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Characterization of volatile compounds and sensory properties of spine grape (*Vitis davidii* Foex) brandies aged with different toasted wood chips

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ABSTRACT

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1. Introduction

Spine grape (Vitis davidii Foex) is a wild Vitis species native to East Asia with hundreds of years of cultivation history in South China (Meng et al., 2013). This grape variety is densely covered with 1–2 mm thorns on one- or two-year-old canes, and the plants have very strong stress tolerance, such as disease resistance, low light resistance, and high humidity and heat resistance (Gutiérrez-Gamboa, Liu, Sun, & Fang, 2020; Meng et al., 2013). In addition, spine grape berries are characterized by a neutral aroma, low sugar content, and high acidity and are suitable for producing wines and distilled spirit-based beverages (Kong et al., 2019; Xiang et al., 2020). Aroma is a crucial quality parameter of brandies and is derived mainly from the grape variety, fermentation process, and storage in oak barrels; thus, these technological parameters directly determine the key flavor characteristics of the resulting products (Tao, García-Martín, & Sun, 2014). Numerous methods, such as cold maceration, enzyme application, and wine aging, have been developed for aroma improvement and to obtain a desirable and pleasant wine flavor.

ethyl decanoate, ethyl 2-methylbutanoate, ethyl octanoate, methyl salicylate, (*Z*)-2-hexenol, and furfural, contributed to the floral, fruity, and roasted/smoky attributes of the brandies aged with FO, AO, and chestnut chips. The 1-butanol, 1-propanol, phenylethanol, phenylethyl acetate, isoamyl acetate, and linalool contributed to the fruity, honey, and floral attributes of the brandies aged with JO and cherry chips. These findings are

extremely useful for the production of differentiated and high-quality spine grape brandies.

Aging is an important process for improving wine and brandy quality. In this study, the chemical characterization

and sensory properties of spine grape brandies were compared after aging with various species of wood chips,

including French oak (FO), American oak (AO), Mongolian oak (MO), Japanese blue oak (JO), chestnut, catalpa,

and cherry. The results showed that high color intensity and significant concentrations of tannins and poly-

phenols were observed in the brandies aged with FO, AO, and chestnut chips. The volatile compounds, such as

Previous studies have reported the key odor-active volatile compounds in spine grape berries and distilled spine grape spirits (Meng et al., 2013; Xiang et al., 2020). However, research on the aroma characteristics of spine grape wine and its derivative products during the aging process has not been reported.

Aging is a crucial process for improving the quality of final products and the organoleptic characteristics of alcoholic beverages. Traditionally, aging is carried out by storing wines and brandies in oak barrels for micro-oxygen aging, during which the beverages undergo important modifications that improve their physicochemical stability and increase the complexity of their mouthfeel and flavor (Canas et al., 2016; Tao et al., 2014). Oak-responsible aromatic compounds and astringencyrelated phenolic compounds are gradually released and transferred into wines or spirits during the aging process and are enhanced or modified by synergistic or masking effects. In addition, certain compounds are gently oxidized and degraded by atmospheric oxygen permeation through the walls of oak barrels, resulting in delicate mouthfeel and changes in color (Cerdán & Ancín-Azpilicueta, 2006; Tao

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et al., 2014). Specifically, the brandies aged in wood chips presented the highest intensities of greenish and topaz color, toasted and coffee odors, whereas the brandies aged in wooden barrels presented the highest intensities of golden color, alcohol odor and bitter taste (Caldeira, Anjos, Portal, Belchior, & Canas, 2010). Thus, the application of wood materials in the aging of spirit-based beverages, including fruit brandies, has a great influence on their final color, taste and aroma characteristics (Coldea et al., 2020; Yan et al., 2024). However, owing to the high cost, long production cycle, and complexity of barrel stock management in traditional natural aging processes, new materials and several alternative technologies have been used to shorten the aging period of wine products and achieve a satisfying color, taste and aroma (Coelho, Teixeira, Tavares, Domingues, & Oliveira, 2021; Tao et al., 2014). Among them, wood alternatives such as sessile oak (Q. petraea), chestnut (C. sativa), mulberry (M. alba and M. nigra), cherry (P. avium), fir (A. alba), and walnut (J. regia) have shown potential for replacing traditional *Q. robur* and *Q. alba* (Coldea et al., 2020). Although several species of wood chips, as alternatives to oak barrels, can be considered suitable for aging, the specific effects of these woods on the chemical characteristics and sensory properties of wines and brandies are need to be further elucidated.

Several studies have reported the evolution of volatile and phenolic compounds in wines and spirits following aging with oak chips (Coelho et al., 2021; Han, Tian, Zheng, Jiang, & Bian, 2024; Schumacher, Alañón, Castro-Vázquez, Pérez-Coello, & Díaz-Maroto, 2013). In most cases, the application of wood chips or staves can accelerate the extraction of wood-derived phenolic and volatile compounds and shorten the aging period. Indeed, since wines can completely soak and penetrate wood chips, the extraction efficiency and rate of wood-related compounds, such as volatiles, tannins, and phenols, are accelerated during the aging process in the presence of wood chips (Alencar et al., 2019; Yan et al., 2024). These compounds present important correlations with several sensory attributes, such as smoke and toasted odors, astringency, and bitter taste, which are positively correlated with the overall quality of the brandies (Caldeira et al., 2010). In apple brandies, the main esters, such as ethyl acetate, isobutyl acetate, ethyl decanoate, isoamyl-2-methylbutyrate, ethyl benzoate and ethyl nonanoate were accentuated by aging with mulberry, chestnut and cherry chips (Coldea et al., 2020). Toasted French oak chips enhanced the astringency, sweetness and richness of apple brandy and reduced the acidity and bitterness (Yan et al., 2024). In terms of organoleptic properties, wood chips or staves are desirable alternatives to oak barrels for producing short-term aged wines and brandies with satisfactory sensory quality (Tao et al., 2014). However, the effects of various species of wood chips on the chemical characterization and sensory properties in the spine grape brandies are still unclear.

Spine grape spirits are produced and widely appreciated by local consumers (Xiang et al., 2020). In this study, seven different species of toasted wood chips were used for spine grape distilled spirit aging to produce spine grape brandies. The color intensities, total tannin and polyphenol contents, volatile compound contents, and sensory properties of wood-aged brandies were investigated to comprehensively assess the potential of wood chips in spine grape brandy during the aging process. The results of this study may provide potential technology for the development of differentiated and high-quality spine grape brandies by aging with wood chips.

2. Materials and methods

2.1. Chemicals

Sodium chloride (NaCl), sodium bicarbonate (NaHCO₃), and anhydrous sodium sulfate (Na₂SO₄) were obtained from Tianjin Chemical Works (Tianjin, China). GC-grade ethanol (\geq 99.8 %), methanol (\geq 99.9 %), dichloromethane (\geq 99.8 %), and chemical standards for quantification and identification analyses (Supplementary Table S1) were

purchased from Sigma–Aldrich (Shanghai, China). 4-Methyl-2-pentanol (\geq 98 %) was employed as an internal standard.

2.2. Oak and substitute wood materials

Seven species of wood chips, including French oak (*Q. robur*, FO), American oak (*Q. alba*, AO), Mongolian oak (*Q. mongolica*, MO), Japanese blue oak (*Q. glauca*, JO), chestnut (*C. sativa*), catalpa (*C. bignonioides*), and cherry (*P. avium*), were obtained from Changyu Winery (Changyu Group Co., Ltd., Yantai, China). All wood chips were $20 \times 5 \times 5$ mm³ and were toasted at 180 °C for 40 min in an oven before being used as toasted chips.

2.3. Fermentation, distillation and aging modalities

The fermentation and distillation of spine grape (*Vitis davidii* Foex, 'Xiangzhenzhu') wines were performed according to previous studies (Duan et al., 2021; Xiang et al., 2020). Briefly, 1500 kg of healthy spine grapes (reducing sugar content, 177.33 g L⁻¹; titratable acidity, 5.61 g L⁻¹) were harvested on Sep. 20, 2021, from a commercial vineyard (31°54′90″ N, 105°02′35″ E; elevation 458 m; subtropical monsoon climate) of Liangqi Winery in Xujia County, Mianyang City, Sichuan Province, China. These grapes were immediately destemmed, crushed, and cold macerated overnight. The must was obtained by squeezing and then fermented at 20–22 °C with an industrial *Saccharomyces cerevisiae* yeast strain (200 mg L⁻¹, AC, Laffort Inc., France). The spine grape wines were fermented to 'dryness' (reducing sugar content <4 g L⁻¹) for approximately 7 to 8 days. After alcoholic fermentation, the basic chemical parameters (pH, 3.12; total acidity, 8.35 g L⁻¹; alcohol content, 8.50 % vol) of the spine grape base wine were determined.

The distillation of base wine was carried out by double distillation with the Charente pot distillation method to obtain the distilled spirit. On the basis of alcohol content, the heads (65 % vol), brouillis (40 % vol) and tails (15 % vol) distillates were obtained from 50 L of the base wine. The heads and tails were subsequently mixed with another portion of base wine (50 L) for further distillation. A total of 20 first-stage distillations were performed. The condenser temperature was maintained at 24 °C. After the first-stage distillations, the total collected brouillis were secondly distilled to obtain the heads (85 % vol), hearts (66 % vol), seconds (15 % vol) and tails (5 % vol) distillates. The hearts and seconds were proportionally mixed until the alcohol content of the blend reached to 40 % vol.

Aging with various species of wood chips was performed according to a previous report (Coelho et al., 2021), with slight modifications. A total of seven groups were applied in this study, i.e., toasted FO, AO, MO, JO, chestnut, catalpa, or cherry chips were added to the distilled spirit at a proportion of 4 g L^{-1} for the aging process. Aging was conducted in glass containers (10*L*) at 16–18 °C, and the mixtures were agitated daily during the aging period. Each experimental group was tested in triplicate. The seven groups of spine grape brandies were aged with different species of wood chips for 6 months, and distilled spirits without aging were used as the control samples (Control). All aged spine grape brandies were filtered and bottled for further chemical and sensory analyses.

2.4. Determination of physicochemical indicators of the spine grape brandies

The titratable acidity (expressed as tartaric acid equivalent, g L⁻¹), volatile acidity (expressed as acetic acid equivalent, g L⁻¹), pH, and alcohol content (% vol) of the spine grape brandies were determined according to the OIV standards (OIV, 2017). The concentrations of total tannins and polyphenols were determined via a UV–Vis spectrophotometer (Cary 60 UV–Vis, Agilent Technologies, Santa Clara, CA, USA) according to previous methods (Alencar et al., 2019; Yan et al., 2024). The total tannin content (expressed as epicatechin equivalents, mg L⁻¹)

was measured using the methyl cellulose precipitation method at 280 nm. The total polyphenol content (expressed as gallic acid equivalent, mg L⁻¹) was measured using the Folin–Ciocalteu method at 765 nm. The CIELab parameters of brandy chromaticity, such as lightness (*L**), redness–greenness (*a**), yellowness–blueness (*b**), chroma (C*), and hue angle (h), were measured using a colorimeter (CM-5, Konica-Minolta, Tokyo, Japan), and the total color difference (Δ E) was calculated as follows: $\Delta E = [(\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2]^{1/2}$.

2.5. Determination of volatile compounds

Volatile compounds of spine grape brandies were analyzed by a headspace solid–phase microextraction system combined with gas chromatography–mass spectrometry (HS–SPME–GC–MS) following previously reported methods (Li, Yang, Tian, Zou, & Li, 2020; Xiang et al., 2020). For sample analysis, 5 mL of each brandy sample was added to 1 g of NaCl and 10 μ L of an internal standard solution (4-methyl-2-pentanol, 1.0083 g L⁻¹) in a 15 mL headspace bottle. The vial was immediately sealed with a lid with a polytetrafluoroethylene (PTFE)-silicon septum. The mixture was homogenized with a magnetic stirrer (1 cm) at 400 rpm, and the volatile compounds were extracted at 40 °C for 30 min. An automatic SPME fiber (Supelco, Bellefonte, PA, USA; 50/30 μ m DVB/CAR/PDMS) was then immersed in the headspace, and the mixture was extracted for 30 min. The volatiles trapped in the fiber were thermally desorbed in the GC injector (250 °C) for 8 min.

GC-MS analysis was performed on an HP-INNOWAX column (60 m \times 0.25 mm, 0.25 μ m) using an Agilent 6890 GC coupled with a 5975C MS detector (Agilent Technologies, Wilmington, USA). The oven temperature was initially set to 40 °C for 2 min, increased to 210 °C at a speed of 3 °C min⁻¹, and increased again to 250 °C with an increase of 5 $^{\circ}$ C min⁻¹, after which the temperature was maintained at 250 $^{\circ}$ C for 5 min. Ultrapure helium (> 99.999 %) was used as a carrier gas at a constant flow rate of 1.0 mL min⁻¹. The temperatures of the injector and ionization source were maintained at 250 °C. The ionization source conditions were an electron energy of 70 eV and a mass scan ranging from m/z 30 to 350. The identities of the volatile compounds were carefully confirmed by comparing the retention times, MS fragmentation patterns of the existing standards, and mass spectra in the National Institute of Standards and Technology Mass Spectral Library. Standards of aroma compounds were dissolved in ethanol (GC-grade), and fifteen different concentrations of standards were serially diluted in alcohol solution (40 % vol: pH 3.8-4.0) to establish standard curves of volatile compounds for the HS-SPME-GC-MS analysis. The quantitative determination of the identified compounds was carried out using an internal standard method with 4-methyl-2-pentanol, and the results were calculated from the calibration curves in Supplementary Table S1. All analyses were conducted in triplicate for each sample.

The odor activity value (OAV), which measures the contribution of each volatile compound to the characteristic aroma, was calculated by dividing the relative concentration of a certain compound by its absolute odor threshold (Bowen & Reynolds, 2012).

2.6. Sensory analysis

The tasting panel consisted of sixteen panelists (aged 20 to 35 years; eight males and eight females), and they were recruited from the Professional Tasting Panel of the College of Enology, Northwest A&F University. All panelists were involved in flavor chemistry research on wines and had the winetasting experience. Before tasting, each panelist received eight consecutive sensory tasting trainings, including clarity, color, aroma, taste, style, and aftertaste, for 1 h each time to standardize the criteria among these panel members, according to our previous descriptions (Duan et al., 2021). The spine grape brandies were presented in ISO standard tasting glasses, numbered and provided to the panelists in a random order in an isolated tasting room (20–22 $^{\circ}$ C). The sensory evaluation was performed in triplicate for each brandy sample. The

brandy samples were evaluated for several quality attributes, i.e., clarity, color, aroma, taste, style, and aftertaste, using an 11-point quantitative scale, where 0 indicated no perceived descriptor (absence) and values from 1 to 10 corresponded to intensities ranging from low to maximum. The specific aroma descriptors used were divided into seven classes: fusel/solvent, green/plant, fruity, sweet, roasted/ smoky, sweaty/fatty, and floral.

2.7. Statistical analysis

The data are shown as the mean \pm standard error (n = 3). Statistically significant differences in the concentrations of physicochemical indicators or volatile compounds among the spine grape brandy samples were evaluated by one-way analysis of variance (ANOVA) and Duncan's multiple range tests at p < 0.05 using SPSS 22.0 (IBM, New York, USA). Calibration curves of volatile compounds were obtained using standard solutions of fifteen different concentrations, and the linear correlation coefficients (\mathbb{R}^2) and linear ranges were calculated. All descriptors of the sensory profile were mean-centered per panelist and scaled to unit variance. The clustered heatmap and principal component analysis (PCA) were performed using R 3.6.1 software.

3. Results and discussion

3.1. Acidity, tannin and polyphenol contents, and chromaticity of spine grape brandies

To determine the physicochemical changes in spine grape brandies aged with different species of wood chips (wood-aged brandies), the acidity, pH, alcohol content, tannin content, polyphenol content, and chromaticity of the resulting spine grape brandies were determined (Table 1). The brandies aged with FO and AO chips had higher values of titratable acidity (0.59 and 0.52 g L^{-1} , respectively) than that of other brandy groups. The values of volatile acidity and pH of all the brandy samples were in accordance with OIV standards. The alcohol contents were similar (~40 % vol) among all the wood-aged brandies, as they were produced from the same fermentation and distillation processes and under similar aging modality.

Compared with those in the control samples, the concentrations of tannins and polyphenols were substantially greater in the wood-aged spine grape brandies. Specifically, the brandies aged with FO, AO, and chestnut chips contained the most tannins (302.27, 311.01, and 304.40 g L^{-1} , respectively), followed by the brandies aged with cherry (117.68 g L^{-1}) and MO (35.99 g L^{-1}) chips, but the lowest tannin contents were detected in the brandies aged with catalpa (14.12 g L^{-1}) and JO (11.27 g) L^{-1}) chips. With respect to the polyphenol contents, spine grape brandies aged with FO and AO chips contained the most polyphenols (325.50 and 331.45 g L^{-1} , respectively), followed by the brandies aged with chestnut (315.57 g L^{-1}), cherry (153.73 g L^{-1}), catalpa (94.18 g L^{-1}), and MO (50.43 g L^{-1}) chips, and the lowest polyphenol content was detected in the brandies aged with JO (25.39 g L^{-1}) chips. During the aging process, several wood-derived compounds are gradually extracted from the hydroalcoholic matrix, which has a positive impact on the properties and final quality of alcoholic beverages (Coldea et al., 2020; Tao et al., 2014). The chestnut as an alternative wood induced a significantly greater accumulation of phenolic compounds, showing a high potential for the aging of wine spirits, confirming previous results that higher levels of gallic acid and ellagitannins are distinctive features of chestnut wood (Canas, Caldeira, Anjos, & Belchior, 2019; de Simón et al., 2014). Previous studies have shown that increasing the concentrations of condensed tannins and polyphenols in wines can affect the perception of astringency by enhancing the sensation of dryness and grip in the mouth (Coldea et al., 2020; Ortega-Heras, Pérez-Magariño, Cano-Mozo, & González-San José, 2010; Watrelot, Kuhl, & Waterhouse, 2019), indicating that spine grape brandies aged with FO, AO or chestnut chips obtained in the present study would be driven to more

Table 1

Ph	vsicochemical	parameters	of spine	grape	brandies	after	aging	with	wood	chips.
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Control	FO	AO	МО	JO	Chestnut	Catalpa	Cherry
$\textbf{0.41} \pm \textbf{0.04b}$	$\textbf{0.59}\pm\textbf{0.04a}$	$0.52\pm0.02a$	$0.33\pm0.03b$	$\textbf{0.34} \pm \textbf{0.04b}$	$0.37\pm0.03b$	$\textbf{0.42}\pm\textbf{0.03b}$	$\textbf{0.40} \pm \textbf{0.02b}$
$\textbf{0.30} \pm \textbf{0.02ab}$	$0.34\pm0.03a$	$0.21\pm0.05b$	$\textbf{0.33} \pm \textbf{0.01a}$	0.29 ± 0.03ab	$0.34\pm0.02a$	$0.26\pm0.02b$	$\textbf{0.26} \pm \textbf{0.01b}$
$\textbf{3.99} \pm \textbf{0.01b}$	$3.88\pm0.05b$	$4.01\pm0.08b$	$\textbf{4.23} \pm \textbf{0.05a}$	$\textbf{4.42} \pm \textbf{0.12a}$	$3.91\pm0.10b$	$\textbf{4.23} \pm \textbf{0.03a}$	$\textbf{4.10} \pm \textbf{0.08b}$
$39.97 \pm \mathbf{0.12a}$	$39.47 \pm \mathbf{0.37a}$	$39.03 \pm \mathbf{0.21a}$	$39.60 \pm \mathbf{0.24a}$	38.83 ± 0.12a	$39.73 \pm \mathbf{0.19a}$	$39.83 \pm \mathbf{0.09a}$	$39.73 \pm \mathbf{0.25a}$
tr	302.27 ± 15.17a	311.01 ± 17.53a	$\textbf{35.99} \pm \textbf{4.76c}$	11.27 ± 5.87d	$\begin{array}{l}\textbf{304.40} \pm \\ \textbf{8.66a} \end{array}$	$14.12 \pm 1.96 \text{d}$	$117.68\pm8.77b$
tr	$325.50\pm4.57a$	$331.45\pm9.27a$	$\textbf{50.43} \pm \textbf{3.27e}$	$25.39 \pm 1.22 f$	315.57 ± 3.75b	$\textbf{94.18} \pm \textbf{2.04d}$	153.73 ± 10.51c
$\begin{array}{c} 100.03 \pm \\ 0.81a \end{array}$	$96.18 \pm 0.12 c$	$\textbf{95.97} \pm \textbf{0.06d}$	98.79 ± 0.21ab	99.36 ± 0.50a	$\textbf{97.18} \pm \textbf{0.15b}$	98.73 ± 0.54ab	$\textbf{97.48} \pm \textbf{0.20b}$
$\textbf{0.12}\pm\textbf{0.01a}$	$-1.44\pm0.16d$	$-1.70\pm0.12e$	$-0.49\pm0.01c$	$\begin{array}{c} -0.29 \pm \\ 0.03b \end{array}$	$-1.56\pm0.07\text{d}$	$-2.03\pm0.16f$	$-0.23\pm0.02b$
$0.05\pm0.01~\text{g}$	$9.06\pm0.06b$	$10.38\pm0.28a$	$1.89 \pm 0.02 e$	$0.65\pm0.01f$	$\textbf{7.17} \pm \textbf{0.39c}$	$\textbf{4.89} \pm \textbf{0.53d}$	$6.39\pm0.05c$
$0.13\pm0.02~\text{g}$	$9.18\pm0.09b$	$10.52\pm0.28a$	$1.95\pm0.02e$	$0.71\pm0.01 f$	$\textbf{7.33} \pm \textbf{0.39c}$	$\textbf{4.40} \pm \textbf{0.05d}$	$7.18\pm0.56c$
$1.16 \pm 0.05 a$	$-1.41\pm0.02\text{d}$	$-1.41\pm0.01\text{d}$	$-1.32\pm0.05c$	$-1.15 \pm 0.01b$	$-1.36\pm0.01c$	$-1.28 \pm 0.02 \mathrm{b}$	$-1.52\pm0.01e$
0.00	$\textbf{9.89} \pm \textbf{0.19b}$	$11.24\pm0.16a$	$2.25\pm0.25 \mathrm{f}$	$0.92\pm0.35~g$	$\textbf{7.80} \pm \textbf{0.19c}$	$5.36\pm0.21e$	$6.83\pm0.12\text{d}$
	Control $0.41 \pm 0.04b$ $0.30 \pm 0.02ab$ $3.99 \pm 0.01b$ $39.97 \pm 0.12a$ tr tr $100.03 \pm$ 0.81a $0.12 \pm 0.01a$ 0.05 ± 0.01 g 0.13 ± 0.02 g $1.16 \pm 0.05a$ 0.00	Control FO $0.41 \pm 0.04b$ $0.59 \pm 0.04a$ $0.30 \pm 0.02ab$ $0.34 \pm 0.03a$ $3.99 \pm 0.01b$ $3.88 \pm 0.05b$ $39.97 \pm 0.12a$ $39.47 \pm 0.37a$ tr $302.27 \pm 15.17a$ tr $325.50 \pm 4.57a$ $100.03 \pm 0.81a$ $96.18 \pm 0.12c$ $0.12 \pm 0.01a$ $-1.44 \pm 0.16d$ 0.05 ± 0.01 g $9.06 \pm 0.06b$ 0.13 ± 0.02 g $9.18 \pm 0.09b$ $1.16 \pm 0.05a$ $-1.41 \pm 0.02d$ 0.00 $9.89 \pm 0.19b$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

FO, French oak; AO, American oak; MO, Mongolian oak; JO, Japanese blue oak. 'tr' means trace. Different letters in the same row means significant differences among the spine grape brandy samples according to Duncan's multiple range tests (p < 0.05).

intensities of mouthfeel attributes than spine grape brandies aged with other wood chips.

After aging, the brandy has a yellowish brown or amber color, reflecting its ripens aging and its desirable quality (Canas et al., 2019; Yan et al., 2024). In this study, the L^* , a^* , and h values of the color parameters were significantly greater in the brandies aged with MO, JO, catalpa, and cherry chips than in the other brandies, except for the a^* value in the catalpa aged brandies and the h value in the cherry aged brandies, whereas the b^* , C^* , and ΔE values were significantly greater in the brandies aged with FO and AO chips, followed by those aged with chestnut and cherry chips. These results, which are in accordance with those of a previous study (Caldeira et al., 2010), indicate that the brandies aged in French oak, American oak, and chestnut present a more discriminant and more mature color. During the aging process, the differences in the chromatic characteristics of the wine distillate might be due to the melanoidins produced by the Maillard reactions during wood roasting being transferred to the liquor (Herzfeld et al., 2011), the condensation reactions between tannins mediated by acetaldehyde (Picariello, Gambuti, Picariello, & Moio, 2017), and the oxidative phenomena of phenols, such as flavan-3-ol monomers ((+)-catechin and (-)-epicatechin), flavan-3-ols polymers (proanthocyanidins, also known as condensed tannins) and ellagitannins, leading to the formation of new brown color pigments (Canas et al., 2019; Flamini, Panighel, & De Marchi, 2021), indicating that the intensities of color and astringency perception of brandies are markedly modified after aging with different wood chips.

3.2. Quantification of volatile compounds and OAV analysis

Identification of the volatile compounds in spine grape brandies aged with toasted wood chips was achieved by HS–SPME–GC–MS analysis (Table 2). A total of 38 main volatile compounds in the spine grape brandies were tentatively identified and were significantly affected after aging with wood chips. The most abundant volatile compound in the wood-aged brandies was isoamyl alcohol (from 2412.10 to 2601.11 mg L⁻¹), followed by isobutanol (from 1342.76 to 1547.95 mg L⁻¹), 1-propanol (from 712.17 to 866.29 mg L⁻¹), and ethyl acetate (from 612.56 to 875.07 mg L⁻¹). Similarly, the quantification results revealed that the relative contents of isoamyl alcohol, isobutanol, and ethyl acetate were the highest among the aroma compounds in alcoholic beverages, especially in fruit brandies (Coldea et al., 2020; Xiang et al., 2020; Yan et al., 2024).

Difference tests confirmed that the concentrations of volatile

compounds were significantly differ among wood-aged brandies because of differences in inherent wood characteristics (Coldea et al., 2020; Picard, Nonier, Vivas, & Vivas, 2021). Compared with those in the control samples, the concentrations of isoamyl alcohol, isobutanol, ethyl lactate, ethyl hexanoate, ethyl 2-methylbutanoate, and linalool were significantly greater in all the wood-aged brandies, and there were no significant differences among these brandies; however, the concentrations of (E)-3-hexenol, ethyl acetate, ethyl butanoate, ethyl octanoate, α -terpineol, acetic acid, benzaldehyde, nonanal, and octanal were markedly lower in these wood-aged brandies. Specifically, the concentrations of 3-methyl-1-pentanol, 4-methyl-1-pentanol, furfural, and octanoic acid were significantly greater in the brandies aged with FO and AO, chestnut, catalpa, and cherry chips than in the other brandies, except for 4-methyl-1-pentanol in the chestnut aged brandies. The concentrations of 1-propanol and isoamyl acetate were significantly greater in the brandies aged with MO, JO, catalpa, and cherry chips than in the other brandies. In addition, the concentrations of ethyl decanoate, methyl salicylate, furfural and phenol were very high in the brandies aged with FO and AO chips. Interestingly, the concentrations of 1butanol were the highest in the brandies aged with JO chips, followed by cherry chips; moreover, the concentrations of (Z)-2-hexenol and propanoic acid were the highest in the brandies aged with AO, FO, and cherry chips. A recent study shows that alcohols, including propanol and 2,3-butanediol, can further affect the complexity of the aroma and taste of mulberry brandy by oxidation reactions to produce aldehydes, such as acetaldehyde, furfural and vanillin, and acids (Han et al., 2024). During the wood aging process, the inconsistency in the increase of ester concentration may be attributed to the direct transfer of wood-derived compounds (e.g., vanillin, guaiacol, and eugenol) from different species of wood and the reactions between volatile compounds and distillates (Caldeira et al., 2010; Yan et al., 2024). However, Ramirez et al. (2001) reported that the concentrations of volatile compounds did not depend on the solubility of these compounds but rather on the acid, base and polarity properties of the solution matrix. These previous findings provide some explanations for our results concerning the differences in volatile compounds in different wood-aged spine grape brandies.

The OAVs of the volatile compounds were analyzed to determine the potential contributions of the specific volatile compounds to the aroma or flavor characteristics of the wood-aged spine grape brandies (Table 3). In the wood-aged brandy samples, ethyl octanoate had the highest OAVs (from 882.01 to 1119.79), followed by (Z)-2-hexenol (from 106.78 to 136.61), ethyl hexanoate (from 110.53 to 123.09), isoamyl acetate (from 43.77 to 53.97), and ethyl acetate (from 19.98 to

Table 2

Concentrations of volatile compounds in the spine grape brandies after aging with wood chips by HS-SPME-GC-MS.

contom form form <thord< th=""> form form <t< th=""><th>Number</th><th>Volatile</th><th></th><th>Concentration (µ</th><th>$\log L^{-1}$)</th><th></th><th colspan="7"></th></t<></thord<>	Number	Volatile		Concentration (µ	$\log L^{-1}$)								
Ishuma Selection Selecition Selection <th< th=""><th></th><th>compounds</th><th>Control</th><th>FO</th><th>AO</th><th>МО</th><th>JO</th><th>Chestnut</th><th>Catalpa</th><th>Cherry</th></th<>		compounds	Control	FO	AO	МО	JO	Chestnut	Catalpa	Cherry			
Jumana Jumana <thjumana< th=""> <thjumana< th=""> <thjumana< td="" th<=""><th>1</th><td>Isobutanol</td><td>904,939.04 \pm</td><td>1,344,817.44</td><td>1,453,403.74</td><td>1,508,217.60</td><td>1,445,964.77</td><td>1,342,762.68</td><td>1,441,225.81</td><td>1,547,950.63</td></thjumana<></thjumana<></thjumana<>	1	Isobutanol	904,939.04 \pm	1,344,817.44	1,453,403.74	1,508,217.60	1,445,964.77	1,342,762.68	1,441,225.81	1,547,950.63			
basepi alcoha 1,580,794,31 2,440,105.30 2,280,385.17 2,411,20,37 2,421,30,37 2,421,30,37 2,421,30,37 2,421,30,37 2,421,30,37 2,420,30,34 2,431,30,36 3 Joctand 70,26,5,370 2,31,37 2,400,37 1,500,37	1	isobutanoi	3413.13b	\pm 17,541.07a	\pm 48,783.42a	\pm 23,539.15a	\pm 132,553.47a	\pm 27,080.96a	\pm 12,585.36a	\pm 72,009.07a			
110.148600 110.24800 110.24800 110.24800 110.24800 120.258000 120.258000 120.258000 120.258000 120.258000 120.258000	2	Isoamyl alcohol	1,508,784.31	2,440,105.82	2,526,385.17	2,412,101.37	2,428,208.41	2,441,542.28	2,562,875.97	2,601,107.98			
3 0.00000000000000000000000000000000000			\pm 15,318.080	\pm 4752.49a 170 54 \pm	$\pm 108,034.02a$ 157.03 \pm	\pm 16,219.56a 141.67 \pm	\pm 58,739.17a 157.98 \pm	\pm 53,199.49a 164 74 \pm	\pm 3990.70a 154 66 \pm	\pm 83,003.40a 156 36 \pm			
1-Prognal S38,114/70 ± S38,202/41 Y 72,306,20 ± Y 72,306 ±	3	1-Octanol	$70.26 \pm \mathbf{3.70c}$	2.81a	2.66ab	3.32b	6.07ab	5.12a	1.49ab	7.50ab			
a brongenom 22.02.23d 12.04.42c 24.107.28b 92.88.06a 518.38a 17.972-1 90.81 1: 97.975 b definition 24.12c 27.46a 34.24b 57.1b 18.78ab 11.28c 24.1ca 99.38 c bernard 10.70c 17.38c 07.03 + 07.03 + 07.05 + 07.04 + 08.03 + 07.05 + 07.04 + d bernard 01.07c 17.38c 03.07 + 27.07 + 07.04 + 07.05 + 07.05 +		1. Durana 1	538,114.70 \pm	735,606.29 ±	772,368.15 ±	858,221.71 ±	801,185.74 ±	712,170.97 ±	805,760.31 ±	866,285.59 ±			
SMethylin Sind Si, L Sind K	4	1-Propanoi	27,202.52d	12,914.42c	24,107.28b	39,288.06a	5185.38a	17,739.90c	17,994.94a	26,717.47a			
pentanol 24.12 27.46a 34.24a 57.1b 18.79ab 18.79ab 18.79ab 18.79ab 347.81a 37.271 0 pentanol 10.70c 17.39a 15.65a 071b 077b 17.98a 18.69a 31.29a	5	3-Methyl-1-	510.35 \pm	932.75 \pm	907.61 \pm	862.10 \pm	884.65 ±	915.72 \pm	908.31 \pm	937.68 \pm			
		pentanol	24.12c	27.46a	34.24a	5.71b	18.79ab	11.28a	24.16a	49.93a			
pensyletham in tr	6	4-Methyl-1-	$221.92 \pm 10.70c$	$3/1.38 \pm 1730_{2}$	375.95 ± 10.60a	351.39 ± 0.71 b	$358.70 \pm 0.17b$	349.81 ± 4 79b	381.29 ± 13.632	$3/2./1 \pm$			
7 Pheneylethanol fr 34/17b 352/2ab 126/165 H 1670:169 H 94/125e 14/23e 1380:39a 3770:53a 8 1-8utanol 460.73d 88.43c 428.87b 75.22c 410.75a 192.49c 105.2cc 109.98a 9 1-4kexanol 0.946 95.56a 155.25a 0.19b 55.74a 30.99a 1.10c 118.28a 10 (2)-2-4kexanol 49.967.25 ± 10.080.90 ± 17.724.51 ± 150.057.92 ± 137.929.84 ± 148.344.99 ± 139.006.55 ± 118.28a 11 (G)-3-4kexanol 456.76a ± 259.35 ± 360.65 ± 212.12a 229.24 ± 220.42a 229.41 20.046 33.26d 0.40b 23.212a 40.93.2a 13.50 ± 13.940.51 13.345.5 ± 23.12a 40.93.2a 12.815.5 ± 23.12a 40.93.2a 12.815.5 ± 23.12a 40.93.2a 12.815.5 ± 23.12a 40.93.2a 12.815.5 ± 12.81.7a ± 13.315.5 ± 13.81 ± 12.91.7a ± 12.91.7a ± 12.91.7a ±	_		10.700	$19.121.66 \pm$	22.092.28 ±	$22.190.69 \pm$	26.943.31 ±	28.630.37 ±	$24.564.13 \pm$	$35.357.18 \pm$			
1-Butanol 405.04 + 40.04 - 50.07 + 40.02 - 50.25 + 40.07 - 50.25 + 40.57 - 40.05 - 50.08 - 50.05 + 50.08 - 50.	7	Phenylethanol	tr	347.17b	352.72ab	1264.54ab	1698.19a	5447.26a	1350.39a	3770.63a			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	8	1-Butanol	4016.04 \pm	5701.70 \pm	6925.24 \pm	$6021.65 \ \pm$	9521.58 \pm	5445.54 \pm	5825.25 \pm	9353.15 \pm			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	0	1 Dutanor	460.75d	88.43c	482.87b	75.22c	419.76a	122.49c	315.26c	190.98a			
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	9	1-Hexanol	$181.20 \pm 0.84c$	$2111.98 \pm$	2041.12 ± 156.252	1839.81 ± 0.10b	1953.49 ±	$2069.26 \pm$	2087.22 ±	2043.43 ± 118.282			
			119.496.72 +	162.089.09 +	171.724.51 +	150.067.92 +	137.929.84 +	148.344.89 +	139.066.55 +	$134.223.71 \pm$			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	10	(Z)-2-Hexenol	13,318.96d	6305.69b	244.40a	460.51c	13,446.33c	4190.07c	3769.08c	50.87c			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	11	(F) 3 Hevenol	467.80 \pm	$259.35~\pm$	366.95 \pm	$205.86~\pm$	198.90 \pm	353.70 \pm	227.13 \pm	$\textbf{229.28} \pm$			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	11	(E)-3-Hexelloi	54.51a	13.81 cd	29.91b	11.90d	33.26d	0.40b	72.91d	20.04d			
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	12	Isoamyl acetate	11,926.03 ±	11,177.34 ±	11,175.21 ±	13,221.85 ±	12,240.36 ±	$10,722.43 \pm$	$12,005.82 \pm$	12,631.56 ±			
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		-	256.54a	447.86D 716.89 +	4/3.01D 717 22 +	248.91a 926 87 +	991.98a 790.60 +	243.74D 659.40 +	232.12a 819 23 +	463.32a 848 30 +			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	13	Isobutyl acetate	tr	11.94c	40.88c	30.52a	790.00 ⊥ 68.58b	58.85c	6.52b	17.60b			
$ \begin{array}{ c c c c } & \operatorname{acctate} & 21.84 & 13.19 & 10.30b & 6.87a & 20.25a & 1.35a & 11.59a & 28.51a \\ 1.015.555 & 56.11.98.65 & 67.111.21 & 87.0666 & 72.3062 & 12.357 & 74.241.85 & 80.050.2 & 1.746.80b \\ 1.010 & \operatorname{Buly} \operatorname{batanoste} & 1.015.555 & 56.10 & 10.809.59c & 17.877.96d & 20.926.73c & 17.426.80b \\ 1.020 & 330.94 & 1566 & 2.08bc & 5.51b & 5.67b & 24.04b & 0.71c & 4.13b & 87.15b \\ 1.566 & 2.08bc & 5.51b & 5.67b & 24.04b & 0.71c & 4.13b & 87.15b \\ 1.566 & 2.08bc & 5.51b & 5.67b & 24.04b & 0.71c & 4.13b & 87.15b \\ 1.010 & 1.010 & 1.020 & 0.010$	14	Phenylethyl	177.26 \pm	$213.51 \pm$	273.74 ±	$319.15 \pm$	324.39 ±	295.98 ±	303.54 ±	$315.41 \pm$			
$ \begin{array}{ c c c c c c } & \mbox{Err} & Err$	14	acetate	21.84d	13.19c	10.30b	6.87a	20.25a	1.35a	11.59a	28.51a			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	15	Ethyl acetate	1,015,555.75	651,198.63 ±	677,111.24 ±	875,066.60 ±	723,964.28 ±	612,561.73 ±	764,281.85 ±	803,603.20 ±			
$ \begin{array}{ c c c c c } \hline Ethyl butanoatec is 0.6.9 \pm 10.6.2 \pm 10.7.2 \pm 10.7.2 \pm 10.7.4 \pm 17.5.4 \pm 17.5.4 \pm 17.5.4 \pm 17.5.4 \pm 17.5.2 \pm 10.7.5.7 \pm 10.7.5.7 \pm 0.5.2.7 \pm 0.5.2.7 \pm 0.5.7 \pm 0.5$		2	\pm 81,572.36a	2750.08d	44,200.32d	25,650.67b	10,899.59c	17,877.96d	20,926.73c	17,426.80b			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	16	Ethyl butanoatec	15 66a	$130.42 \pm 2.08 \text{ bc}$	5.51b	$102.04 \pm 5.67b$	$173.41 \pm 24.04b$	$147.20 \pm 0.71c$	$102.79 \pm 4.13b$	$180.32 \pm$ 8 47b			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		Pd 11	101000	20,465.81 ±	$17,983.95 \pm$	$15,568.26 \pm$	$13,247.65 \pm$	$11,075.51 \pm$	9886.06 ±	9261.27 ±			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	17	Ethyl decanoate	tr	1044.51a	870.13b	1901.89b	729.99c	80.48d	1435.60d	264.52d			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	18	Ethyl hexanoate	1978.66 \pm	3494.91 \pm	$\textbf{3651.08} \pm$	3570.20 \pm	3315.79 \pm	3520.24 \pm	3692.70 \pm	3548.71 \pm			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	10	Etherland and the	159.74b	199.07a	79.11a	94.80a	220.96a	373.51a	71.80a	91.60a			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	19	Etnyi nonanoate	$2.32 \pm 0.40c$ 18 041 24 +	$8.44 \pm 0.40a$ 13 653 84 +	$9.59 \pm 0.39a$ 14 445 30 +	$10.01 \pm 0.85a$ 11.664.74 +	$6.97 \pm 1.72D$ 11.652.82 +	6.53 ± 0.060 12 501 25 +	5.81 ± 0.680 11 787 72 +	5.56 ± 0.280 11 377 01 +			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	20	Ethyl octanoate	458.62a	486.61c	24.42b	28.54d	1147.30d	355.18d	356.48d	35.61d			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	01	Ethyl lourate	$72.24~\pm$	$360.68~\pm$	434.98 \pm	744.27 \pm	356.35 \pm	195.25 \pm	$170.22~\pm$	129.63 \pm			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	21	Euryi laurate	35.98d	74.46b	45.13b	156.53a	141.67b	1.41c	39.75c	6.37c			
23Methyl salicylate1.30 ± 0.4623.99,458a1504,72a491,68a3.202,2ba1436,81a1523,13a3583,32a24Methyl octanoate6.84 ± 0.72d11.40 ± 0.64b12.73 ± 0.10b1.75 ± 0.15c1.37 ± 0.12c1.33 ± 0.01c1.18 ± 0.14c1.21 ± 0.17c25Diethyl376,56 ±535,00 ±611.67 ±666.95 ±694.73 ±630.81 ±627.92 ±785.18 ±26Ethyl 2- methylbutanoate10.00 ± 0.76b15.38 ± 0.08a14.99 ± 0.19a13.05 ± 0.01a10.95 ± 1.85a14.09 ± 3.86a12.78 ± 0.29a14.21 ± 1.15a27 α -Terpineol39.13 ± 6.64a4.72 ± 0.70b3.47 ± 0.42b5.33 ± 0.21b5.09 ± 0.14b4.10 ± 0.25b5.24 ± 0.24b5.55 ± 1.48b28Linalool4.26 ± 1.94b10.80 ± 0.19a10.08 ± 1.06a11.42 ± 0.07a12.47 ± 0.32a9.62 ± 0.13a11.95 ± 0.18a12.58 ± 0.89a29Acetic acid2707.37 ±1095.79 ±1074.68 ±908.83 ±970.46 ±1114.54 ±1441.42 ±2176.47 ±30Decanoic acid1020.52 ±1155.62 ±93.12 ±466.48 ±592.24 ±1115.51 ±813.53 ±726.02 ±31Hexanoic acid891.00 ±1100.74 ±1269.37 ±970.35 ±1320.25 ±1427.48 ±141.04 ±1276.47 ±32Octanoic acid891.00 ±1100.74 ±1269.37 ±950.81 ±957.61 ±86.63 ±23.62a23.60a33.62a31Hexanoic acid891.00 ±1100.74 ±<	22	Ethyl lactate	$13,385.62 \pm$	21,951.78 ±	22,046.12 \pm	23,955.44 ±	23,445.67 ±	18,699.44 ±	21,887.75 ±	25,549.06 ±			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	23	Methyl salicylate	894.620 1 30 ± 0.46c	379.68a $3.02 \pm 0.44a$	1504.72a 2 15 \pm 0 10b	491.68a 1 75 ± 0 15c	3202.26a 1 37 \pm 0 12c	1456.81a $1.33 \pm 0.01c$	1323.13a $1.18 \pm 0.14c$	3583.32a 1 21 \pm 0 17c			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	23	Methyl octanoate	$6.84 \pm 0.72d$	$11.40 \pm 0.64b$	12.73 ± 0.100 $12.73 \pm 0.26a$	11.50 ± 0.100	$9.86 \pm 0.02c$	$11.76 \pm 0.51b$	$11.29 \pm 0.06b$	$10.83 \pm 0.09b$			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	25	Diethyl	376.56 \pm	535.00 \pm	$611.67~\pm$	666.95 \pm	694.73 \pm	630.81 \pm	627.92 \pm	785.18 \pm			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	25	butanedioate	14.72c	16.25b	10.25a	30.12a	76.33a	44.90a	44.10a	168.46a			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	26	Ethyl 2-	$10.00\pm0.76b$	$15.38\pm0.08\mathrm{a}$	$14.99\pm0.19a$	$13.05\pm0.01a$	$10.95 \pm 1.85 a$	$14.09\pm3.86a$	$12.78\pm0.29a$	$14.21 \pm 1.15 \mathrm{a}$			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	27	methylbutanoate	30.13 ± 6.642	4.72 ± 0.70 b	3.47 ± 0.42 b	5.33 ± 0.21 b	5.09 ± 0.14 b	$4.10 \pm 0.25b$	5.24 ± 0.24 b	$5.55 \pm 1.48b$			
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	28	Linalool	$4.26 \pm 1.94b$	$10.80 \pm 0.99a$	$10.08 \pm 1.06a$	$11.42 \pm 0.07a$	$12.47 \pm 0.32a$	$9.62 \pm 0.13a$	$11.95 \pm 0.18a$	$12.58 \pm 0.89a$			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	20	A posti o poi d	2707.37 \pm	1095.79 \pm	1074.68 \pm	908.83 \pm	970.46 \pm	1114.54 \pm	1441.42 \pm	2176.47 \pm			
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	29	Acetic acid	233.07a	6.09 cd	69.17 cd	28.03d	77.15d	78.63 cd	263.59c	131.87b			
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	30	Decanoic acid	1020.52 ±	1155.62 ±	993.12 ±	466.48 ±	592.24 ±	1115.51 ±	813.53 ±	726.02 ±			
31 Hexanoic acid 30.00 ± 901.21 ± 770.30 ± $300.77 \pm$ $907.71 \pm$ $900.32 \pm$ $920.00 \pm$ $1200.33 \pm$ 32 Octanoic acid 891.00 ± 1100.74 ± 1269.37 ± 909.82 ± 1320.25 ± 1427.48 ± 1419.64 ± 1870.17 ± 33 Propanoic acid 79.14b 108.22a 94.59a 58.62b 1.83a 288.45a 231.68a 332.62a 33 Propanoic acid tr 22,079.23 ± 15,416.94 ± 13,725.39 ± 18,459.15 ± 18,416.19 ± 16,943.49 ± 18,741.35 ± 34 Benzaldehyde $\frac{162.84 \pm}{34.95a}$ 59.20 ± 0.45d 80.85 ± 1.52b 76.15 ± 1.68c 77.30 ± 2.67c 79.02 ± 0.65c 83.70 ± 0.51b 84.93 ± 1.07b 35 Furfural 1933.00 ± 18,978.76 ± 22,914.09 ± 6528.30 ± 5350.37 ± 14,291.14 ± 7873.77 ± 7641.65 ±			25.44a	316.34a	109.05a	22.29b	46.66b	373.49a	157.60a	53.65a			
32Octanoic acid891.00 \pm 1100.74 \pm 1269.37 \pm 909.82 \pm 1320.25 \pm 1427.48 \pm 1419.64 \pm 1870.17 \pm 33Propanoic acidtr22,079.23 \pm 15,416.94 \pm 13,725.39 \pm 18,459.15 \pm 18,416.19 \pm 16,943.49 \pm 18,741.35 \pm 34Benzaldehyde162.84 \pm 34.95a59.20 \pm 0.45d80.85 \pm 1.52b76.15 \pm 1.68c77.30 \pm 2.67c79.02 \pm 0.65c83.70 \pm 0.51b84.93 \pm 1.07b35Eurfural1933.00 \pm 18,978.76 \pm 22,914.09 \pm 6528.30 \pm 5350.37 \pm 14,291.14 \pm 7873.77 \pm 7641.65 \pm	31	Hexanoic acid	28 78ab	46.40a	770.30 ± 16 79b	75.03b	937.01 ± 72.65a	41.86b	920.00 ± 194.66a	$1200.03 \pm 173.35a$			
32 Octanoic acid 79.14b 108.22a 94.59a 58.62b 1.83a 288.45a 231.68a 332.62a 33 Propanoic acid tr 22,079.23 ± 15,416.94 ± 13,725.39 ± 18,459.15 ± 18,416.19 ± 16,943.49 ± 18,741.35 ± 34 Benzaldehyde $\frac{162.84 \pm}{34.95a}$ 59.20 ± 0.45d 80.85 ± 1.52b 76.15 ± 1.68c 77.30 ± 2.67c 79.02 ± 0.65c 83.70 ± 0.51b 84.93 ± 1.07b 35 Furfural 1933.00 ± 18,978.76 ± 22,914.09 ± 6528.30 ± 5350.37 ± 14,291.14 ± 7873.77 ± 7641.65 ±	00	Ostansis ssid	891.00 ±	1100.74 \pm	$1269.37 \pm$	909.82 ±	$1320.25 \pm$	1427.48 \pm	$1419.64 \pm$	$1870.17 \pm$			
33 Propanoic acid tr $22,079.23 \pm \\ 1456.67a$ $15,416.94 \pm \\ 110.50b$ $13,725.39 \pm \\ 449.87c$ $18,459.15 \pm \\ 1052.59b$ $18,416.19 \pm \\ 36.60b$ $16,943.49 \pm \\ 1852.34b$ $18,741.35 \pm \\ 1541.05b$ 34 Benzaldehyde $\frac{162.84 \pm }{34.95a}$ $59.20 \pm 0.45d$ $80.85 \pm 1.52b$ $76.15 \pm 1.68c$ $77.30 \pm 2.67c$ $79.02 \pm 0.65c$ $83.70 \pm 0.51b$ $84.93 \pm 1.07b$ 35 Furfural 1933.00 \pm $18,978.76 \pm $ $22,914.09 \pm $ $6528.30 \pm $ $5350.37 \pm $ $14,291.14 \pm $ $7873.77 \pm $ $7641.65 \pm $	32	Octanoic acid	79.14b	108.22a	94.59a	58.62b	1.83a	288.45a	231.68a	332.62a			
34 Benzaldehyde $162.84 \pm \\ 34.95a$ $59.20 \pm 0.45d$ $80.85 \pm 1.52b$ $76.15 \pm 1.68c$ $77.30 \pm 2.67c$ $79.02 \pm 0.65c$ $83.70 \pm 0.51b$ $84.93 \pm 1.07b$ 35 Furfural 1933.00 \pm 18,978.76 \pm 22,914.09 \pm 6528.30 \pm 5350.37 \pm 14,291.14 \pm 7873.77 \pm 7641.65 \pm 168c $77.30 \pm 2.67c$ $79.02 \pm 0.65c$ $83.70 \pm 0.51b$ $84.93 \pm 1.07b$	33	Propanoic acid	tr	22,079.23 ±	15,416.94 \pm	13,725.39 ±	18,459.15 \pm	18,416.19 \pm	16,943.49 ±	18,741.35 \pm			
34 Benzaldehyde $102.64 \pm 34.95a$ $59.20 \pm 0.45d$ $80.85 \pm 1.52b$ $76.15 \pm 1.68c$ $77.30 \pm 2.67c$ $79.02 \pm 0.65c$ $83.70 \pm 0.51b$ $84.93 \pm 1.07b$ 35 Furfural $1933.00 \pm 18,978.76 \pm 22,914.09 \pm 6528.30 \pm 5350.37 \pm 14,291.14 \pm 7873.77 \pm 7641.65 \pm 1.68c$ $77.30 \pm 2.67c$ $79.02 \pm 0.65c$ $83.70 \pm 0.51b$ $84.93 \pm 1.07b$		I	160.04	1456.67a	110.50b	449.87c	1052.59b	365.60b	1852.34b	1541.05b			
$\begin{array}{c} 35 \\ \text{Furfural} \end{array} \begin{array}{c} 35 \\ 1933.00 \pm \\ 18,978.76 \pm \\ 22,914.09 \pm \\ 6528.30 \pm \\ 5350.37 \pm \\ 14,291.14 \pm \\ 7873.77 \pm \\ 7641.65 \pm \\ $	34	Benzaldehyde	102.84 ± 34.95a	$59.20\pm0.45d$	$80.85 \pm 1.52 b$	$76.15 \pm \mathbf{1.68c}$	$77.30 \pm \mathbf{2.67c}$	$\textbf{79.02} \pm \textbf{0.65c}$	$83.70\pm0.51b$	$\textbf{84.93} \pm \textbf{1.07b}$			
35 Furtural			1933.00 ±	18,978.76 \pm	$22,914.09 \pm$	$6528.30 \pm$	5350.37 ±	14,291.14 \pm	7873.77 ±	7641.65 \pm			
159.78e 383.13a 375.31a 246.56d 168.45d 192.53b 283.23c 159.11c	35	Furtural	159.78e	383.13a	375.31a	246.56d	168.45d	192.53b	283.23c	159.11c			
	36	Nonanal	$26.92 \pm \mathbf{3.27a}$	$\textbf{3.19} \pm \textbf{0.11b}$	$\textbf{2.48} \pm \textbf{0.31b}$	$\textbf{2.00} \pm \textbf{0.27b}$	$2.82\pm0.14\text{b}$	$\textbf{3.02} \pm \textbf{0.25b}$	$\textbf{2.59} \pm \textbf{0.29b}$	$1.93 \pm 0.25 b$			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	37	Octanal	82.96 ±	$12.81\pm3.91\mathrm{b}$	$7.70 \pm 1.10 \mathrm{b}$	$8.16\pm2.32b$	$5.27\pm0.73c$	$3.55\pm0.05c$	$4.71\pm0.34c$	$9.26\pm0.80b$			
25.02a 66.88 +			25.02a	66 88 +									
38 Phenol $2.48 \pm 1.44d$ 30.00 ± 1 $43.71 \pm 4.02b$ $35.49 \pm 0.74c$ $34.90 \pm 0.38c$ $30.27 \pm 0.76c$ $31.96 \pm 0.91c$ $30.40 \pm 2.67c$	38	Phenol	$\textbf{2.48} \pm \textbf{1.44d}$	11.44a	$43.71\pm4.02b$	$\textbf{35.49} \pm \textbf{0.74c}$	$34.90 \pm \mathbf{0.38c}$	$30.27 \pm \mathbf{0.76c}$	$31.96 \pm \mathbf{0.91c}$	$30.40 \pm \mathbf{2.67c}$			

FO, French oak; AO, American oak; MO, Mongolian oak; JO, Japanese blue oak. 'tr' means trace. Different letters in the same row means significant differences among the spine grape brandy samples according to Duncan's multiple range tests (p < 0.05).

Table 3

Odor activity values (OAVs) of volatile compounds in the spine grape brandies after aging with wood chips.

Number	Volatile	Class	Aroma characteristics	Aroma series	Threshold ($\mu g L^{-1}$)	OAVs							
	compounds					Control	FO	AO	MO	JO	Chestnut	Catalpa	Cherry
1	Isobutanol	HA	fusel, solvent	1	28300 ⁵	31.98	47.52	51.36	53.29	51.09	47.45	50.93	54.70
2	Isoamyl alcohol	HA	pungent	1	179000 ⁵	8.43	13.63	14.11	13.48	13.57	13.64	14.32	14.53
3	1-Octanol	HA	alcoholic, fruity	1,3	900 ³	0.08	0.19	0.17	0.16	0.18	0.18	0.17	0.17
4	1-Propanol	HA	fusel, solvent	1	54000 ⁵	9.97	13.62	14.30	15.89	14.84	13.19	14.92	16.04
_	3-Methyl-1-			1.0	5005	1.00	1.05	1 00	1 50	1	1.00	1.00	1.00
5	pentanol 4-Methyl-1-	HA	alcoholic, harsh alcoholic, almond	1, 2	500°	1.02	1.87	1.82	1.72	1.77	1.83	1.82	1.88
6	pentanol	НА	roasted	1.2.5	1000^{3}	0.22	0.37	0.38	0.35	0.36	0.35	0.38	0.37
7	Phenylethanol	HA	floral	7	2600 ⁵	_	7.38	8.51	8.56	10.34	11.04	9.42	13.62
)		alcoholic.										
8	1-Butanol	C6	solvent	1	2730^{2}	1.47	2.09	2.54	2.21	3.49	1.99	2.13	3.43
9	1-Hexanol	C6	green, grass	2	8000 ⁵	0.02	0.26	0.26	0.23	0.24	0.26	0.26	0.26
10	(Z)-2-Hexenol	C6	grass, green	2	1257 ⁵	95.07	128.95	136.61	119.84	109.73	118.02	110.63	106.78
11	(E)-3-Hexenol	C6	grass, green	2	10005	0.47	0.26	0.37	0.21	0.20	0.35	0.23	0.23
			fruity, sweet,										
12	Isoamyl acetate	AE	honey	3, 4	245 ⁵	38.68	45.62	45.61	53.97	49.96	43.77	49.00	51.56
			fruity, pear,										
13	Isobutyl acetate Phenylethyl	AE	banana	3	1600 ³	-	0.45	0.45	0.58	0.49	0.41	0.51	0.53
14	acetate	AE	rose, honey pineapple,	3, 4, 7	250 ³	0.71	0.85	1.09	1.28	1.30	1.18	1.21	1.26
15	Ethyl acetate	EE	fruity, sweet	1, 3	32600^2	31.15	19.98	20.77	26.84	22.21	18.79	23.44	24.65
16	Ethyl butanoate	EE	fruity, sweet	3	9.51 ⁵	35.43	16.45	18.62	19.14	18.23	15.48	17.12	19.61
17	Ethyl decanoate	EE	floral	7	1120^{2}	-	18.27	16.06	13.90	11.83	9.89	8.83	8.27
18	Ethyl hexanoate	EE	fruity	3	305	65.96	116.50	121.70	120.01	110.53	117.34	123.09	118.29
19	Ethyl nonanoate	EE	fruity	3	3150^{2}	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01
20	Ethyl octanoate	EE	fruity	3	12.90^{2}	1398.55	1058.44	1119.79	904.24	903.32	969.09	913.78	882.01
21	Ethyl laurate	OE	fruity	3	-	-	-	-	-	-	-	-	-
22	Ethyl lactate	OE	fruity	3	128000^2	0.10	0.17	0.17	0.19	0.18	0.15	0.17	0.20
23	Methyl salicylate	OE	mint	-	-	-	-	-	-	-	-	-	-
24	Methyl octanoate Diethyl	OE	waxy, fruity	3	100 ⁴	0.07	0.11	0.13	0.12	0.10	0.12	0.11	0.11
25	butanedioate	OE	fruity	3	6000 ⁴	0.06	0.09	0.10	0.11	0.12	0.11	0.10	0.13
26	methylbutanoate	OF	fruity	3	$57 50^2$	0.17	0.27	0.26	0.23	0.10	0.25	0.22	0.25
20	a-Ternineol	Ter	floral lilac	7	250^3	0.17	0.02	0.01	0.23	0.12	0.02	0.22	0.23
27	Lipalool	Ter	rose floral	7	230 23 ⁵	0.10	0.02	0.01	0.02	0.02	0.02	0.02	0.02
20	Acetic acid	EA	vinegar	6	25 75521 ⁵	0.14	0.00	0.04	0.00	0.42	0.02	0.41	0.43
30	Decanoic acid	FA	fatty, sweaty	6	2800 ⁵	0.36	0.41	0.35	0.01	0.21	0.40	0.29	0.26
			sweaty,		-								
31	Hexanoic acid	FA	pungent	6	2520^{2}	0.35	0.38	0.31	0.34	0.38	0.34	0.37	0.51
32	Octanoic acid	FA	fatty, sweaty	6	$2700^{3,5}$	0.33	0.41	0.47	0.34	0.49	0.53	0.53	0.69
33	Propanoic acid	FA	fatty, sweaty	6	8100 ⁵	-	2.73	1.90	1.69	2.28	2.27	2.09	2.31
34	Benzaldehyde	Car	sharp, cherry roasted, smoky,	7	4200 ⁵	0.04	0.01	0.02	0.02	0.02	0.02	0.02	0.02
35	Furfural	Car	sweet, almond green, floral,	4, 5	5800 ²	0.30	2.27	3.95	1.38	0.93	2.43	1.37	1.30
36	Nonanal	Car	fruity	2, 3, 7	15^{3}	1.79	0.21	0.17	0.13	0.19	0.20	0.17	0.13
37	Octanal	Car	fatty	6	39.60 ¹	2.09	0.32	0.19	0.21	0.13	0.09	0.12	0.23
38	Phenol	VP	phenol, medicinal		18900 ²	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01

The categories for volatile compounds: HA, higher alcohols; C6, C6 alcohols; AE, acetate esters; EE, ethyl esters; OE, other esters; Ben, benzenes; Ter, terpenes; FA, fatty acids; Car, carbonyl compounds; VP, volatile phenols. The classification standard of aroma series referenced according to Xiang et al., 2020. 1, fusel/solvent; 2, green/ plant; 3, fruity; 4, sweet; 5, roasted/smoky; 6, sweaty/fatty; 7, floral. The references for odor thresholds: 1, Fan et al. (2015); 2, Gao et al. (2014); 3, Kong et al. (2019); 4, Peng, Wen, Tao, and Lan (2013); 5, Xiang et al. (2020). OAVs, Odor activity values. FO, French oak; AO, American oak; MO, Mongolian oak; JO, Japanese blue oak.

26.84). However, the OAVs of ethyl octanoate, ethyl butanoate, ethyl acetate, nonanal, and octanal were greater in the control samples than in the wood-aged brandies. These results were consistent with previous observations that ethyl butanoate (OAV: 1894) had the highest OAV in the head of distilled spirits from spine grape wines, followed by ethyl hexanoate (OAV: 1086), ethyl octanoate (OAV: 425), and isoamyl acetate (OAV: 106) (Xiang et al., 2020), although some of these compounds were present at relatively low concentrations in spine grape brandies after wood aging. Several different groups of volatile compounds, such as higher alcohols, esters, terpenes, fatty acids, carbonyl compounds, and volatile phenols, have significant effects on the final aroma characteristics of the resulting brandy products (Tao et al., 2014; Xiang et al.,

2020).

Alcohols are produced primarily through the process of ethanolic fermentation and provide alcohol products with alcoholic odors and some grass/green attributes. C6 compounds (especially 2-hexenal) are characteristic flavor components of various spine grape (*V. davidii* Foex) clones (Meng et al., 2013). In this study, the OAVs of higher alcohols and C6 alcohols, including (*Z*)-2-hexenol (from 106.78 to 136.61), isobutanol (from 47.45 to 54.70), isoamyl alcohol (from 13.48 to 14.53), 1-propanol (from 13.19 to 16.04), phenylethanol (from 7.38 to 13.62), 1-butanol (from 1.99 to 3.49), and 3-methyl-1-pentanol (from 1.72 to 1.88), markedly increased in brandies aged with wood chips. These alcohols have also been identified as key odor-active compounds in many

wines, such as brandy (Coldea et al., 2020), baijiu (Fan, Fan, & Xu, 2015; Gao, Fan, & Xu, 2014), and spine grape spirits (Xiang et al., 2020). Compared to those of the control samples, the OAVs of isoamyl alcohol and 3-methyl-1-pentanol were greater in the brandies aged with FO and AO, chestnut, catalpa, and cherry chips, whereas the OAVs of isobutanol and 1-propanol were greater in the brandies aged with MO and cherry chips. Phenylethanol and 1-butanol were relatively high OAVs in the brandies aged with cherry and JO chips. (Z)-2-Hexenol had the highest OAVs in the brandies aged with AO chips, followed by those aged with FO chips. These results indicated that the concentration and proportion of odor-active compounds in alcohol products are affected not only by fermentation and distillation technologies but also by the type of oak wood used in the aging process. Previous studies have shown that higher alcohols, such as 1-propanol, 1-butanol, and 2-methyl-1-propanol, positively influence the aroma complexity of wines or distillates in a specific proportion, whereas at relatively high concentrations, these compounds are characterized by penetrating odors that mask the aromatic finesse (Tsakiris, Kallithraka, & Kourkoutas, 2014; Xiang et al., 2020)

Ester compounds are also particularly important since they mainly provide fruity and floral attributes. The OAVs of most acetate esters, ethyl esters, and other esters, such as ethyl hexanoate (from 110.53 to 123.09), isoamyl acetate (from 43.77 to 53.97), ethyl decanoate (from 8.27 to 18.27), and phenylethyl acetate (from 0.85 to 1.30), markedly increased after the brandies were aged with wood chips. The OAVs of isoamyl acetate, phenylethyl acetate, and ethyl acetate were greater in the brandies aged with MO, cherry, catalpa, and JO chips than in those aged with other chips. However, ethyl decanoate, ethyl octanoate, and ethyl 2-methylbutanoate were more abundant in the brandies aged with FO and AO chips. Moreover, the OAVs of ethyl hexanoate were the highest in the brandies aged with catalpa chips, followed by those aged with AO, MO, and cherry chips. The esters contributed to a pleasant fruity and floral aroma at a specific quantity in brandies aged with French and American oak chips (Bozalongo, Carrillo, Torroba, & Tena, 2007; Yan et al., 2024). In apple brandies, ethyl acetate significantly decreased in cherry and mulberry aged brandies, but a 2.3 % increase was observed in sessile aged brandies (Coldea et al., 2020), indicating that the type of wood also has a decisive effect on ester compounds.

In addition, fatty acids, such as octanoic acid, propanoic acid, and decanoic acid, provide fatty and sweaty attributes (Tsakiris et al., 2014; Xiang et al., 2020). In this study, the OAVs of propanoic acid were the highest in spine grape brandies aged with FO chips, followed by those aged with JO, chestnut, and cherry chips. Although fatty acids are often thought to confer an undesirable aroma, these compounds produce these aroma attributes only at concentrations above 20 mg L^{-1} (Zhao et al., 2012). In the spine grape brandies, the OAVs of linalool were greater in the brandies aged with JO, chestnut, and cherry chips than in the other brandy groups, and the OAVs of furfural were greater in the brandies aged with AO, FO, and chestnut chips. Previous studies have shown that the extraction of furfural from wood is greater than the conversion of furfural to form the corresponding alcohols during short aging periods; thus, furfural tends to accumulate in wine and confers a pleasant wood, almond, caramel and vanilla-like aroma (Bautista-Ortín et al., 2008; Flamini et al., 2021). In addition, the volatile profiles of apple brandies aged with different wood chips revealed that cherry wood caused an \sim 15 % decrease in furfural content, but sessile oak caused an \sim 5 % increase, suggesting that the effects of different woods on the concentration of furfural differ depending mainly on the wood species (Coldea et al., 2020).

Collectively, these findings establish the aroma potential of oaks or substitute woods in spine grape brandy during the aging process.

3.3. Effect of wood chip addition on volatile compounds in spine grape brandies

Clustered heatmap and principal component analysis (PCA) were

constructed and independently performed to obtain an overall view of the effects of various wood chips on volatile compounds in spine grape brandies (Fig. 1 and Fig. 2). Hierarchical clustering analysis classified the five sets of samples into two major groups (Fig. 1). The first group included ethyl esters (ethyl acetate, ethyl butanoate, and ethyl octanoate), carbonyl compounds (nonanal, octanal, and benzaldehyde), terpenes (α -terpineol), fatty acids (acetic acid), and alcohols ((E)-3hexenol), which corresponded to high levels in the control samples and low levels in the wood-aged brandy samples. The second group was subdivided into three subgroups due to the relatively high and different contents of most volatile compounds in the wood-aged brandy samples as a result of aging with different species of wood chips. Specifically, high levels of ethyl esters (ethyl laurate, ethyl decanoate, and ethyl nonanoate), other esters (ethyl 2-methylbutanoate and methyl salicylate), carbonyl compounds (furfural), C6 alcohols ((Z)-2-hexenol), and phenol were observed in the brandies aged with FO and AO chips, followed by those aged with MO chips. High levels of higher alcohols (1octanol, 3-methyl-1-pentanol, isoamyl alcohol, 4-methyl-1-pentanol, isobutanol, 1-propanol, and phenylethanol), fatty acids (propanoic acid), C6 alcohols (1-hexanol and 1-butanol), ethyl esters (ethyl hexanoate, isobutyl acetate, and phenylethyl acetate), other esters (methyl octanoate, ethyl lactate, and diethyl butanedioate), and terpenes (linalool) were detected in the brandies aged with FO, AO, MO, JO, chestnut, catalpa, and cherry chips, although the concentrations of these compounds varied in these brandy samples. Additionally, high levels of acetate esters (isoamyl acetate) and fatty acids (hexanoic acid and octanoic acid) were observed in the brandies aged with cherry chips; and their levels were relatively low after aging with FO, AO, and chestnut chips, indicating that the volatile composition of different species of wood chips differentiated the chemical characteristics of spine grape brandies, in agreement with previous studies (Coldea et al., 2020; Tao et al., 2014; Yan et al., 2024).

Unsupervised PCA was also performed on those volatile compounds that presented significant differences in wood-aged brandy samples to determine which compounds produced the greatest variability among the brandies aged with various species of wood chips (Fig. 2). The first



Fig. 1. Hierarchical clustering analysis of the volatile compounds obtained from spine grape brandies after aging with different wood chips.



Fig. 2. Principal component analysis of the volatile compounds (represented by the numbers presented in Table 2) from the spine grape brandies aged with different wood chips.

two components (PC1 and PC2) explained for 70.1 % of the data variance (PC1: 57.2 % and PC2: 12.9 %), representing the largest fraction of variability. The wood-aged brandy samples were distributed in the positive direction of PC1 despite these brandies being aged with different types of wood, while the control samples were distributed in the negative direction of PC1. The high loading values obtained for phenylethyl acetate, isoamyl acetate, octanoic acid, 1-butanol, ethyl lactate, isobutanol, phenylethanol, and 1-propanol in the positive directions of both PC1 and PC2 were the main contributors to the brandies aged with cherry, JO, and chestnut samples, with more pronounced fruity, honey, and floral attributes. In contrast, furfural, methyl salicylate, phenol, ethyl decanoate, ethyl 2-methylbutanoate, ethyl decanoate, 3-methyl-1-pentanol, and decanoic acid in the positive direction of PC1 and the negative direction of PC2 corresponded to the brandies aged with FO, AO, and chestnut chips with more pronounced roasted/ smoky, floral, and fruity attributes. In addition, isobutanol, 1-propanol, isobutyl acetate, and isoamyl alcohol in the positive direction of PC1 and the origin of PC2 corresponded to the brandies aged with MO chips, resulting in a greater fusel, solvent, and fruity aroma.

As mentioned above, previous studies have shown that due to differences in the volatile components of different wood species, the concentrations and compositions of aromatic compounds extracted from wood during the aging process differ among brandies (Caldeira et al., 2010; Coldea et al., 2020; Tao et al., 2014). Compared to the brandies aged with Portuguese chestnut wood (C. sativa), the wines-brandies aged with French oak (Q. robur) chips or staves extracted significant quantities of ethyl 2-methylpropanoate, ethyl butyrate and ethyl octanoate and lower levels of butanoic acid, syringol and cis- β -methyl- γ -octalactone during the aging period (Caldeira et al., 2010). In apple brandies, some researchers found that the main ester contributors, such as ethyl acetate, isobutyl acetate, and isoamyl-2-methylbutyrate, significantly increased in content with age in the presence of mulberry (M. alba), chestnut (C. sativa), and cherry (P. avium) chips; hexanal, ethyl-4-decenoate, and homovanilic acid were solubilized only in fir (A. alba) wood-aged brandy (Coldea et al., 2020).

3.4. Sensory properties of spine grape brandies

The transfer of compounds from wood chips to beverages through wood utilization modifies not only the chemical composition but also the sensory properties of alcoholic beverages (Tao et al., 2014). To identify the discrimination power of the sensory attributes of spine grape

brandies aged with wood chips, a descriptive sensory analysis was additionally applied for the characterization test. Spine grape brandies aged with different species of wood chips presented distinct sensory attributes compared to those of the control samples (Fig. 3). In this study, the color of all wood-aged brandies was observed noticeably yellow or amber color (Fig. 3A). The wood-aged brandies were perceived to have significantly greater color intensity, aroma quality (purity, intensity, and persistence), and taste quality (purity, intensity, and persistence) than the control samples (Fig. 3B). Among these woodaged brandy samples, the brandies aged with MO (8.18), JO (8.00), and catalpa (8.54) chips presented lower color intensities and differed from the brandies aged with FO (9.64), AO (9.46), chestnut (9.62), and cherry (9.10) chips, which presented higher color intensities, indicating that the wood chip additives could enhance the color of the spine grape brandies during the aging period. In addition, the aroma and taste intensity of the brandies aged with AO (9.10 and 8.30) and FO (9.15 and 8.80) chips were the perceived high average scores, followed by those of the samples aged with chestnut (9.05 and 8.10), catalpa (8.95 and 7.90), MO (8.80 and 8.05), and cherry (8.70 and 7.75) chips, whereas the aroma and taste intensity of the brandies aged with JO chips were the lowest, with values of 7.80 and 7.00, respectively. These results are consistent with those of Coldea et al. (2020), who reported that volatile and phenolic compounds extracted from different types of wood could enhance the aroma and taste characteristics of wood-aged apple brandies during a short aging period. The toasted chip-treated wine produced more taste sensation and had more grassy/vegetal and roasted/smoky odors in the wines because of the extraction of wood-derived volatile and phenolic compounds during the exchange of oak/substitute wood and wine or its liquor (Coldea et al., 2020; García-Carpintero, Gallego, Sánchez-Palomo, & Viñas, 2012; Ortega-Heras et al., 2010). Moreover, the evolution of the redox potential, which reflects the oxidation-reduction reaction in red wines during the aging process, has also been shown to be different in different aging systems, such as oak chips, oak staves, and barrels (del Alamo, Nevares, & Cárcel, 2006), resulting in notable differences in wine style and characteristics. Previous studies have shown that wood botanical species are more discriminant than aging systems (Caldeira et al., 2010; Canas et al., 2019). Among the wood species, the brandies aged with chestnut wood presented higher intensities of topaz color and lower intensities of yellow-straw color than did the brandies aged with Limousin oak (Caldeira et al., 2010). Generally, the sensory attributes of aged wines and brandies, such as color, aroma, and taste, depend upon a combination of several factors, such as the wood species, type of oak toasting, and wine-wood contact time (Cadahía, De Simón, & Jalocha, 2003; Cerdán & Ancín-Azpilicueta, 2006; Tao et al., 2014).

Compared with the absence of wine chips, the addition of oak chips improved aroma complexity and modified sensory descriptors (Alencar et al., 2019; Coldea et al., 2020). According to aroma descriptors of the main volatile compounds, including fusel/solvent, green/plant, fruity, sweet, roasted/smoky, sweaty/fatty, and floral attributes, the total OAVs logarithmic values of the different aromatic series were calculated separately, standardized, and showed significant differences between the brandies aged with different species of wood chips (Fig. 3C). Compared with the control samples, the brandies aged with AO, FO, and chestnut chips, as expected, presented the three highest intensities of fruity, floral, and roasted/smoky characters, which might be ascribed to the high levels of (Z)-2-hexenol, ethyl decanoate, ethyl 2-methylbutanoate, ethyl hexanoate, and ethyl octanoate, and furfural, according to the OAV results (Table 3); the brandies aged with JO and cherry chips with more fruity, honey, and floral attributes might be ascribed to the high levels of 1-butanol, 1-propanol, phenylethanol, phenylethyl acetate, isoamyl acetate, and linalool. However, the control samples showed the lowest intensity of roasted/smoky, sweet, and floral aromas but were still dominated by fruity aroma characters, which might be attributed to the high levels of ethyl esters (ethyl acetate, ethyl butanoate, ethyl octanoate) and carbonyl compounds (nonana and octanal).



Fig. 3. Sensory properties of spine grape brandies after aging with different wood chips. (A) Samples of spine grape brandies. FO, French oak; AO, American oak; MO, Mongolian oak; JO, Japanese blue oak. (B) Sensory evaluation. (C) Aroma attributes. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

These findings are in agreement with those of Alencar et al. (2019), who reported that the addition of oak chips to wine also imparts the sensations of a vegetal and spicy aroma, sweetness/caramelization, taste persistence, and an alcoholic flavor. Syrah wines aged with mediumtoasted American oak chips seem to have relatively exotic aromas of coffee and woody attributes, whereas French oak chips impart more of a perception of sweetness and provide a more elegant and balanced aroma with vanilla, nut, and spice notes (Alencar et al., 2019). In addition, previous studies have shown that chestnut chips significantly increase the vanillin content in wood-aged apple brandy and showed excellent qualities for this type of oenological product (Canas et al., 2019; Coldea et al., 2020).

4. Conclusions

Aging with seven species of wood chips, including FO, AO, MO, JO, chestnut, catalpa, and cherry, markedly increased the levels of tannins, polyphenols and most volatile compounds in spine grape brandies. The chemical and sensory profiles of brandies aged with chestnut chips were more similar to those of brandies aged with FO and AO chips than to those of brandies aged with other wood species. The brandies aged with FO, AO and chestnut chips presented high color intensities (low L* values and high b^{*}, C^{*} and ΔE values), tannin and polyphenol contents, and aroma and taste qualities. Concerning the volatile composition of the spine grape brandies, the volatile compounds represented by ethyl decanoate, ethyl 2-methylbutanoate, ethyl octanoate, methyl salicylate, (Z)-2-hexenol, and furfural contributed to the brandies aged with FO, AO, and chestnut chips with more floral, fruity, and roasted/smoky attributes, while 1-butanol, 1-propanol, phenylethanol, phenylethyl acetate, isoamyl acetate, and linalool contributed to the brandies aged with JO and cherry chips with more fruity, honey, and floral attributes. Therefore, these findings could provide potential technology for choosing a range of wood chips to produce differentiated and highquality spine grape brandies.

CRediT authorship contribution statement

Bingbing Duan: Writing – original draft, Visualization, Investigation, Data curation. **Wei Chang:** Methodology, Investigation, Data curation. **Leqi Zhang:** Investigation, Data curation. **Mingyuan Zheng:** Visualization, Formal analysis. **Chenxing Su-Zhou:** Software. **Hasmik Merkeryan:** Writing – review & editing. **Meilong Xu:** Validation, Supervision. **Xu Liu:** Writing – review & editing, Resources, Project administration, Funding acquisition, Conceptualization.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

Data will be made available on request.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.fochx.2024.101777.

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