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## Structure Reports

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# Propan-2-yl *r*-4-(4-fluorophenyl)-3-hydroxy-*c*-6-methyl-2-phenyl-4,5-dihydro-2*H*-indazole-*t*-5-carboxylate

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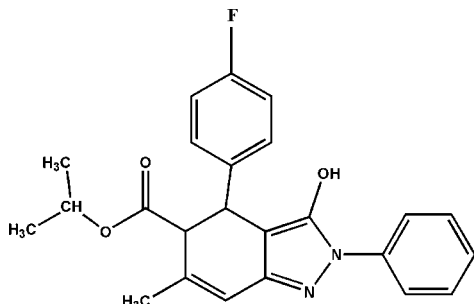
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Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å; disorder in main residue;  $R$  factor = 0.051;  $wR$  factor = 0.135; data-to-parameter ratio = 13.8.

In the title compound,  $\text{C}_{24}\text{H}_{23}\text{FN}_2\text{O}_3$ , the cyclohexene ring adopts a screw-boat conformation. The fluorobenzene ring attached to the cyclohexene ring and the phenyl ring attached to the indazole moiety are inclined to one another by  $57.77$  ( $13$ )°. In the crystal, molecules are linked by  $\text{O}-\text{H}\cdots\text{N}$  and  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds, forming chains with  $C(5)$  and  $C(10)$  graph-set motifs. There are also  $\text{C}-\text{H}\cdots\pi$  interactions present. The isopropoxycarbonyl group undergoes considerable thermal motion.

## Related literature

For examples of the biological activities of indazole derivatives, see: Jain *et al.* (1987); Palazzo *et al.* (1966); Popat *et al.* (2003); Beylin *et al.* (1991); George *et al.* (1998); Roman (1990). For the crystal structure of a similar compound, namely 4,6-bis(4-fluorophenyl)-2-phenyl-1*H*-indazol-3(2*H*)-one, see: Butcher *et al.* (2011). For information on graph-set motifs, see: Bernstein *et al.* (1995). For information on ring-puckering parameters, see: Cremer & Pople (1975).



## Experimental

## Crystal data

$\text{C}_{24}\text{H}_{23}\text{FN}_2\text{O}_3$	$V = 2185.9$ (2) Å <sup>3</sup>
$M_r = 406.44$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 17.640$ (1) Å	$\mu = 0.09$ mm <sup>-1</sup>
$b = 11.0295$ (6) Å	$T = 293$ K
$c = 11.3791$ (6) Å	$0.35 \times 0.25 \times 0.25$ mm
$\beta = 99.133$ (1)°	

## Data collection

Bruker SMART APEX CCD area-detector diffractometer	3843 independent reflections
20548 measured reflections	2970 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.034$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.051$	279 parameters
$wR(F^2) = 0.135$	H-atom parameters constrained
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.34$ e Å <sup>-3</sup>
3843 reflections	$\Delta\rho_{\text{min}} = -0.22$ e Å <sup>-3</sup>

## Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the N1/N2/C3/C8/C9 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O1}-\text{H1}\cdots\text{N1}^{\text{i}}$	0.82	1.82	2.6143 (18)	162
$\text{C22}-\text{H22}\cdots\text{O52}^{\text{ii}}$	0.93	2.56	3.229 (6)	129
$\text{C24}-\text{H24}\cdots\text{Cg1}^{\text{iii}}$	0.93	2.77	3.694 (3)	174

 Symmetry codes: (i)  $x, -y + \frac{1}{2}, z + \frac{1}{2}$ ; (ii)  $x, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (iii)  $-x, y + \frac{1}{2}, -z - \frac{1}{2}$ .

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1999) and *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*, *PLATON* and *publCIF* (Westrip, 2010).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2500).

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## supplementary materials

*Acta Cryst.* (2012). E68, o3021–o3022 [doi:10.1107/S1600536812039955]

## Propan-2-yl *r*-4-(4-fluorophenyl)-3-hydroxy-*c*-6-methyl-2-phenyl-4,5-dihydro-2*H*-indazole-*t*-5-carboxylate

S. Rizwana Begum, R. Hema, K. Pandiarajan, Sridhar Balasubramanian and A. G. Anitha

### Comment

Indazole derivatives possess a wide spectrum of pharmacological activities, such as analgesic, anti-inflammatory, anti-depressant, anti-hypertensive, anti-viral and anti-cancer (Jain *et al.*, 1987; Palazzo *et al.*, 1966; Popat *et al.*, 2003; Beylin *et al.*, 1991; George *et al.*, 1998; Roman, 1990). In the view of these important attributes, the title compound was synthesized and its crystal structure is described herein.

In the title molecule, Fig. 1, the cyclohexene ring adopts a scew-boat conformation with puckering parameters:  $Q = 0.391(2) \text{ \AA}$ ,  $\theta = 61.7(3)^\circ$  and  $\varphi = 28.7(3)^\circ$  (Cremer & Pople, 1975). The methyl group attached at C6 is substituted in the  $\beta$  equatorial position [ $C8-C7-C6-C61 = 176.7(2)^\circ$ ]. The pyrazole ring (N1/N2/C3/C8/C9) is almost planar with a maximum deviation of  $0.010(2) \text{ \AA}$  for atom N2. This five membered ring and the attached phenyl (C21-C26) ring make a dihedral angle of  $37.51(12)^\circ$ . The same phenyl ring makes a dihedral angle of  $57.77(13)^\circ$  with the fluorophenyl ring (C41-C46). This is similar to the situation in a related structure, 4,6-Bis(4-fluorophenyl)-2-phenyl-1*H*-indazol-3(2*H*)-one (Butcher *et al.*, 2011), where the same angle is  $57.69(10)^\circ$ .

In the crystal, molecules are linked by  $O-H\cdots N$  and weak  $C-H\cdots O$  hydrogen bonds, forming chains with C(5) and C(10) graph-set motifs (Bernstein *et al.*, 1995) [Table 1 and Fig. 2]. There are also  $C-H\cdots\pi$  interactions present (Table 1).

### Experimental

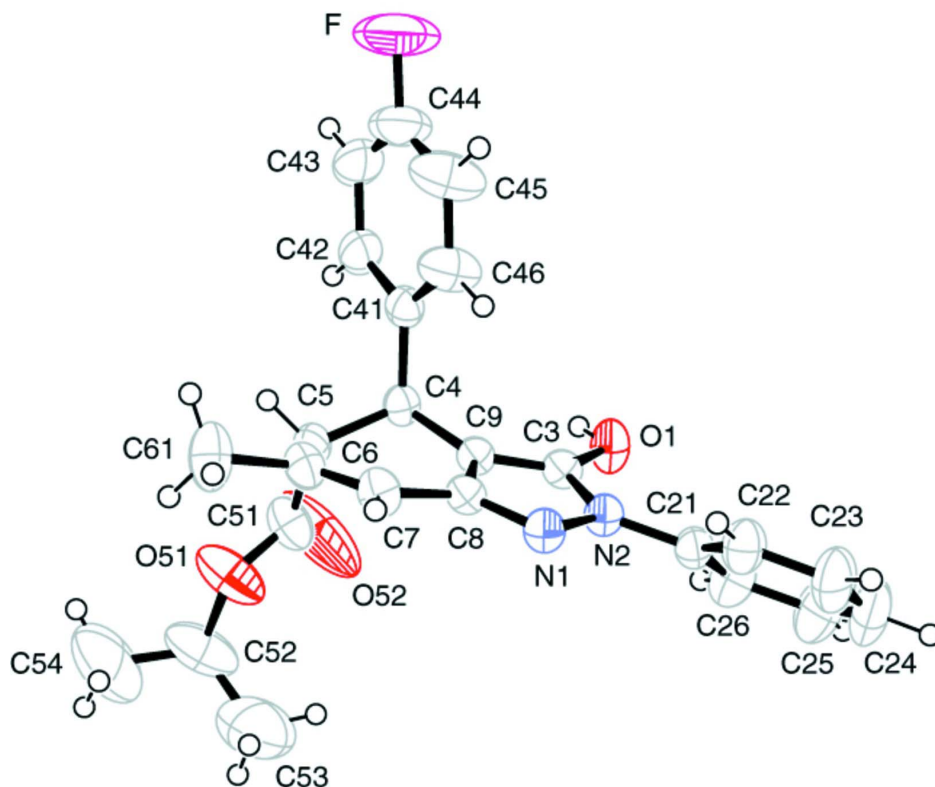
A mixture of isopropyl acetoacetate (14.4 g, 100 mmol), 4-fluorobenzaldehyde (6.2 g, 50 mmol) and methylamine (1.55 g, 50 mmol) in ethanol (50 ml) was heated to boiling. The reaction mixture was kept overnight and the solid that separated out was filtered off and purified by recrystallization from ethanol. 9.5 g (25 mmol) of this product, *r*(2),*c*(4)-bis-(isopropoxycarbonyl)-*c*(5)-hydroxy-*t*(5)-methyl-*t*(3)-(4'-fluorophenyl)cyclohexanone (yield 14.8 g, 78%, M.p. 452 K), was dissolved in acetic acid and after the addition of phenylhydrazine (4.3 g, 40 mmol) in the presence of sodium acetate (2.9 g, 50 mmol), the reaction mixture was refluxed for 4 h. The solution was cooled and then poured into crushed ice. The precipitated solid was filtered off by suction (yield 70%, M.p. 493 K) and recrystallized from ethanol giving block-like colourless crystals.

### Refinement

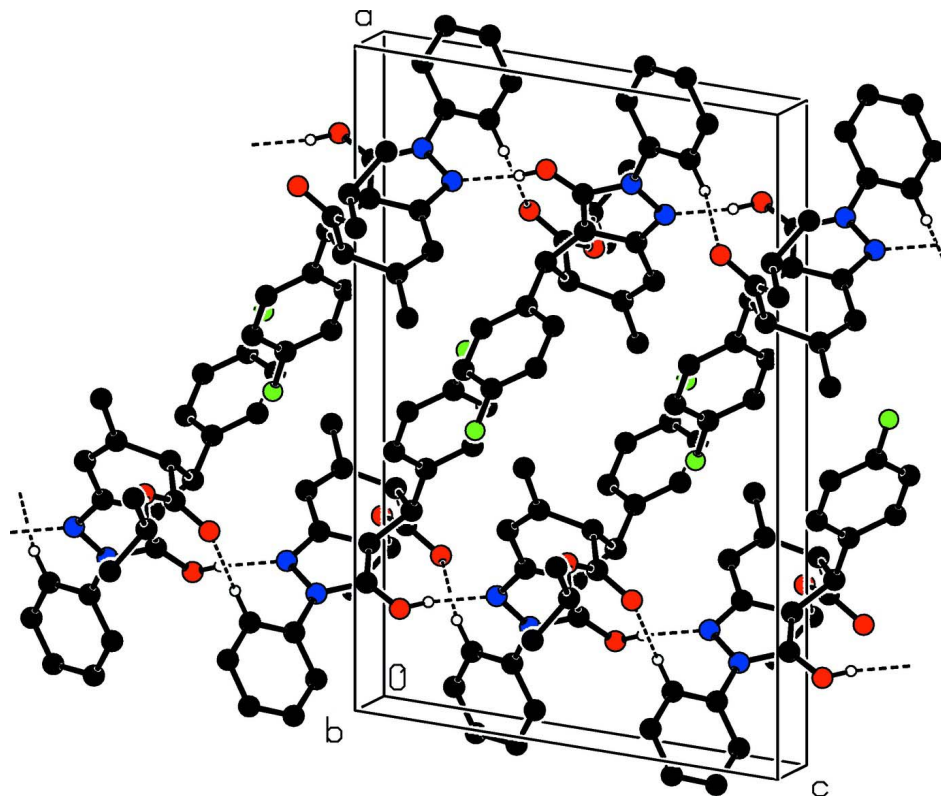
The isopropylcarboxylate group undergoes considerable thermal motion. Atom O52 was refined as disordered over two positions (O52/O52') and had a final refined occupancy ratio of 0.724(15):0.276(15). The C-bound H-atoms were included in calculated positions and treated as riding atoms: O-H =  $0.82 \text{ \AA}$ , C-H = 0.93, 0.98 and  $0.96 \text{ \AA}$  for CH(aromatic), CH and CH<sub>3</sub> H-atoms, respectively, with  $U_{iso}(H) = k \times U_{eq}(\text{parent C-atom})$ , where  $k = 1.5$  for OH and CH<sub>3</sub> H-atoms and  $= 1.2$  for other H-atoms. Reflection 1 0 0 was partially obscured by the beam stop and was omitted.

**Computing details**

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE* (Bruker, 2001); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1999) and *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008), *PLATON* (Spek, 2009) and *publCIF* (Westrip, 2010).

**Figure 1**

View of the molecule structure of the title molecule, showing the atom-labelling. Displacement ellipsoids are drawn at the 50% probability level. Only the major component of the disordered atom O52 is shown.



**Figure 2**

A view along the *b* axis of the crystal packing diagram of the title compound. Dashed lines indicate hydrogen bonds (see Table 1 for details).

**Propan-2-yl *r*-4-(4-fluorophenyl)-3-hydroxy-*c*-6-methyl-2-phenyl- 4,5-dihydro-2*H*-indazole-*t*-5-carboxylate**

*Crystal data*

$C_{24}H_{23}FN_2O_3$

$M_r = 406.44$

Monoclinic,  $P2_1/c$

Hall symbol:  $-P\ 2_1/c$

$a = 17.640\ (1)\ \text{\AA}$

$b = 11.0295\ (6)\ \text{\AA}$

$c = 11.3791\ (6)\ \text{\AA}$

$\beta = 99.133\ (1)^\circ$

$V = 2185.9\ (2)\ \text{\AA}^3$

$Z = 4$

$F(000) = 856$

$D_x = 1.235\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 3844 reflections

$\theta = 2\text{--}25^\circ$

$\mu = 0.09\ \text{mm}^{-1}$

$T = 293\ \text{K}$

Block, colourless

$0.35 \times 0.25 \times 0.25\ \text{mm}$

*Data collection*

Bruker SMART APEX CCD area-detector  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\omega$  scan

20548 measured reflections

3843 independent reflections

2970 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.034$

$\theta_{\text{max}} = 25.0^\circ$ ,  $\theta_{\text{min}} = 2.2^\circ$

$h = -20 \rightarrow 20$

$k = -13 \rightarrow 13$

$l = -13 \rightarrow 13$

Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.051$	H-atom parameters constrained
$wR(F^2) = 0.135$	$w = 1/[\sigma^2(F_o^2) + (0.0586P)^2 + 0.9373P]$
$S = 1.03$	where $P = (F_o^2 + 2F_c^2)/3$
3843 reflections	$(\Delta/\sigma)_{\max} < 0.001$
279 parameters	$\Delta\rho_{\max} = 0.34 \text{ e } \text{\AA}^{-3}$
0 restraints	$\Delta\rho_{\min} = -0.22 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
F	0.55203 (11)	0.4061 (2)	0.2222 (2)	0.1378 (9)	
O1	0.14456 (9)	0.32293 (13)	0.05959 (11)	0.0505 (4)	
H1	0.1609	0.2941	0.1252	0.076*	
O51	0.26507 (13)	-0.19468 (16)	-0.01810 (17)	0.0885 (6)	
O52	0.2223 (4)	-0.1001 (5)	0.1280 (9)	0.116 (2)	0.724 (15)
O52'	0.1999 (12)	-0.0778 (13)	0.0673 (18)	0.116 (2)	0.276 (15)
N2	0.14598 (9)	0.28868 (14)	-0.13981 (12)	0.0369 (4)	
N1	0.18599 (9)	0.22532 (14)	-0.21440 (12)	0.0386 (4)	
C3	0.17400 (10)	0.26286 (16)	-0.02333 (14)	0.0356 (4)	
C4	0.28913 (11)	0.12354 (17)	0.07402 (16)	0.0376 (4)	
H4	0.2638	0.1076	0.1431	0.045*	
C5	0.31565 (12)	0.00027 (17)	0.02803 (17)	0.0430 (5)	
H5	0.3642	-0.0202	0.0788	0.052*	
C6	0.33235 (12)	0.00401 (18)	-0.09910 (18)	0.0453 (5)	
C7	0.29229 (12)	0.07767 (18)	-0.17831 (17)	0.0443 (5)	
H7	0.3004	0.0767	-0.2571	0.053*	
C8	0.23623 (10)	0.15914 (16)	-0.14217 (15)	0.0362 (4)	
C9	0.23134 (10)	0.17912 (16)	-0.02211 (15)	0.0351 (4)	
C21	0.08425 (11)	0.36638 (17)	-0.18736 (16)	0.0396 (4)	
C22	0.09024 (13)	0.4340 (2)	-0.28730 (18)	0.0538 (6)	
H22	0.1342	0.4292	-0.3226	0.065*	
C23	0.03079 (16)	0.5085 (2)	-0.3344 (2)	0.0722 (7)	
H23	0.0347	0.5539	-0.4021	0.087*	
C24	-0.03366 (17)	0.5169 (3)	-0.2835 (3)	0.0826 (9)	
H24	-0.0738	0.5671	-0.3165	0.099*	
C25	-0.03905 (15)	0.4505 (3)	-0.1829 (3)	0.0794 (8)	

H25	-0.0828	0.4571	-0.1472	0.095*
C26	0.01950 (13)	0.3744 (2)	-0.1345 (2)	0.0589 (6)
H26	0.0154	0.3289	-0.0669	0.071*
C41	0.35810 (11)	0.20446 (17)	0.11368 (17)	0.0424 (5)
C42	0.40331 (13)	0.1842 (2)	0.2224 (2)	0.0575 (6)
H42	0.3898	0.1232	0.2715	0.069*
C43	0.46809 (15)	0.2525 (3)	0.2599 (3)	0.0768 (8)
H43	0.4980	0.2382	0.3335	0.092*
C44	0.48710 (16)	0.3400 (3)	0.1879 (3)	0.0832 (9)
C45	0.44411 (19)	0.3651 (3)	0.0811 (3)	0.0933 (10)
H45	0.4582	0.4270	0.0335	0.112*
C46	0.37878 (15)	0.2966 (2)	0.0444 (2)	0.0690 (7)
H46	0.3485	0.3134	-0.0284	0.083*
C51	0.25938 (15)	-0.0996 (2)	0.0456 (2)	0.0570 (6)
C52	0.2159 (2)	-0.3005 (3)	-0.0069 (3)	0.0973 (11)
H52	0.2039	-0.3045	0.0742	0.117*
C53	0.1431 (3)	-0.2891 (4)	-0.0942 (4)	0.1460 (17)
H53A	0.1171	-0.2157	-0.0786	0.219*
H53B	0.1105	-0.3574	-0.0861	0.219*
H53C	0.1550	-0.2868	-0.1736	0.219*
C54	0.2622 (3)	-0.4083 (3)	-0.0300 (5)	0.176 (2)
H54A	0.2719	-0.4060	-0.1106	0.264*
H54B	0.2344	-0.4809	-0.0177	0.264*
H54C	0.3101	-0.4076	0.0235	0.264*
C61	0.39422 (15)	-0.0780 (2)	-0.1293 (2)	0.0732 (8)
H61A	0.3978	-0.0699	-0.2122	0.110*
H61B	0.3821	-0.1605	-0.1127	0.110*
H61C	0.4424	-0.0562	-0.0823	0.110*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
F	0.0942 (13)	0.1364 (18)	0.175 (2)	-0.0579 (13)	-0.0032 (13)	-0.0562 (16)
O1	0.0724 (10)	0.0546 (9)	0.0250 (7)	0.0222 (7)	0.0091 (6)	0.0010 (6)
O51	0.1363 (17)	0.0548 (11)	0.0841 (13)	-0.0338 (11)	0.0475 (12)	-0.0155 (9)
O52	0.190 (5)	0.060 (2)	0.130 (5)	-0.038 (2)	0.120 (5)	-0.015 (3)
O52'	0.190 (5)	0.060 (2)	0.130 (5)	-0.038 (2)	0.120 (5)	-0.015 (3)
N2	0.0446 (9)	0.0414 (9)	0.0246 (7)	0.0058 (7)	0.0059 (6)	0.0001 (6)
N1	0.0469 (9)	0.0430 (9)	0.0267 (8)	0.0047 (7)	0.0086 (7)	-0.0009 (7)
C3	0.0464 (11)	0.0372 (10)	0.0232 (9)	-0.0012 (8)	0.0061 (8)	-0.0001 (7)
C4	0.0446 (11)	0.0376 (10)	0.0310 (9)	0.0030 (8)	0.0069 (8)	0.0035 (8)
C5	0.0473 (11)	0.0386 (11)	0.0419 (11)	0.0041 (9)	0.0038 (9)	0.0050 (8)
C6	0.0508 (12)	0.0374 (11)	0.0502 (12)	0.0023 (9)	0.0156 (10)	-0.0006 (9)
C7	0.0554 (12)	0.0447 (11)	0.0358 (10)	0.0026 (10)	0.0167 (9)	-0.0001 (9)
C8	0.0424 (10)	0.0357 (10)	0.0308 (9)	-0.0002 (8)	0.0069 (8)	0.0005 (8)
C9	0.0421 (10)	0.0347 (10)	0.0278 (9)	0.0000 (8)	0.0039 (8)	0.0007 (7)
C21	0.0466 (11)	0.0391 (10)	0.0318 (9)	0.0043 (9)	0.0020 (8)	-0.0020 (8)
C22	0.0616 (14)	0.0564 (13)	0.0436 (12)	0.0109 (11)	0.0086 (10)	0.0102 (10)
C23	0.0874 (19)	0.0681 (17)	0.0581 (15)	0.0247 (15)	0.0020 (14)	0.0201 (12)
C24	0.0769 (19)	0.0825 (19)	0.084 (2)	0.0399 (16)	-0.0019 (15)	0.0140 (16)

C25	0.0602 (16)	0.097 (2)	0.0829 (19)	0.0294 (15)	0.0176 (14)	0.0055 (17)
C26	0.0548 (13)	0.0712 (16)	0.0519 (13)	0.0120 (12)	0.0126 (11)	0.0084 (11)
C41	0.0474 (11)	0.0405 (11)	0.0393 (10)	0.0046 (9)	0.0064 (9)	-0.0047 (9)
C42	0.0580 (14)	0.0562 (14)	0.0531 (13)	0.0080 (11)	-0.0070 (11)	-0.0061 (11)
C43	0.0652 (17)	0.0789 (19)	0.0766 (18)	0.0082 (15)	-0.0182 (14)	-0.0249 (16)
C44	0.0624 (17)	0.084 (2)	0.100 (2)	-0.0204 (15)	0.0017 (16)	-0.0396 (19)
C45	0.106 (2)	0.086 (2)	0.089 (2)	-0.0484 (19)	0.0201 (19)	-0.0070 (17)
C46	0.0818 (17)	0.0691 (16)	0.0535 (14)	-0.0267 (14)	0.0024 (12)	0.0025 (12)
C51	0.0759 (16)	0.0414 (12)	0.0585 (14)	0.0012 (11)	0.0251 (12)	0.0012 (10)
C52	0.142 (3)	0.0652 (19)	0.088 (2)	-0.047 (2)	0.029 (2)	-0.0090 (16)
C53	0.155 (4)	0.104 (3)	0.172 (4)	-0.040 (3)	0.003 (3)	-0.014 (3)
C54	0.198 (5)	0.054 (2)	0.274 (7)	-0.019 (3)	0.033 (5)	-0.004 (3)
C61	0.0776 (17)	0.0674 (16)	0.0804 (18)	0.0276 (14)	0.0305 (14)	0.0079 (14)

*Geometric parameters (Å, °)*

F—C44	1.361 (3)	C23—H23	0.9300
O1—C3	1.325 (2)	C24—C25	1.375 (4)
O1—H1	0.8200	C24—H24	0.9300
O51—C51	1.288 (3)	C25—C26	1.376 (3)
O51—C52	1.472 (3)	C25—H25	0.9300
O52—C51	1.225 (7)	C26—H26	0.9300
O52'—C51	1.142 (18)	C41—C46	1.371 (3)
N2—C3	1.369 (2)	C41—C42	1.380 (3)
N2—N1	1.377 (2)	C42—C43	1.378 (4)
N2—C21	1.423 (2)	C42—H42	0.9300
N1—C8	1.328 (2)	C43—C44	1.343 (4)
C3—C9	1.368 (3)	C43—H43	0.9300
C4—C9	1.502 (2)	C44—C45	1.354 (4)
C4—C41	1.519 (3)	C45—C46	1.386 (4)
C4—C5	1.555 (3)	C45—H45	0.9300
C4—H4	0.9800	C46—H46	0.9300
C5—C51	1.517 (3)	C52—C54	1.489 (6)
C5—C6	1.522 (3)	C52—C53	1.499 (5)
C5—H5	0.9800	C52—H52	0.9800
C6—C7	1.330 (3)	C53—H53A	0.9600
C6—C61	1.499 (3)	C53—H53B	0.9600
C7—C8	1.444 (3)	C53—H53C	0.9600
C7—H7	0.9300	C54—H54A	0.9600
C8—C9	1.400 (2)	C54—H54B	0.9600
C21—C26	1.376 (3)	C54—H54C	0.9600
C21—C22	1.378 (3)	C61—H61A	0.9600
C22—C23	1.372 (3)	C61—H61B	0.9600
C22—H22	0.9300	C61—H61C	0.9600
C23—C24	1.358 (4)		
C3—O1—H1	109.5	C25—C26—H26	120.5
C51—O51—C52	119.8 (2)	C46—C41—C42	117.8 (2)
C3—N2—N1	110.45 (14)	C46—C41—C4	122.59 (18)
C3—N2—C21	129.06 (15)	C42—C41—C4	119.56 (19)



N1—N2—C21	120.48 (14)	C43—C42—C41	121.5 (2)
C8—N1—N2	104.75 (13)	C43—C42—H42	119.3
O1—C3—C9	134.69 (16)	C41—C42—H42	119.3
O1—C3—N2	117.61 (16)	C44—C43—C42	118.6 (3)
C9—C3—N2	107.64 (15)	C44—C43—H43	120.7
C9—C4—C41	113.30 (15)	C42—C43—H43	120.7
C9—C4—C5	108.49 (15)	C43—C44—C45	122.4 (3)
C41—C4—C5	110.14 (15)	C43—C44—F	119.3 (3)
C9—C4—H4	108.3	C45—C44—F	118.3 (3)
C41—C4—H4	108.3	C44—C45—C46	118.7 (3)
C5—C4—H4	108.3	C44—C45—H45	120.7
C51—C5—C6	111.98 (17)	C46—C45—H45	120.7
C51—C5—C4	110.64 (16)	C41—C46—C45	121.0 (3)
C6—C5—C4	114.21 (15)	C41—C46—H46	119.5
C51—C5—H5	106.5	C45—C46—H46	119.5
C6—C5—H5	106.5	O52'—C51—O52	38.2 (9)
C4—C5—H5	106.5	O52'—C51—O51	116.8 (8)
C7—C6—C61	122.76 (19)	O52—C51—O51	121.7 (3)
C7—C6—C5	119.90 (17)	O52'—C51—C5	121.2 (7)
C61—C6—C5	117.34 (18)	O52—C51—C5	123.0 (3)
C6—C7—C8	120.09 (17)	O51—C51—C5	114.0 (2)
C6—C7—H7	120.0	O51—C52—C54	105.7 (3)
C8—C7—H7	120.0	O51—C52—C53	109.3 (3)
N1—C8—C9	112.28 (16)	C54—C52—C53	112.7 (3)
N1—C8—C7	125.88 (16)	O51—C52—H52	109.7
C9—C8—C7	121.81 (17)	C54—C52—H52	109.7
C3—C9—C8	104.85 (15)	C53—C52—H52	109.7
C3—C9—C4	134.26 (16)	C52—C53—H53A	109.5
C8—C9—C4	120.43 (16)	C52—C53—H53B	109.5
C26—C21—C22	120.27 (19)	H53A—C53—H53B	109.5
C26—C21—N2	120.68 (18)	C52—C53—H53C	109.5
C22—C21—N2	119.04 (18)	H53A—C53—H53C	109.5
C23—C22—C21	119.5 (2)	H53B—C53—H53C	109.5
C23—C22—H22	120.2	C52—C54—H54A	109.5
C21—C22—H22	120.2	C52—C54—H54B	109.5
C24—C23—C22	120.9 (2)	H54A—C54—H54B	109.5
C24—C23—H23	119.5	C52—C54—H54C	109.5
C22—C23—H23	119.5	H54A—C54—H54C	109.5
C23—C24—C25	119.5 (2)	H54B—C54—H54C	109.5
C23—C24—H24	120.3	C6—C61—H61A	109.5
C25—C24—H24	120.3	C6—C61—H61B	109.5
C24—C25—C26	120.8 (2)	H61A—C61—H61B	109.5
C24—C25—H25	119.6	C6—C61—H61C	109.5
C26—C25—H25	119.6	H61A—C61—H61C	109.5
C21—C26—C25	119.1 (2)	H61B—C61—H61C	109.5
C21—C26—H26	120.5		
C3—N2—N1—C8	1.81 (19)	N1—N2—C21—C22	-38.0 (3)
C21—N2—N1—C8	-177.34 (16)	C26—C21—C22—C23	-0.7 (3)

N1—N2—C3—O1	175.77 (16)	N2—C21—C22—C23	179.5 (2)
C21—N2—C3—O1	-5.2 (3)	C21—C22—C23—C24	0.3 (4)
N1—N2—C3—C9	-1.8 (2)	C22—C23—C24—C25	0.5 (5)
C21—N2—C3—C9	177.29 (17)	C23—C24—C25—C26	-1.1 (5)
C9—C4—C5—C51	-83.40 (19)	C22—C21—C26—C25	0.1 (4)
C41—C4—C5—C51	152.07 (17)	N2—C21—C26—C25	180.0 (2)
C9—C4—C5—C6	44.0 (2)	C24—C25—C26—C21	0.7 (4)
C41—C4—C5—C6	-80.5 (2)	C9—C4—C41—C46	-21.0 (3)
C51—C5—C6—C7	94.1 (2)	C5—C4—C41—C46	100.8 (2)
C4—C5—C6—C7	-32.6 (3)	C9—C4—C41—C42	160.39 (18)
C51—C5—C6—C61	-85.5 (2)	C5—C4—C41—C42	-77.9 (2)
C4—C5—C6—C61	147.7 (2)	C46—C41—C42—C43	-1.2 (3)
C61—C6—C7—C8	-176.7 (2)	C4—C41—C42—C43	177.5 (2)
C5—C6—C7—C8	3.7 (3)	C41—C42—C43—C44	-0.2 (4)
N2—N1—C8—C9	-1.2 (2)	C42—C43—C44—C45	1.4 (5)
N2—N1—C8—C7	-179.28 (17)	C42—C43—C44—F	-178.3 (2)
C6—C7—C8—N1	-170.79 (19)	C43—C44—C45—C46	-0.9 (5)
C6—C7—C8—C9	11.3 (3)	F—C44—C45—C46	178.7 (3)
O1—C3—C9—C8	-176.0 (2)	C42—C41—C46—C45	1.6 (4)
N2—C3—C9—C8	1.0 (2)	C4—C41—C46—C45	-177.0 (2)
O1—C3—C9—C4	-4.1 (4)	C44—C45—C46—C41	-0.6 (5)
N2—C3—C9—C4	172.85 (19)	C52—O51—C51—O52'	-33.0 (12)
N1—C8—C9—C3	0.2 (2)	C52—O51—C51—O52	10.8 (7)
C7—C8—C9—C3	178.33 (17)	C52—O51—C51—C5	177.8 (2)
N1—C8—C9—C4	-173.10 (16)	C6—C5—C51—O52'	-114.5 (13)
C7—C8—C9—C4	5.1 (3)	C4—C5—C51—O52'	14.1 (13)
C41—C4—C9—C3	-80.1 (3)	C6—C5—C51—O52	-160.0 (6)
C5—C4—C9—C3	157.3 (2)	C4—C5—C51—O52	-31.3 (7)
C41—C4—C9—C8	90.8 (2)	C6—C5—C51—O51	33.2 (3)
C5—C4—C9—C8	-31.8 (2)	C4—C5—C51—O51	161.8 (2)
C3—N2—C21—C26	-36.8 (3)	C51—O51—C52—C54	-148.1 (3)
N1—N2—C21—C26	142.2 (2)	C51—O51—C52—C53	90.4 (4)
C3—N2—C21—C22	143.1 (2)		

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

Cg1 is the centroid of the N1/N2/C3/C8/C9 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1—H1 $\cdots$ N1 <sup>i</sup>	0.82	1.82	2.6143 (18)	162
C22—H22 $\cdots$ O52 <sup>ii</sup>	0.93	2.56	3.229 (6)	129
C24—H24 $\cdots$ Cg1 <sup>iii</sup>	0.93	2.77	3.694 (3)	174

Symmetry codes: (i)  $x, -y+1/2, z+1/2$ ; (ii)  $x, -y+1/2, z-1/2$ ; (iii)  $-x, y+1/2, -z-1/2$ .