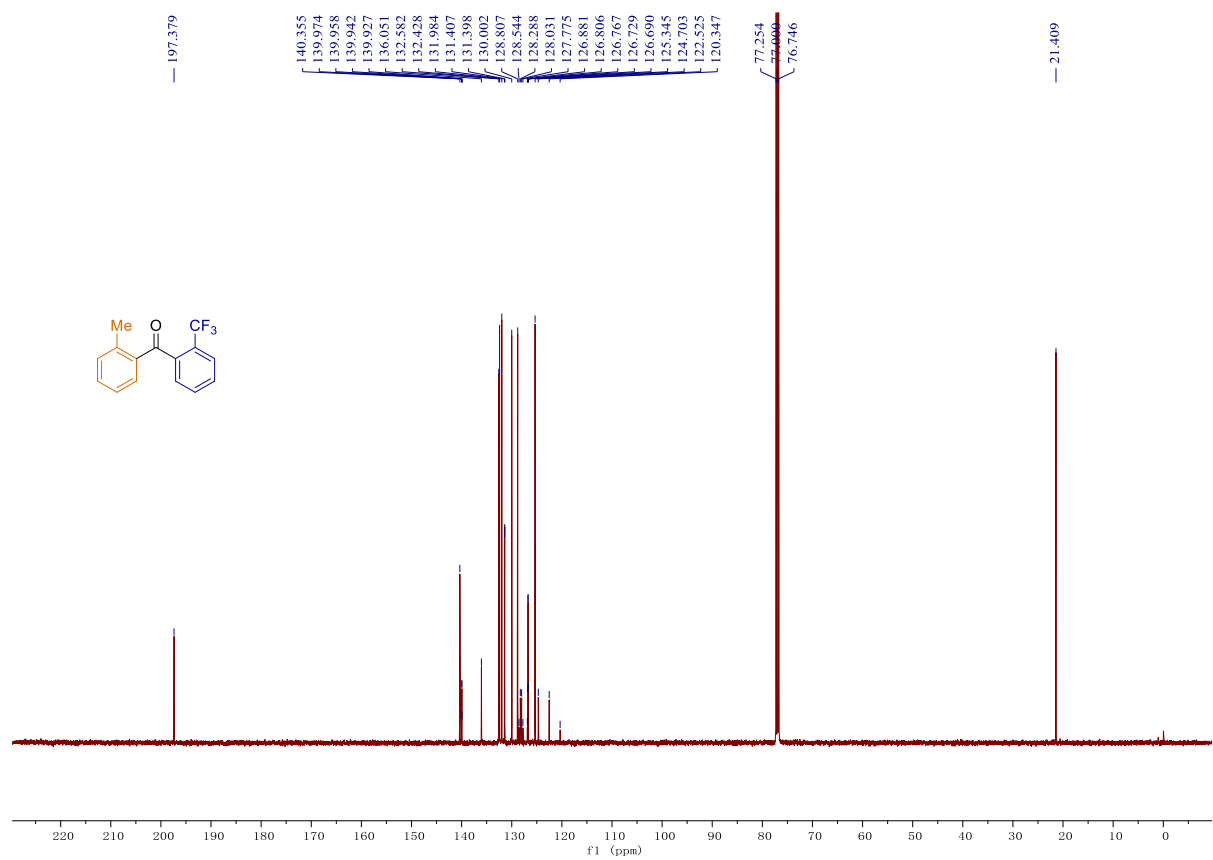
Supplementary Fig. 132 ¹H NMR (500 MHz, Chloroform-*d*) of (3-fluoro-5-methoxyphenyl)(4-methoxyphenyl)methanone (1v).



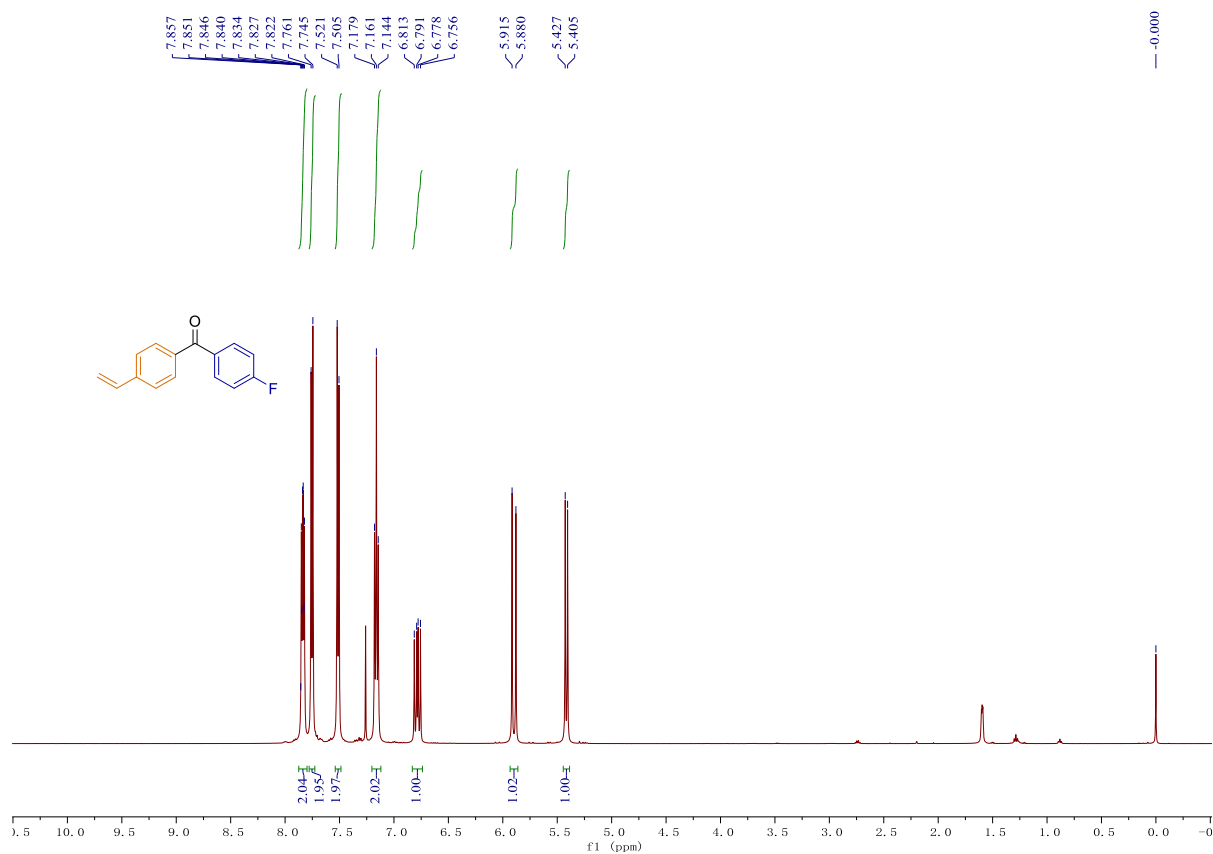
Supplementary Fig. 133 ¹³C NMR (126 MHz, Chloroform-*d*) of (3-fluoro-5-methoxyphenyl)(4-methoxyphenyl)methanone (1v).



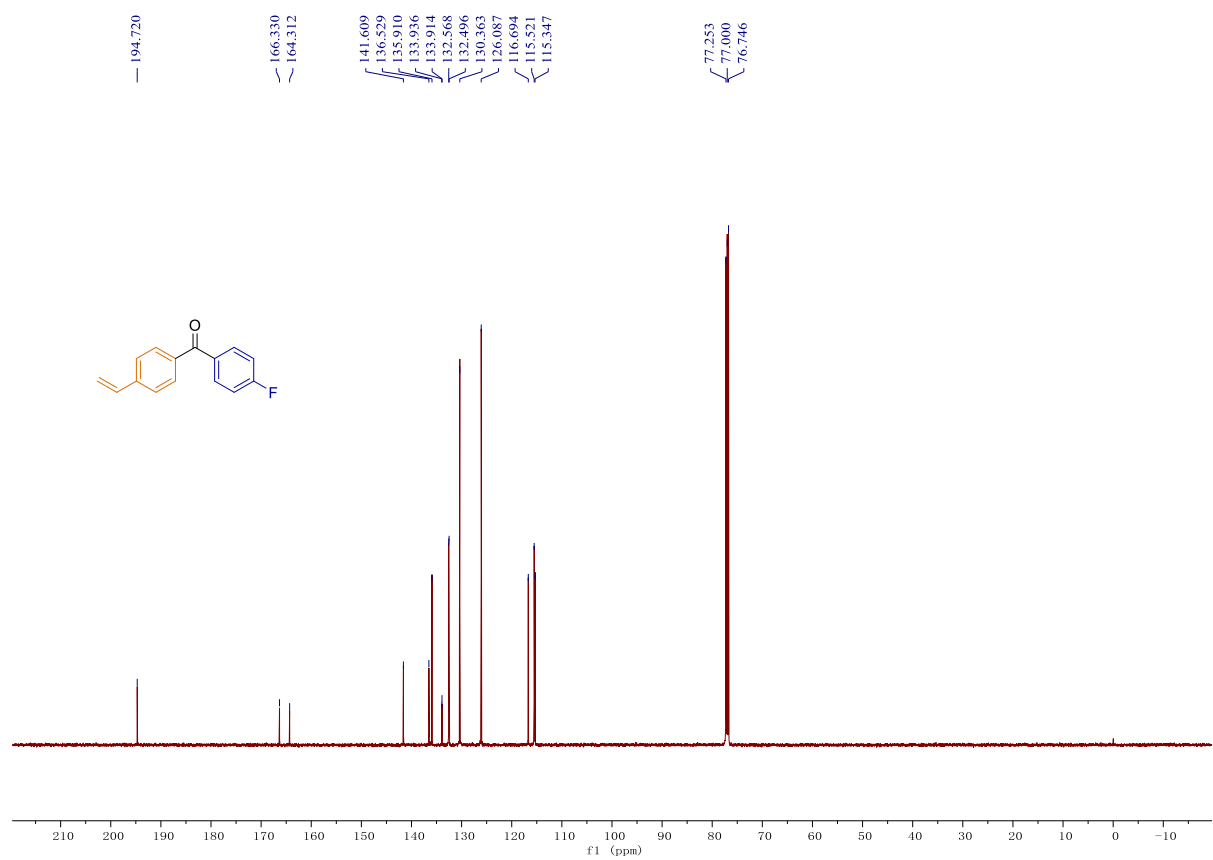
Supplementary Fig. 134 ¹H NMR (500 MHz, Chloroform-*d*) of *o*-tolyl(2-(trifluoromethyl)phenyl)methanone (1u).



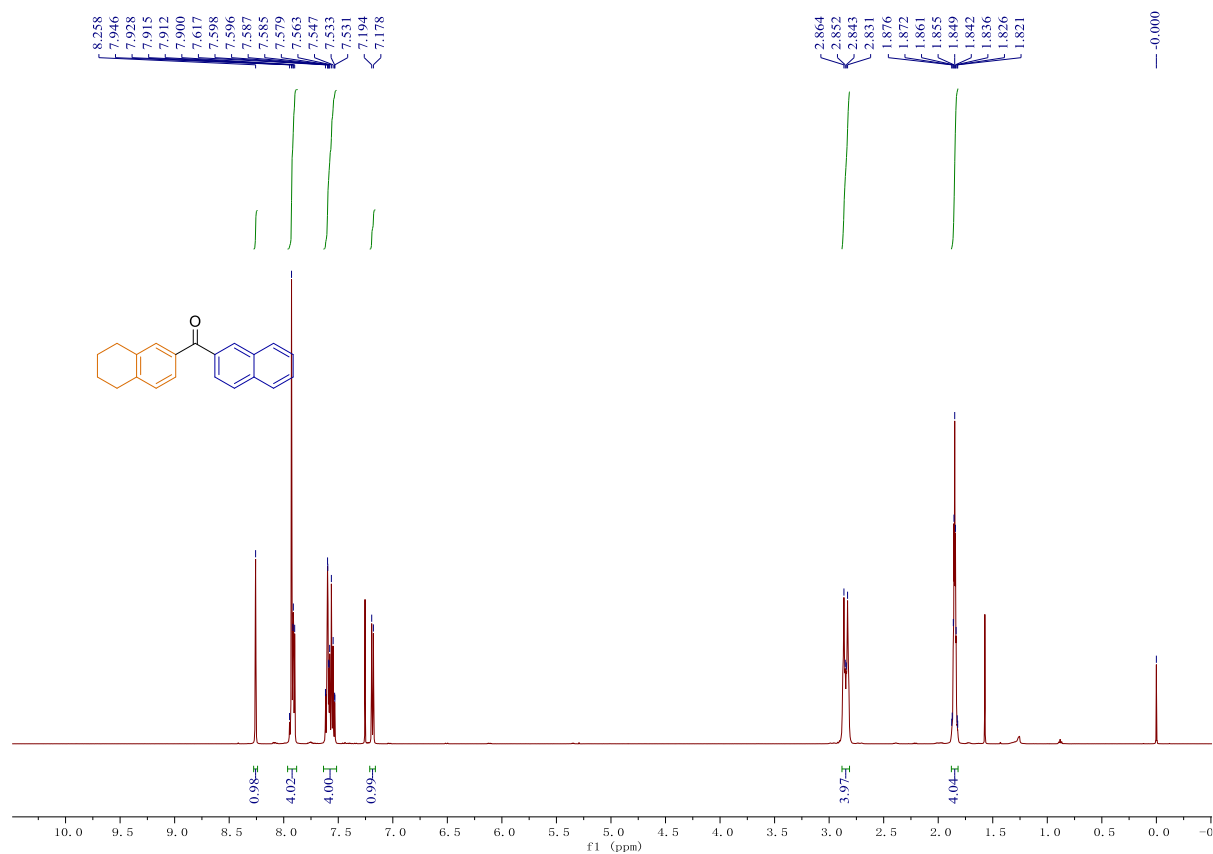
Supplementary Fig. 135 ¹³C NMR (126 MHz, Chloroform-*d*) of *o*-tolyl(2-(trifluoromethyl)phenyl)methanone (1u).



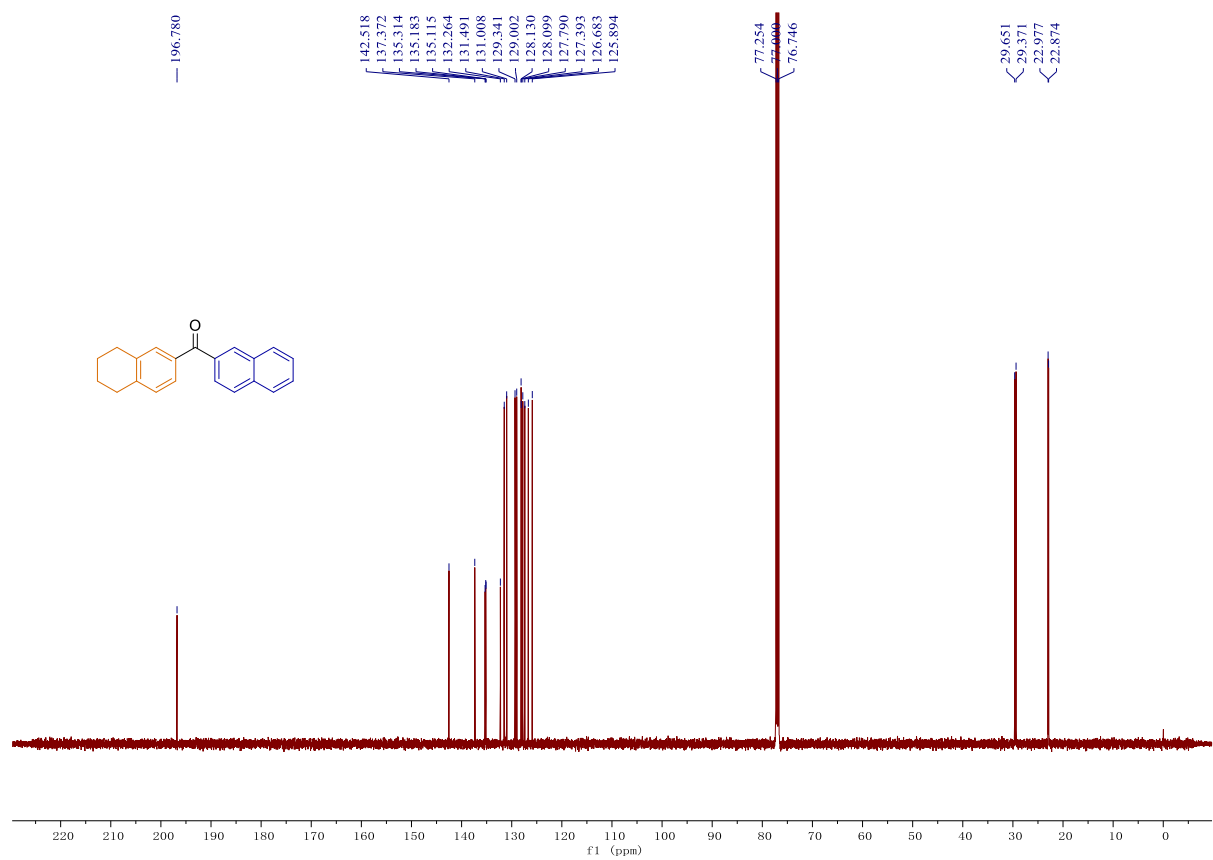
Supplementary Fig. 136 ¹H NMR (500 MHz, Chloroform-*d*) of (4-fluorophenyl)(4-vinylphenyl)methanone (1ac).



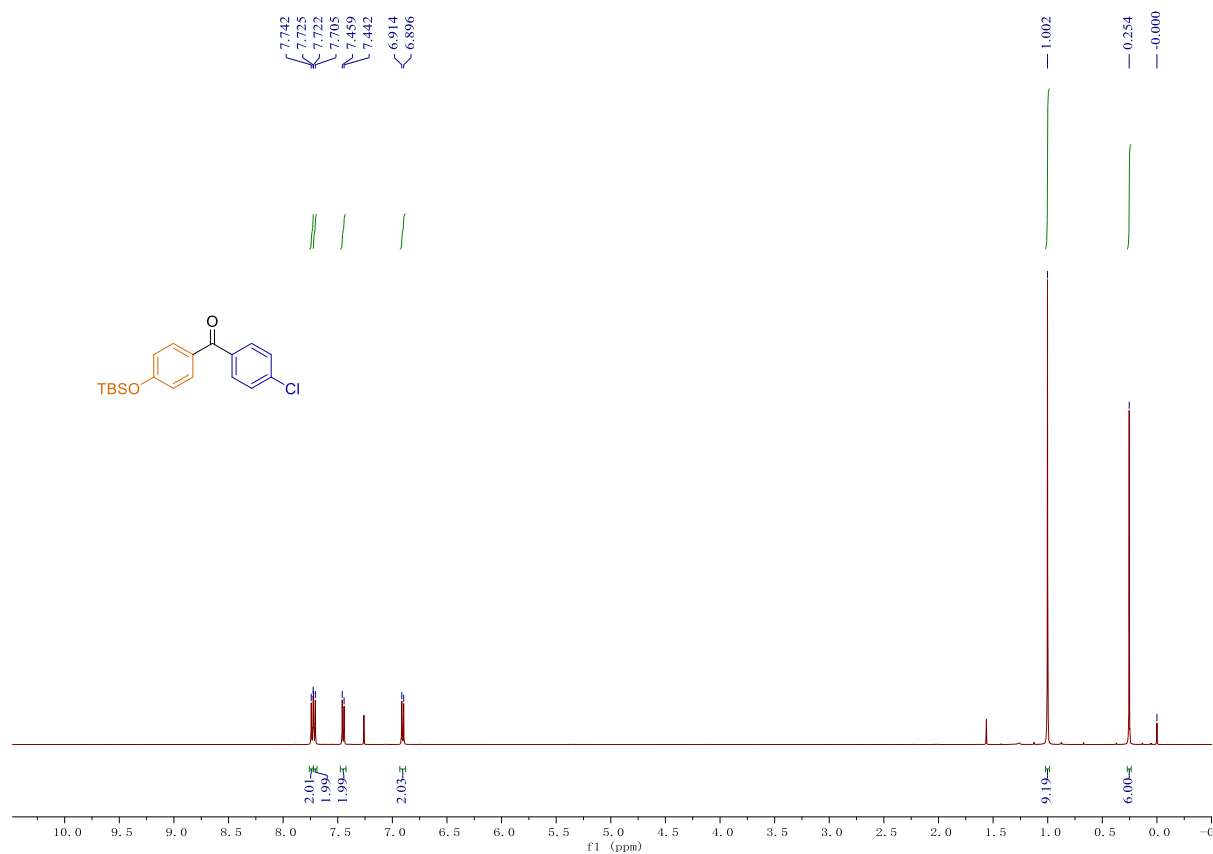
Supplementary Fig. 137 ¹³C NMR (126 MHz, Chloroform-*d*) of (4-fluorophenyl)(4-vinylphenyl)methanone (1ac).



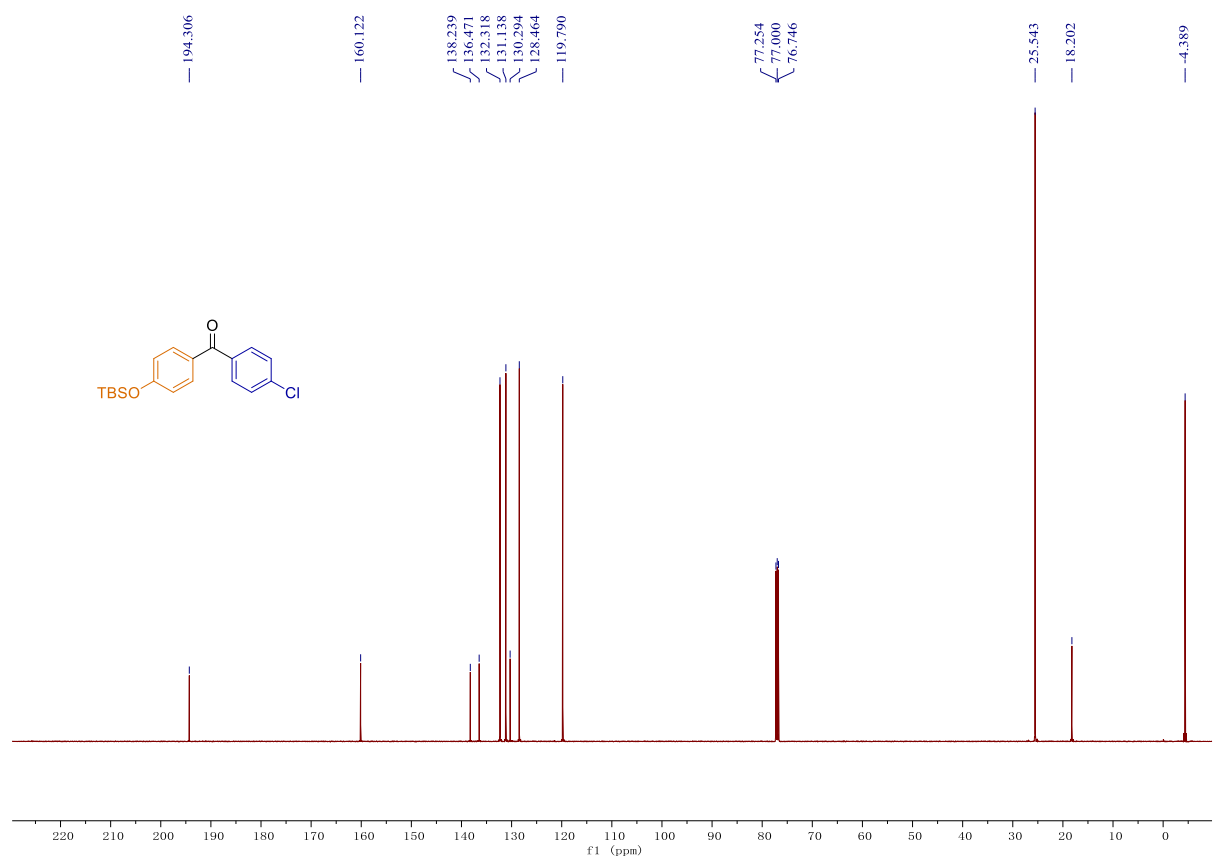
Supplementary Fig. 138 ¹H NMR (500 MHz, Chloroform-*d*) of naphthalen-2-yl(5,6,7,8-tetrahydronaphthalen-2-yl)methanone (1w).



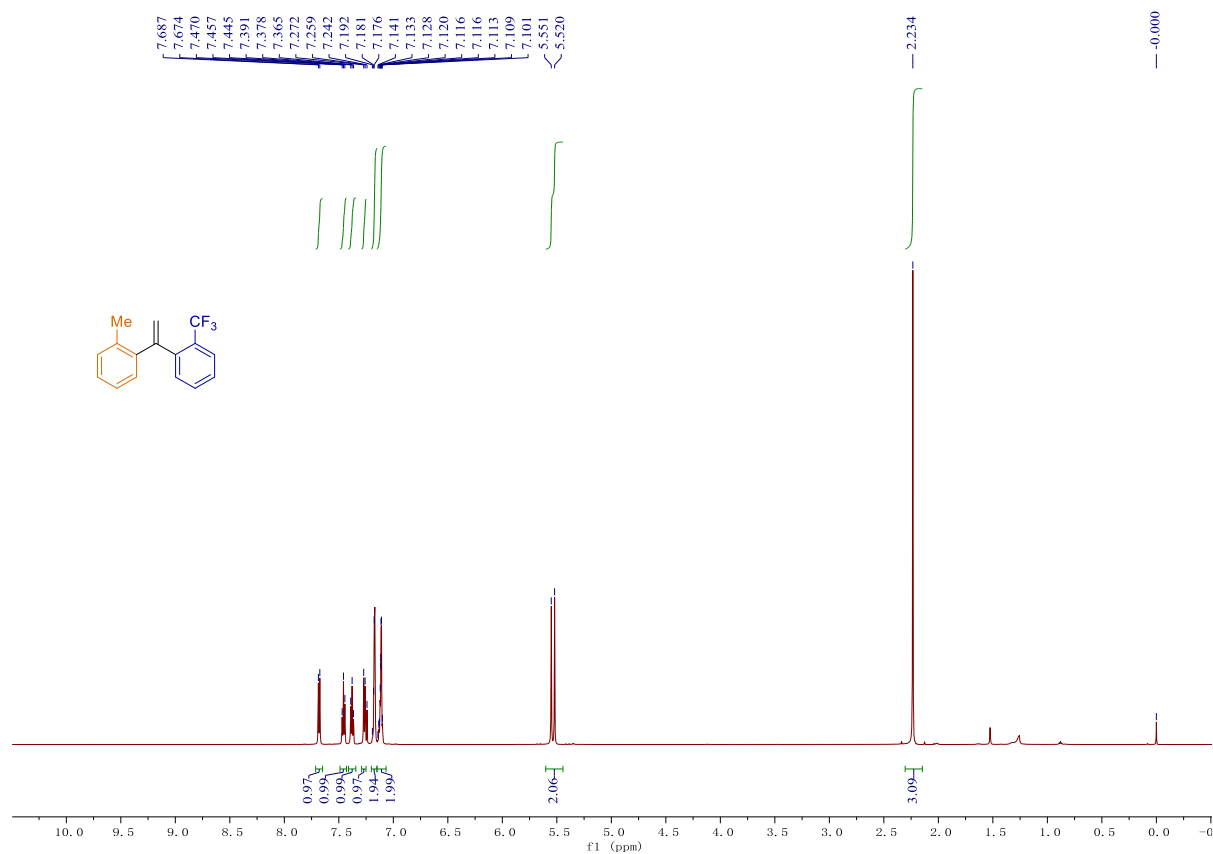
Supplementary Fig. 139 ¹³C NMR (126 MHz, Chloroform-*d*) of naphthalen-2-yl(5,6,7,8-tetrahydronaphthalen-2-yl)methanone (1w).



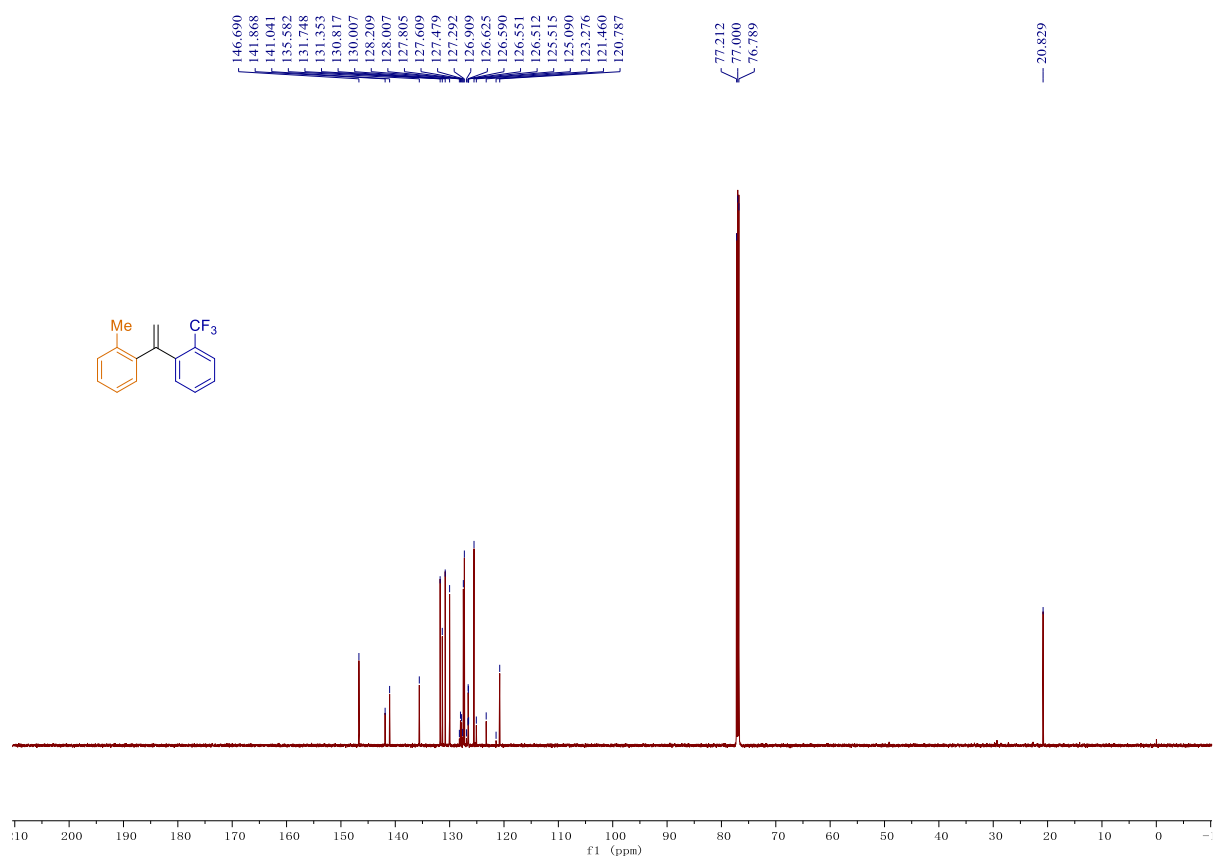
Supplementary Fig. 140 ¹H NMR (500 MHz, Chloroform-*d*) of (4-((tert-butyldimethylsilyl)oxy)phenyl)(4-chlorophenyl)methanone (1q).



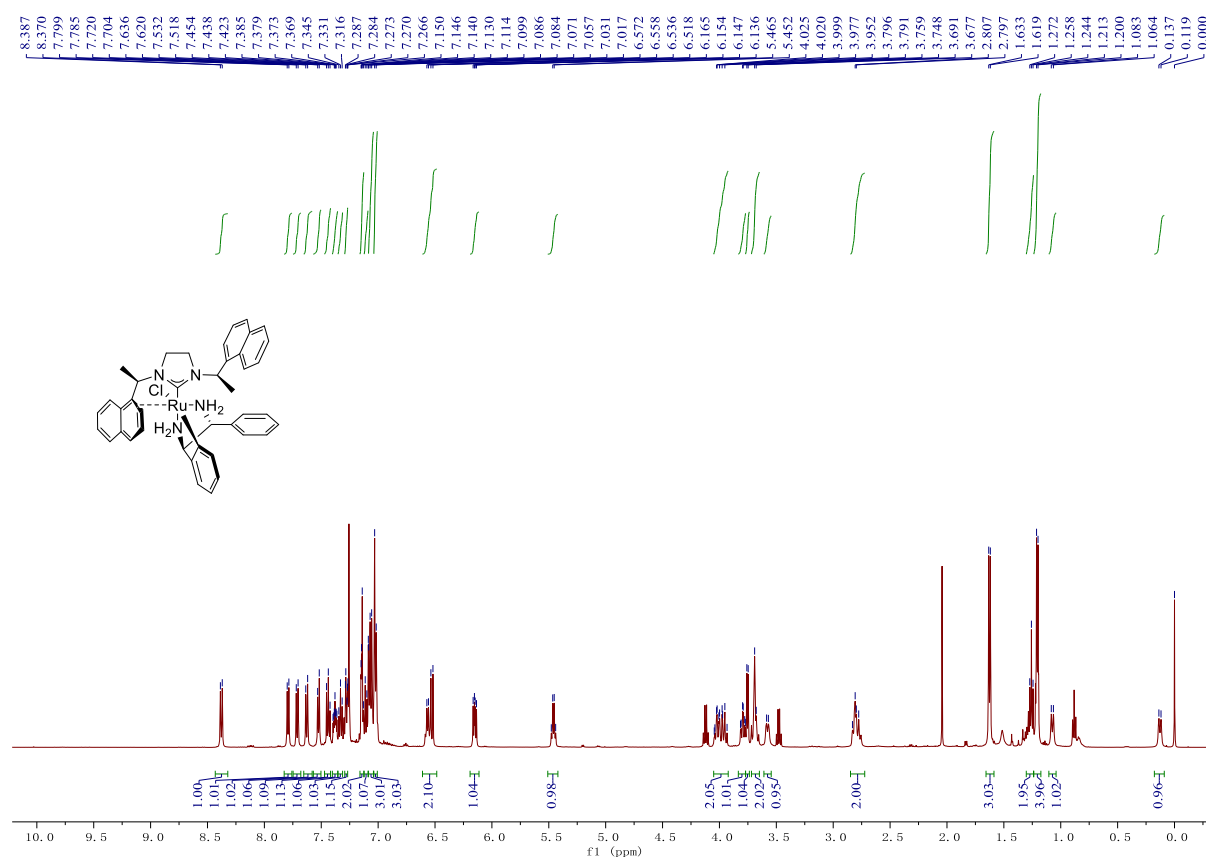
Supplementary Fig. 141 ¹³C NMR (126 MHz, Chloroform-*d*) of (4-((tert-butyldimethylsilyl)oxy)phenyl)(4-chlorophenyl)methanone (1q).



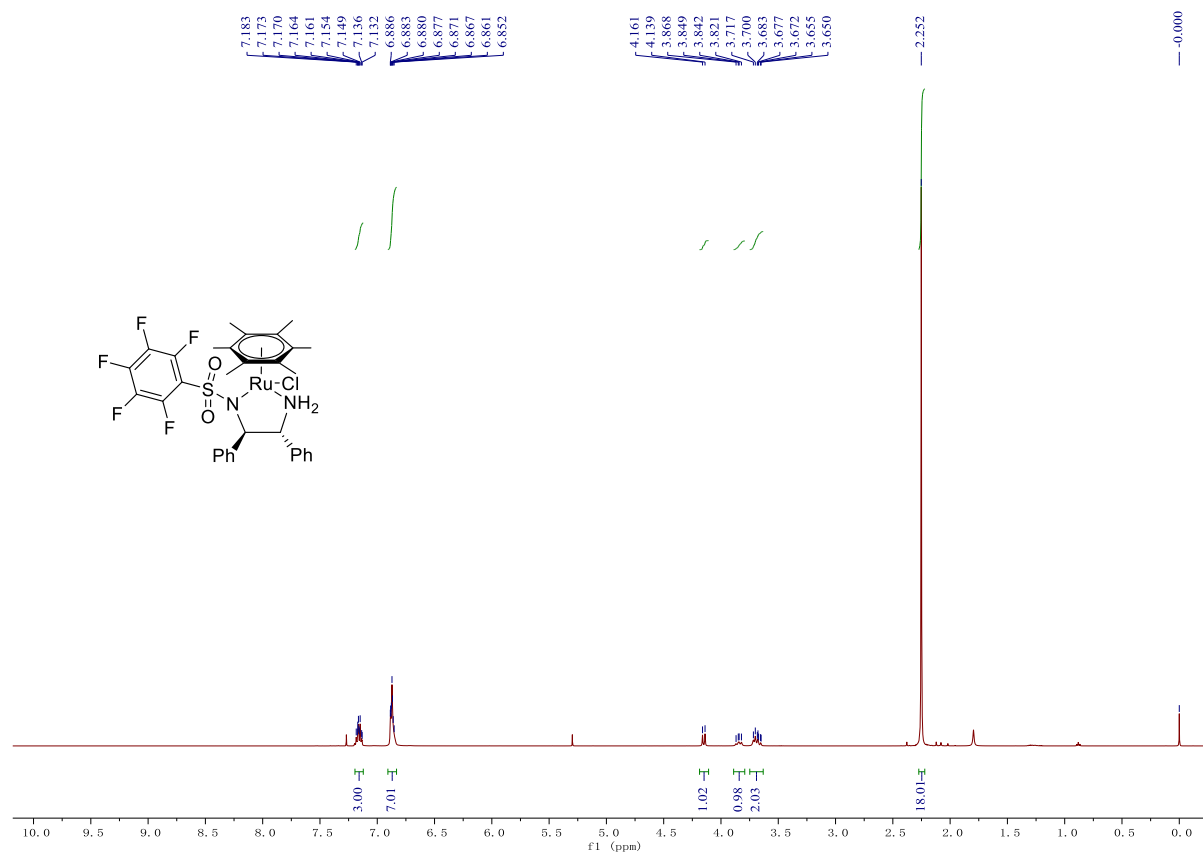
Supplementary Fig. 142 ¹H NMR (600 MHz, Chloroform-*d*) of 1-methyl-2-(1-(2-(trifluoromethyl)phenyl)vinyl)benzene (3d).



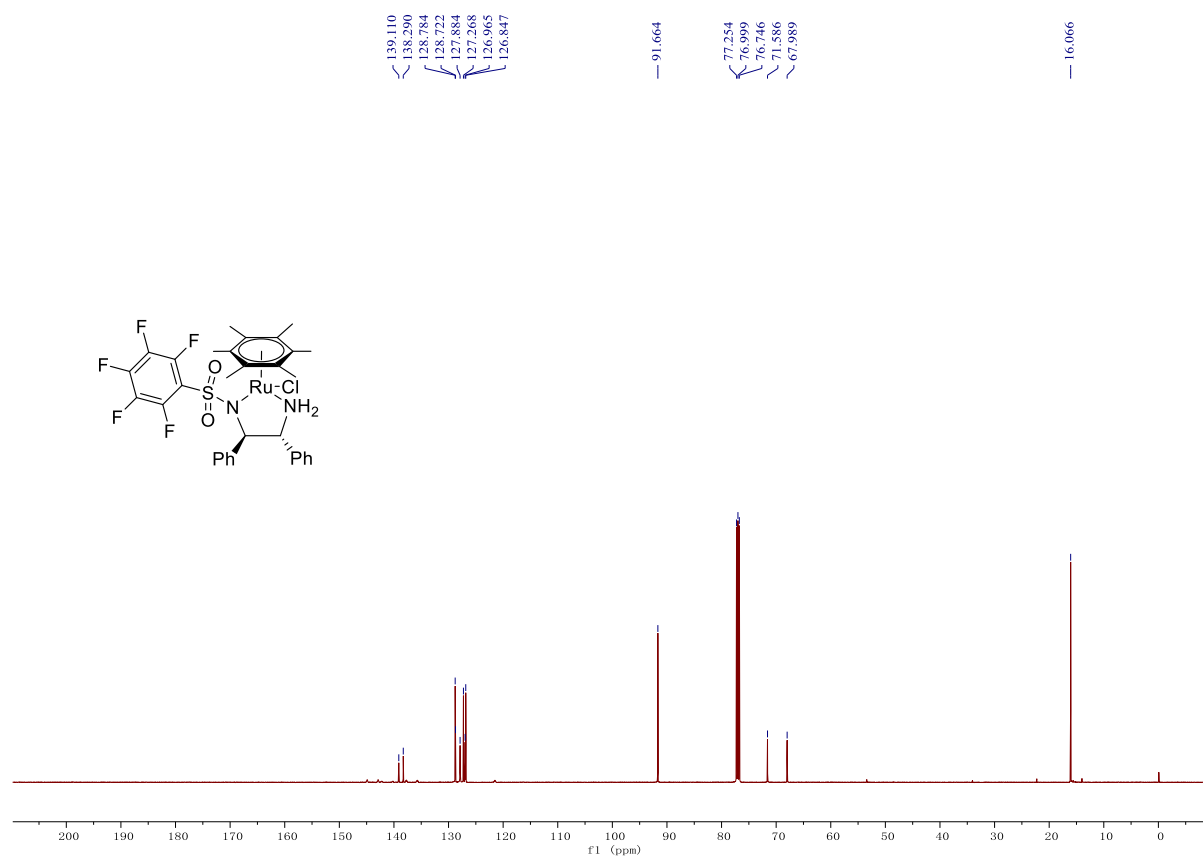
Supplementary Fig. 143 ¹³C NMR (151 MHz, Chloroform-*d*) of 1-methyl-2-(1-(2-(trifluoromethyl)phenyl)vinyl)benzene (3d).



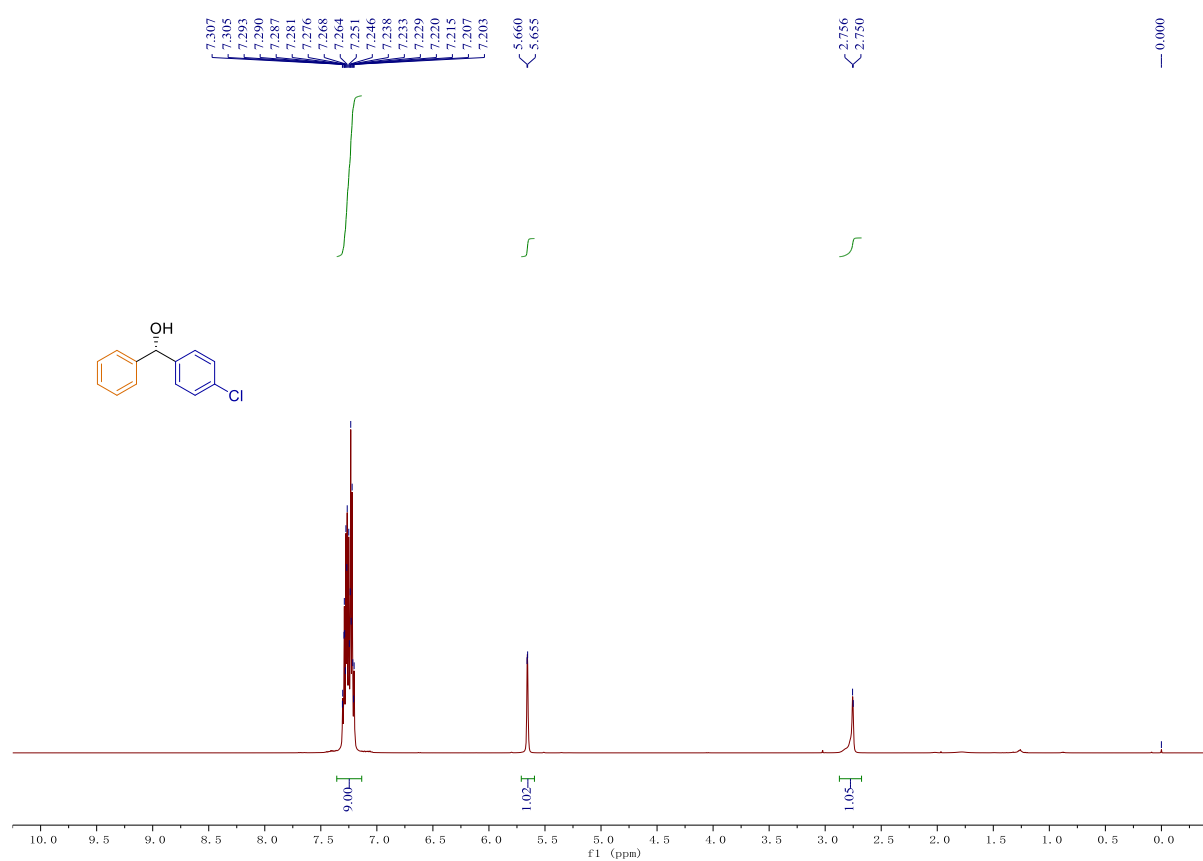
Supplementary Fig. 144 ¹H NMR (500 MHz, Chloroform-*d*) of Ru-1.



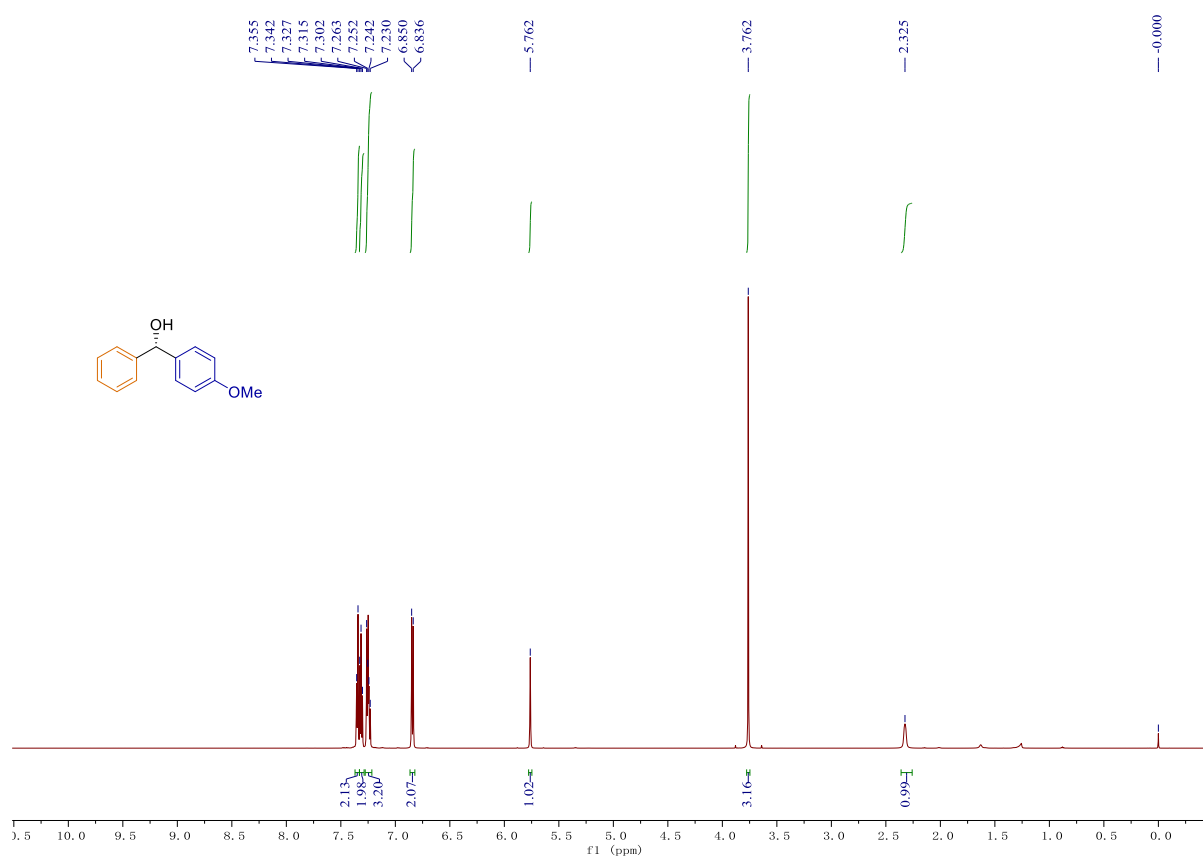
Supplementary Fig. 145 ¹H NMR (500 MHz, Chloroform-*d*) of Ru-2.



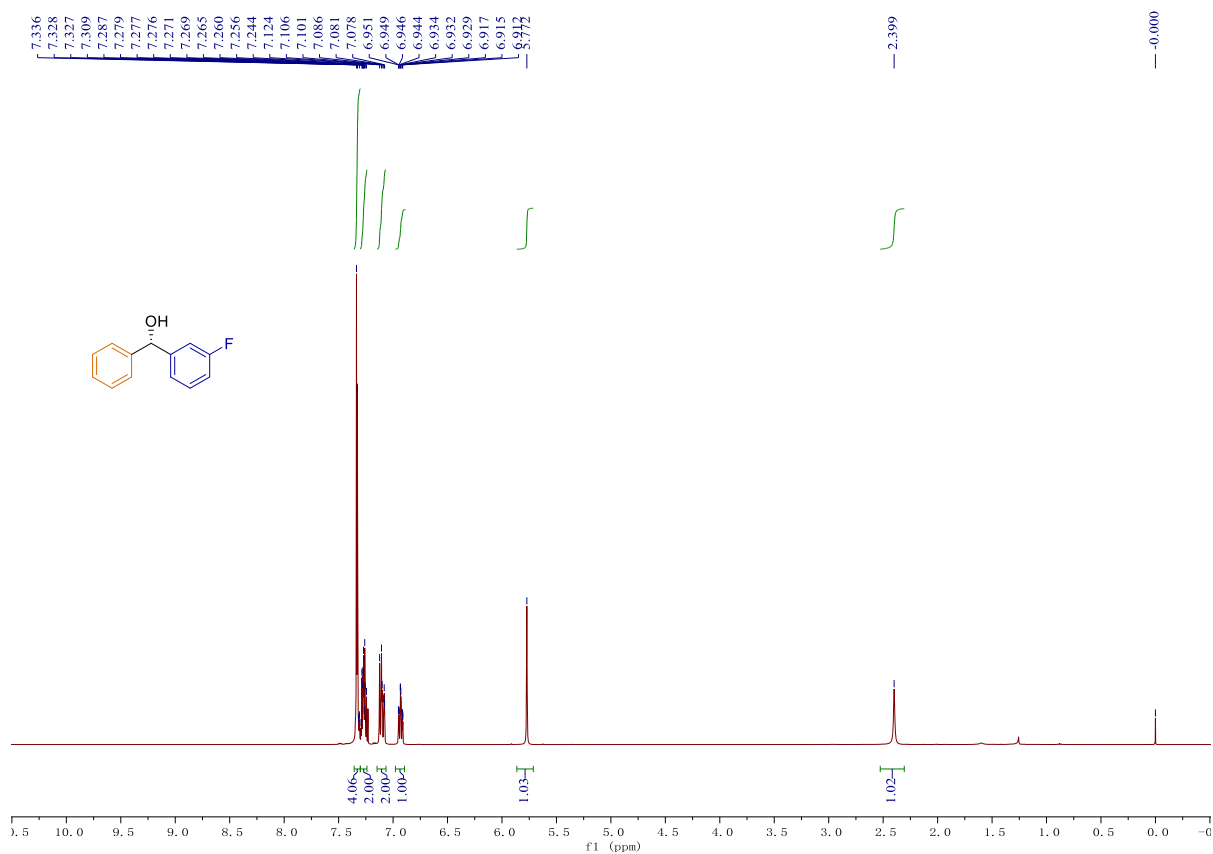
Supplementary Fig. 146 ¹³C NMR (126 MHz, Chloroform-*d*) of Ru-2.



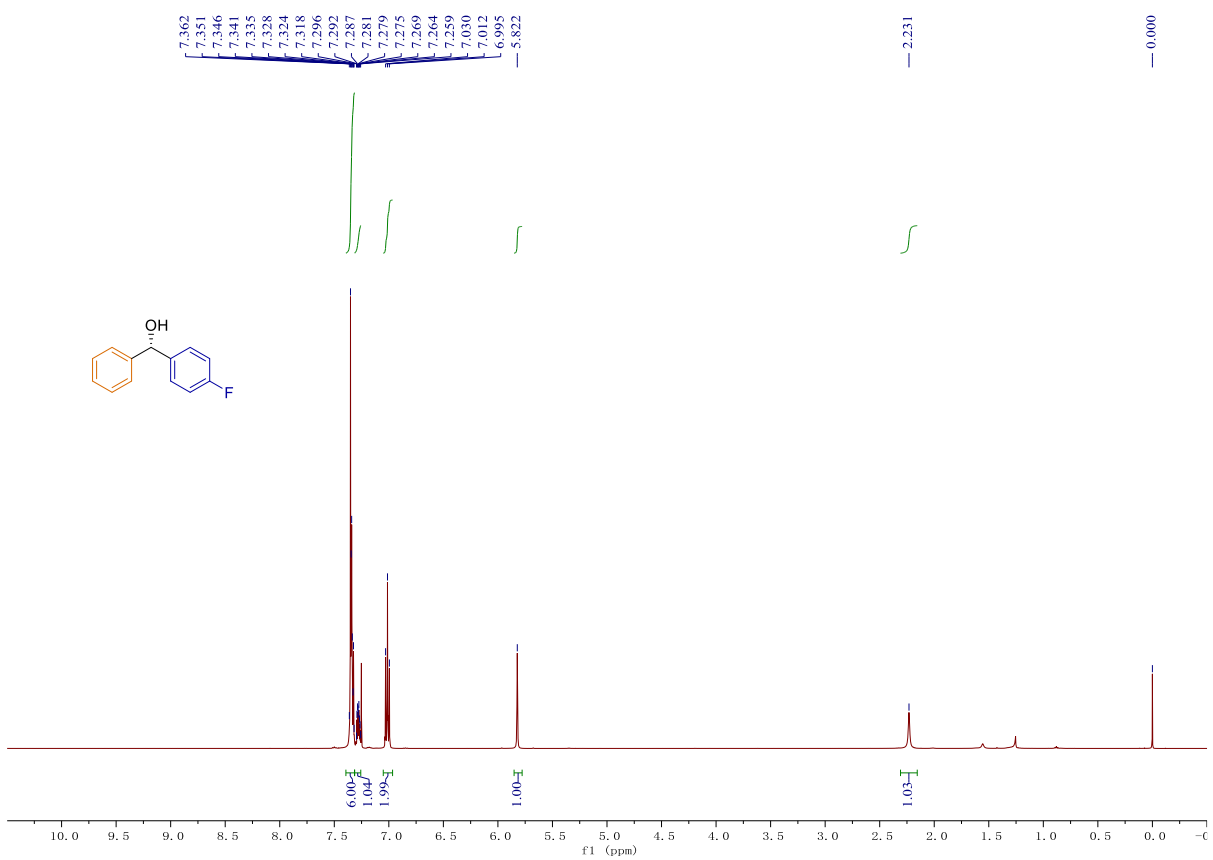
Supplementary Fig. 147 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-(4-chlorophenyl)(phenyl)methanol (4a).



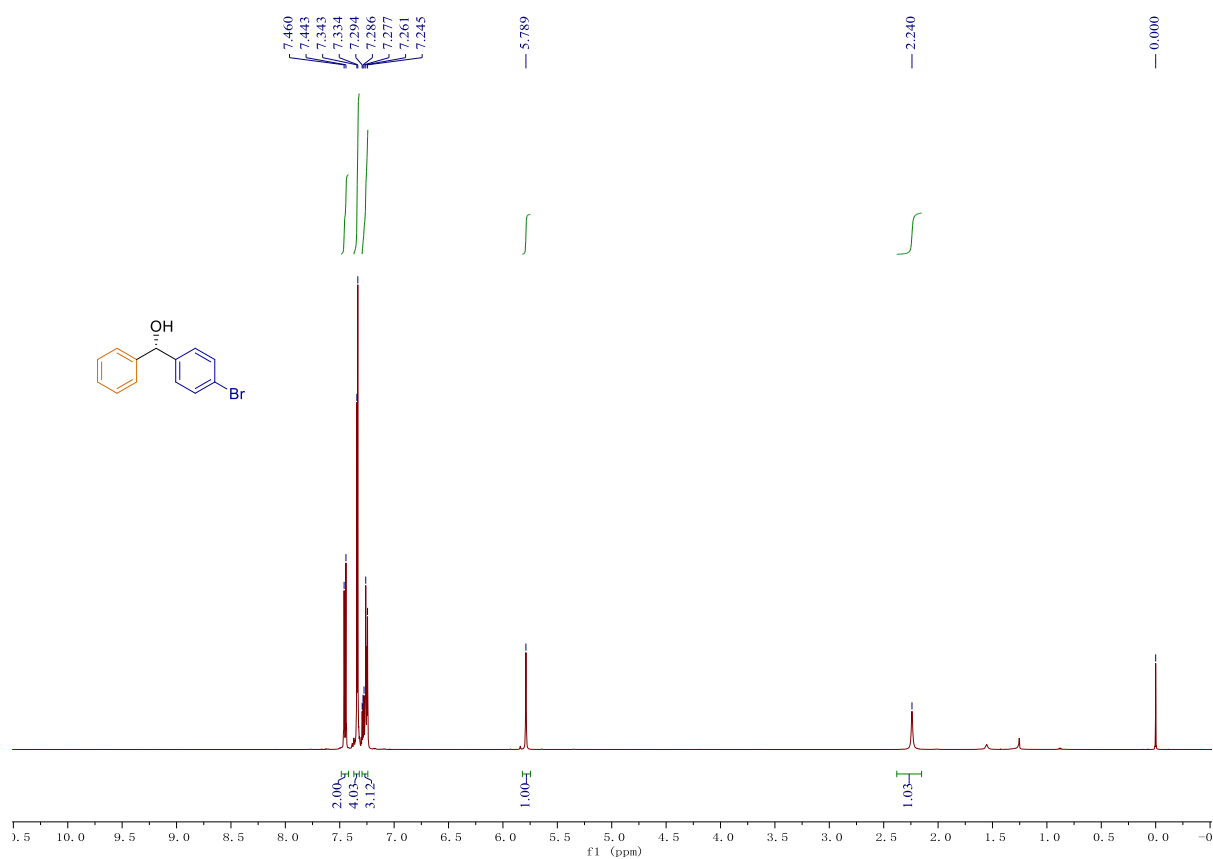
Supplementary Fig. 148 ¹H NMR (600 MHz, Chloroform-*d*) of (*R*)-(4-methoxyphenyl)(phenyl)methanol (4b).



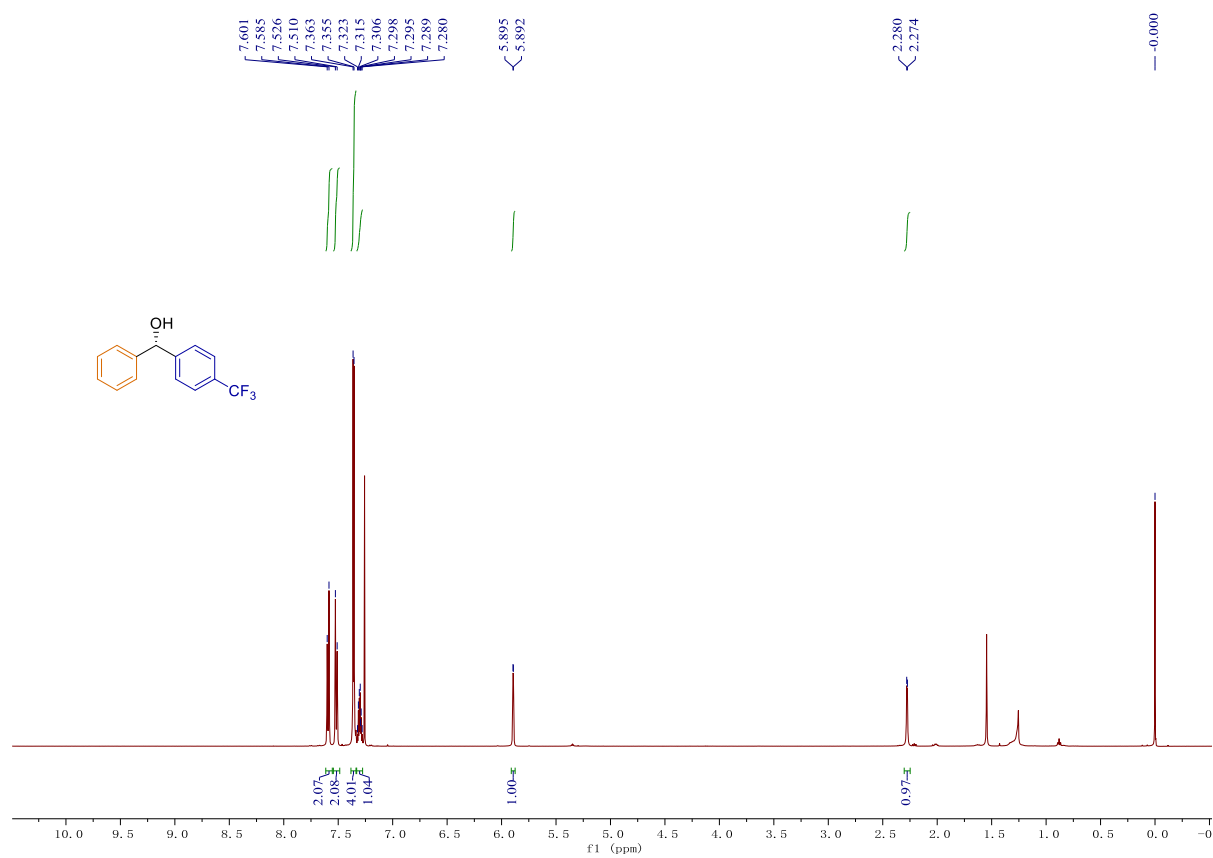
Supplementary Fig. 149 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-(3-fluorophenyl)(phenyl)methanol (4c).



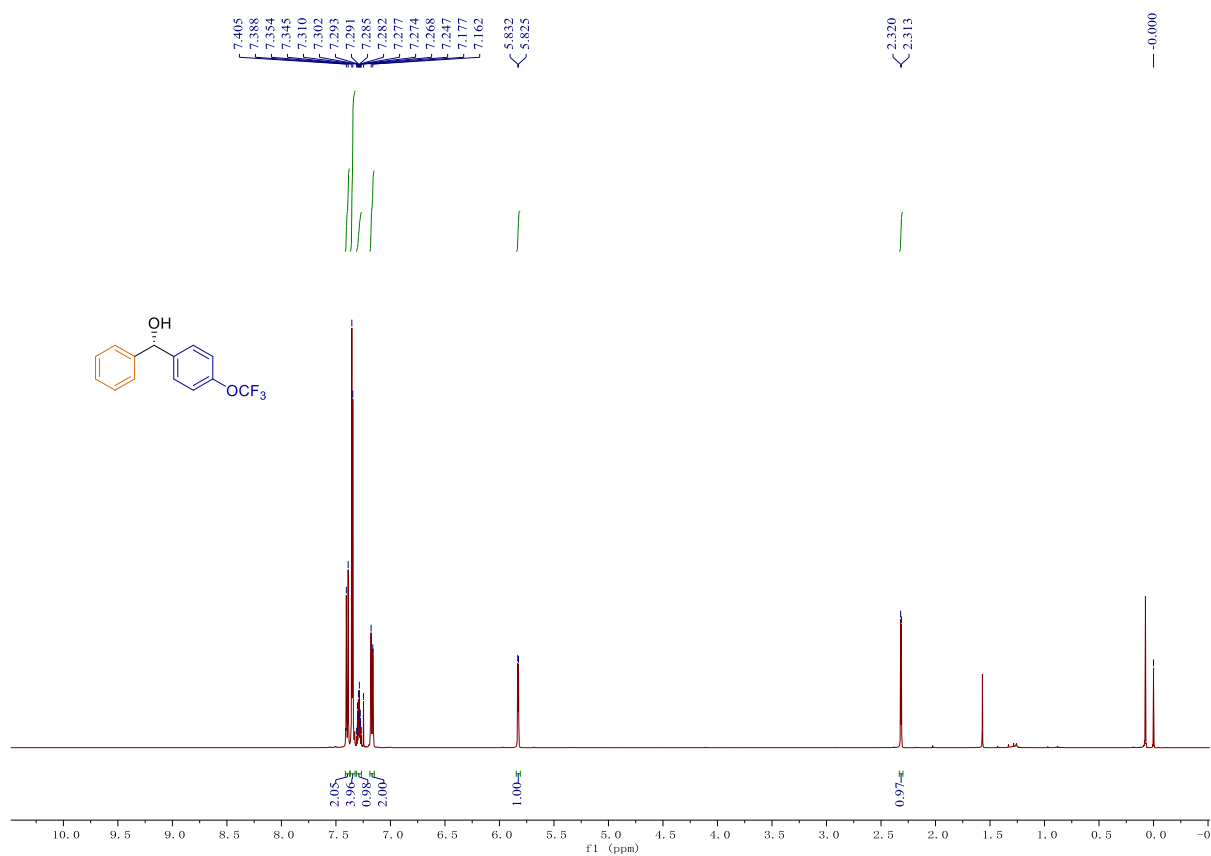
Supplementary Fig. 150 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-(4-fluorophenyl)(phenyl)methanol (4d).



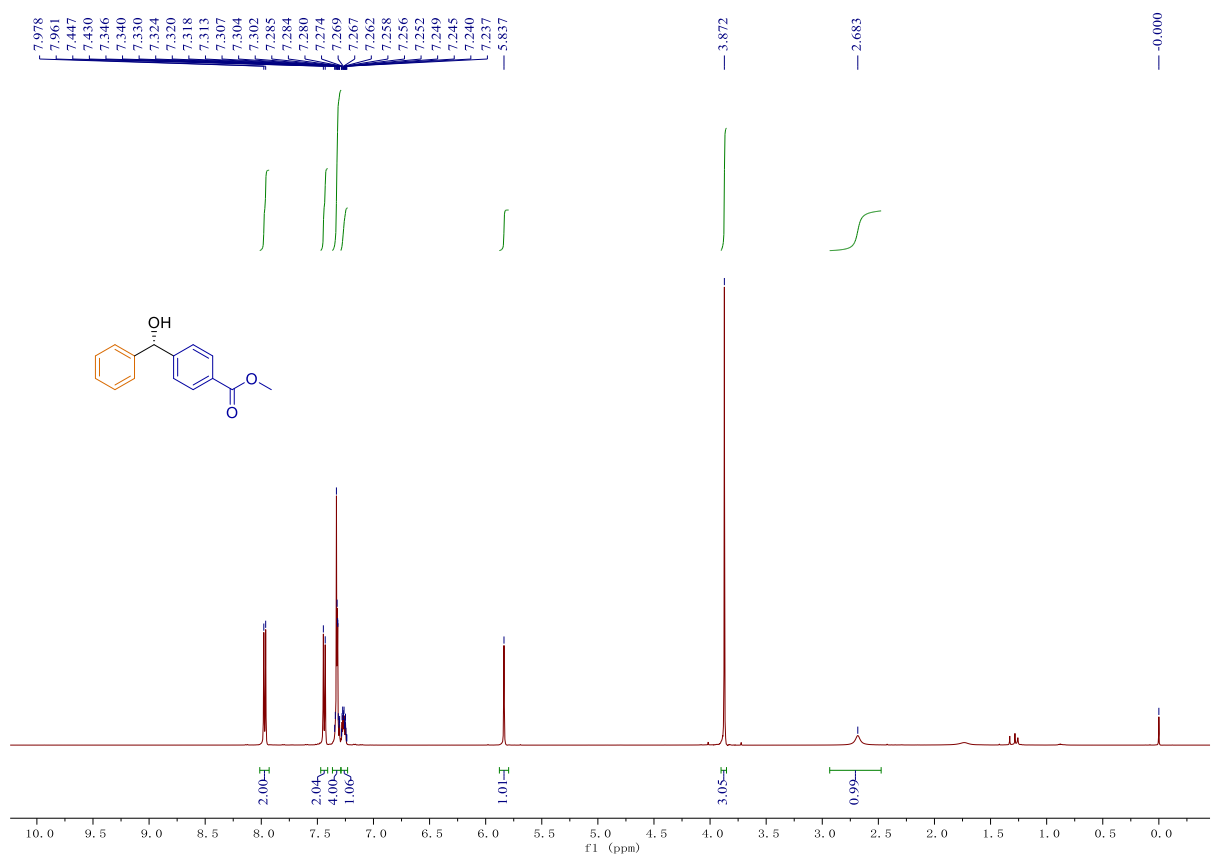
Supplementary Fig. 151 ^1H NMR (500 MHz, Chloroform- d) of (*S*)-(4-bromophenyl)(phenyl)methanol (4e).



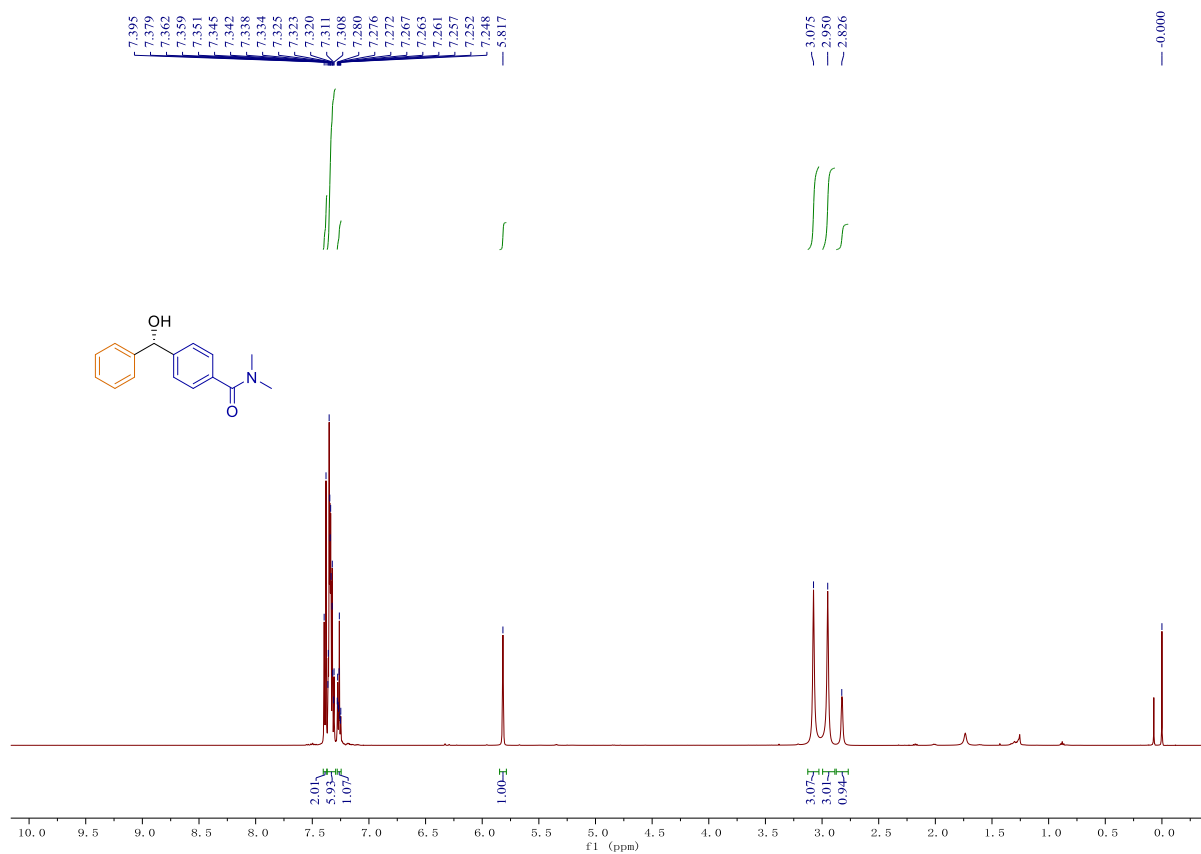
Supplementary Fig. 152 ^1H NMR (500 MHz, Chloroform- d) of (*S*)-phenyl(4-(trifluoromethyl)phenyl)methanol (4f).



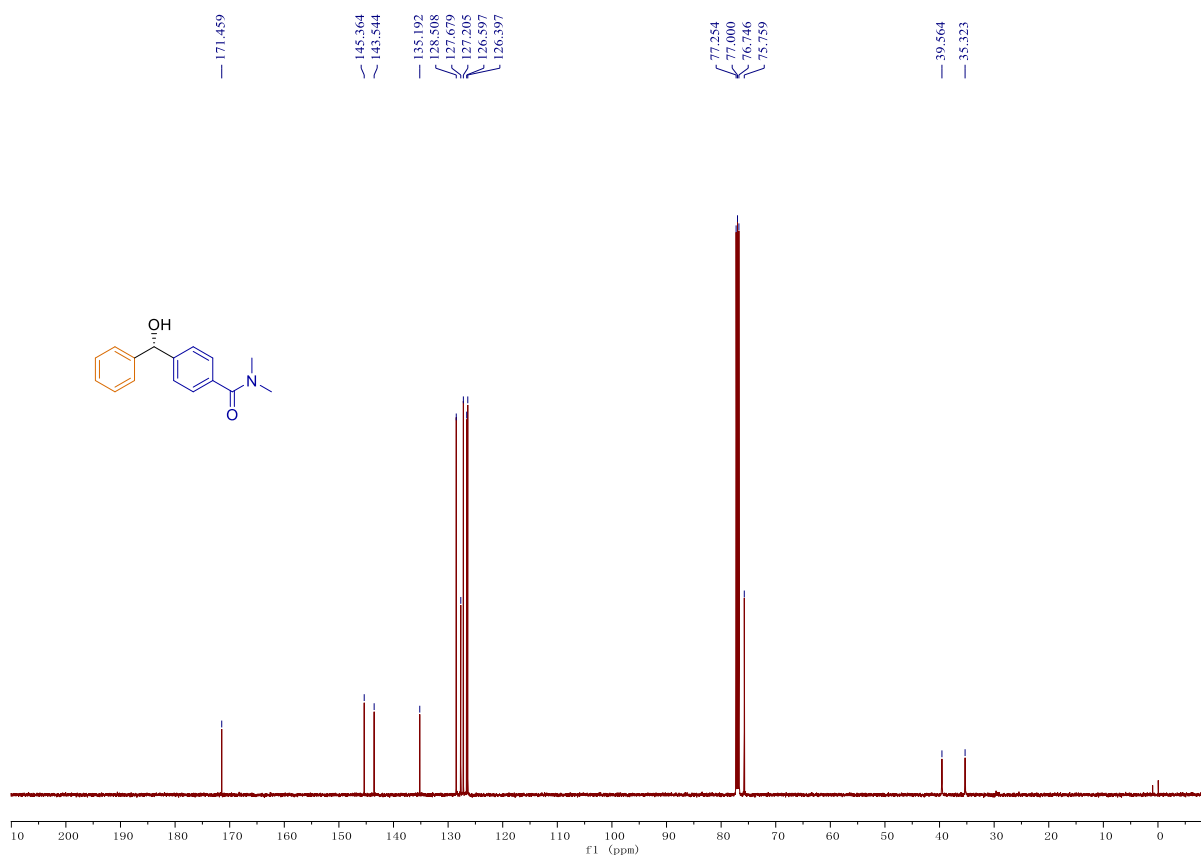
Supplementary Fig. 153 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-phenyl(4-(trifluoromethoxy)phenyl)methanol (4g).



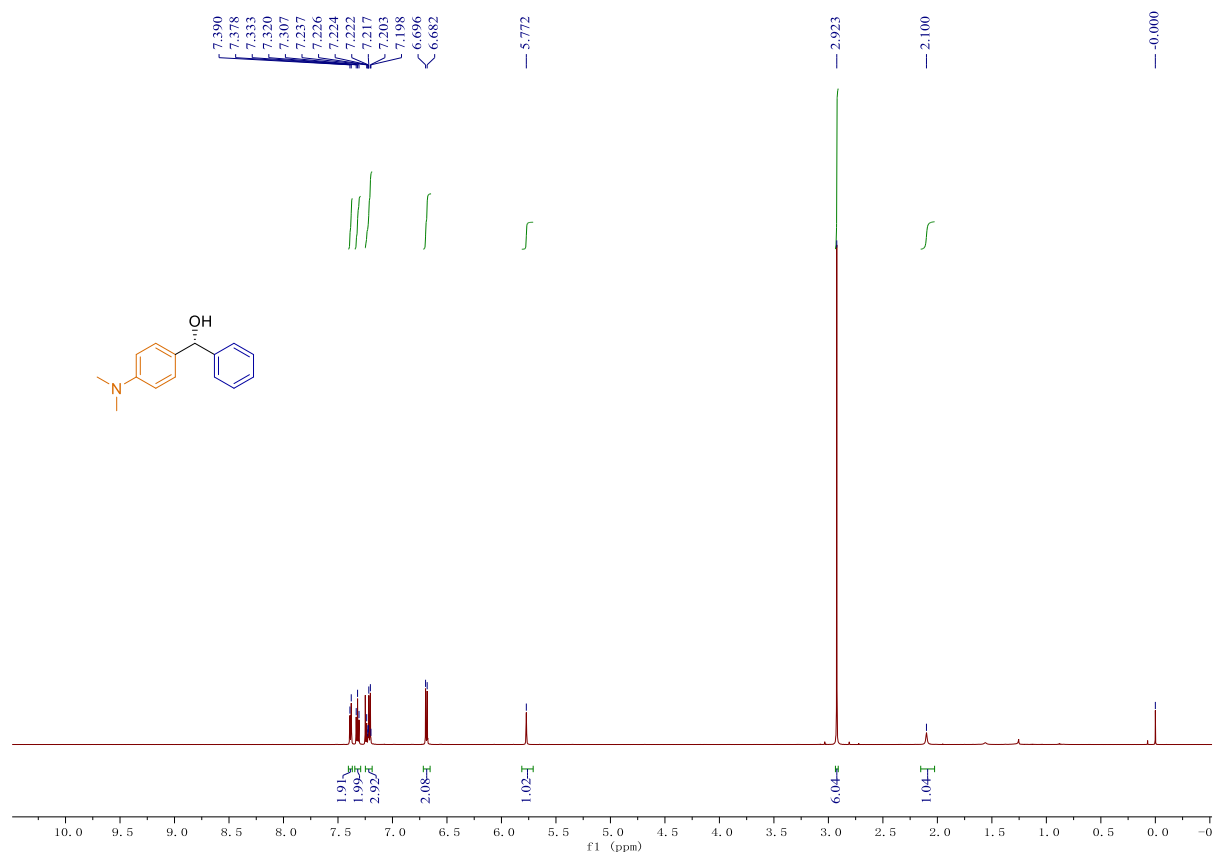
Supplementary Fig. 154 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-4-(hydroxy(phenyl)methyl)benzoate (4h).



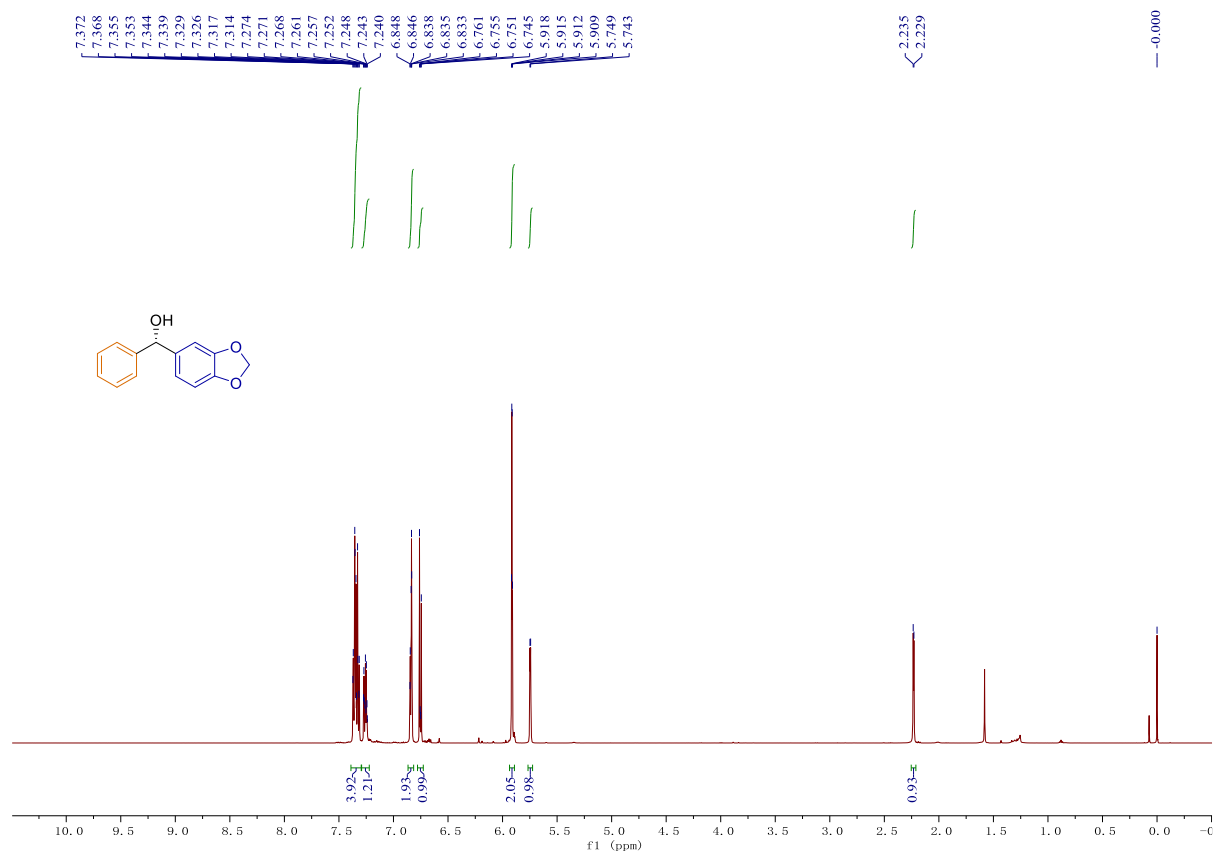
Supplementary Fig. 155 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-4-(hydroxy(phenyl)methyl)-*N,N*-dimethylbenzamide (4i).



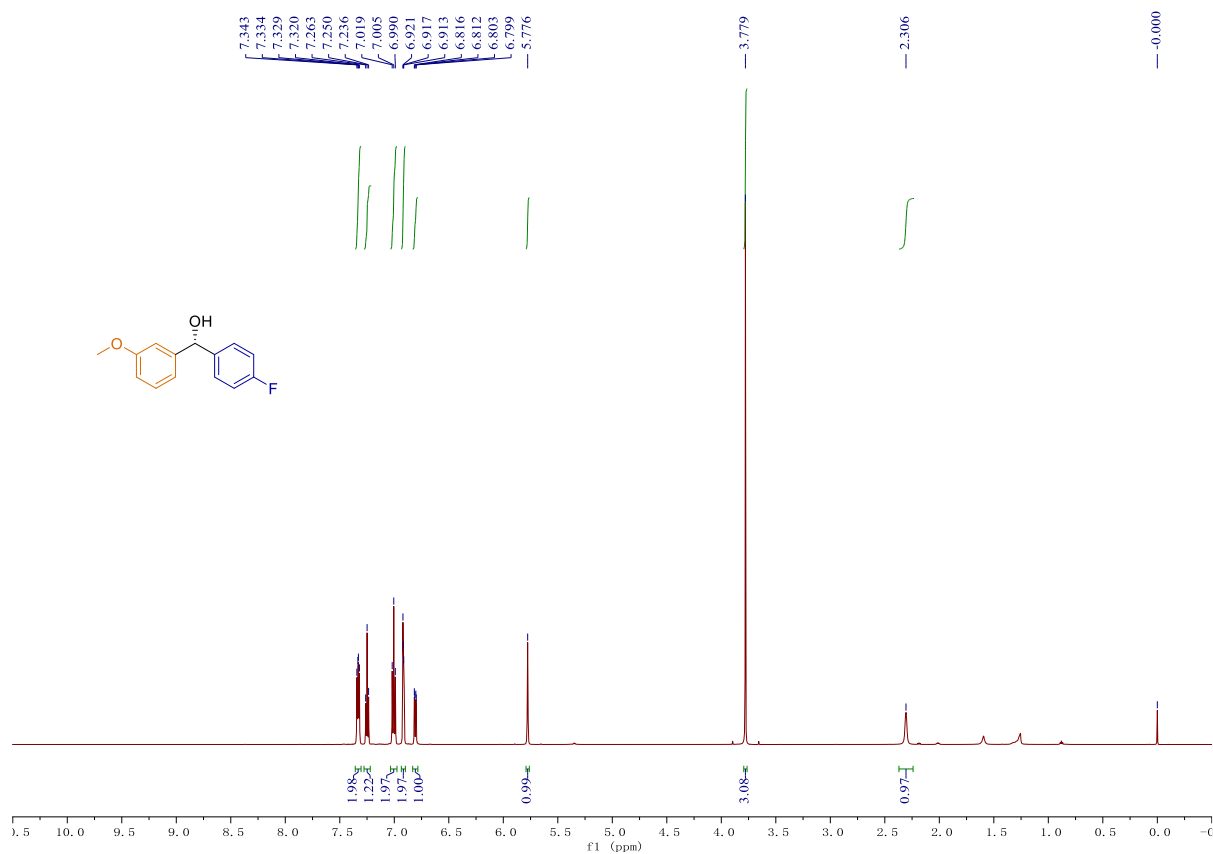
Supplementary Fig. 156 ¹³C NMR (126 MHz, Chloroform-*d*) of (S)-4-(hydroxy(phenyl)methyl)-*N,N*-dimethylbenzamide (4i).



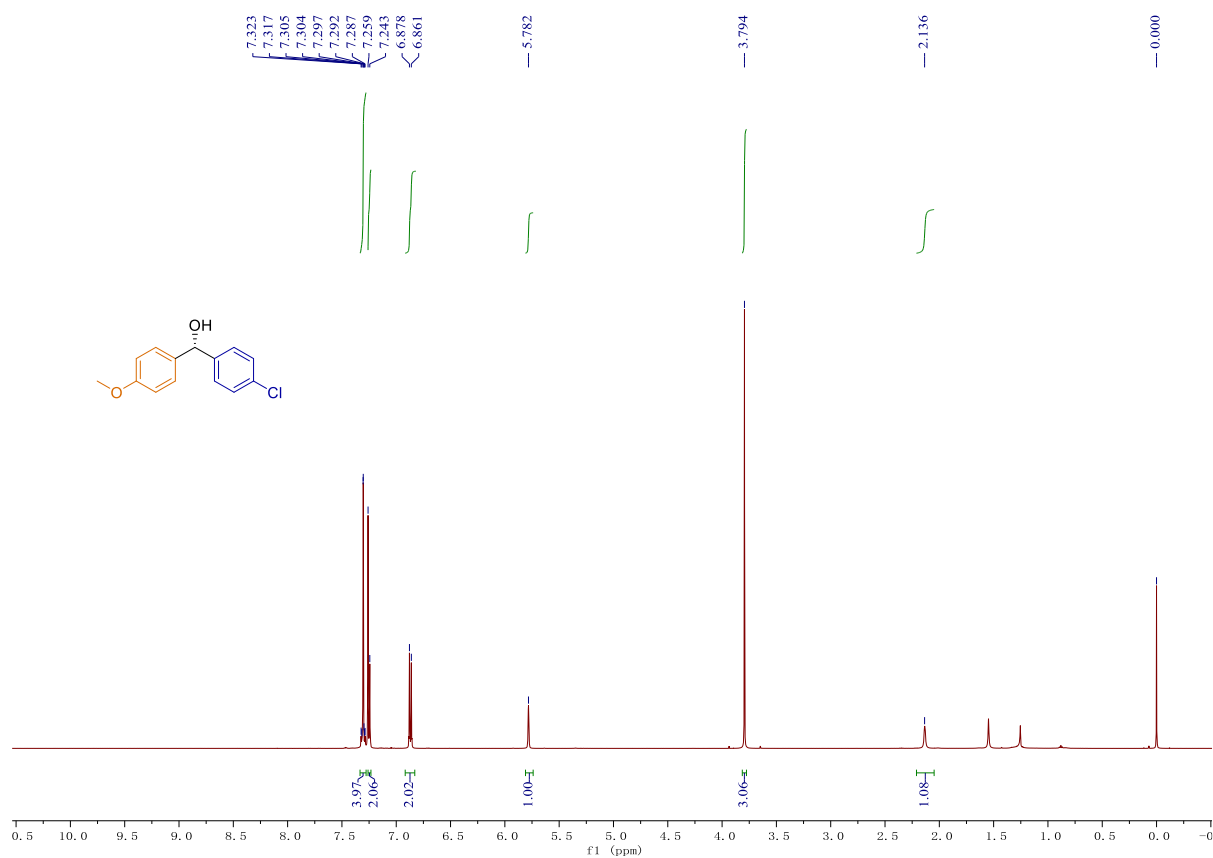
Supplementary Fig. 157 ^1H NMR (500 MHz, Chloroform- d) of (R)-4-(dimethylamino)phenyl(phenyl)methanol (4j).



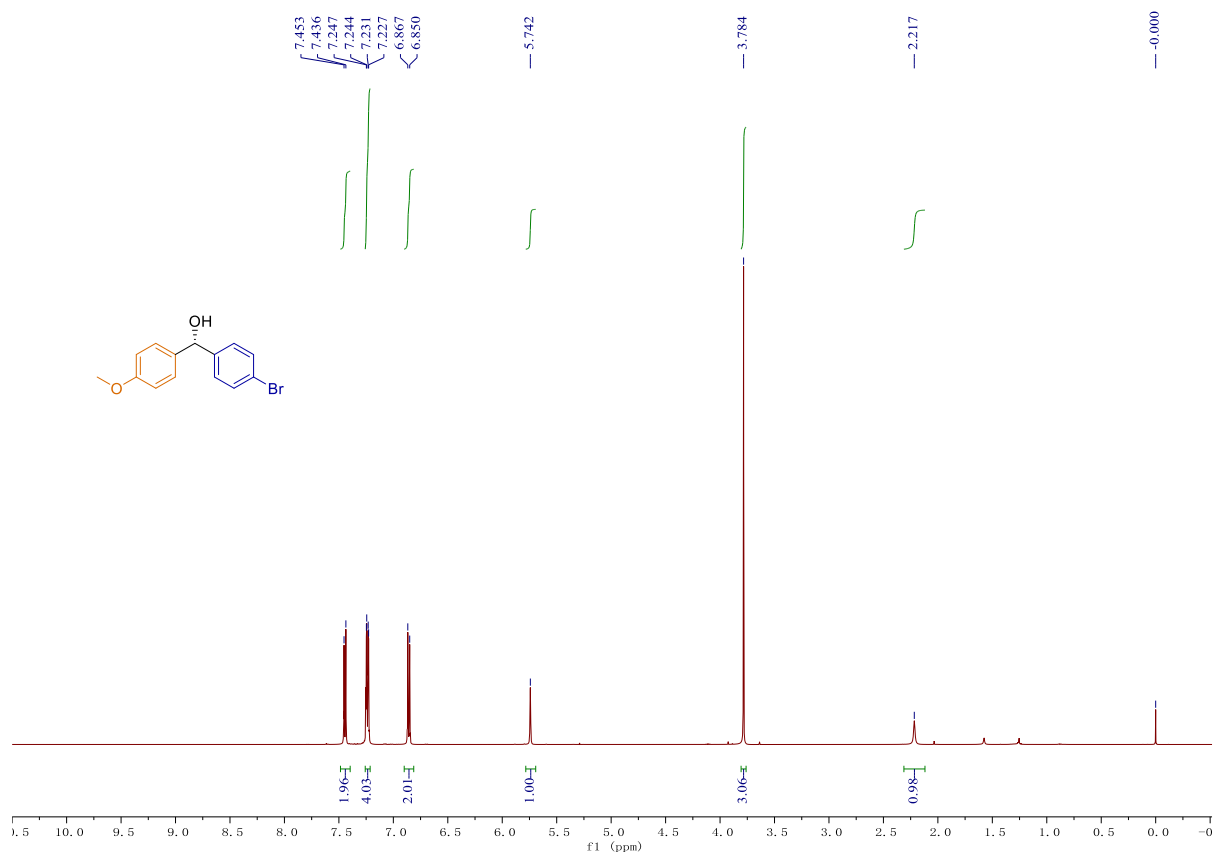
Supplementary Fig. 158 ^1H NMR (500 MHz, Chloroform- d) of (S)-benzo[d][1,3]dioxol-5-yl(phenyl)methanol (4k).



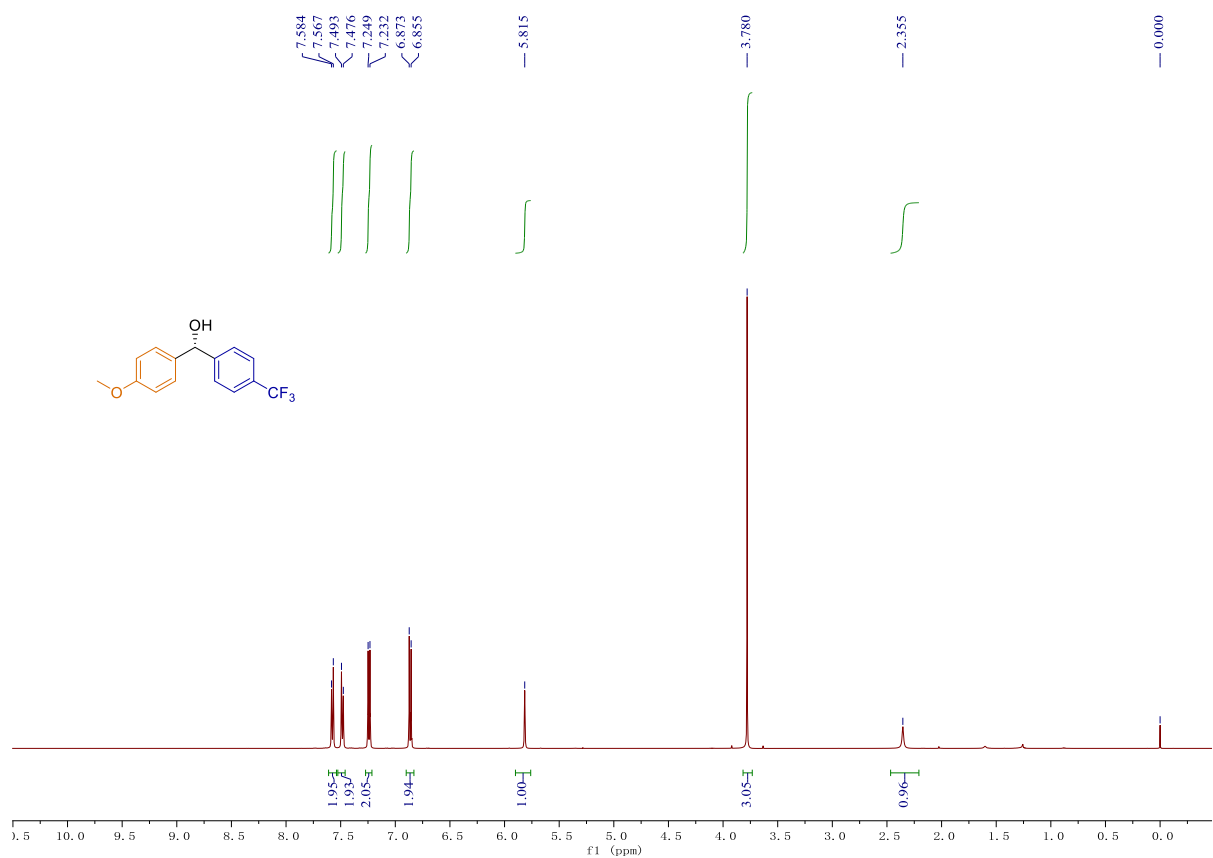
Supplementary Fig. 159 ¹H NMR (600 MHz, Chloroform-*d*) (*R*)-(4-fluorophenyl)(3-methoxyphenyl)methanol (4l).



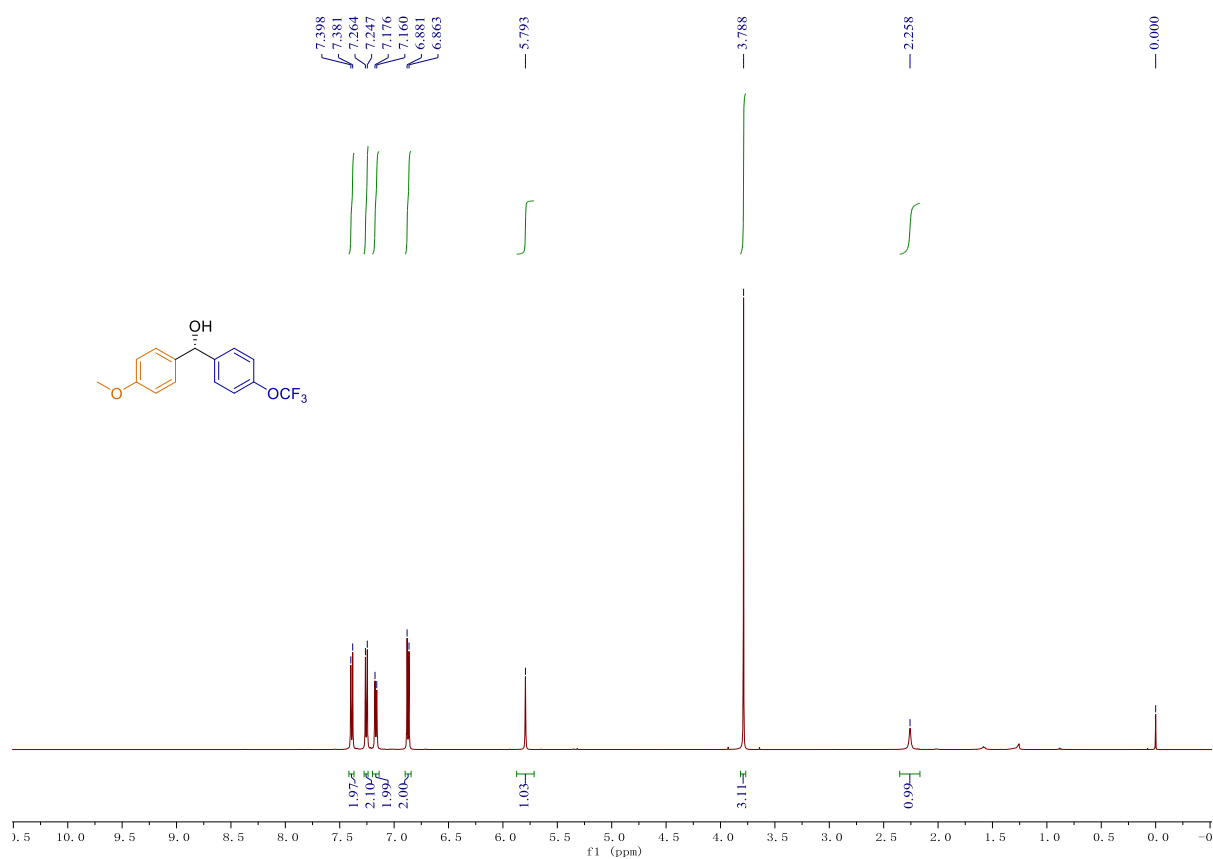
Supplementary Fig. 160 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-(4-chlorophenyl)(4-methoxyphenyl)methanol (4m).



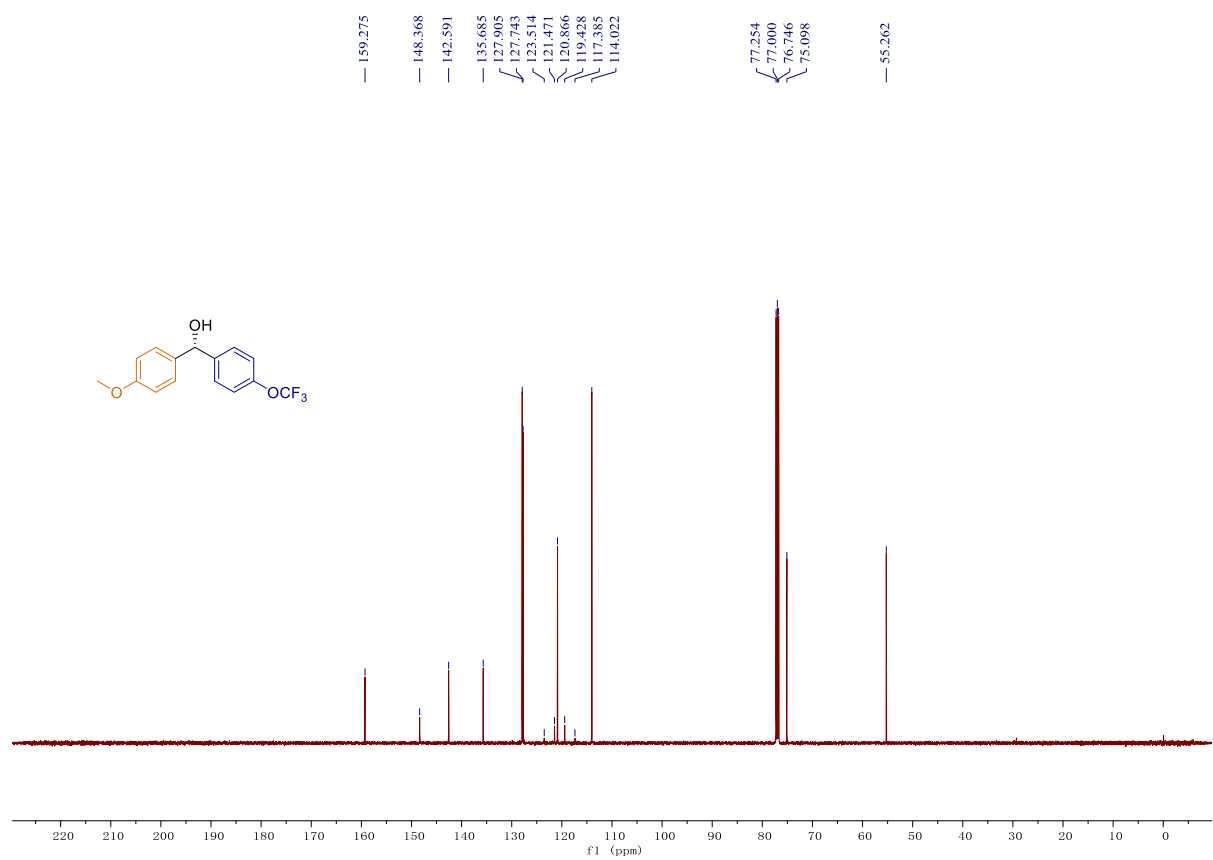
Supplementary Fig. 161 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-(4-bromophenyl)(4-methoxyphenyl)methanol (4n).



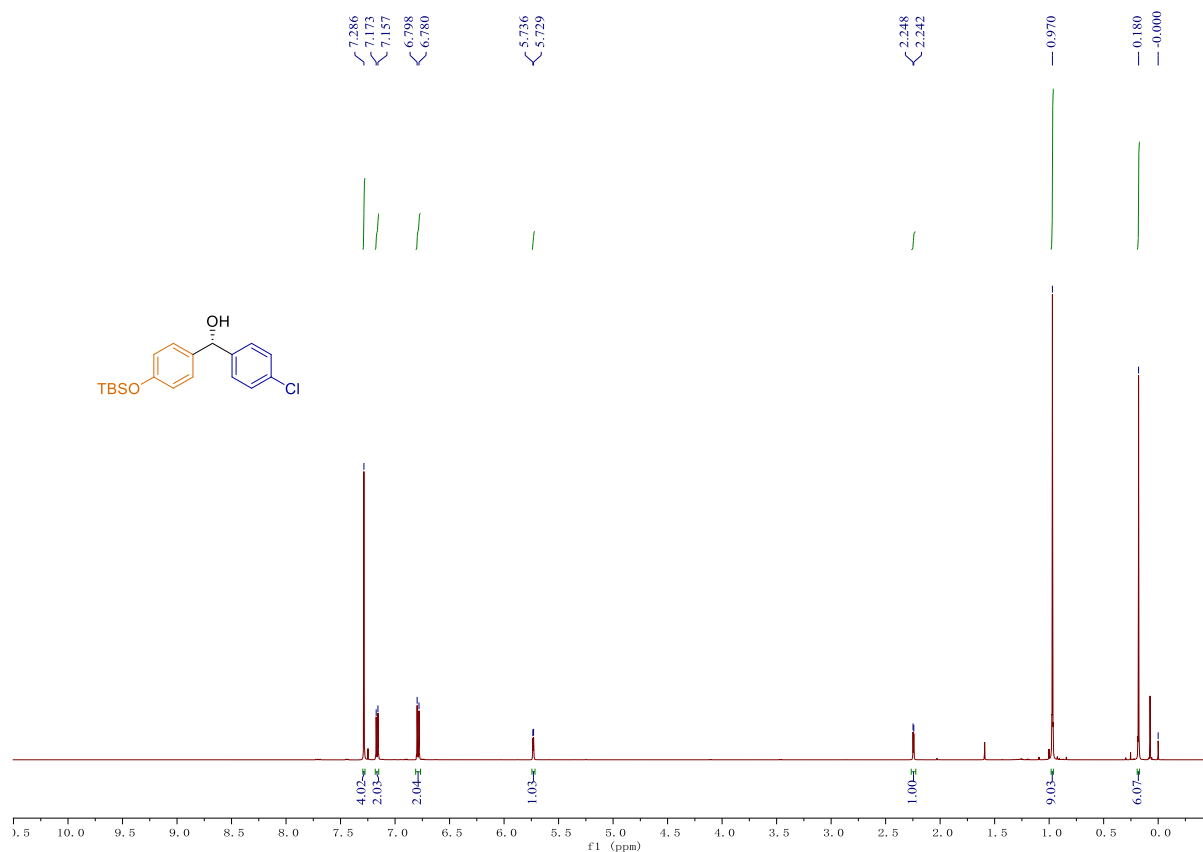
Supplementary Fig. 162 ¹H NMR (500 MHz, Chloroform-*d*) of (*R*)-(4-methoxyphenyl)(4-(trifluoromethyl)phenyl)methanol (4o).



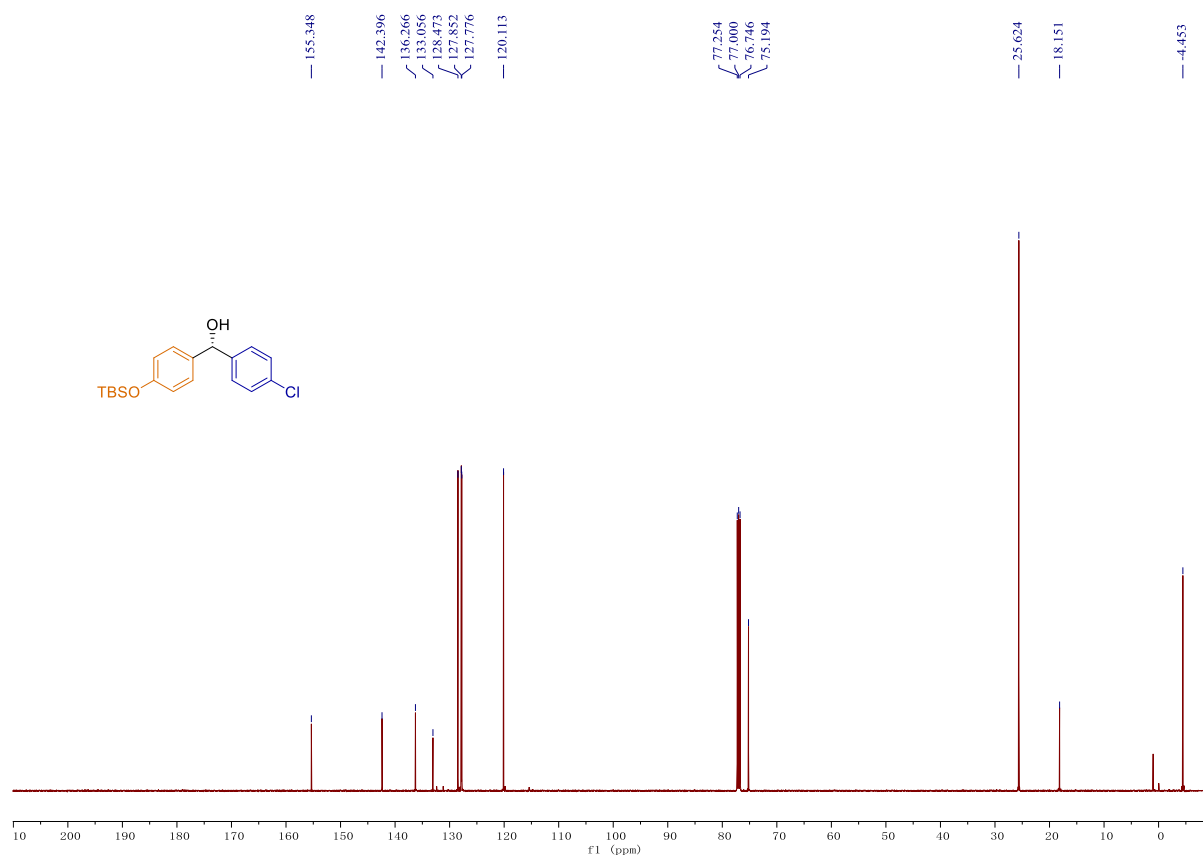
Supplementary Fig. 163 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-(4-methoxyphenyl)(4-(trifluoromethoxy)phenyl)methanol (4p).



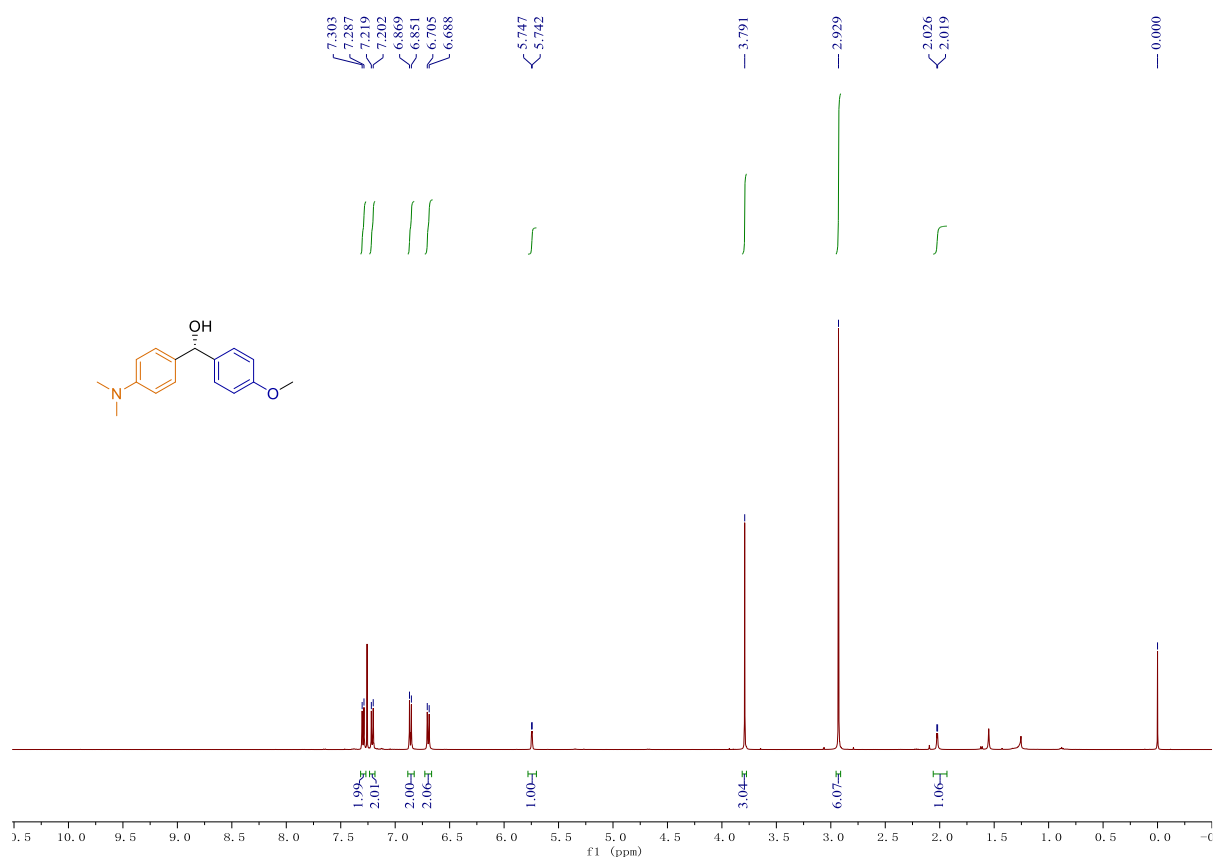
Supplementary Fig. 164 ¹³C NMR (126 MHz, Chloroform-*d*) of (S)-(4-methoxyphenyl)(4-(trifluoromethoxy)phenyl)methanol (4p).



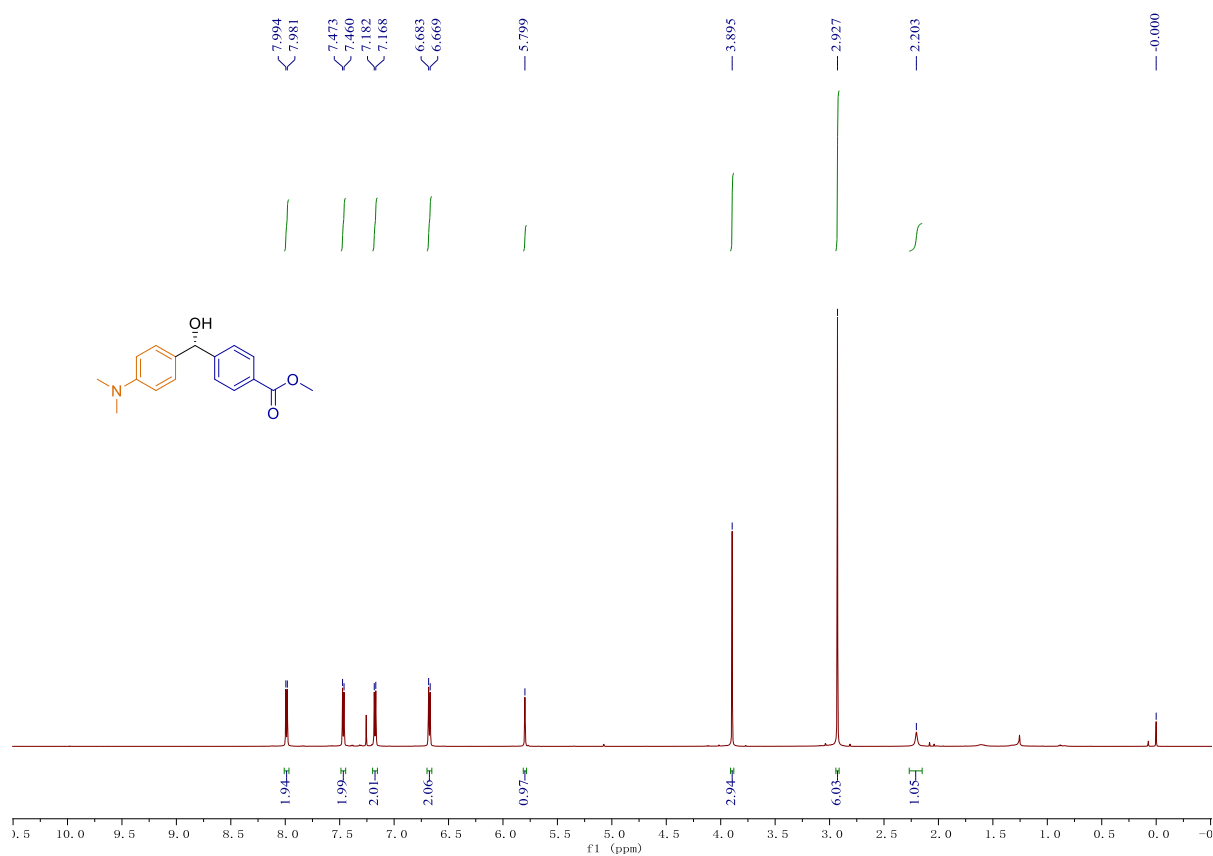
Supplementary Fig. 165 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-4-((tert-butyldimethylsilyl)oxy)phenyl(4-chlorophenyl)methanol (4q).



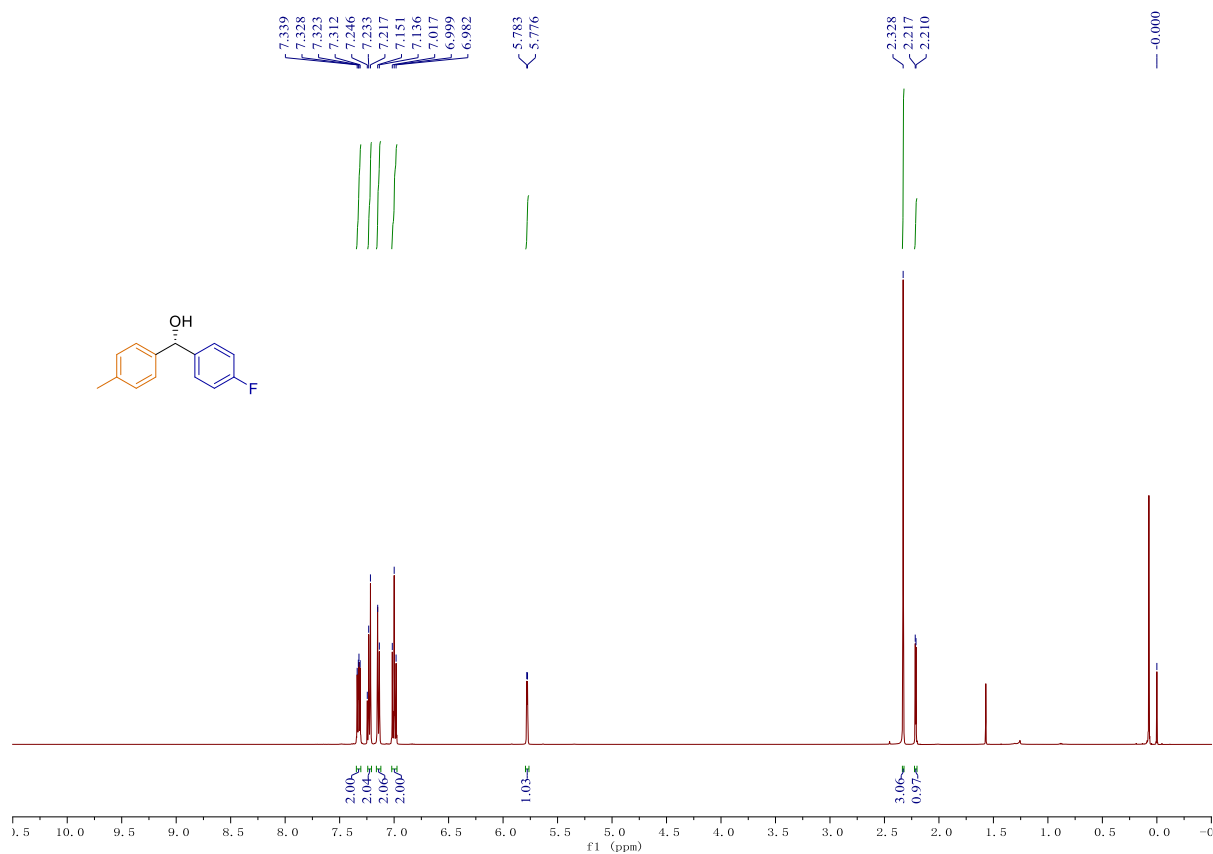
Supplementary Fig. 166 ¹³C NMR (126 MHz, Chloroform-*d*) of (*S*)-4-((tert-butyldimethylsilyl)oxy)phenyl(4-chlorophenyl)methanol (4q).



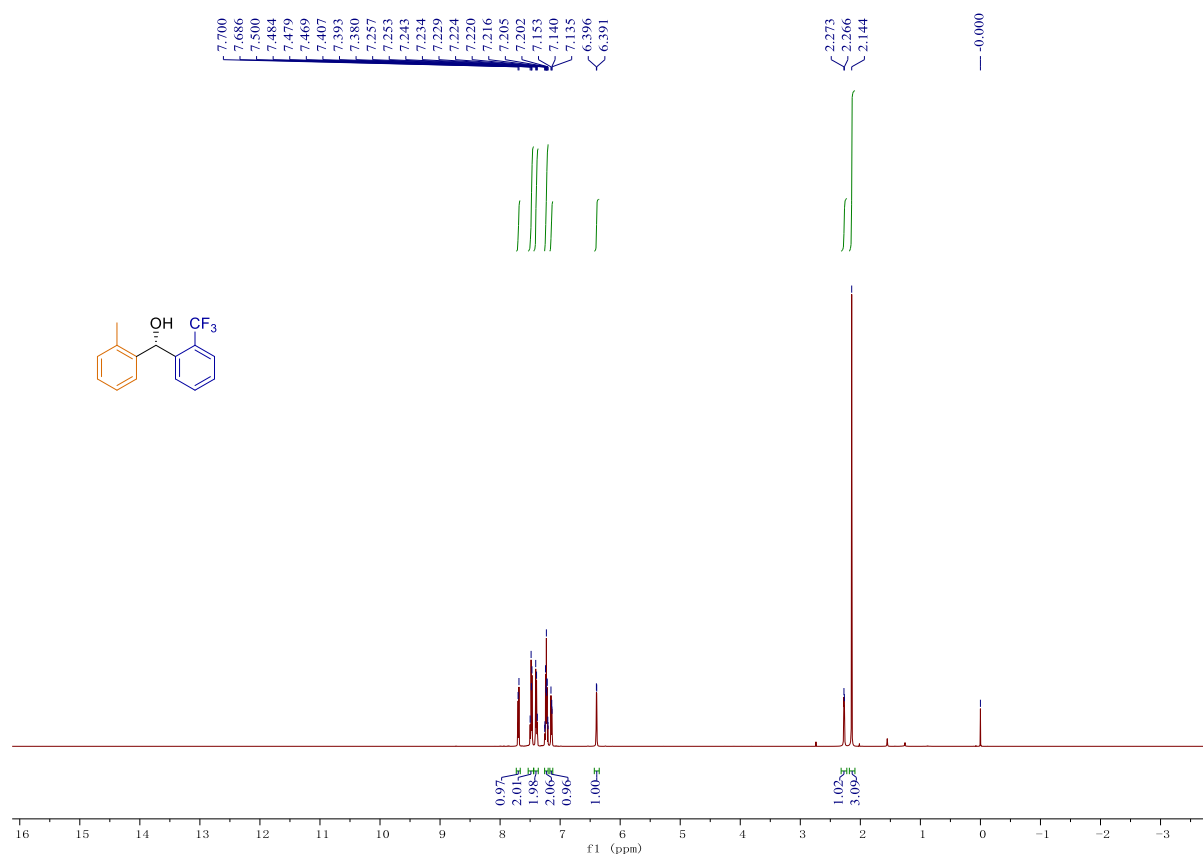
Supplementary Fig. 167 ^1H NMR (500 MHz, Chloroform- d) of (S)-4-(dimethylamino)phenyl(4-methoxyphenyl)methanol (4r).



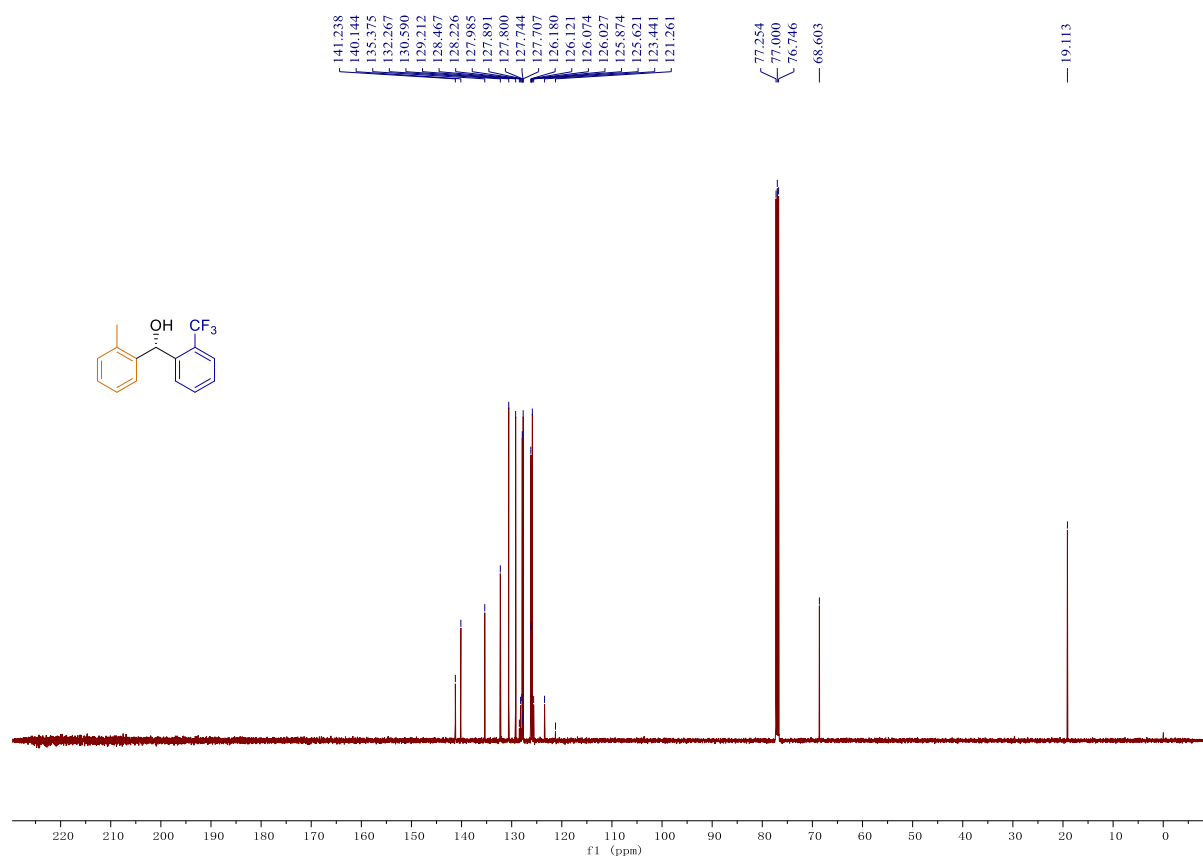
Supplementary Fig. 168 ^1H NMR (500 MHz, Chloroform- d) of methyl (R)-4-((4-(dimethylamino)phenyl)(hydroxymethyl)benzoate (4s).



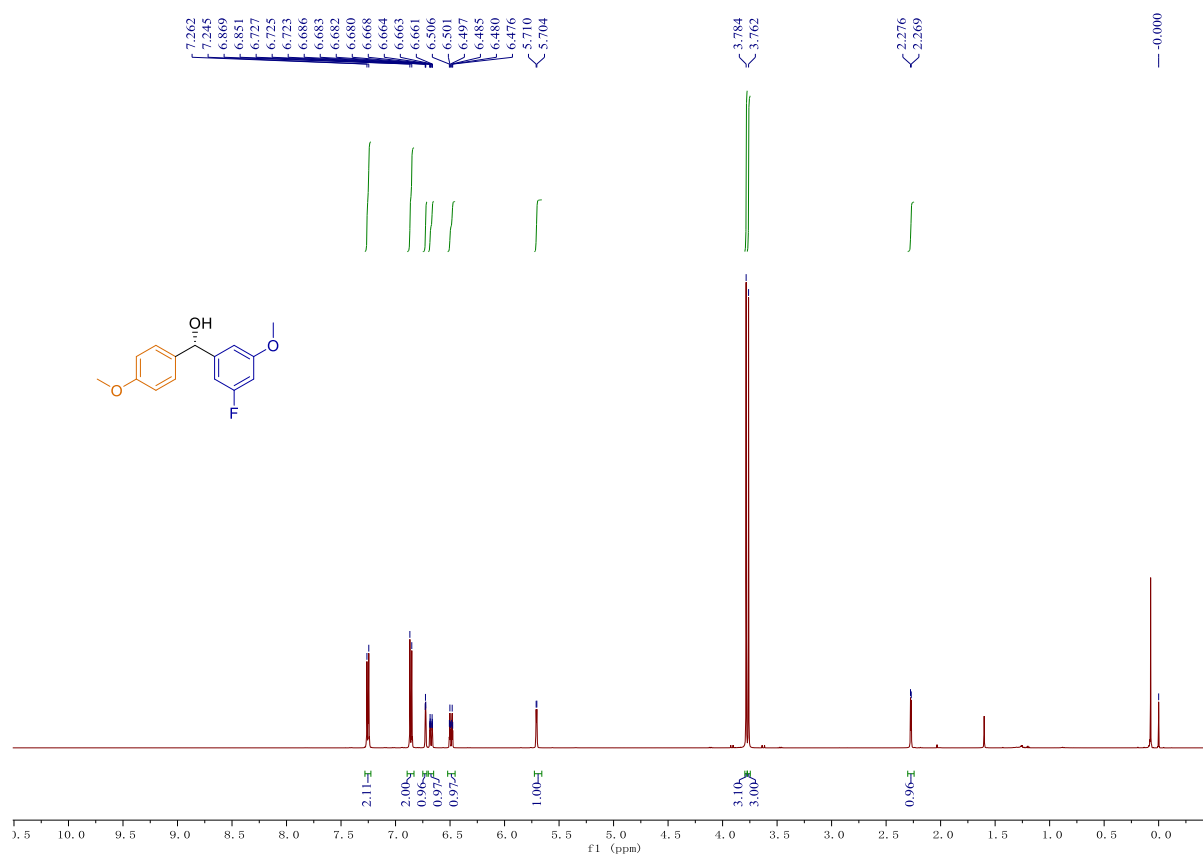
Supplementary Fig. 169 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-(4-fluorophenyl)(p-tolyl)methanol (4t).



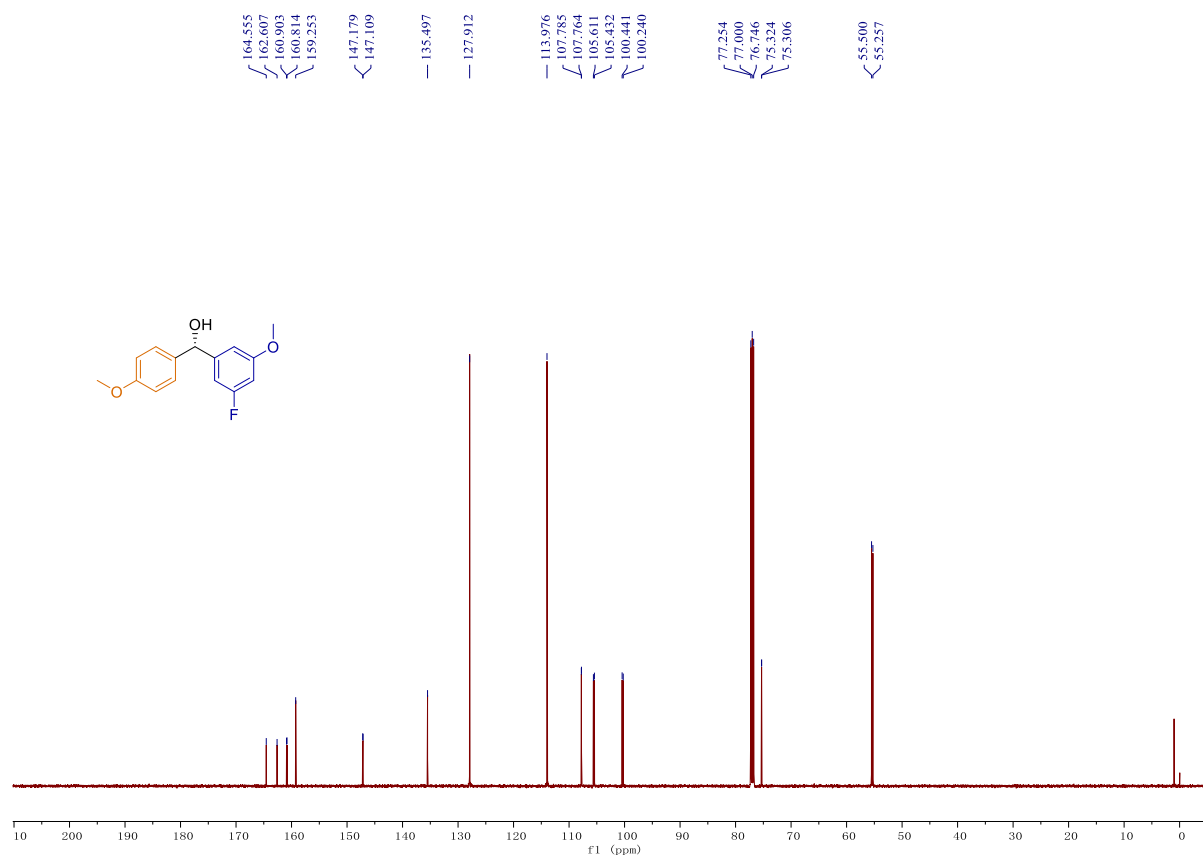
Supplementary Fig. 170 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-o-tolyl(2-(trifluoromethyl)phenyl)methanol (4u).



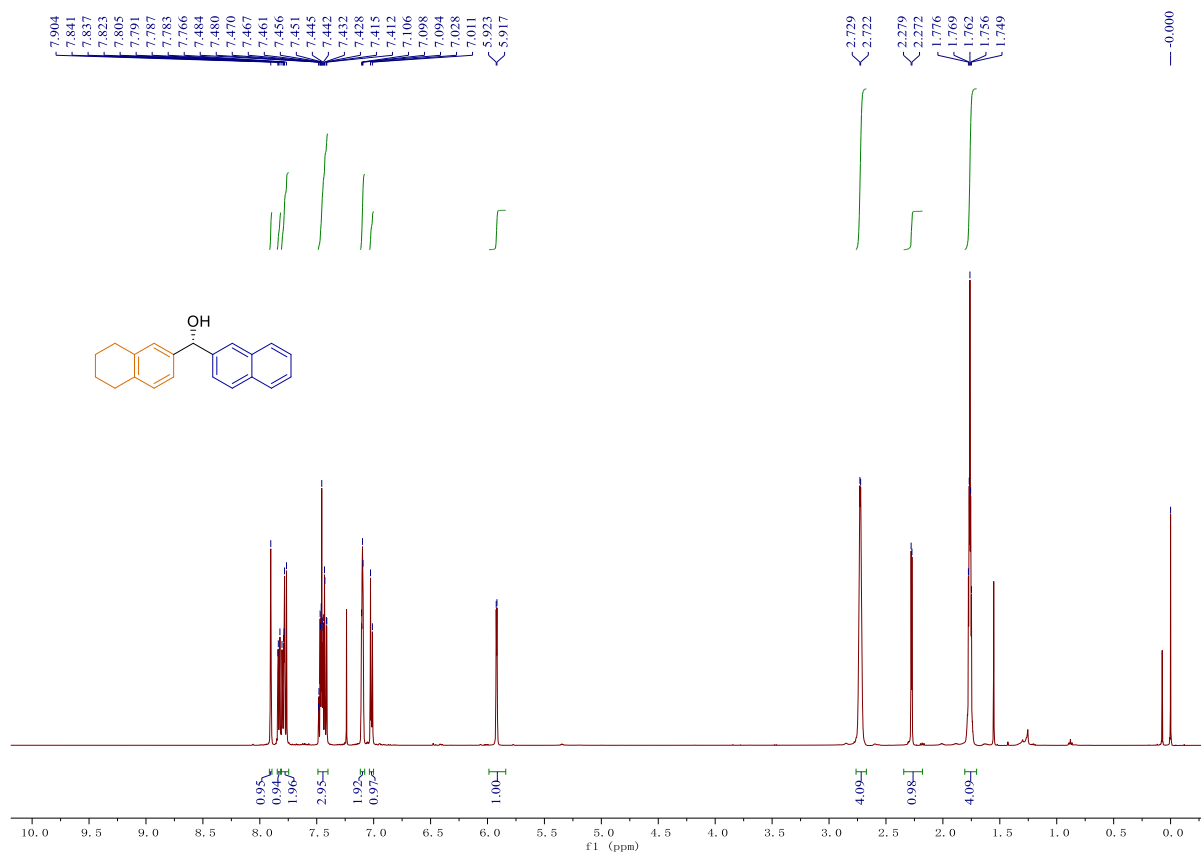
Supplementary Fig. 171 ¹³C NMR (126 MHz, Chloroform-*d*) of (S)-o-tolyl(2-(trifluoromethyl)phenyl)methanol (4u).



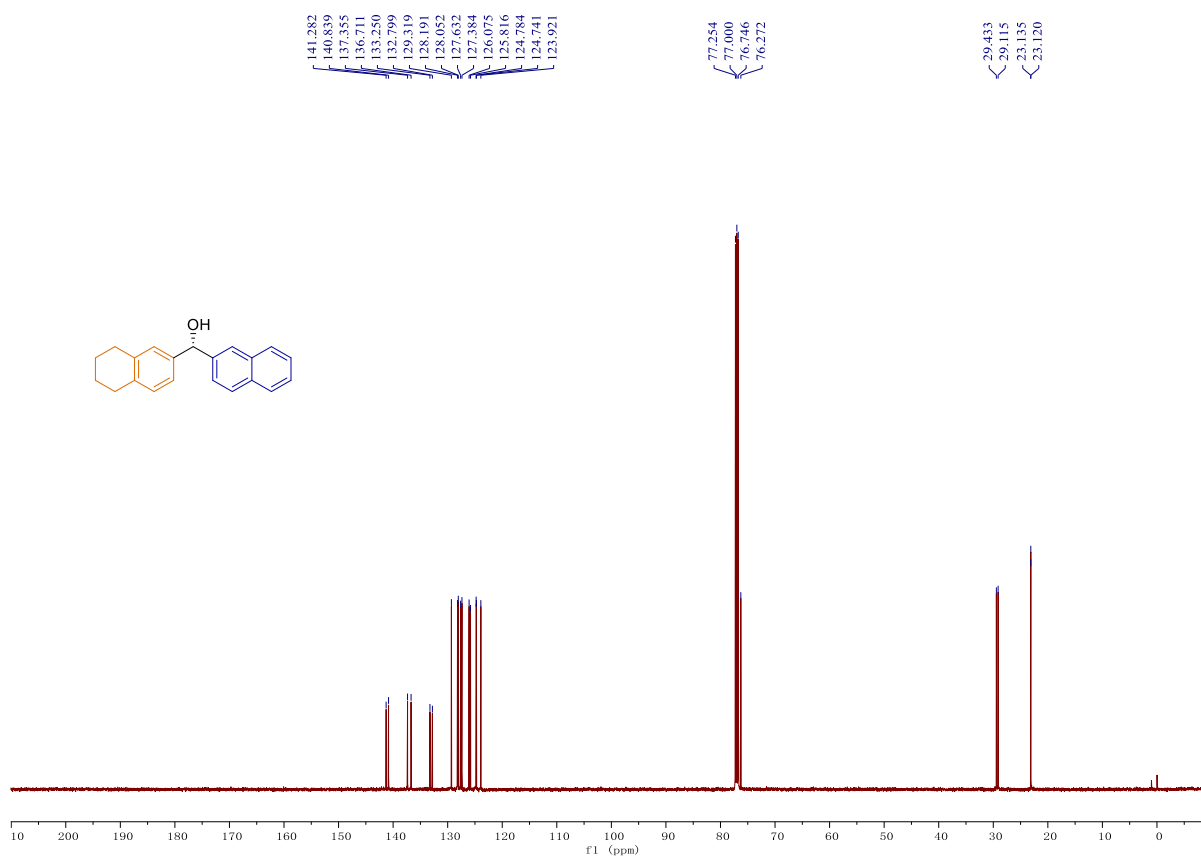
Supplementary Fig. 172 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-(3-fluoro-5-methoxyphenyl)(4-methoxyphenyl)methanol (4v).



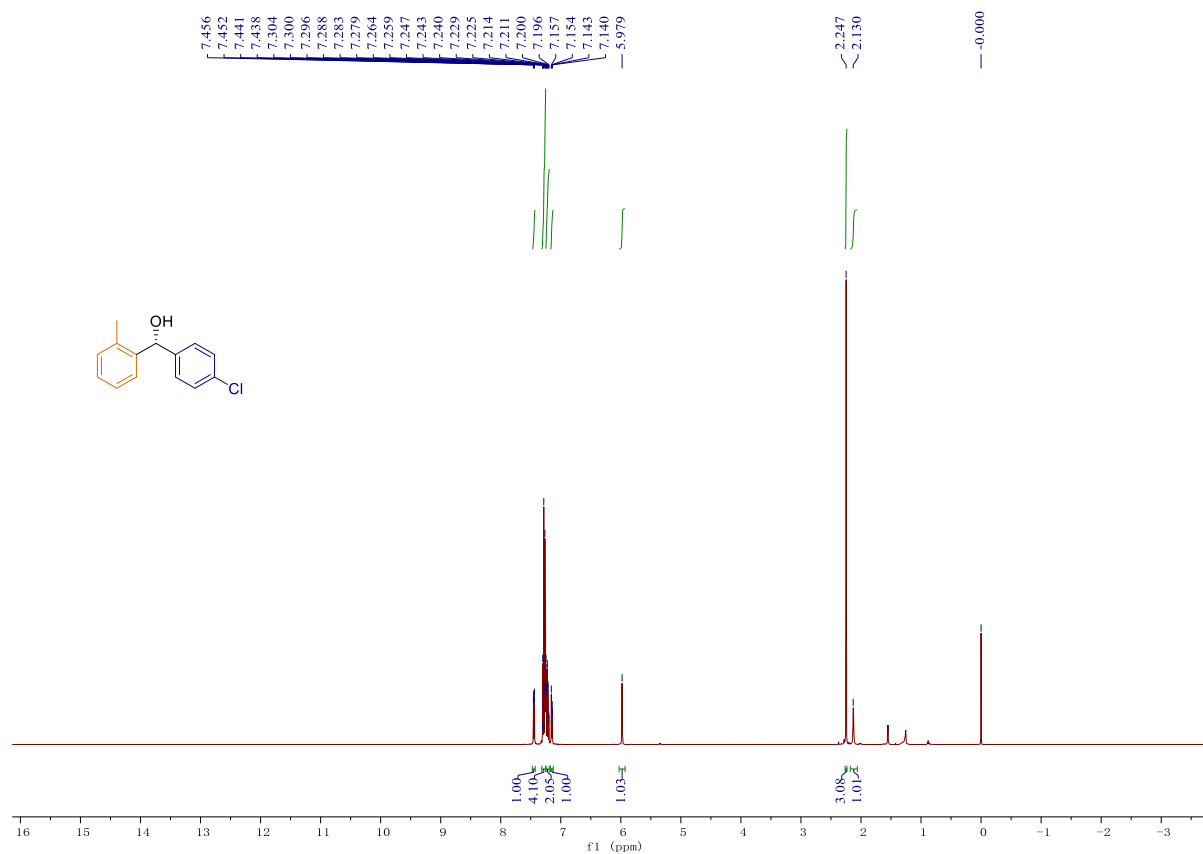
Supplementary Fig. 173 ¹³C NMR (126 MHz, Chloroform-*d*) of (S)-(3-fluoro-5-methoxyphenyl)(4-methoxyphenyl)methanol (4v).



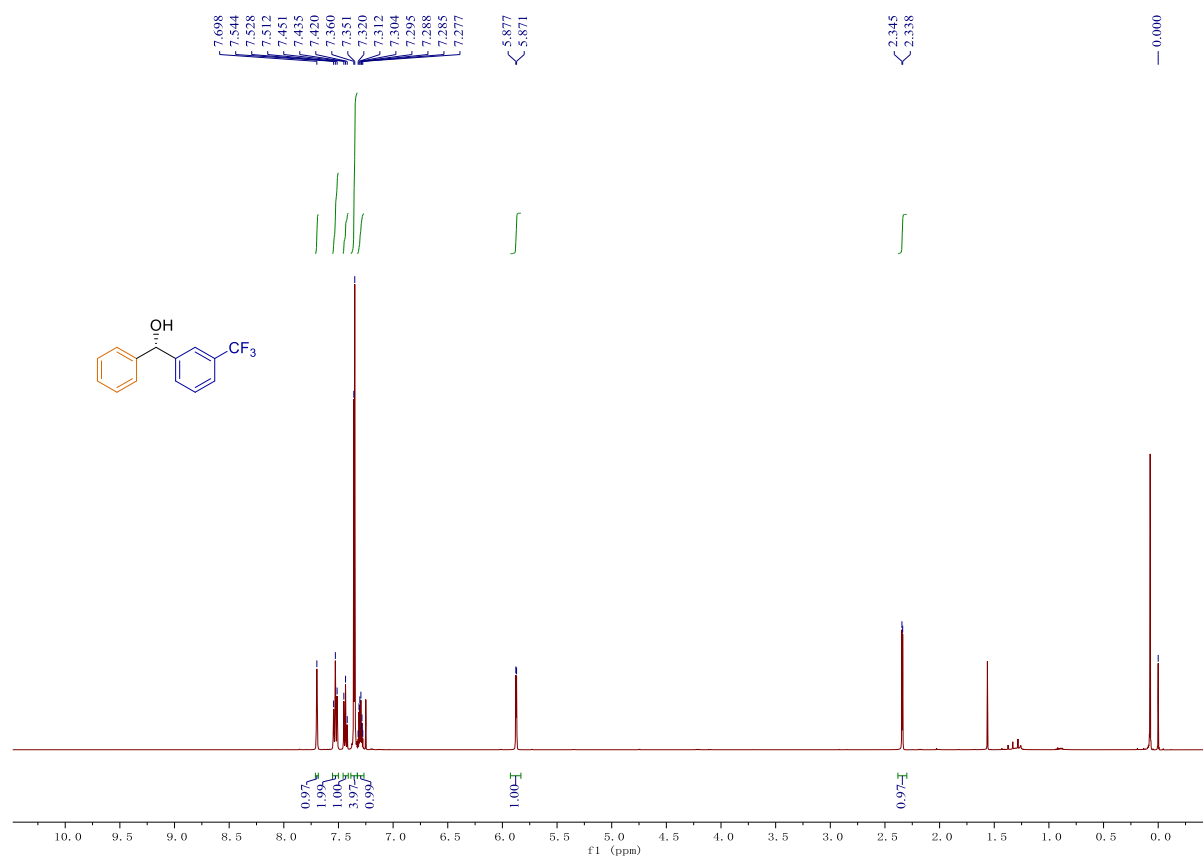
Supplementary Fig. 174 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-naphthalen-2-yl(5,6,7,8-tetrahydronaphthalen-2-yl)methanol (4w).



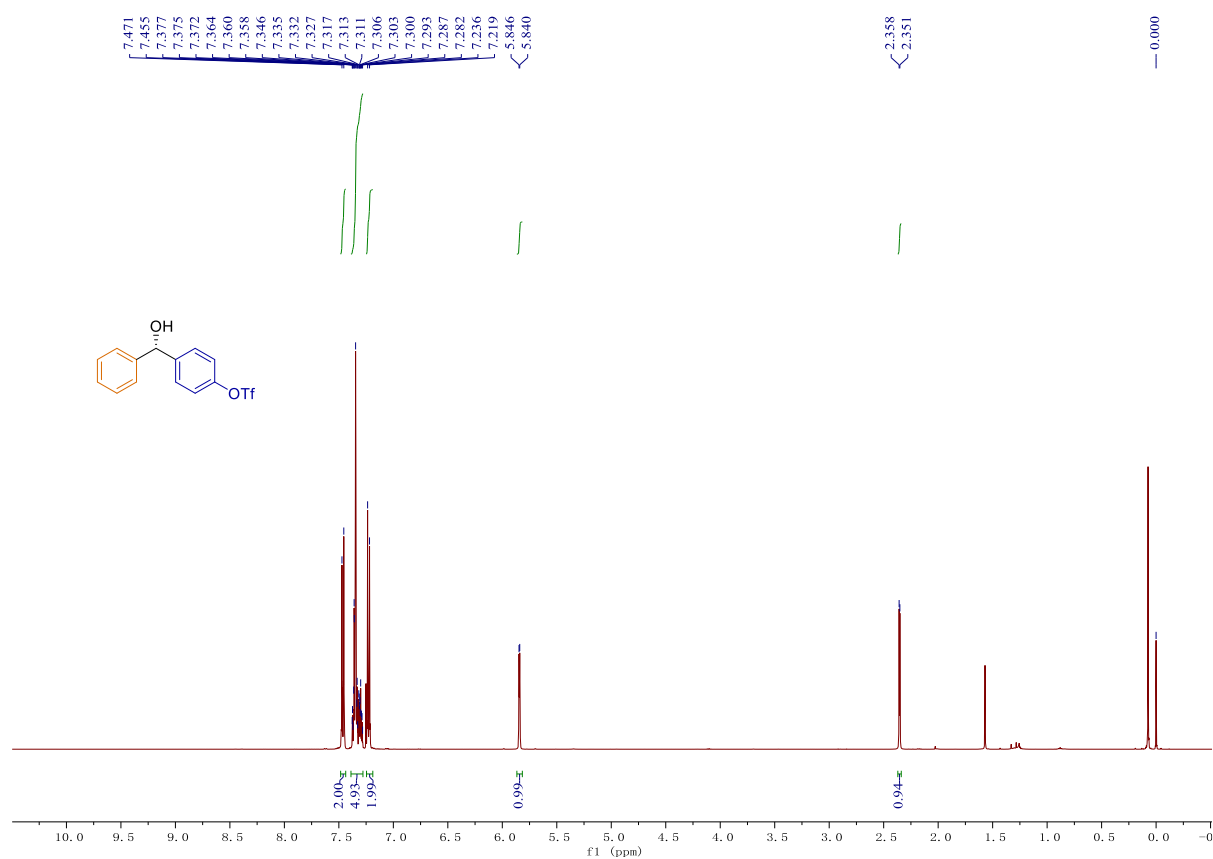
Supplementary Fig. 175 ¹³C NMR (126 MHz, Chloroform-*d*) of (*S*)-naphthalen-2-yl(5,6,7,8-tetrahydronaphthalen-2-yl)methanol (4w).



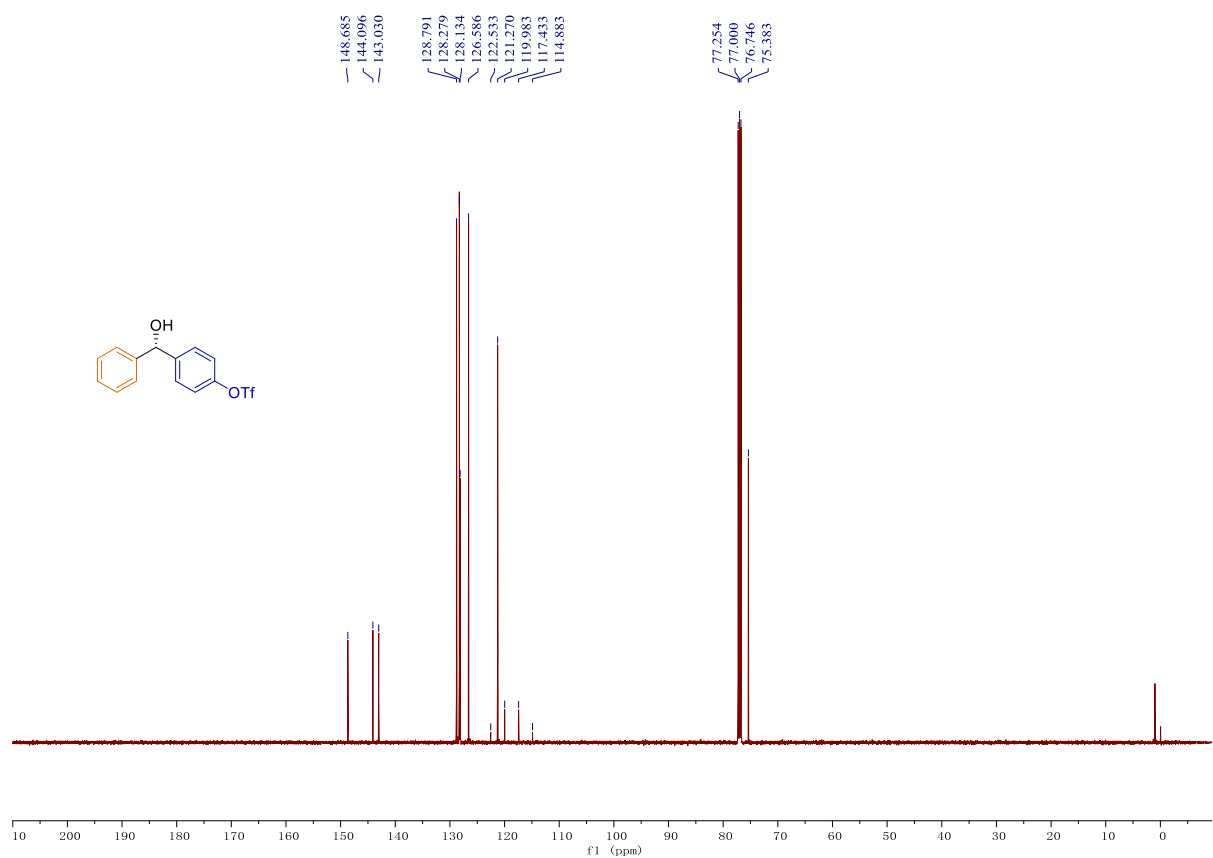
Supplementary Fig. 176 ¹H NMR (500 MHz, Chloroform-*d*) of (*R*)-(4-chlorophenyl)(*o*-tolyl)methanol (4x).



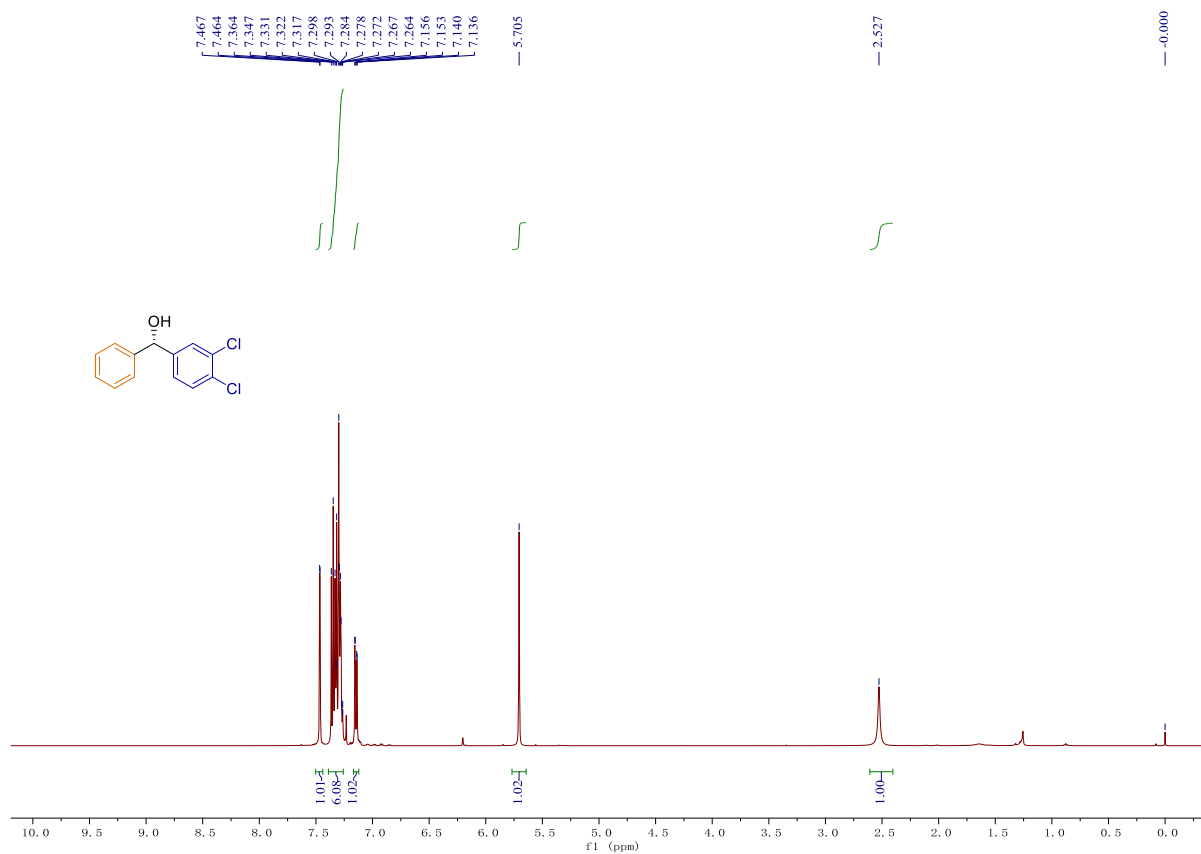
Supplementary Fig. 177 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-phenyl(3-(trifluoromethyl)phenyl)methanol (4y).



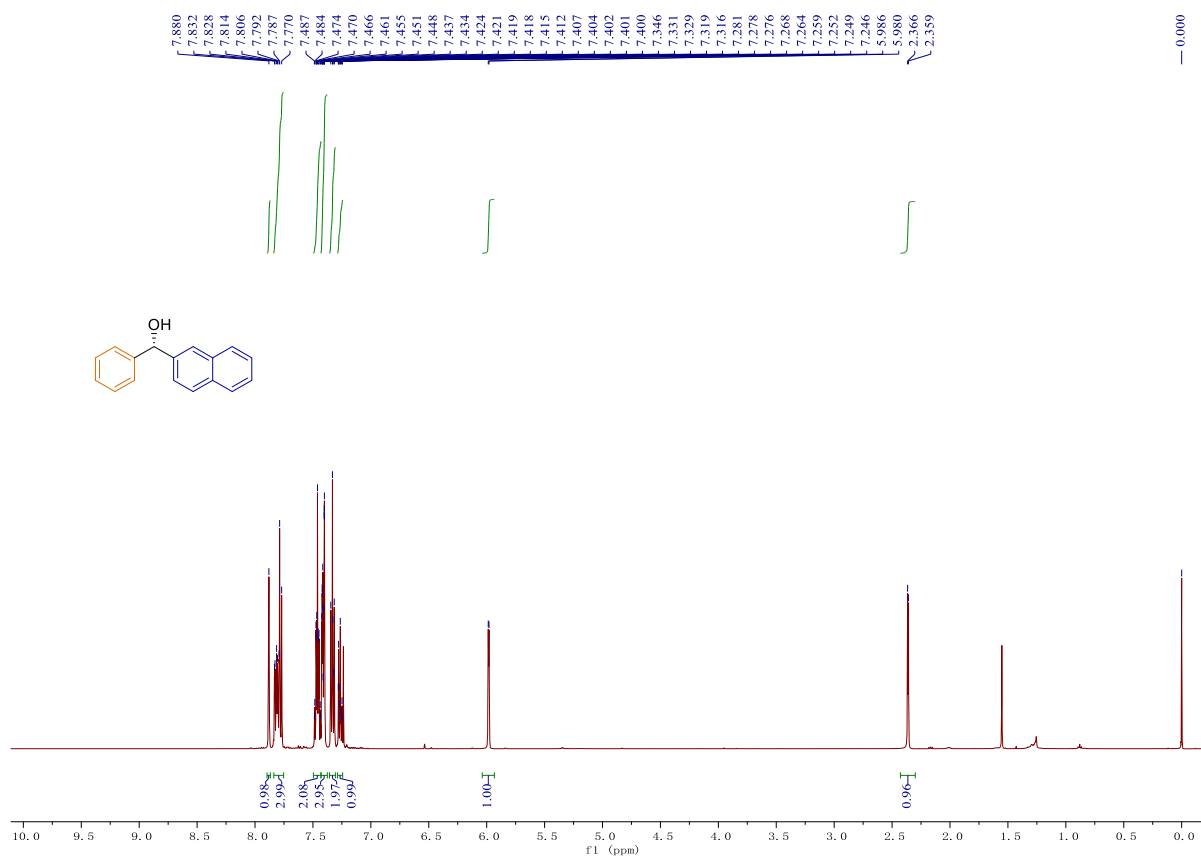
Supplementary Fig. 178 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-4-(hydroxy(phenyl)methyl)phenyl trifluoromethanesulfonate (1z).



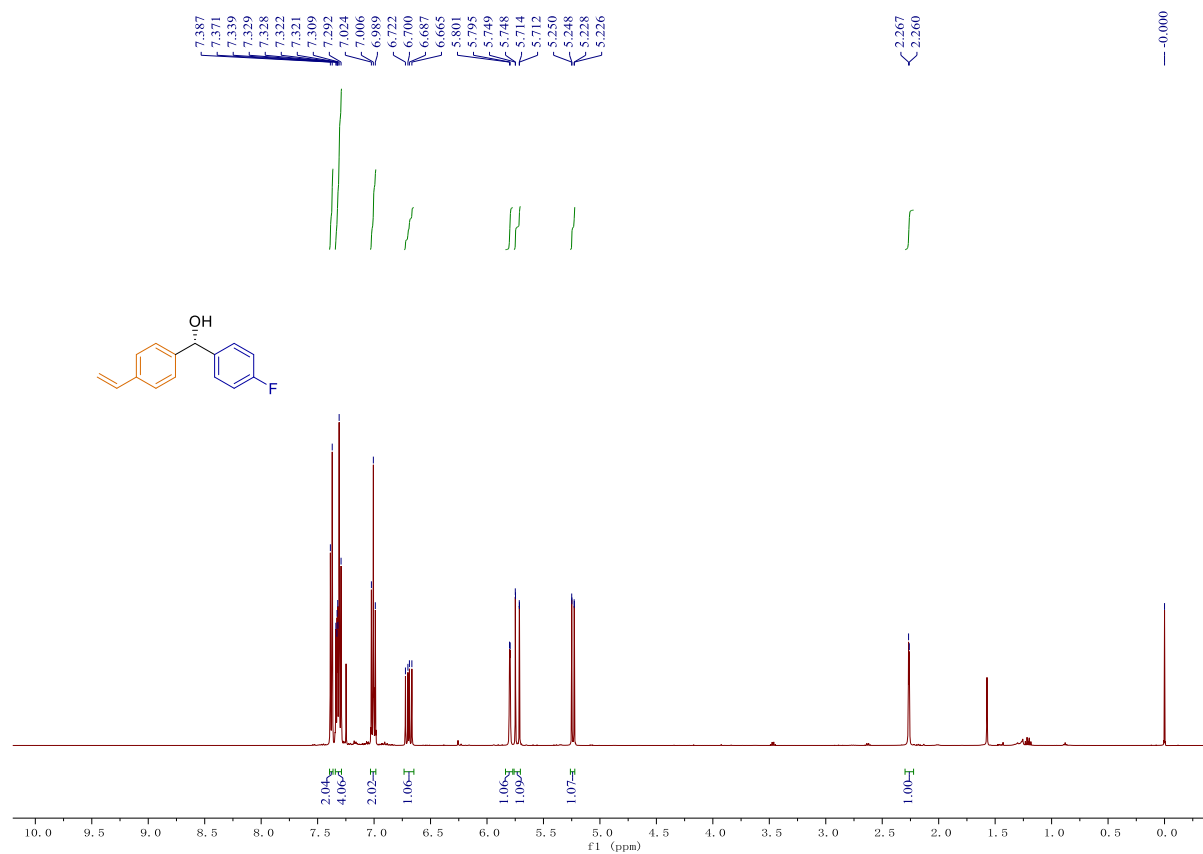
Supplementary Fig. 179 ¹³C NMR (126 MHz, Chloroform-*d*) of (S)-4-(hydroxy(phenyl)methyl)phenyl trifluoromethanesulfonate (1z).



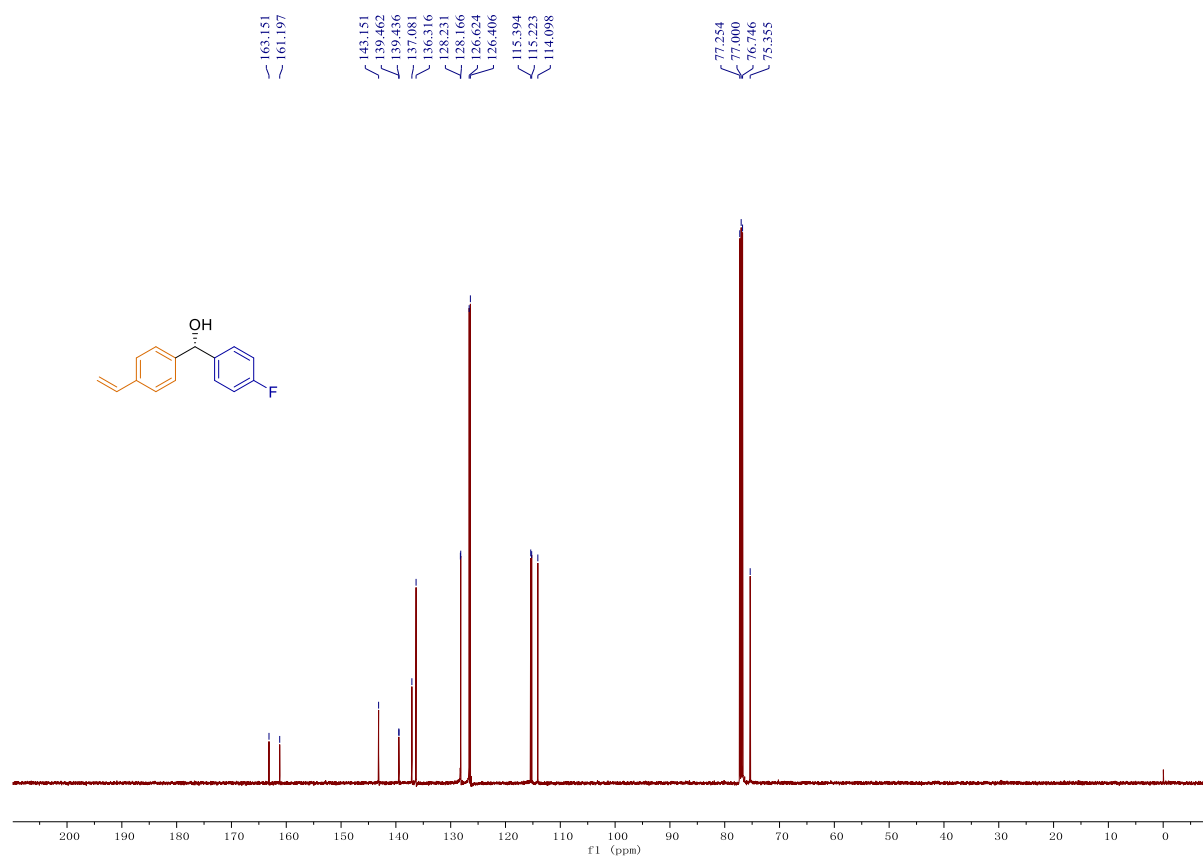
Supplementary Fig. 180 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-(3,4-dichlorophenyl)(phenyl)methanol (4aa).



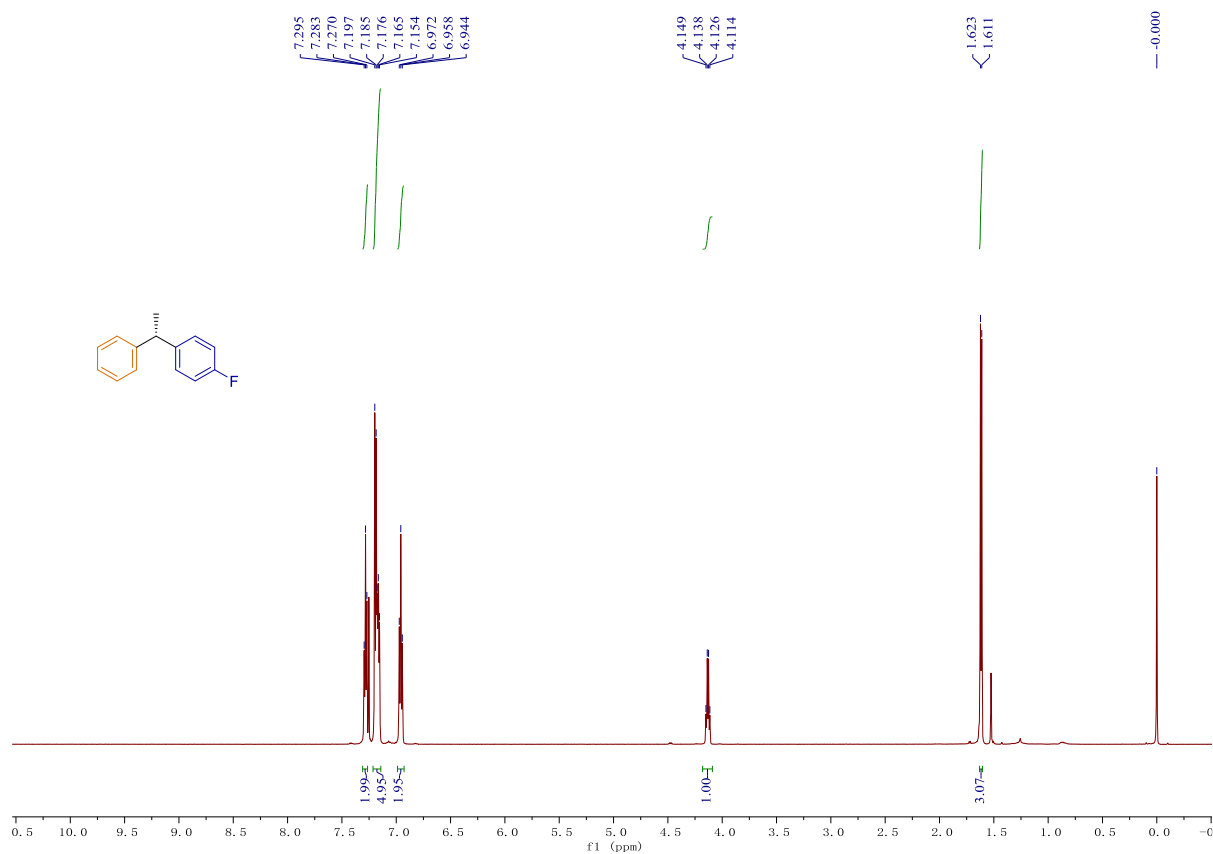
Supplementary Fig. 181 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-naphthalen-2-yl(phenyl)methanol (4ab).



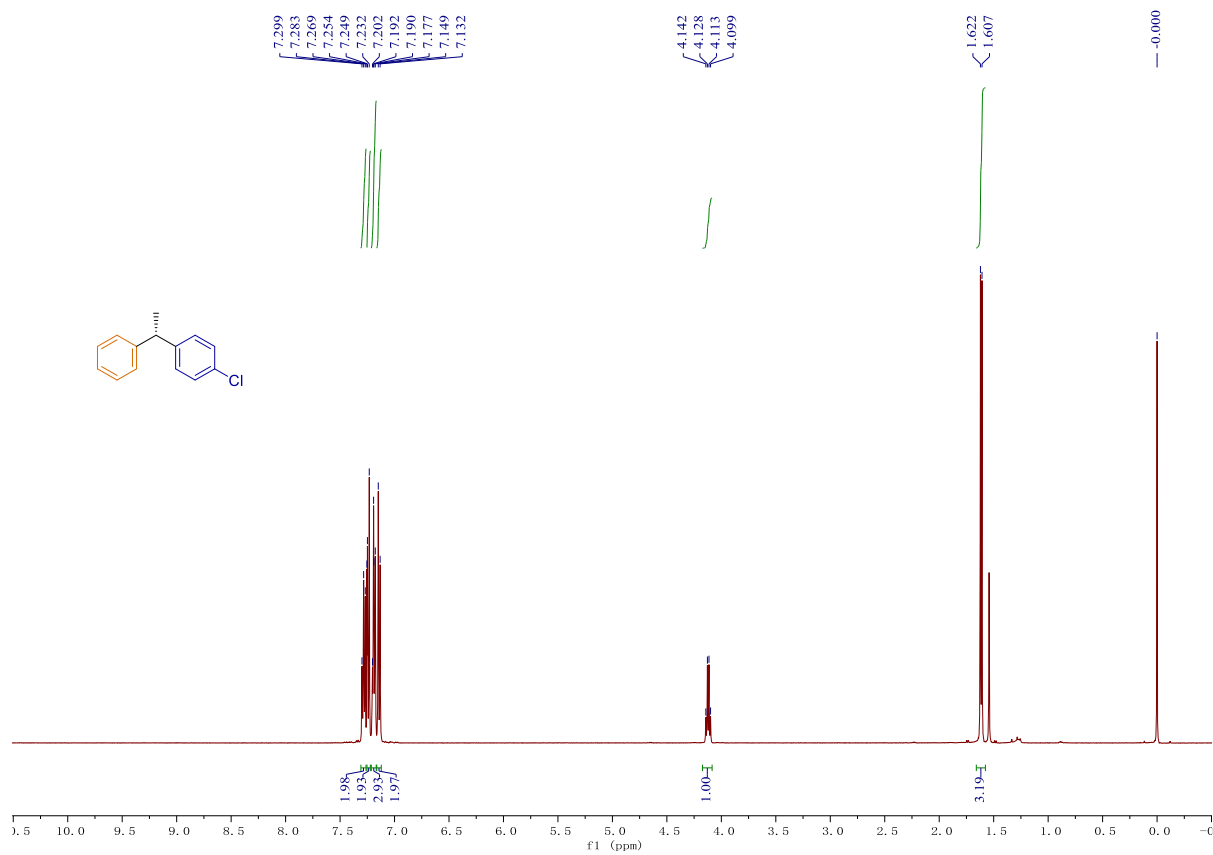
Supplementary Fig. 182 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-(4-fluorophenyl)(4-vinylphenyl)methanol (4ac).



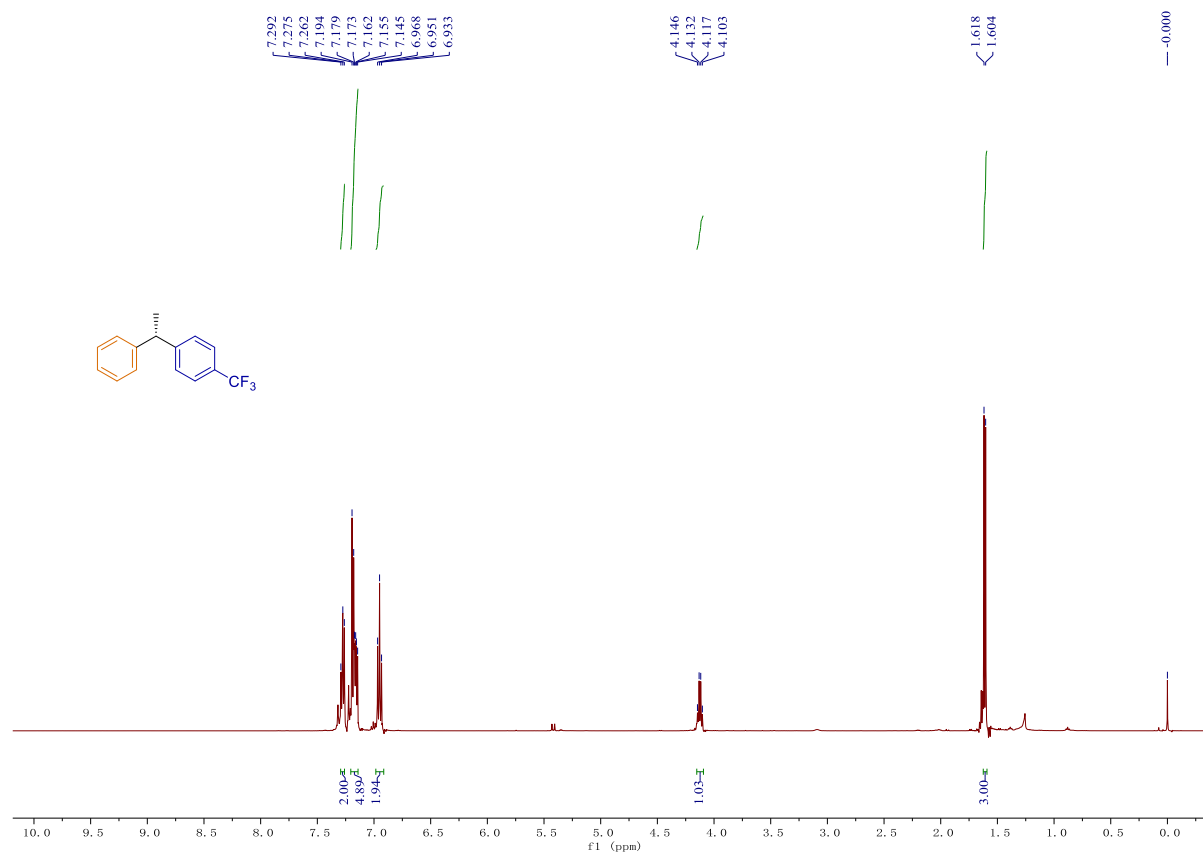
Supplementary Fig. 183 ¹³C NMR (126 MHz, Chloroform-*d*) of (S)-(4-fluorophenyl)(4-vinylphenyl)methanol (4ac).



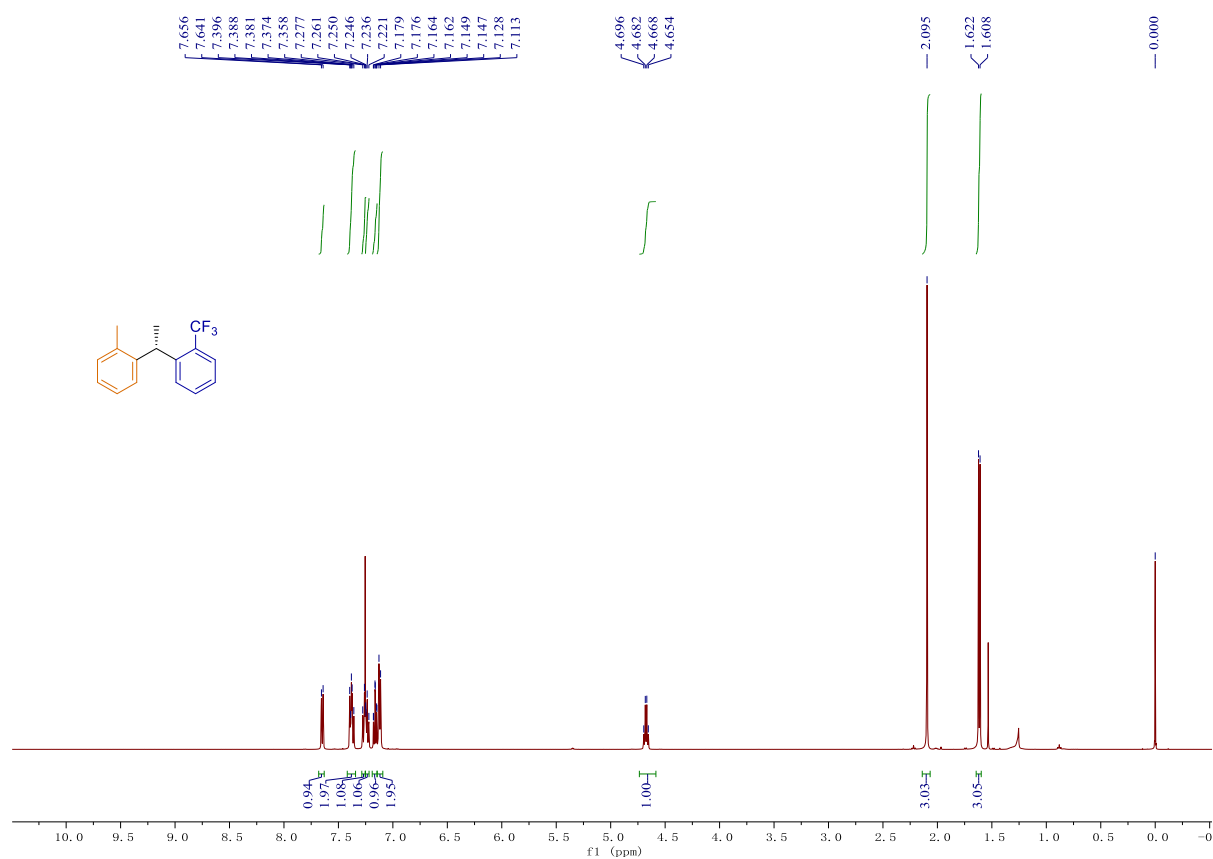
Supplementary Fig. 184 ^1H NMR (500 MHz, Chloroform- d) of (*S*)-1-fluoro-4-(1-phenylethyl)benzene (5a).



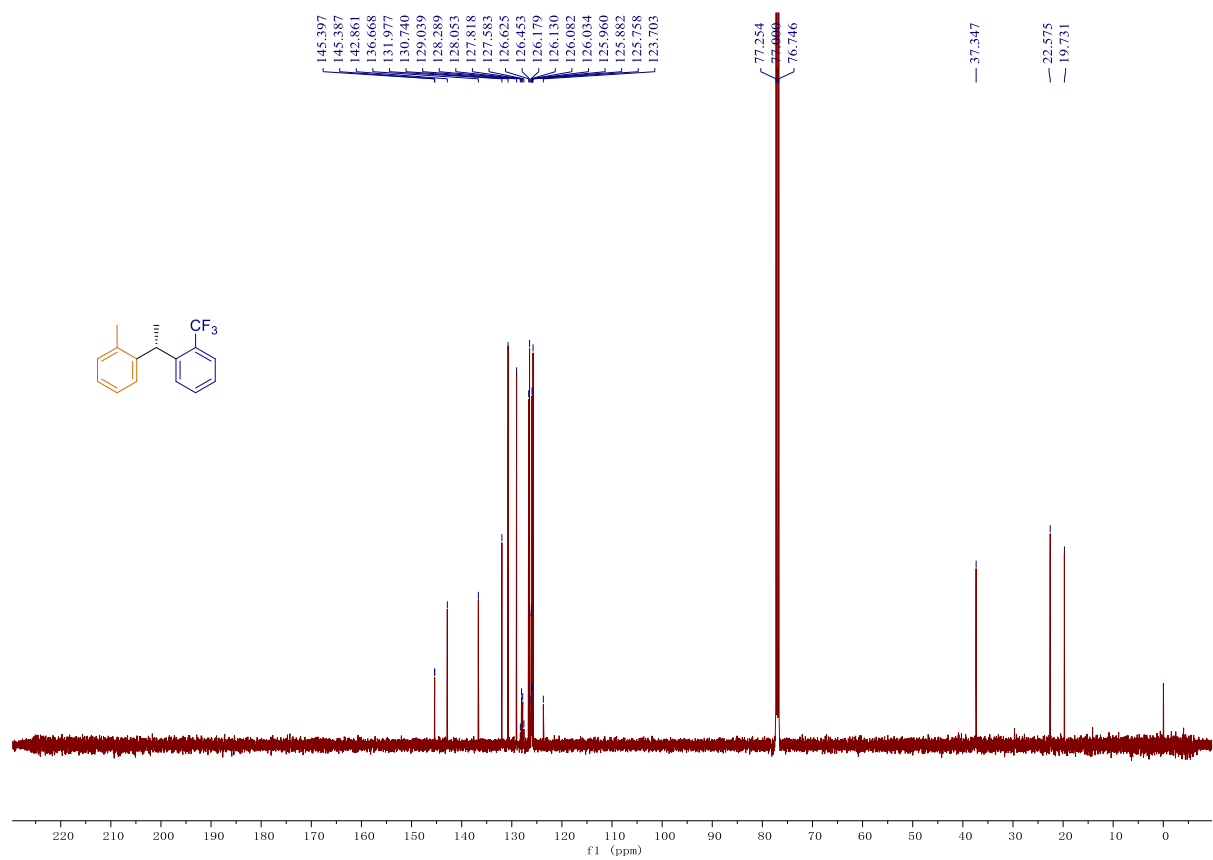
Supplementary Fig. 185 ^1H NMR (500 MHz, Chloroform- d) of (*S*)-1-chloro-4-(1-phenylethyl)benzene (5b).



Supplementary Fig. 186 ¹H NMR (500 MHz, Chloroform-*d*) of (*S*)-1-(1-phenylethyl)-4-(trifluoromethyl)benzene (5c).



Supplementary Fig. 187 ¹H NMR (500 MHz, Chloroform-*d*) of (S)-1-methyl-2-(1-(2-(trifluoromethyl)phenyl)ethyl)benzene (5d).



Supplementary Fig. 188 ¹³C NMR (126 MHz, Chloroform-*d*) of (S)-1-methyl-2-(1-(2-(trifluoromethyl)phenyl)ethyl)benzene (5d).

Supplementary References

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