# Effects of Changes of Intracellular pH on Contraction in Sheep Cardiac Purkinje Fibers

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ABSTRACT Intracellular pH (pHi) was measured with a pH-sensitive microelectrode in voltage-clamped sheep cardiac Purkinje fibers while tension was simultaneously measured. All solutions were nominally CO<sub>2</sub>/HCO<sub>3</sub> free and were buffered with Tris. The addition of NH<sub>4</sub>Cl (5–20 mM) produced an initial intracellular alkalosis that was associated with an increase of twitch tension. At the same time, a component of voltage-dependent tonic tension developed. Prolonged exposure (>5 min) to NH<sub>4</sub>Cl resulted in a slow recovery of pH<sub>i</sub> accompanied by a decrease of tension. Removal of NH4Cl produced a transient acidosis that was accompanied by a fall of force. In some experiments, there was then a transient recovery of force. If extracellular pH (pH<sub>o</sub>) was decreased, then pH<sub>i</sub> decreased slowly. Tension also fell slowly. An increase of pH<sub>o</sub> produced a corresponding increase of both force and pH<sub>i</sub>. The application of strophanthidin (10  $\mu$ M) increased force and produced an intracellular acidosis. The addition of NH<sub>4</sub>Cl, to remove this acidosis partially, produced a significant increase of force. The above results show that contraction is sensitive to changes of intracellular but not extracellular pH. This pH dependence will therefore modify the contractile response to inotropic maneuvers that also affect pH<sub>i</sub>.

#### INTRODUCTION

It has been known for many years that changes of extracellular pH (pH<sub>o</sub>) affect the contraction of cardiac muscle: acidification decreases and alkalinization increases the force of contraction (cf. Gaskell, 1880). More recent work has attempted to determine whether these effects are a direct consequence of changes in pH<sub>o</sub> (Vaughan Williams and Whyte, 1967) or whether they are secondary to changes in intracellular pH (pH<sub>i</sub>) (Fry and Poole-Wilson, 1981; Allen and Orchard, 1983; Poole-Wilson and Seabrooke, 1985). Evidence for the role played by pH<sub>i</sub> has come from the work of Fabiato and Fabiato (1978) on skinned cardiac fibers. These authors showed that changing the pH of the bathing solution modified the sensitivity of the contractile proteins to Ca<sup>2+</sup>: acidification reduced and alkalinization increased their Ca sensitivity. However, in intact fibers, ma-

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neuvers that alter pH<sub>i</sub> also modify the diastolic level of intracellular  $Ca^{2+}$  ( $[Ca^{2+}]_i$ ) (Bers and Ellis, 1982; Allen et al., 1984) and the magnitude of the  $Ca^{2+}$  transient that underlies the twitch (Allen and Orchard, 1983). Hence, in intact cells, a change in pH<sub>i</sub> is likely to affect contraction via more than one mechanism.

 $pH_i$  is changed by many experimental maneuvers and may have significant effects on contraction. For example, ischemia or hypoxia produce an intracellular acidification (e.g., De Hemptinne et al., 1982; Allen et al., 1985b), which has been suggested to account for the observed contractile failure (Katz and Hecht, 1969; Poole-Wilson, 1978). Furthermore, many inotropic maneuvers that increase [Ca<sup>2+</sup>]<sub>i</sub>, and thereby directly increase contraction, also produce an intracellular acidification (Deitmer and Ellis, 1980; Vaughan-Jones et al., 1983), which would be expected to decrease contraction indirectly.

The present experiments were designed to answer three questions. (a) Can the effects of  $pH_o$  on contraction in cardiac fibers be accounted for by the subsequent changes of  $pH_i$ ? (b) What is the quantitative relationship between  $pH_i$  and contraction in intact fibers? (c) To what extent do changes of  $pH_i$  modify the inotropic actions of other experimental maneuvers?

The experiments were carried out on Purkinje fibers isolated from sheep hearts.  $pH_i$  was measured directly with a recessed-tip, pH-sensitive glass microelectrode (Thomas, 1978) and tension was recorded simultaneously. Twitch and tonic tension in this tissue are voltage dependent (e.g., Eisner and Lederer, 1979). Consequently, it was necessary to control the membrane potential using a voltage-clamp apparatus so that depolarizing pulses of constant amplitude and duration could be used to elicit force development. Preliminary accounts of some of these results have appeared already (Eisner et al., 1983*a*; Vaughan-Jones et al., 1985).

#### METHODS

The experimental methods were similar to those described previously (Eisner and Lederer, 1979; Eisner et al., 1981, 1983b). Briefly, sheep hearts were obtained from a local abattoir. Purkinje fibers were isolated, shortened to 1-2 mm in length, and attached to a tension transducer in the experimental bath. Voltage-clamp control was maintained with a twomicroelectrode clamp. pH; was measured with a recessed-tip, pH-sensitive glass microelectrode (see Thomas, 1978; Vaughan-Jones, 1979). The pH electrodes were calibrated in the experimental bath before and after each experiment using a modified Tyrode's solution with the pH adjusted to 7.4 and 6.4 at 37°C. Acceptable electrodes displayed a response of 54-60 mV per pH unit, with a 90% response occurring within 20 s of changing the pH of the bathing solution. Because of this rather slow response time, the relationship between tension and pH<sub>i</sub> could not be quantified during rapid changes in pH<sub>i</sub>, such as were produced immediately after the addition or removal of external NH<sub>4</sub>Cl (see, e.g., Fig. 1). Slower changes of pH<sub>i</sub> (over several minutes) can be measured accurately. Consequently, the quantitative effect of pH<sub>i</sub> upon tension was measured during the slow secondary changes in pH<sub>i</sub> that occur after the addition and removal of NH<sub>4</sub>Cl and during the slow changes in  $pH_i$  after a change in  $pH_o$ .

In several of the figures, the tension traces have been high-pass filtered (as indicated in the legends) to remove changes of baseline tension in order to resolve changes in twitch tension.

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Solutions

The modified Tyrode's solution consisted of (mM): 145 NaCl, 4 KCl, 5 CaCl<sub>2</sub>, 1 MgCl<sub>2</sub>, 10 glucose, 10 Tris-HCl. The pH was adjusted to 7.4 at 37°C. All solutions were bubbled with air.

#### RESULTS

## Effects on Tension of Changes of pH<sub>i</sub>

The first experiments were designed to investigate the effects on tension of changing pH<sub>i</sub> under conditions where the pH of the superfusing solution (pH<sub>o</sub>) was constant. This was achieved by exposing the preparation to NH4Cl. Fig. 1A shows the effects of an 18-min exposure to NH<sub>4</sub>Cl (20 mM). The application of NH<sub>4</sub>Cl initially produced an intracellular alkalinization because the highly permeant NH<sub>3</sub> crosses the membrane more quickly than does NH $_4^+$  (e.g., Boron and De Weer, 1976). Once inside the cell, NH<sub>3</sub> combines with H ions to form NH<sup>‡</sup>, thus raising pH<sub>i</sub>. However, during prolonged application of NH<sub>4</sub>Cl, as in Fig. 1A, the initial alkalosis is not maintained, and  $pH_i$  gradually recovers to a less alkaline level. This is caused by the slow entry of NH4 ions into the cell (see Roos and Boron, 1981, for a discussion of this process). The eventual removal of external NH<sub>4</sub>Cl leads to a large intracellular acidification and this occurs because NH4 ions that have accumulated intracellularly now convert to NH3, which escapes from the cell, leaving behind H ions. Finally, these H ions are extruded from the cell via Na-H exchange (Deitmer and Ellis, 1980), so that pH<sub>i</sub> slowly recovers from the acidification. The addition and removal of NH4Cl thus leads to large transient changes in  $pH_i$ . Fig. 1A shows that these changes in  $pH_i$  affect tension. The initial alkalosis caused by the addition of NH<sub>4</sub>Cl is associated with an increase in twitch tension. However, as pH<sub>i</sub> becomes less alkaline again on prolonged exposure to NH<sub>4</sub>Cl, twitch tension also declines. Similarly, the transient acidification on removal of NH<sub>4</sub>Cl is accompanied by a transient decrease of twitch tension. Specimen tension records are shown on an expanded time scale in Fig. 1B. Alkalinization increased not only the twitch component of tension but also revealed a component of voltage-dependent tonic tension that was maintained throughout the depolarizing voltage-clamp step. Conversely, intracellular acidification, as well as reducing twitch tension, also abolished this voltage-dependent component of tonic tension.

A similar effect of pH<sub>i</sub> upon tonic tension was also evident at the holding potential, which, in Fig. 1A, was -60 mV. It is clear that after the addition of NH<sub>4</sub>Cl, there was a transient increase in resting force, which, like that seen for the twitch, followed the transient rise in pH<sub>i</sub>. The experiment illustrated in Fig. 2 was designed to investigate further the effect of pH<sub>i</sub> on the voltage-dependent component of tonic tension. In order to increase this component of tension, the preparation was exposed to strophanthidin (10  $\mu$ M; see Eisner et al., 1983). The specimen record (Fig. 2B, trace a) shows that under these conditions, depolarization produced a tonic component of tension as well as an aftercontraction. Next, NH<sub>4</sub>Cl was added, first at a concentration of 10 mM and then at 20 mM. This produced a graded intracellular alkalinization (Fig. 2A) accompanied by an



FIGURE 1. The effects of prolonged exposure to NH<sub>4</sub>Cl on pH<sub>i</sub> and tension. (A) Time course of effects. Traces show (from top to bottom): pH<sub>i</sub>, tension (band-pass filtered: 0.1-10 Hz), tension (DC to 10 Hz), and membrane potential. The tension trace recorded from DC to 10 Hz illustrates the slow changes in resting tension that occur during the application of NH<sub>4</sub>Cl. However, after removal of NH<sub>4</sub>Cl, disturbances in the solution flow produced artifactual disturbances of resting tension, which are evident as spontaneous, downward baseline displacements. The membrane potential was held at -60 mV and 2-s pulses were applied to -30 mV. NH<sub>4</sub>Cl (10 mM) was applied for the period shown above. (B) Specimen records displayed on a faster time base, obtained at the points indicated by the letters in A. Note the appearance of voltage-dependent tonic tension when pH<sub>i</sub> is alkaline.

increase of tonic tension, twitch tension, and the aftercontraction (Fig. 2B, traces b and c). Finally, the removal of NH<sub>4</sub>Cl produced an intracellular acidosis and a corresponding decrease of all components of tension (Fig. 2B, trace d). Therefore, an intracellular alkalosis increased and an intracellular acidosis decreased twitch tension, resting tonic tension, voltage-dependent tonic tension, and, when present, the aftercontraction.



FIGURE 2. The effects of NH<sub>4</sub>Cl in the presence of strophanthidin. (A) Time course. The traces show (top to bottom) membrane potential (V), pH<sub>i</sub>, and tension. [NH<sub>4</sub>Cl] was varied as shown above. The membrane potential was held at -70 mV and 2-s pulses were applied to -30 mV at 0.1 Hz. Strophanthidin ( $10 \mu$ M) had been present for 20 min before the record began. (B) Specimen records obtained at the points shown in A.

The data from seven experiments with NH<sub>4</sub>Cl are shown in Fig. 3, which plots twitch tension as a function of pH<sub>i</sub> on linear coordinates and semilogarithmic coordinates (inset). It is apparent that changes in pH<sub>i</sub> of  $\pm 0.3$  units produced graded changes in tension. Furthermore, in this range, the plot of logarithmic force as a function of pH<sub>i</sub> is roughly linear, with a slope of ~2. However, at values of pH<sub>i</sub> lower than 6.8, force sometimes increased again. Such an increase of the twitch at low pH<sub>i</sub> is illustrated in Fig. 4. In this experiment, the fiber had been exposed to 20 mM NH<sub>4</sub>Cl for 23 min. After the removal of NH<sub>4</sub>Cl, pH<sub>i</sub> declined rapidly by 0.5 units to 6.7 and then recovered more slowly. The large and rapid fall in pH<sub>i</sub> was accompanied initially by a decrease in twitch tension.



FIGURE 3. The relationship between tension and  $pH_i$ . (A) Linear scale. The ordinate shows twitch tension normalized to the level in control solution (mean  $pH_i$  = 7.24 ± 0.04 SEM, n = 7). The abscissa is the change of  $pH_i$  from the control level during the application of NH<sub>4</sub>Cl (positive values) or after removal of NH<sub>4</sub>Cl (negative, acidic values). No attempt was made to measure the relationship during the rapid development of the alkalosis (because of the slow response time of the pH electrode; see Methods). The points at alkaline pH were obtained during the recovery of pH<sub>i</sub> from its most alkaline level. The points at acidic pH were obtained during the recovery of pH<sub>i</sub> from its most acid level after the removal of NH<sub>4</sub>Cl. The different symbols correspond to different preparations, except for the open squares, which were produced by a brief exposure to NH<sub>4</sub>Cl. Each point represents a different preparation.

However, twitch tension then increased transiently, such that, at a  $pH_i$  of 6.68, it was comparable to that observed at the start of the record, when  $pH_i$  was 7.25. Such transient increases in force after the removal of NH<sub>4</sub>Cl were observed in two out of seven fibers, but only after prolonged periods of exposure to NH<sub>4</sub>Cl, when the removal of NH<sub>4</sub>Cl resulted in a large acidosis. This paradoxical increase in force at low  $pH_i$  will be considered in the Discussion.

#### Effects on Tension of Changes in pH<sub>o</sub>

In these experiments,  $pH_i$  was changed by altering the pH of the superfusing solution. Fig. 5A shows that an external acidification (from pH 7.4 to 5.9) produced a slow reversible fall in  $pH_i$  and twitch tension. Opposite effects on

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pH<sub>i</sub> and tension were observed after a rise in pH<sub>o</sub>, as shown in Fig. 5*B*. As in the experiments with NH<sub>4</sub>Cl, the time courses of the change in twitch tension paralleled the changes in pH<sub>i</sub>. The quantitative relationship between tension and pH<sub>i</sub> observed in Fig. 5 is plotted in Fig. 6. Thus, regardless of whether pH<sub>i</sub> is altered by varying pH<sub>o</sub> or by exposure to NH<sub>4</sub>Cl, changes in twitch tension correlate well with changes in pH<sub>i</sub>. It is therefore appropriate to ask whether tension can be expressed as a unique function of pH<sub>i</sub>. To do this, we have compared the effects of changing pH<sub>o</sub> with those of adding NH<sub>4</sub>Cl. Fig. 7*A* shows that an increase of pH<sub>o</sub> from 7.4 to 8.4 produced about the same change of pH<sub>i</sub> as did the application of 5 mM NH<sub>4</sub>Cl. It is apparent that the relationship between tension and pH<sub>i</sub> was similar when pH<sub>i</sub> was recovering from these two



FIGURE 4. Dissociation between changes of pH<sub>i</sub> and tension. The traces show (top to bottom) pH<sub>i</sub>, tension (band-pass filtered at 0.1-10 Hz), membrane potential. The membrane potential was held at -60 mV and depolarized to -30 mV at 0.1 Hz. NH<sub>4</sub>Cl (20 mM) had been present for 23 min before the record began and was removed as indicated above the record.

types of alkalosis (Fig. 7*B*, filled circles and open triangles). However, this relationship differs quantitatively from that seen between tension and  $pH_i$  during the onset of the effects of pH 8.4 (Fig. 7*B*, open circles). Such hysteresis between force and  $pH_i$  was observed consistently after reversible alterations in  $pH_o$  (six experiments on four fibers). For example, inspection of the data shown in Fig. 6 also reveals a hysteresis, although in this case it is less pronounced than that seen in Fig. 7. It should also be noted that hysteresis did not occur consistently in a given direction; e.g., that shown in Fig. 6 is in the opposite direction to that shown in Fig. 7*B*. The conclusion we draw from these experiments, therefore, is that after a change of  $pH_i$ , although there is usually a good qualitative correlation between  $pH_i$  and tension, the quantitative relationship depends on the direction in which  $pH_i$  was changed. Unfortunately, it was not possible to



FIGURE 5. The effects on pH<sub>i</sub> and tension of changing pH<sub>o</sub>. In both panels, the traces show (top to bottom) pH<sub>i</sub>, tension (band-pass filtered at 0.1-10 Hz), and membrane potential. (A) The effects of external acidification. pH<sub>o</sub> was changed as indicated above the figure. The membrane potential was held at -70 mV and 2-s depolarizing pulses were applied to -35 mV at 0.1 Hz. (B) The effects of external alkalinization. Both A and B are taken from the same experiment; B was recorded 15 min after A.

discover whether a similar hysteresis between  $pH_i$  and tension occurred with  $NH_4Cl$ . This is because the rather slow response time of the pH electrode precluded accurate measurements during fast changes in  $pH_i$  (see Methods).

## Changes of pH<sub>i</sub> Modify Other Inotropic Maneuvers

In this section, we consider briefly two common experimental maneuvers that result in large changes of tension: (a) the addition of cardioactive steroids and (b) variations in  $[Ca^{2+}]_o$ . Both maneuvers exert their inotropic effects via changes in  $[Ca^{2+}]_i$ , but both maneuvers also produce changes in pH<sub>i</sub> that may modify the contractile response.



FIGURE 6. The quantitative effect on tension of a change in pH<sub>i</sub> produced by altering pH<sub>o</sub>. Data were taken from the experiment shown in Fig. 5. The relationship between tension and pH<sub>i</sub> has been plotted after external acidosis (pH<sub>o</sub> 7.4–5.9, filled squares), recovery from external acidosis (pH<sub>o</sub> 5.9–7.4, open squares), external alkalosis (pH<sub>o</sub> 7.4–8.9, open circles), and recovery from external alkalosis (pH<sub>o</sub> 8.9–7.4, filled circles). The line was drawn by eye.

In the experiment of Fig. 8A, strophanthidin (10  $\mu$ M) was added to the superfusing solution. It has been shown previously that this increases intracellular Na activity ( $a_{Na}^{i}$ ), intracellular Ca activity ( $a_{Ca}^{i}$ ), and the various components of tension, including the appearance of an oscillatory aftercontraction and a transient inward current, both of which occur after repolarization (e.g., Eisner et al., 1983b). Fig. 8A also shows that pH<sub>i</sub> decreased by 0.1 unit (cf. Deitmer and Ellis, 1980; Vaughan-Jones et al., 1983). The results of the experiments of Figs. 1–7 suggest that a change in pH<sub>i</sub> of this magnitude will have significant effects on contraction. This point was tested directly by adding 10 mM NH<sub>4</sub>Cl to the superfusate after strophanthidin had been applied for 15 min. The application



FIGURE 7. A comparison of the effects of NH<sub>4</sub>Cl with those of changing pH<sub>0</sub>. (A) Original record. The traces show (top to bottom) pH<sub>i</sub>, tension (band-pass filtered at 0.1-10 Hz), and membrane potential. The membrane potential was held at -60 mV and 2-s depolarizing pulses were applied to -30 mV at 0.1 Hz. The control solution had a pH<sub>0</sub> of 7.4 and, as indicated above the record, either pH<sub>0</sub> was changed to 8.4 or 5 mM NH<sub>4</sub>Cl was added (at constant pH<sub>0</sub>). (B) Relationship between pH<sub>i</sub> and twitch tension. The graph shows twitch tension as a function of pH<sub>i</sub> during either the development of the effects of external alkalinization (open circles), the recovery from the effects of extracellular alkalinization (filled circles), and during the recovery of pH during the exposure to NH<sub>4</sub>Cl (triangles). The symbols correspond to those shown in A.

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FIGURE 8. The effects of changes of pH<sub>i</sub> on the inotropic response to Na-K pump inhibition. (A) Time course. The traces show (top to bottom) membrane potential, pH<sub>i</sub>, and tension (band-pass filtered at 0.1–10 Hz): The membrane potential was held at -60 mV and 2-s pulses were applied at 0.1 Hz. Strophanthidin (10  $\mu$ M) and NH<sub>4</sub>Cl (5 mM) were applied for the periods indicated above the record. (B) Specimen records of membrane potential and tension (DC to 10 Hz) obtained at the points indicated in A.

of NH<sub>4</sub>Cl transiently returned pH<sub>i</sub> to a less acid level and simultaneously produced a large increase of tension. The effects on tension are shown on an expanded time scale in Fig. 8*B*. Alkalinization increased all components of tension (the increases in Fig. 8*B* are 50% for twitch, 120% for voltage-dependent tonic tension, and 21% for the aftercontraction). These results therefore make it clear that the acidification produced by strophanthidin will have greatly attenuated the positive inotropic effects.

The effects on pH<sub>i</sub> and tension of elevating [Ca<sup>2+</sup>], have been examined previously (Vaughan-Jones et al., 1983) but it is necessary to make some additional observations here. Under control conditions, raising the [Ca<sup>2+</sup>]<sub>o</sub> from 0.5 to 5.0 mM produces a rise in twitch tension but has little effect on pH<sub>i</sub> (not shown). However, after prolonged inhibition of the Na-K pump, the same rise in [Ca<sup>2+</sup>]<sub>o</sub> leads to a large rise and then, over the next few minutes, a secondary decline in all the components of tension (twitch tension, voltage-dependent tonic tension, and the aftercontraction; Eisner et al., 1983; Vaughan-Jones et al., 1983), and it is notable that this secondary decline in tension is associated with an intracellular acidosis of 0.1-0.2 units. We have observed these effects of [Ca<sup>2+</sup>]<sub>o</sub> on pH<sub>i</sub> and tension in several fibers when the Na-K pump is inhibited with either a cardioactive steroid  $(10^{-5} \text{ M strophanthidin for } 15-30 \text{ min})$  or by superfusion with a K-free solution. Although not illustrated here, the effects can be seen clearly in Fig. 3 of Vaughan-Jones et al. (1983) (Na-K pump inhibited in K-free solution). In that figure, raising [Ca<sup>2+</sup>]<sub>o</sub> from 0.5 to 5.0 mM produced a rise followed by a secondary decline (within 3 min) of tonic tension to  $\sim 60\%$  of its peak value and this was associated with an intracellular acidosis of 0.15 units. The fall in pH<sub>i</sub> has been shown to be secondary to a rise in  $[Ca^{2+}]_i$  (Vaughan-Jones et al., 1983). It is now evident that the intracellular acidosis of 0.15 units can account for much of the secondary decline in tension when reference is made to the effects of pH<sub>i</sub> on tension seen in the present work. Although other factors also contribute to the secondary decline of tension when [Ca<sup>2+</sup>]<sub>o</sub> is raised (see Discussion), the present work emphasizes that the effects of changes in pH<sub>i</sub> cannot be ignored.

#### DISCUSSION

## Effects on Tension of Changes in pH<sub>i</sub> and pH<sub>o</sub>

The present work provides direct measurements of  $pH_i$  coupled with simultaneous measurements of twitch and tonic tension in an isolated cardiac preparation. Most previous studies on the effects of pH on contraction have measured pH<sub>i</sub> in intact hearts using either nuclear magnetic resonance (Jacobus et al., 1982) or dye distribution techniques (Waddell and Bates, 1969). These techniques suffer, however, from very limited time resolution in comparison with the pH-sensitive electrodes used in the present study (the recessed-tip pH electrodes used in the present work gave a 90% response to a pH change within 20 s; see Methods).

Under most circumstances, we find that an intracellular acidosis is associated with a fall in twitch and tonic tension. In addition, the time course of change in tension follows, at least qualitatively, the change in pH<sub>i</sub>. It is important to note that altering  $pH_o$  does not affect tension until there is a change in  $pH_i$  (Fig. 5). Furthermore, changes in tension induced with  $NH_4Cl$ , when the pH of the bulk solution is constant, also correlate well with the observed changes in  $pH_i$ . These results therefore indicate that acidification and alkalinization affect tension via changes in  $pH_i$  rather than  $pH_o$ . De Marranes et al. (1981) have shown that the addition of external weak acids produces a small transient alkalosis at the extracellular surface of a Purkinje fiber. Similarly, superfusion of an external weak base, e.g.,  $NH_4Cl$ , produces a transient acidosis at the surface (Kaila, K., and R. D. Vaughan-Jones, unpublished observations). While we cannot exclude entirely some influence on tension from alterations in surface pH, we conclude that this effect will be small, simply because changing the bulk  $pH_o$  by 1.0 unit has no immediate effect on force.

## Quantitative Effect of pHi on Tension

Measurements of  $[Ca^{2+}]_i$  using acquorin have shown that the decrease in twitch tension during an intracellular acidosis is not associated with a decrease in the [Ca<sup>2+</sup>]<sub>i</sub> transient (Allen and Orchard, 1983). Indeed, systolic [Ca<sup>2+</sup>]<sub>i</sub> may even increase during acidification. Furthermore, diastolic [Ca2+], also rises rather than falls during an intracellular acidosis (Bers and Ellis, 1983; Allen et al., 1984; but compare Hess and Weingart, 1982). It therefore appears that the fall in twitch and tonic tension during an intracellular acidosis is not due to a fall in either systolic or diastolic  $[Ca^{2+}]_i$  and must therefore be due to a decrease in the  $Ca^{2+}$ sensitivity of the contractile proteins. Direct evidence for this is provided by work on skinned cardiac muscle: acidification shifts the tension-pCa curve to higher Ca concentrations (Fabiato and Fabiato, 1978). However, a quantitative comparison of our results with those observed in skinned fibers is complicated by the fact that we have not measured  $[Ca^{2+}]_i$ . The data shown in Fig. 3 indicate that, when bulk pH<sub>o</sub> is constant, a fall in resting pH<sub>i</sub> of 0.17 units is sufficient to reduce twitch tension by 50%. Such a large effect of pH<sub>i</sub> on force can be observed in the skinned fiber provided that the contractile proteins are <50% saturated with Ca<sup>2+,1</sup> That this is likely to be the case in the present experiments is indicated by the fact that under control conditions, twitch tension in a Purkinje fiber is normally a small fraction of that which can be produced after, say, exposure to high concentrations of cardiac glycosides (see Eisner et al., 1983b, 1984). Hence, the relatively high sensitivity of tension to variations in pH<sub>i</sub> in the intact Purkinje fiber is consistent with the data obtained from skinned fibers.

<sup>&</sup>lt;sup>1</sup> In skinned cardiac fibers, the active tension-pCa curve is sigmoidal. A fall in pH<sub>i</sub>, in the range 7.4–6.6, produces a shift of this curve along the pCa axis to lower pCa values, i.e., to a higher  $[Ca^{2+}]$ . Because the tension-pCa curve is sigmoidal, the fractional depressant effect on tension of a given fall in pH will be smaller at low pCa values (when the myofibrils are saturated with  $Ca^{2+}$ ) and larger at high pCa values (when the proteins are far from saturated with  $Ca^{2+}$ ). By replotting the data presented in Fig. 1*B* of Fabiato and Fabiato (1978) (a tension-pCa curve determined at different pH values for skinned rat ventricular fibers), one can determine that, in skinned fibers, the mean slope of the relationship between log tension and pH<sub>i</sub> (at a constant pCa) is close to 2.0, provided that tension is <50% of maximum tension (this slope of 2.0 should be compared with the similar slope observed in Fig. 3*B* of the present work). However, as expected, at higher levels of tension, this slope in skinned fibers can decrease considerably.

However, a quantitative analysis of the present work shows that the exact relationship between  $pH_i$  and tension depends on the way in which  $pH_i$  is changed, a finding that is clearly at variance with the work on skinned fibers, where a unique relationship was observed.

The reason for the difference between the pH sensitivity of skinned and intact fibers is not known, but it seems likely that a change of pH<sub>i</sub> in intact fibers has an effect not only on the Ca<sup>2+</sup> sensitivity of the myofilaments, as in skinned fibers, but also on the processes that regulate  $[Ca^{2+}]_i$  and deliver Ca<sup>2+</sup> to the myofilaments. For example, a change in pH<sub>i</sub> could influence (a) Ca<sup>2+</sup> release and reuptake at internal stores such as the sarcoplasmic reticulum or other intracellular "buffer sites" (Fabiato and Fabiato, 1978), (b) Ca<sup>2+</sup> influx via membrane channels (Fry and Poole-Wilson, 1981), and (c) Ca<sup>2+</sup> efflux and/or influx via Na-Ca exchange (Philipson et al., 1982). Such additional influences of pH<sub>i</sub> could then contribute to the observation, mentioned above, that altering pH<sub>i</sub> in intact fibers also changes  $[Ca^{2+}]_i$ . The difference between the pH<sub>i</sub> sensitivity of force in skinned and intact fibers may therefore be due to the additional influence of pH<sub>i</sub> on  $[Ca^{2+}]_i$  in intact fibers.

Another observation in the present work is that in an intact fiber, the quantitative relationship between tension and pH<sub>i</sub> can depend on the direction in which pH<sub>i</sub> changes. Such hysteresis between force and pH<sub>i</sub> has not, as yet, been reported for either skinned or intact fibers. Its origin remains unknown, but there are at least three possible contributing factors. (a) As mentioned above, a change in  $pH_i$  can affect  $[Ca^{2+}]_i$ . Consequently, if the kinetics of the  $pH_i$ - $[Ca^{2+}]_i$  interaction were to depend on the direction in which pH<sub>i</sub> was changed, this would generate a hysteresis between force and pH<sub>i</sub>. (b) We cannot exclude the possibility that, for example, after a change in pH<sub>o</sub>, there is for a few minutes a distribution of cells within the Purkinje strand that possess different values of pH<sub>i</sub>. Since pH<sub>i</sub> in the present work was measured in a single cell, whereas force development was measured from the whole tissue, it is possible that a nonhomogeneity of  $pH_i$ under some non-steady state conditions could contribute to the hysteresis between force and pH<sub>i</sub>. (c) Finally, hysteresis has been reported between  $[Ca^{2+}]_i$ and force development in both cardiac muscle (Harrison et al., 1985) and skeletal muscle (Ridgway et al., 1983), and this phenomenon will further complicate the pH<sub>i</sub> dependence of force. However, the latter effect would be expected to influence both skinned and intact fibers equally.

One puzzling observation in the present work is that, on removal of external NH<sub>4</sub>Cl, after the initial rapid fall of tension, there was sometimes a secondary redevelopment of force. This secondary rise of tension coincided with the intracellular acidosis and seemed to occur when this acidosis was large. It is possible that, under these conditions, the intracellular acidosis promoted a rise in  $[Ca^{2+}]_i$  that was sufficiently large to increase developed tension despite the fact that pH<sub>i</sub> had decreased. It is interesting to note that the peak of the acidosis after NH<sub>4</sub>Cl removal coincided with a transient rise in  $a_{Na}^i$  that appeared to be mediated by Na-H exchange (Deitmer and Ellis, 1980; Piwnica-Worms et al., 1985). The transient increase in  $a_{Na}^i$  would increase  $[Ca^{2+}]_i$  via Na-Ca exchange and could account for the rise of tension at low pH<sub>i</sub>.

Simultaneous measurements of  $pH_i$  and contraction have recently been reported for guinea pig papillary muscle (Poole-Wilson and Seabrooke, 1985). Unlike the results reported here, an intracellular acidosis was always associated with a decrease in tension. Furthermore, a linear relationship between  $pH_i$  and tension was found and no evidence of hysteresis between  $pH_i$  and force was reported. A linear dependence of developed pressure on  $pH_i$  has also been reported for intact rabbit heart (Jacobus et al., 1982). It may therefore seem surprising that a steeper, nonlinear dependence of force on  $pH_i$  can be observed in the Purkinje fiber (e.g., Figs. 3 and 6). However, since the present work shows that the quantitative dependence of force on  $pH_i$  can depend upon the way in which  $pH_i$  is varied, it would be premature to conclude that the differences between our results and those reported previously are due to real differences in the  $pH_i$  sensitivity of contraction in different cardiac tissues.

## Effect of a Change in pH<sub>i</sub> on Tonic Tension

Our experiments show that a change in pH<sub>i</sub> affected not only twitch tension but also resting tension and voltage-dependent tonic tension (see Fig. 1). However, these components of tonic tension are not normally observed in the Purkinje fiber unless  $a_{Na}^{i}$ , and hence  $[Ca^{2+}]_{i}$ , has been elevated (Eisner et al., 1983b). It is therefore interesting to note the tension records in alkaline conditions. Not only was there the appearance of a tonic contracture at the holding potential (Fig. 1), but there was also a component of voltage-dependent tonic tension. These results can be explained if depolarization always produces a tonic increase of  $[Ca^{2+}]_{i}$ , but, under normal conditions, the rise of  $[Ca^{2+}]_{i}$  is too small to increase tension. Therefore, tonic tension would be seen if either  $[Ca^{2+}]_{i}$  was increased or the Ca sensitivity of the contractile proteins was increased.

An alternative explanation for the appearance of voltage-dependent tonic tension at high  $pH_i$  may be given by the observation that a low  $pH_i$  inhibits Na-Ca exchange in cardiac muscle (Philipson et al., 1982). It has been argued that, in cardiac muscle, the rise in tonic tension upon depolarization is controlled by a Ca<sup>2+</sup> influx via Na-Ca exchange (Chapman and Tunstall, 1981; Eisner et al., 1983b). Hence, a larger tonic force at a more alkaline  $pH_i$  could represent a release from inhibition of this exchange mechanism.

## Effects of pH<sub>i</sub> during Inotropic Interventions

It is well established that the positive inotropic effects of Na-K pump inhibition are produced by an increase in  $a_{Na}^{i}$  that acts on Na-Ca exchange to increase  $[Ca^{2+}]_i$ . This rise of  $[Ca^{2+}]_i$  is, however, accompanied by an intracellular acidification, and we have suggested previously that this acidification may attenuate the rise of tension produced by Na-K pump inhibition and will thereby decrease the steepness of the relationship between tension and  $a_{Na}^{i}$  (Eisner et al., 1983b, 1984). This is confirmed in the present experiments, which show that when the acidification was reversed, a large increase in all the components of tension was produced (cf. Fig. 8B, traces c and d). In previous work, we found that tension varied with  $(a_{Na})^n$ , where n, on average, is ~3 (Eisner et al., 1983b, 1984). The present results show that pH<sub>i</sub> fell by ~0.1 unit after inhibition of the Na-K pump (Fig. 8; see also Vaughan-Jones et al., 1983; Kaila and Vaughan-Jones, 1985). Therefore, the dependence of tension on  $a_{Na}^{i}$  will be underestimated, and, in the absence of any pH<sub>i</sub> change, tension will depend more steeply on  $a_{Na}^{i}$ , making n closer to 4 than to 3.

Similar arguments apply to the effects of elevating  $[Ca^{2+}]_o$ . After inhibition of the Na-pump, elevating  $[Ca^{2+}]_o$  produces an increase in twitch and tonic tension, which then declines back toward control levels (Vaughan-Jones et al., 1983; Allen et al., 1984). The transient component of the response may be accounted for partly by the observation that increasing  $[Ca^{2+}]_o$  produces only a transient rise in  $[Ca^{2+}]_i$  (Allen et al., 1984; Eisner and Valdeolmillos, 1986; Kaila, K., J. Voipio, and R. D. Vaughan-Jones, manuscript in preparation). However, the present work indicates that another contributing factor is likely to be the intracellular acidification that occurs after the elevation of  $[Ca^{2+}]_o$ .

A decrease in pH<sub>i</sub> may be relevant to two related observations in the literature. (a) If the Na-K pump is inhibited for a prolonged period, then after the initial rise in force, developed tension gradually decreases. (b) When the Na-K pump is inhibited, increasing [Ca<sup>2+</sup>]<sub>o</sub> can actually decrease developed tension. Both observations have been attributed to the phenomenon of "Ca overload" (Vassalle and Lin, 1979; Allen et al., 1985a). Two explanations have already been presented for this Ca overload. (a) The fall in twitch tension under these conditions may be produced by spontaneous oscillatory release of Ca<sup>2+</sup> from the sarcoplasmic reticulum (Kort and Lakatta, 1984; Cannell et al., 1985; Valdeolmillos and Eisner, 1985; Allen et al., 1985a). This release occurs most prominently at high mean levels of [Ca<sup>2+</sup>]<sub>i</sub> (Allen et al., 1985a). When such [Ca<sup>2+</sup>]<sub>i</sub> oscillations precede the stimulation of a twitch via depolarization, then both the [Ca<sup>2+</sup>]<sub>i</sub> transient and twitch tension are reduced (Allen et al., 1985a). (b) The fall of tension is accompanied by an increase of the intracellular inorganic phosphate concentration (Allen et al., 1986). An increase of inorganic phosphate decreases the Ca sensitivity of the contractile apparatus (Kentish, 1986) and could account for part of the fall of force. While the present data do not conflict with these interpretations of the Ca overload phenomenon, they nevertheless point to a prominent role of changes in pH<sub>i</sub> in reducing tension under these conditions.

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