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Data Article

Data on vibrational spectra of the langasites $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ ($\text{Ln} = \text{La}, \text{Pr}, \text{Nd}$) and *ab initio* calculations



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ABSTRACT

In “Lattice dynamics and structure of the new langasites $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ ($\text{Ln} = \text{La}, \text{Pr}, \text{Nd}$): vibrational spectra and *ab initio* calculations” [1], experimental and calculated results on lattice dynamics of the recently discovered new compounds $\text{La}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$, $\text{Pr}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$, and $\text{Nd}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ are reported. These compounds belong to the langasite series and constitute a new class of low-dimensional antiferromagnets. The data presented in this article includes IR diffuse transmission spectra of powder samples of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ ($\text{Ln} = \text{La}, \text{Pr}, \text{Nd}$) registered at room temperature with a Bruker 125HR Fourier spectrometer, Raman spectra taken in the backscattering geometry (also at room temperature) with a triple monochromator using the line 514, 5 nm of an argon laser as an excitation, results of the DFT calculations with the B3LYP and PBE0 hybrid functionals on the optimized crystal structures, eigenfrequencies and eigenvectors of the normal vibrational modes. These data can be used to analyse electron-phonon interaction and multiferroic properties of the new langasites and to compare the lattice dynamics of different langasites.

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The dataset is available on mendeley data public repository at <https://doi.org/10.17632/32grbb4p82.1>.

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Specifications Table

Subject	Materials Science
Specific subject area	Electronic, Optical and Magnetic Materials
Type of data	Table Figure Text file
How data were acquired	IR spectra were collected in diffuse transmission mode with Bruker 125HR Fourier spectrometer, Raman spectra were collected in the backscattering geometry with a home-made triple monochromator using the line 514, 5 nm of an argon laser as an excitation. The CRYSTAL14 program designed for simulating periodic structures in the MO LCAO approximation was used for DFT <i>ab initio</i> calculations. Quasi-relativistic pseudopotentials ECP46MWB, ECP59MWB, and ECP60MWB with corresponding valence basis sets ECPnMWB were taken for La, Pr, and Nd. All-electron basis sets of TZVP type were used for Cr, Ge, Be, and O.
Data format	Raw Analyzed
Parameters for data collection	Spectra were collected on powder samples at room temperature. Calculations were performed within the framework of MO LCAO approach and the density functional theory, by using B3LYP and PBE0 hybrid functionals which takes into account both the local and nonlocal (in the Hartree–Fock formalism) exchange.
Description of data collection	Infrared diffuse transmission spectra were registered with a Bruker 125HR Fourier spectrometer equipped with a DTGS and a liquid-nitrogen-cooled MCT detectors. Raman spectra were collected in the backscattering geometry with a home-made triple monochromator ($\lambda_{\text{excit}} = 514,5 \text{ nm}$). The frequencies and eigenvectors of the normal vibrational modes were obtained from <i>ab initio</i> calculations. First, the full geometry optimization (atomic positions and all unit-cell parameters) was carried out. Then, the phonon spectrum (at the Γ^- point) was calculated for the crystal structure corresponding to the minimum energy.
Data source location	Institute of Spectroscopy, Russian Academy of Sciences, Troitsk, Moscow, Russian Federation 55.464596°N37.297538°E
Data accessibility	Repository name: Mendeley Data Data identification number: 32grbb4p82.1 Direct URL to data: https://doi.org/10.17632/32grbb4p82.1
Related research article	N.N. Kuzmin et al., Lattice dynamics and structure of the new langasites $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ (Ln = La, Pr, Nd): vibrational spectra and <i>ab initio</i> calculations, Journal of Physics and Chemistry of Solids, In Press

Value of the Data

- These data can be used to compare the lattice dynamics of different langasites.
- These data can be used by researchers working on vibrational and magnetoelectric properties of langasites.
- These data can be used to analyse electron-phonon interaction and multiferroic properties of the new langasites.

1. Data description

The dataset includes 6 text files for our measured infrared (IR) and Raman spectra of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ (Ln = La, Pr, Nd, raw data) [1]. These text files are named by rare-earth (RE) element symbol plus the method used to take the spectrum, e.g., Pr_IR.txt means an IR spectrum of $\text{Pr}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$. Each

text file has two columns which correspond to wave number (unit: cm^{-1}) and IR absorbance or Raman intensity (in arbitrary units). The same data are presented also as Excel files, e.g., Pr_IR.xlsx.

The data of *ab initio* calculations of optimized crystal structures is provided in 5 Excel tables. Table 1 provides the coordinates of atoms in the unit cell for the optimized structures of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ ($\text{Ln} = \text{La}, \text{Pr}, \text{Nd}$), calculated with the B3LYP hybrid functional. Table 2 provides the interatomic distances for the optimized structures of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ ($\text{Ln} = \text{La}, \text{Pr}, \text{Nd}$), calculated with the B3LYP hybrid functional. Table 3 provides the lattice constants for the optimized structures of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$

Table 1

Calculated (B3LYP) coordinates of atoms in the unit cell of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ ($\text{Ln} = \text{La}, \text{Pr}, \text{Nd}$). The experimental data for $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ [14] are shown in square brackets.

Ion	site	Ln = La			Ln = Pr			Ln = Nd		
		x/a	y/b	z/c	x/a	y/b	z/c	x/a	y/b	z/c
Ln	3e	0.42858 [0.42983(4)]	0.	0.	0.42797	0.	0.	0.42776	0.	0.
Cr	1a	0.	0.	0.	0.	0.	0.	0.	0.	0.
Ge	3f	0.74264 [0.74350(8)]	0.	0.5	0.74243	0	0.5	0.74235	0.	0.5
Be	2d	1/3	2/3	0.52202 [0.5260(10)]	1/3	2/3	0.52576	1/3	2/3	0.52735
O1	2d	1/3	2/3	0.20577 [0.1973(9)]	1/3	2/3	0.20729	1/3	2/3	0.20790
O2	6g	0.46708 [0.4671(4)]	0.30292 [0.3049(3)]	0.32707 [0.3251(5)]	0.46748	0.30543	0.32133	0.46770	0.30667	0.31882
O3	6g	0.22247 [0.2256(3)]	0.09505 [0.0966(3)]	0.75926 [0.7571(4)]	0.22352	0.09250	0.75799	0.22402	0.09133	0.75743

Table 2

Calculated (B3LYP) and experimentally determined [14] (in square brackets) M – O distances (Å) in the structure of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ ($\text{Ln} = \text{La}, \text{Pr}, \text{Nd}$).

	Ln = La	Ln = Pr	Ln = Nd
R–polyhedron			
R–O1 × 2	2.637 [2.577(2)]	2.621	2.613
R–O2 × 2	2.520 [2.457(4)]	2.473	2.451
R–O2' × 2	2.850 [2.816(3)]	2.827	2.818
R–O3 × 2	2.488 [2.450(3)]	2.449	2.431
(R–O) _{av}	2.624 [2.575]	2.593	2.578
Cr–octahedron			
Cr–O3 × 6	1.987 [1.979(2)]	1.984	1.983
Ge–tetrahedron			
Ge–O2 × 2	1.784 [1.760(4)]	1.781	1.780
Ge–O3 × 2	1.774 [1.733(3)]	1.775	1.775
(Ge–O) _{av}	1.779 [1.747]	1.778	1.778
Be–tetrahedron			
Be–O1	1.586 [1.622(6)]	1.589	1.590
Be–O2 × 3	1.698 [1.672(3)]	1.694	1.692
(Be–O) _{av}	1.670 [1.660]	1.668	1.667

Table 3

Experimentally determined [14] and calculated (PBE0) lattice constants (Å) of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$.

$\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$		a	c
$\text{La}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$	Exp.	8.033(2)	4.934(2)
	Calc.	8.0622	4.9680
$\text{Pr}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$	Exp.	7.957(2)	4.904(2)
	Calc.	7.9968	4.9433
$\text{Nd}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$	Exp.	7.931(2)	4.894(2)
	Calc.	7.9683	4.9323

(Ln = La, Pr, Nd), calculated with the PBE0 hybrid functional. Table 4 provides the coordinates of atoms in the unit cell for the optimized structures of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ (Ln = La, Pr, Nd), calculated with the PBE0 hybrid functional. Table 5 provides the interatomic distances for the optimized structures of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ (Ln = La, Pr, Nd), calculated with the PBE0 hybrid functional. In Tables 1–5, available experimental data are in square brackets.

The dataset includes 3 text files for the calculated with the B3LYP hybrid functional frequencies of normal modes and their intensities in the IR and Raman spectra. These text files are named by RE element symbol plus the method to get the data, e.g., Pr_abinit.txt means the calculated data for $\text{Pr}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$. Each text file has four columns which correspond to the symmetry of the mode (irreducible representation), wave number (unit: cm^{-1}), IR intensity, Raman intensity (arb. units). First, all A_1 modes are listed, they are followed by the A_2 and, then, E modes. The same data are presented also as Excel files, e.g., Pr_abinit.xlsx. Three Excel Tables, Table 6, Table 7, and Table 8, provide all calculated modes compared with those found from the measured spectra (analyzed data), in increasing order of their frequency for $\text{La}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$, $\text{Pr}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$, and $\text{Nd}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$, respectively. Mode symmetries are indicated.

The data on calculated displacements of different atoms in normal crystal modes of different frequencies for $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ (Ln = La, Pr, Nd) is provided in 3 text files named, e.g., Pr_displ.txt. Each

Table 4

Calculated (PBE0) and experimentally determined [14] (in square brackets) coordinates of atoms in the unit cell of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ (Ln = La, Pr, Nd).

Ion	site	Ln = La			Ln = Pr			Ln = Nd		
		x/a	y/b	z/c	x/a	y/b	z/c	x/a	y/b	z/c
Ln	3e	0.43071 [0.42983(4)]	0	0	0.43021	0	0	0.42976	0	0
Cr	1a	0	0	0	0	0	0	0	0	
Ge	3f	0.74464 [0.74350(8)]	0	0.5	0.74435	0	0.5	0.74428	0	0.5
Be	2d	1/3	2/3	0.52203 [0.5260(10)]	1/3	2/3	0.52539	1/3	2/3	0.52713
O1	2d	1/3	2/3	0.20488 [0.1973(9)]	1/3	2/3	0.20613	1/3	2/3	0.20711
O2	6g	0.46553 [0.4671(4)]	0.30026 [0.3049(3)]	0.32701 [0.3251(5)]	0.46600	0.30279	0.32134	0.46618	0.30408	0.31883
O3	6g	0.22215 [0.2256(3)]	0.09317 [0.0966(3)]	0.75867 [0.7571(4)]	0.22329	0.09093	0.75757	0.22364	0.08958	0.75679

Table 5

Calculated (PBE0) and experimentally determined [14] (in square brackets) M – O distances (Å) in the structure of $\text{Ln}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ (Ln = La, Pr, Nd).

	Ln = La	Ln = Pr	Ln = Nd
R–polyhedron			
R–O1 × 2	2.601 [2.577(2)]	2.584	2.578
R–O2 × 2	2.496 [2.457(4)]	2.449	2.429
R–O2' × 2	2.811 [2.816(3)]	2.788	2.779
R–O3 × 2	2.468 [2.450(3)]	2.430	2.412
(R–O) _{ep}	2.594 [2.575]	2.563	2.550
Cr–octahedron			
Cr–O3 × 6	1.966 [1.979(2)]	1.963	1.963
Ge–tetrahedron			
Ge–O2 × 2	1.769 [1.760(4)]	1.766	1.764
Ge–O3 × 2	1.756 [1.733(3)]	1.756	1.756
(Ge–O) _{ep}	1.762 [1.747]	1.761	1.760
Be–tetrahedron			
Be–O1	1.576 [1.622(6)]	1.578	1.579
Be–O2 × 3	1.683 [1.672(3)]	1.678	1.677
(Be–O) _{ep}	1.656 [1.660]	1.653	1.653

Table 6Experimentally determined [1] and calculated (B3LYP) frequencies in the Raman (R) and infrared (IR) spectra of $\text{La}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$.

Exp, R	Calculated			Exp, IR	Exp, R	Calculated			Exp, IR
	A1	E	A2			A1	E	A2	
Raman – active					Raman – active				
		IR – active				IR – active			
		88				396		400	
108		105		100	406				
108		109			433		425	424	
122	127				462		456	456	
			130	133	488		482		
143		141			544			506	
			156		568		532	551	
159		164		167	586	561			
190		193		189	625			580	
	212				625	583			
			213	216			624	625	
233		235					661		
		241		245			715	722	
259		269		265	730		728	722	
	284				783			741	
			287		783		781	789	
292		290		294	783	785			
292	292						809		
326		328						816	
			332			818			
351		343		344				825	
378		376		384	836	839		821	

Table 7Experimentally determined [1] and calculated (B3LYP) frequencies in the Raman (R) and infrared (IR) spectra of $\text{Pr}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$.

Exp, R	Calculated			Exp, IR	Exp, R	Calculated			Exp, IR
	A1	E	A2			A1	E	A2	
Raman – active					Raman – active				
		IR – active				IR – active			
			83	89			398	404	
		96		89	409	401			
108		108			435		431	430	
123	127				462		459	462	
			131	135	489		485		
145		141			548			509	
			150		575		538	556	
161		164		166	587	565			
193		194		188	631			583	
	215				631	587			
			216				632	633	
235		235					668		
		243		245			721	728	
260		265		263				743	
	284				732		734,8	743	
			286		783		785,6	794	
295		294		297	783	785,7			
295	297						813		
		327				818			
			337					821	
355		348		352				823	
382		378		385	836	843		823	

Table 8Experimentally determined [1] and calculated (B3LYP) frequencies in the Raman (R) and infrared (IR) spectra of Nd₃CrGe₃Be₂O₁₄.

Exp, R	Calculated			Exp, IR	Exp, R	Calculated			Exp, IR
	A1	E	A2			A1	E	A2	
Raman – active				IR – active					
			80					400	405
		92		90	410	404			
108		108			436		434		435
123	127				463		461		463
			131	138	490		486		
146		141						505	512
			148		549		540		560
160		165		166	572	569			
191		194		191				585	
	217				589	590			
			218		626		635		634
236		236					671		
		244		247			723		731
261		264		265	734			735	746
	284						738		
			286		784	786			794
296		295		301	784		787		794
	301						814		
330		327				817			
			340						820
356		351		354					826
382		380		385	838	845			823

text file has eight columns. The first column correspond to the mode frequency (unit: cm⁻¹), the columns 2–8 correspond to the displacements (unit: Å) of the following atoms: Ln, Cr, Ge, Be, O1, O2, O3. The same data are presented also as Excel tables with 8 columns, named, e.g., Pr_displ.xlsx.

Fig. 1 depicts these displacements for all three title compounds, namely, La₃CrGe₃Be₂O₁₄, Pr₃CrGe₃Be₂O₁₄, and Nd₃CrGe₃Be₂O₁₄. It is given as the eps and opj files, Figure 1.eps and Figure 1.opj, respectively. The table in text (and Excel) format Figure 1_table.txt (and Figure 1_table.xlsx) provides the data necessary to create Figure 1.

2. Experimental design, materials, and methods

The main information on the samples and experimental equipment used to take the spectra, as well as on the calculation methods is presented in Ref. [1]. Powder samples of the studied compounds La₃CrGe₃Be₂O₁₄, Pr₃CrGe₃Be₂O₁₄, and Nd₃CrGe₃Be₂O₁₄ were synthesized by a high-temperature solid-state reaction from high-purity La₂O₃, Pr₂O₃, Nd₂O₃ and GeO₂, Cr₂O₃ (reagent grade), and BeO (99.54%). Stoichiometric amounts of oxides were thoroughly ground together, pressed into pellets, placed on a Pt substrate and sintered in air for 5 h at 1350°C (the Nd and Pr compounds) and at 1325°C (the La compound). To reduce the loss of GeO₂ due to evaporation, the pressed samples were encapsulated in the original powdered charges. The phase composition of sintering products was studied by X-ray diffraction using a diffractometer STOE STADI_MP in a transmission mode (CuK_{α1} radiation). The space group P321 was confirmed for all samples.

The infrared diffuse transmission and Raman scattering spectra of Ln₃CrGe₃Be₂O₁₄ (Ln = La, Pr, Nd) powder samples were measured at room temperature. Powders of Ln₃CrGe₃Be₂O₁₄ were mixed with optical-grade KBr powder and pressed into pellets. Far-infrared diffuse transmission spectra were registered in the spectral region 50–1200 cm⁻¹ at a resolution 2 cm⁻¹ using a Fourier spectrometer Bruker IFS 125HR and a DTGS and a liquid-nitrogen-cooled MCT detectors. Raman spectra were taken in the backscattering geometry at a resolution 3 cm⁻¹ with a home-made triple monochromator using the line 514, 5 nm of an argon laser as an excitation.

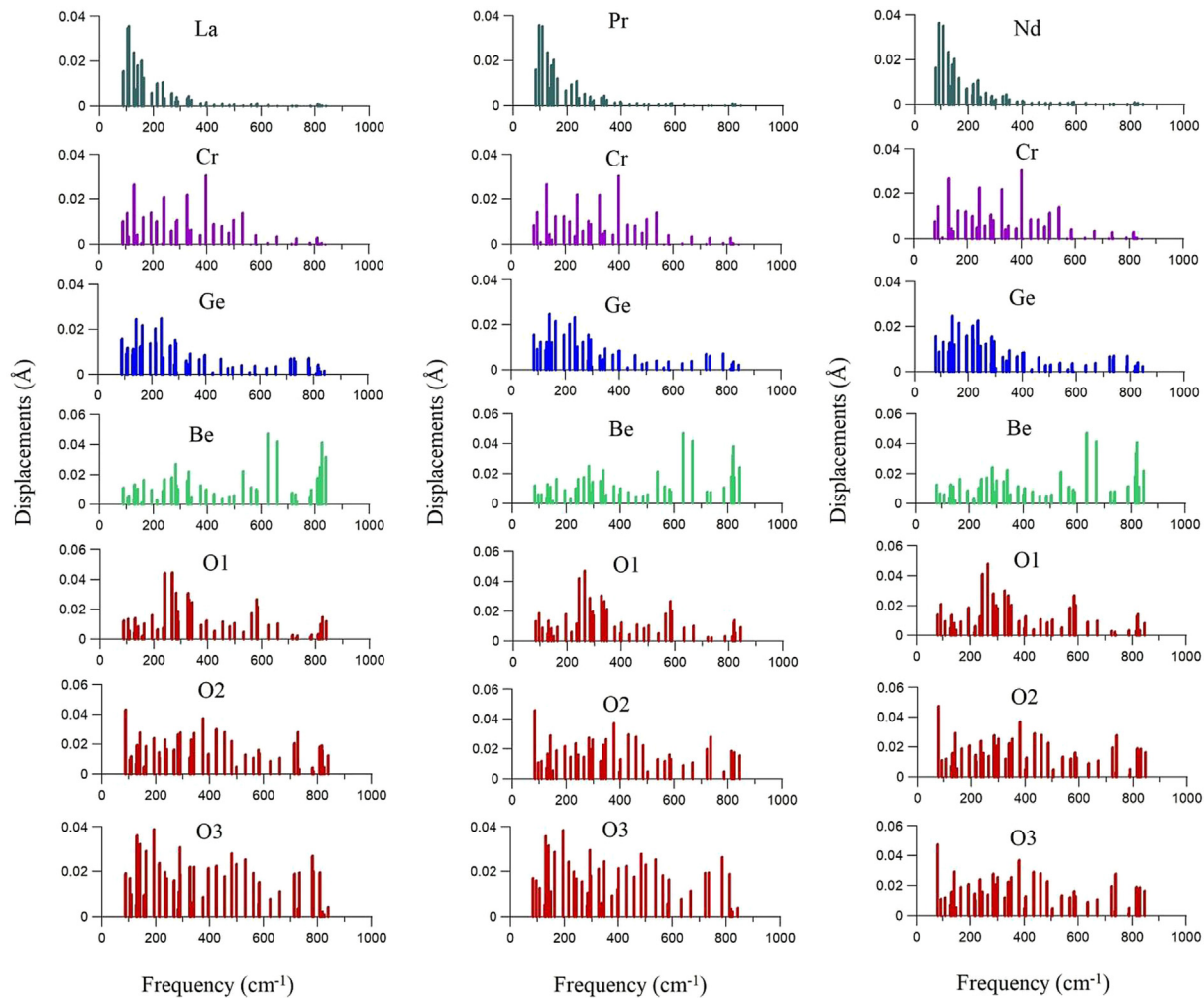


Fig. 1. Displacements of different atoms of $\text{La}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ in normal crystal modes of different frequencies.

Ab initio calculations of phonon frequencies and intensities of the infrared- and Raman-active modes of $\text{La}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$, $\text{Pr}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$, and $\text{Nd}_3\text{CrGe}_3\text{Be}_2\text{O}_{14}$ were performed in a framework of the density functional theory (DFT) with the hybrid functional B3LYP [2], which takes into account both local and nonlocal (in the Hartree-Fock formalism) exchange. The sequence of calculations was as follows. The optimization of the crystal structure was carried out first. After that, the phonon spectrum was calculated for the crystal structure corresponding to the minimum energy. The CRYSTAL14 program [3] designed for simulating periodic structures in the MO LCAO approximation was used for calculations. Quasi-relativistic pseudopotentials ECP46MWB, ECP59MWB, and ECP60MWB [4,5] with corresponding valence basis sets ECP*n*MWB [6] were taken for La, Pr, and Nd. All-electron basis sets of TZVP type were used for Cr, Ge, Be, and O [7]. These basis sets are available at the CRYSTAL website. The reciprocal space sampling was performed by Monkhorst-Pack mesh. The algorithm of calculation of the two-electron Coulomb and exchange integrals is given in Ref. [8]. The tolerance of self-consistently solving of the system of Kohn-Sham equations was 10^{-9} . The phonon spectrum was calculated in the harmonic approximation. In the Hessian matrix, the first (second) derivatives were calculated analytically (numerically). To perform numerical calculations of the second derivatives, the atom was displaced from the equilibrium position by 0.003 Å [8].

We used the Born charges when calculating Raman and infrared intensities in the CRYSTAL code [9]. Electric dipole properties were calculated using the periodic coupled-perturbed Hartree-Fock (CPHF) or Kohn-Sham (CPKS) approach [10–12].

The Placzek approximation was used to calculate the intensity of the Raman modes at a non-resonant excitation [11]. For an oriented single crystal, the intensity associated with the mode ω_k is [3]:

$$I_{ij}^k \propto C \left(\alpha_{ij}^k \right)^2, \quad (1)$$

where α_{ij}^k is an element of the Raman tensor, $i, j = x, y, z$. The value C in (1) is defined by the laser frequency ω_L and the temperature T dependence as follows:

$$C \sim \frac{1 + n(\omega_k)}{30\omega_k} (\omega_L - \omega_k)^4, \quad (2)$$

where

$$1 + n(\omega_k) = \left[1 - \exp\left(-\frac{\hbar\omega_k}{k_B T}\right) \right]^{-1}, \quad (3)$$

$n(\omega_k)$ being the Bose occupation factor.

The simulation of the intensity of Raman modes for powder sample has been done by computing integrals over all possible orientations of ideal bulk crystal. These integrals can be reduced to three rotational invariants [13]:

$$G_k^{(0)} = \frac{1}{3} \left(\alpha_{xx}^k + \alpha_{yy}^k + \alpha_{zz}^k \right)^2 \quad (4)$$

$$G_k^{(1)} = \frac{1}{2} \left[\left(\alpha_{xy}^k - \alpha_{yx}^k \right)^2 + \left(\alpha_{xz}^k - \alpha_{zx}^k \right)^2 + \left(\alpha_{zy}^k - \alpha_{yz}^k \right)^2 \right] \quad (5)$$

$$G_k^{(2)} = \frac{1}{2} \left[\left(\alpha_{xy}^k + \alpha_{yx}^k \right)^2 + \left(\alpha_{xz}^k + \alpha_{zx}^k \right)^2 + \left(\alpha_{zy}^k + \alpha_{yz}^k \right)^2 \right] + \frac{1}{3} \left[\left(\alpha_{xx}^k - \alpha_{yy}^k \right)^2 + \left(\alpha_{xx}^k - \alpha_{zz}^k \right)^2 + \left(\alpha_{yy}^k - \alpha_{zz}^k \right)^2 \right] \quad (6)$$

The intensity for the powder sample can be calculated as [14]:

$$I_{tot,k}^{powder} = I_{\parallel,k}^{powder} + I_{\perp,k}^{powder}, \quad (7)$$

where

$$I_{\parallel,k}^{powder} = C \left(10G_k^{(0)} + 4G_k^{(2)} \right) \quad (8)$$

$$I_{\perp,k}^{powder} = C \left(5G_k^{(1)} + 3G_k^{(2)} \right) \quad (9)$$

and C is given by Eq. (2).

The infrared intensity of the p -th mode can be written as [3]:

$$I_p = \frac{\pi}{3} \frac{N_A}{c^2} d_p |\vec{Z}_p|^2, \quad (10)$$

where N_A is the Avogadro's number, c is the speed of light, d_p is the degeneracy of the mode, \vec{Z}_p is the mass-weighted Born effective charge vector of the mode. The infrared intensity was calculated assuming an isotropic response.

The high-spin ($S = 3/2$) state of the Cr^{3+} ions was set in the calculations. At the simulation, magnetic moments of chromium ions were codirected (along the z axis), hereby the ferromagnetic state was simulated. In this work, we consecutively calculate the crystal structure and, then, the phonon spectrum. The initial structural data were taken from Ref. [14].

When choosing a functional, calculations with the hybrid functional PBE0 [15] were also performed.

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Conflict of Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.dib.2019.104889>.

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